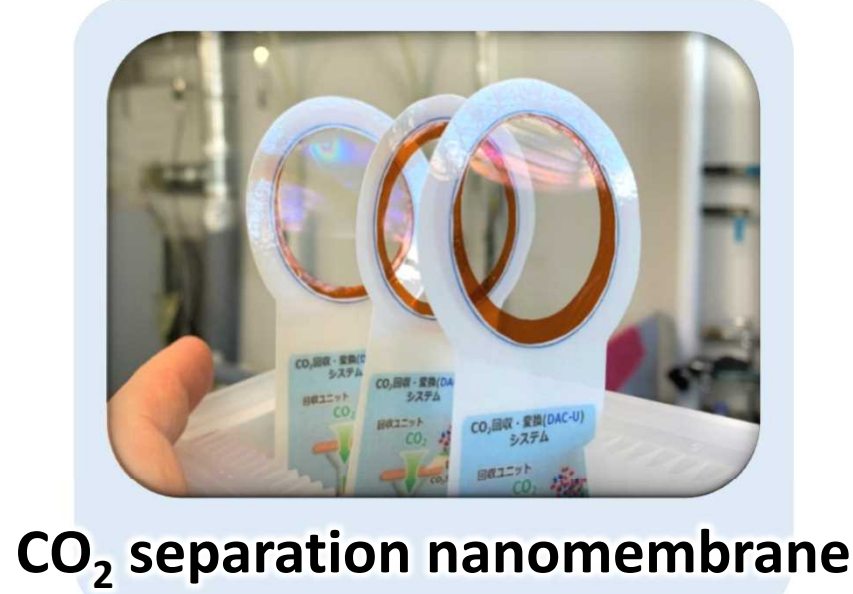


### Toward a Beyond Zero Society through Ubiquitous Carbon Capture and Utilization (CCU)

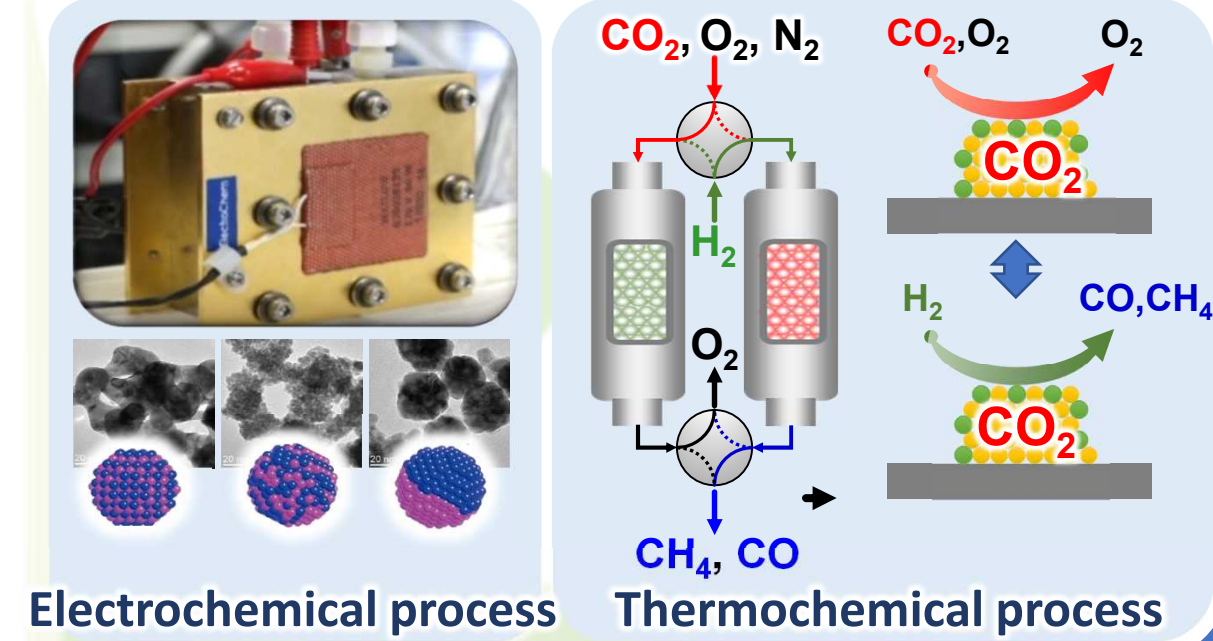
#### CO<sub>2</sub> separation nanomembranes with overwhelming high CO<sub>2</sub> permeability

- Kyushu University has developed a nanomembrane that is only 1/2500 as thin as the diameter of a hair.
- Developed nanomembranes show highest CO<sub>2</sub> permeance, which is 20 times higher than conventional membranes



#### Catalyst technology for CO<sub>2</sub> conversion and small conversion system

- Nanocatalysts and electrochemical process system for highly efficient conversion of CO<sub>2</sub> to useful substances
- Novel thermochemical catalysts and processes for simultaneous O<sub>2</sub> removal and CO<sub>2</sub> conversion

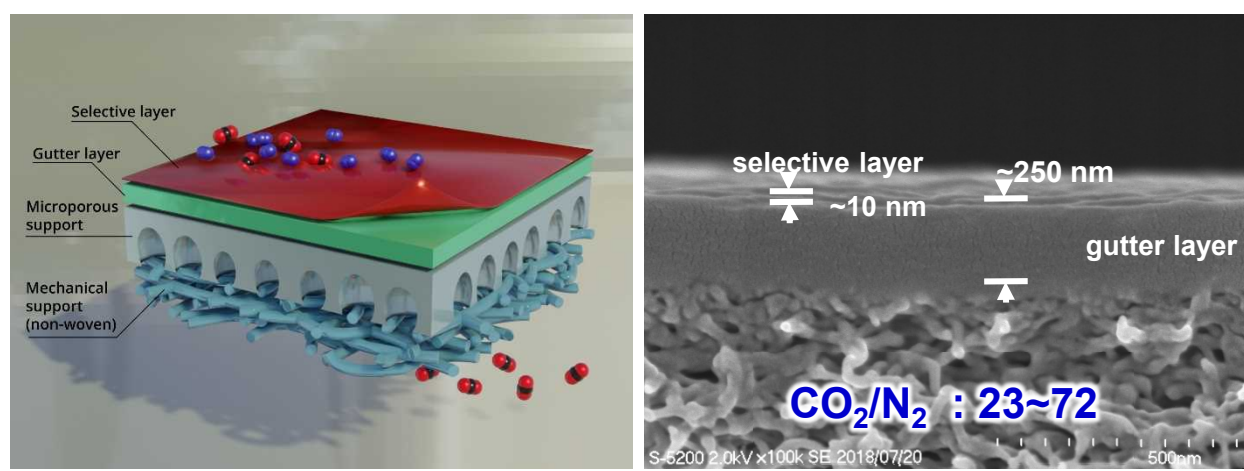


### Goals of the Project

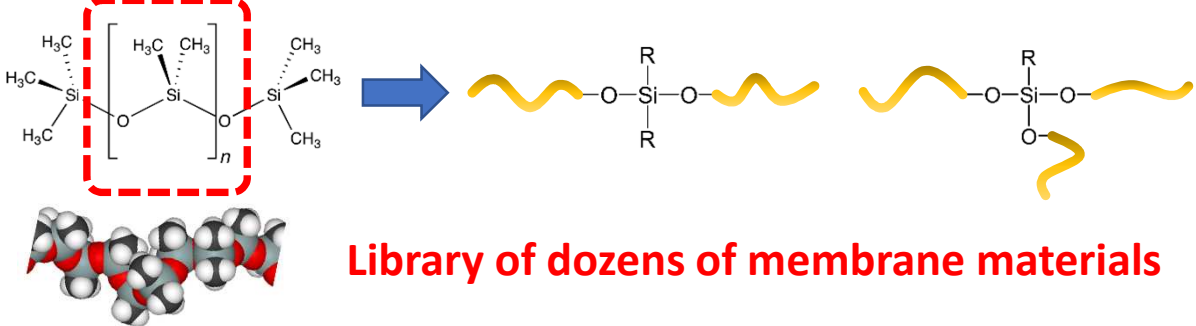
### Compact, Scalable, Distributed Deployable Direct Air Capture-Utilization (DAC-U) System



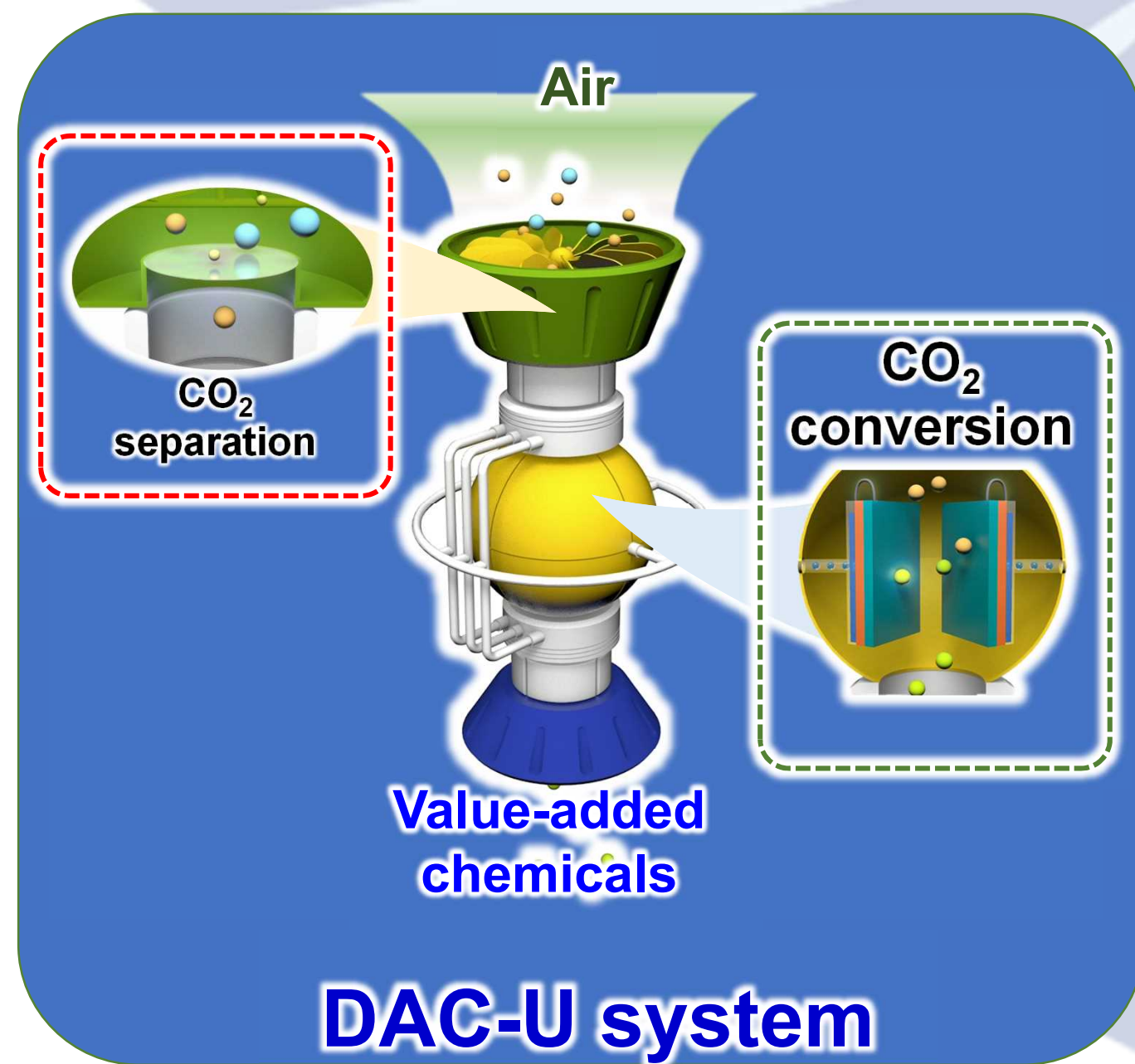
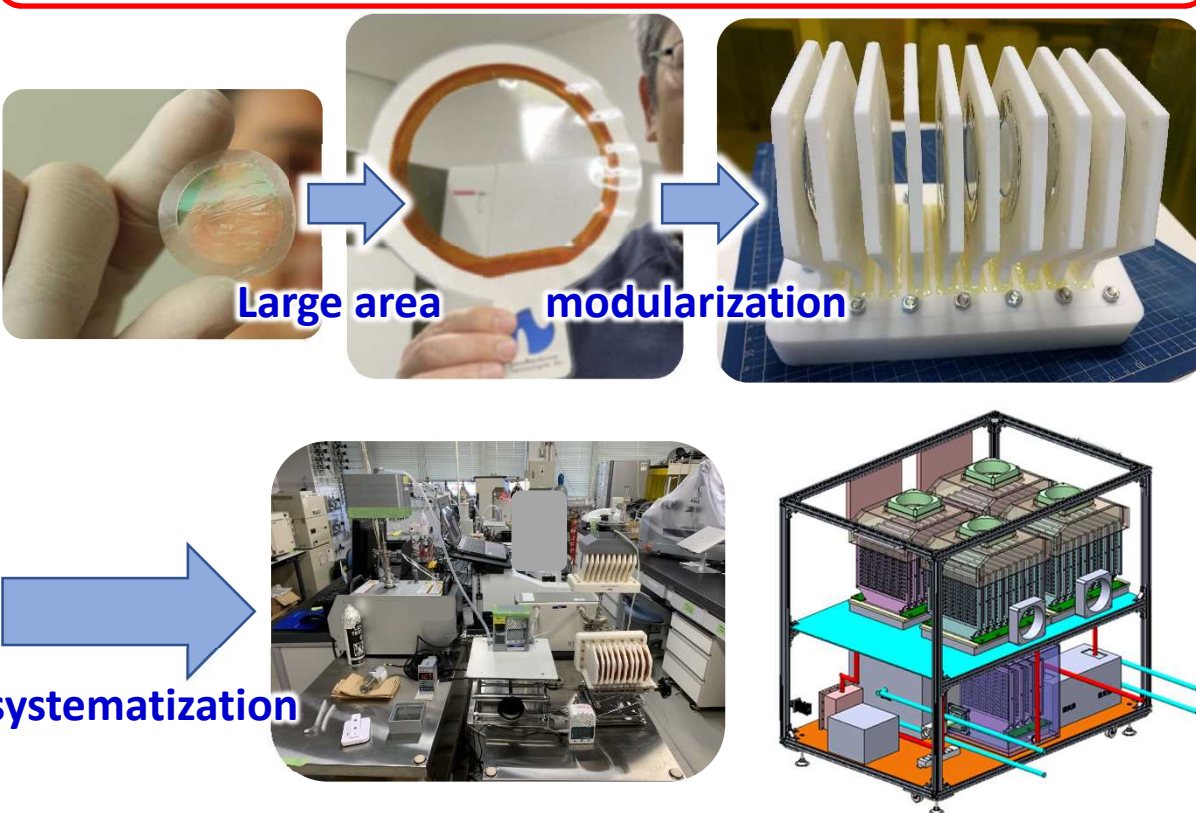
#### Enhancement of CO<sub>2</sub> selectivity



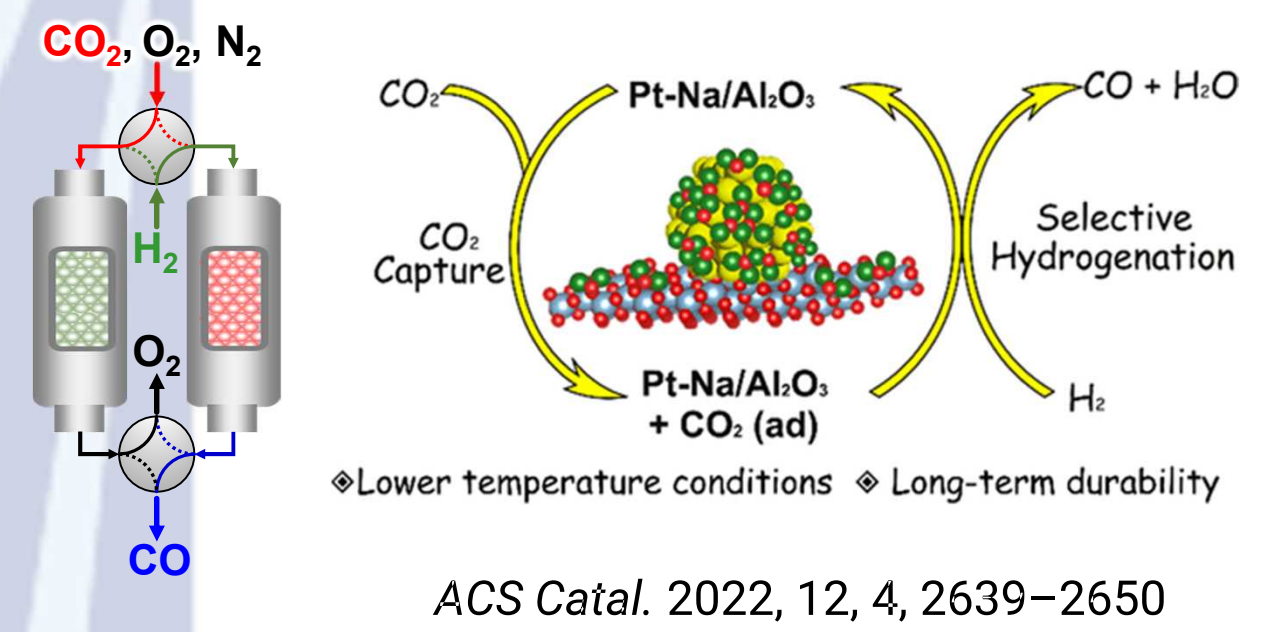
#### Development of membrane materials



#### Large area and modularization of nanomembranes

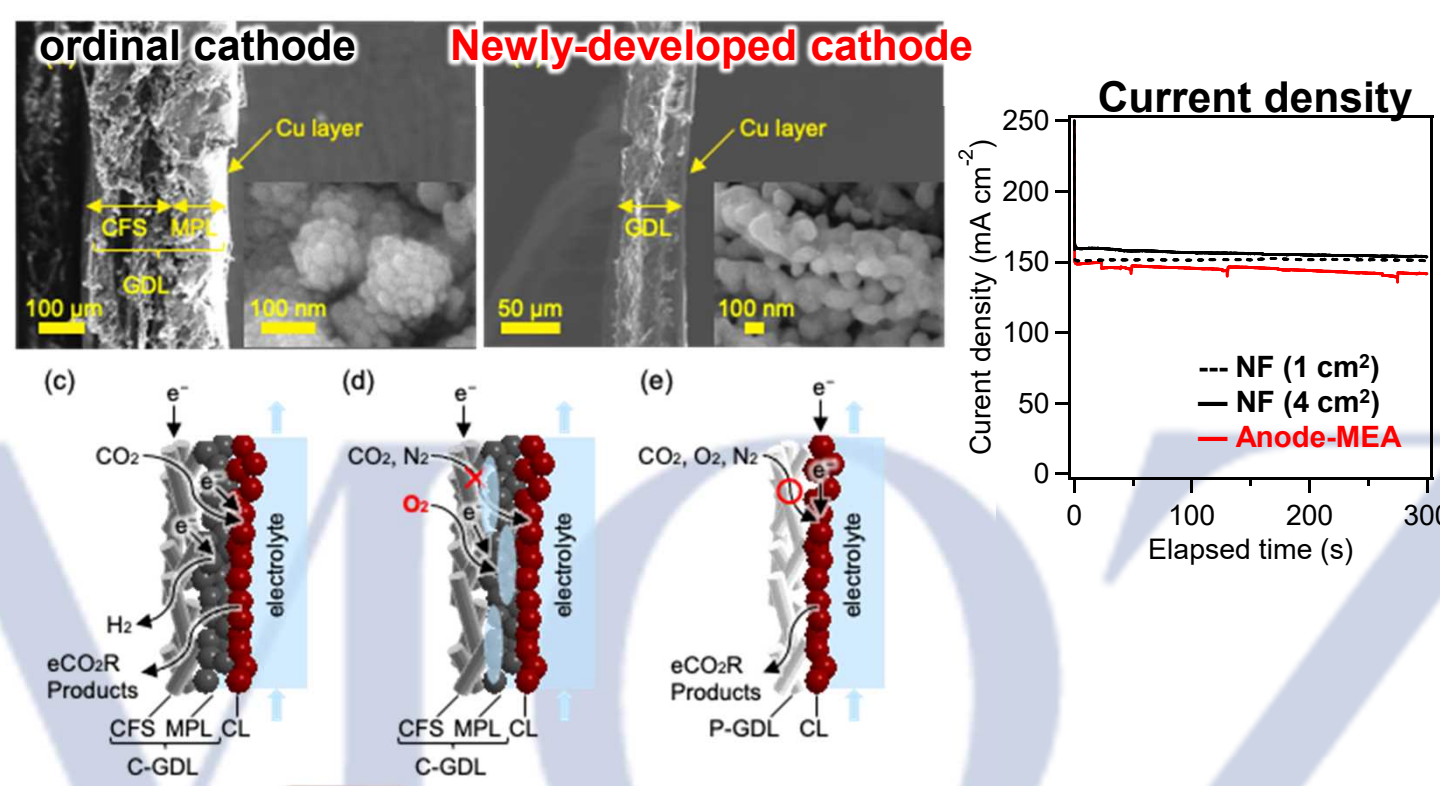


#### CO<sub>2</sub> absorption and thermochemical CO<sub>2</sub> conversion



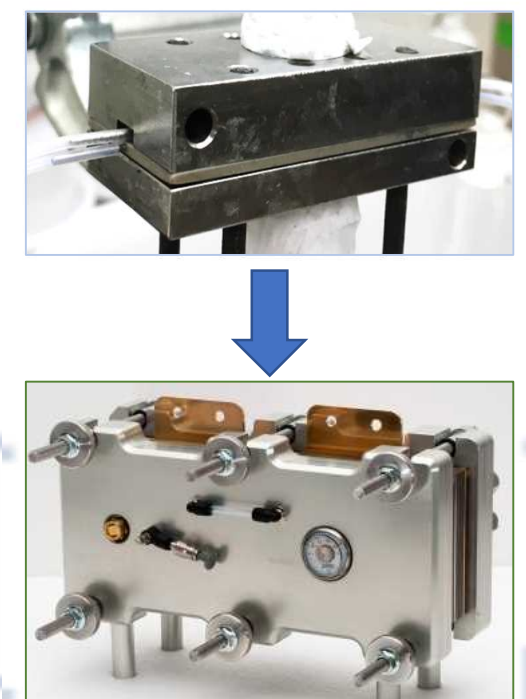
Prof. Ken-ichi SHIMIZU (Hokkaido Univ.)

#### New electrodes for electrochemical CO<sub>2</sub> conversion



Prof. Miho YAMAUCHI (Kyushu Univ.)

#### Development of high-performance electrochemical cells



Univ. Illinois at Urbana Champaign  
Prof. Paul KENIS



Carbon resource recycling society with decentralized deployment of DAC-U systems

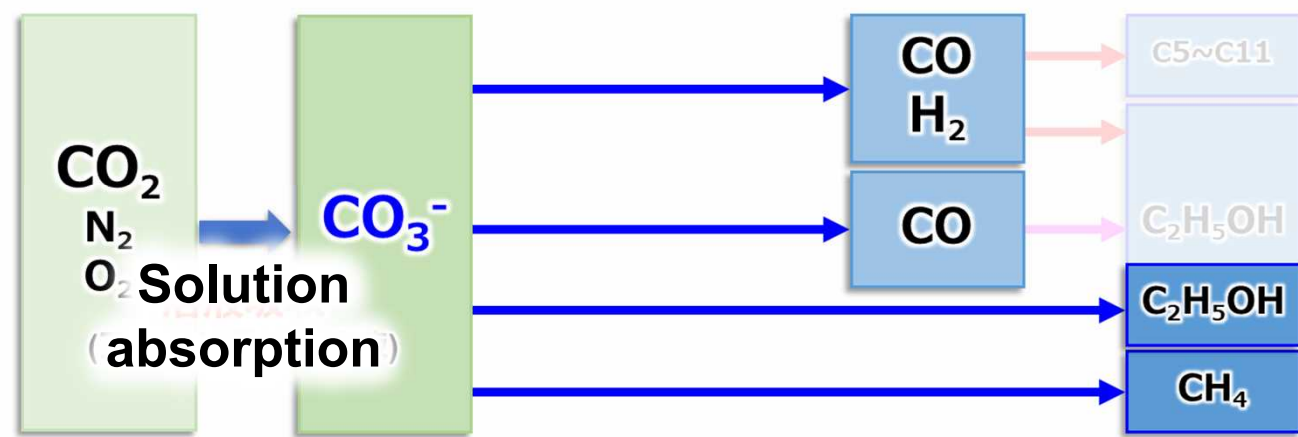


# Electrochemical CO<sub>2</sub> reduction using 60% air mixed 40% CO<sub>2</sub> gas (Air-CO<sub>2</sub>)

目標 : [Air-CO<sub>2</sub>] + H<sub>2</sub>O → CO, CH<sub>4</sub>, C<sub>2</sub>H<sub>4</sub> or α

**Method A**  
40CO<sub>2</sub>, 12%O<sub>2</sub>, 48%N<sub>2</sub>

**Method B**  
electrolyte



○ simple system

× few reports

× N<sub>2</sub>, O<sub>2</sub> separation

○ no N<sub>2</sub>, O<sub>2</sub> separation

× few reports

× predominant HER

× dissolving Air-CO<sub>2</sub>

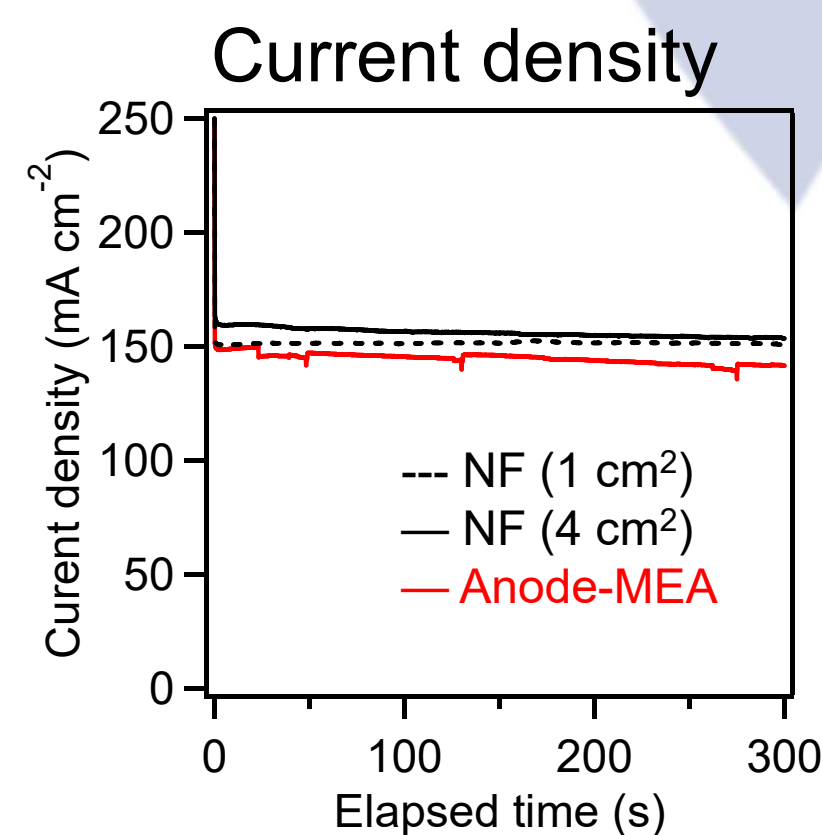
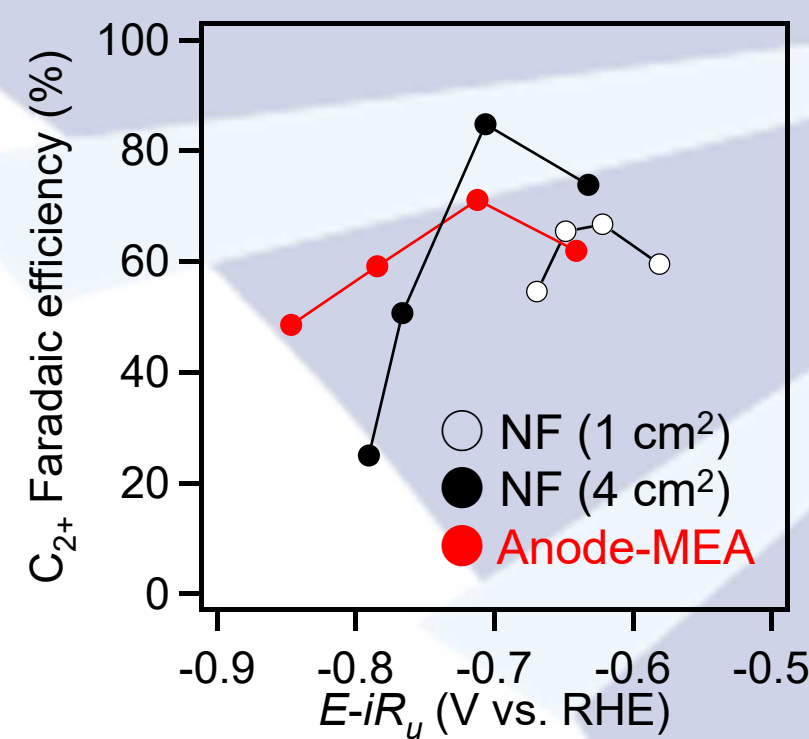
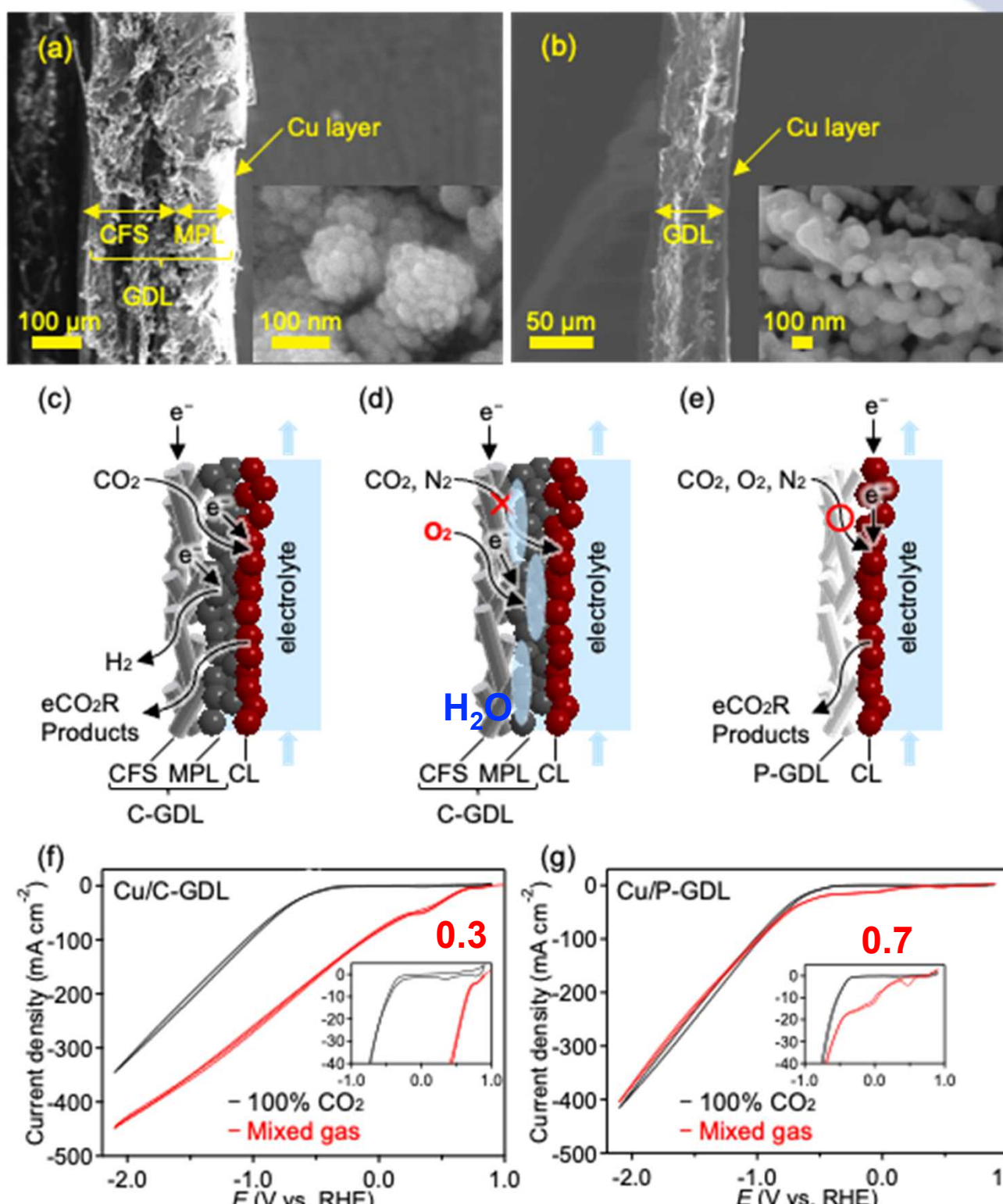
## Method A

## Direct eCO<sub>2</sub>R using Air-CO<sub>2</sub>

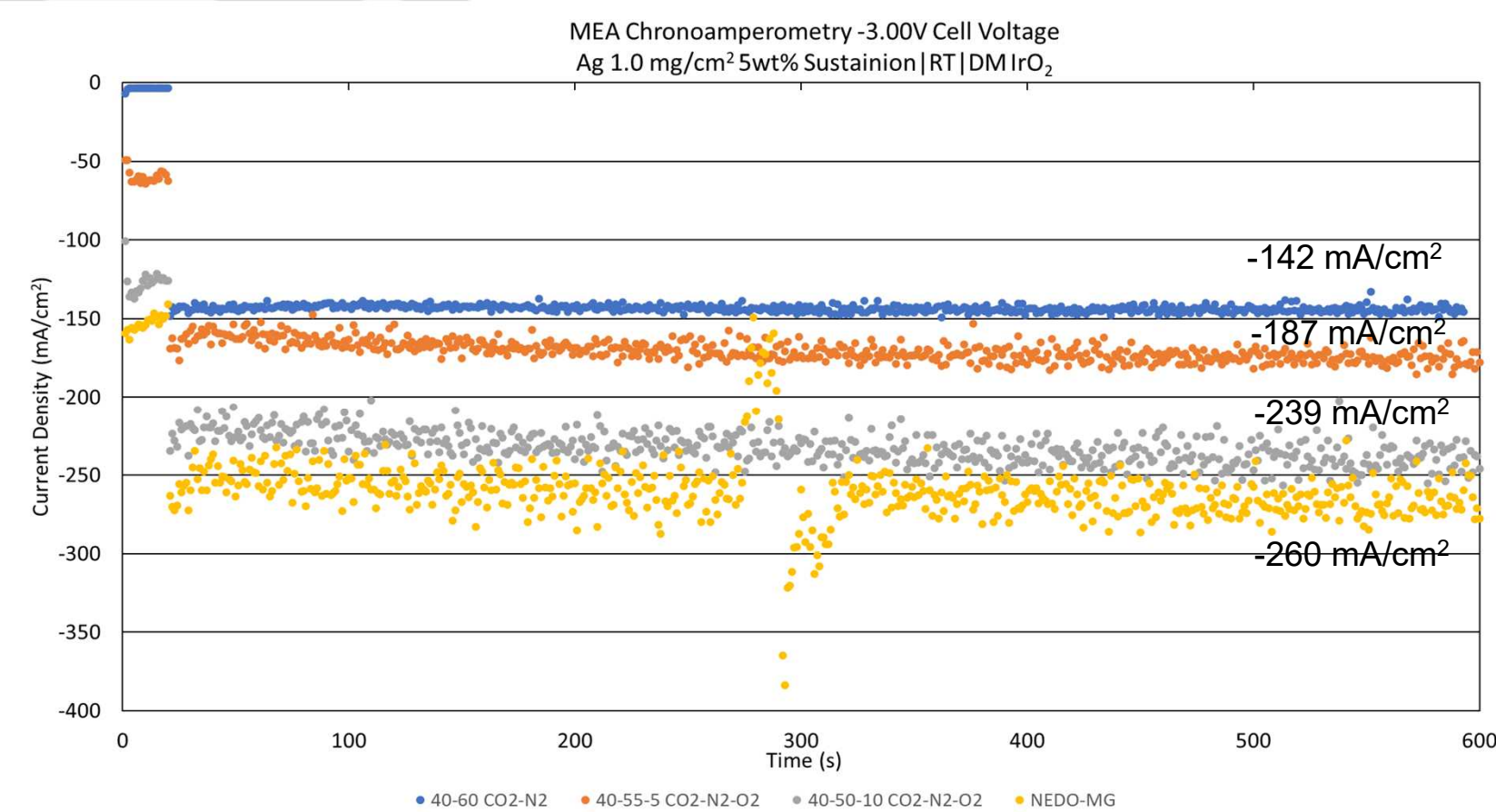
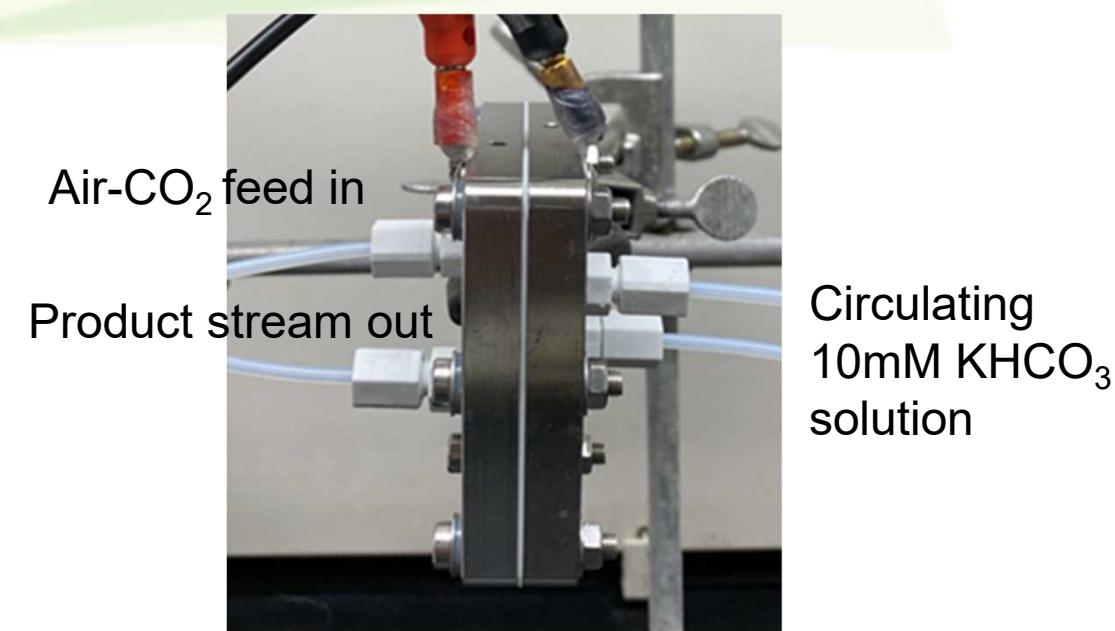
Conventional cathode

Novel cathode

Faradaic efficiency



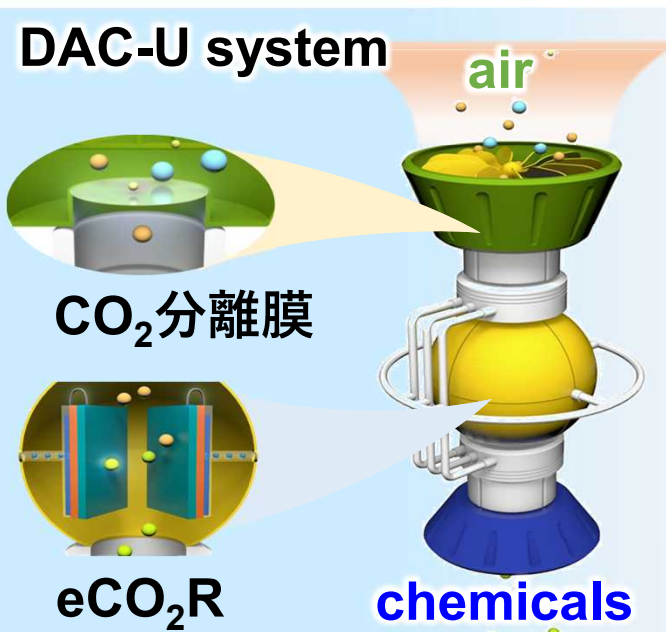
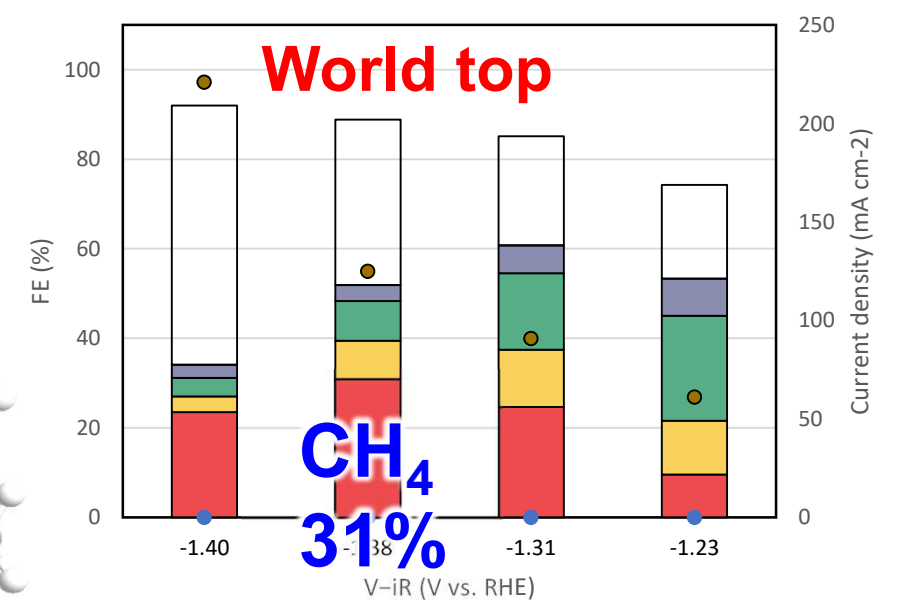
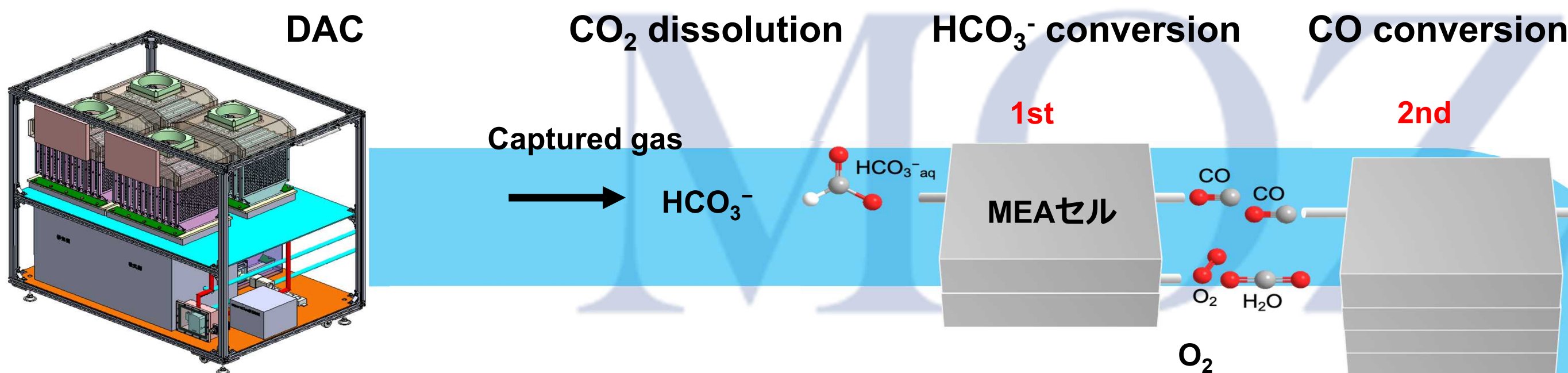
Prof. Kenis (UIUC)



Chem. Comm., 59, 11188-11191 (2023)

## Method B

## Production of O<sub>2</sub>-N<sub>2</sub> free resources by eCO<sub>2</sub>R from Air-CO<sub>2</sub>



Electricity used to create CH<sub>4</sub> from 2.0 kg CO<sub>2</sub> (current)

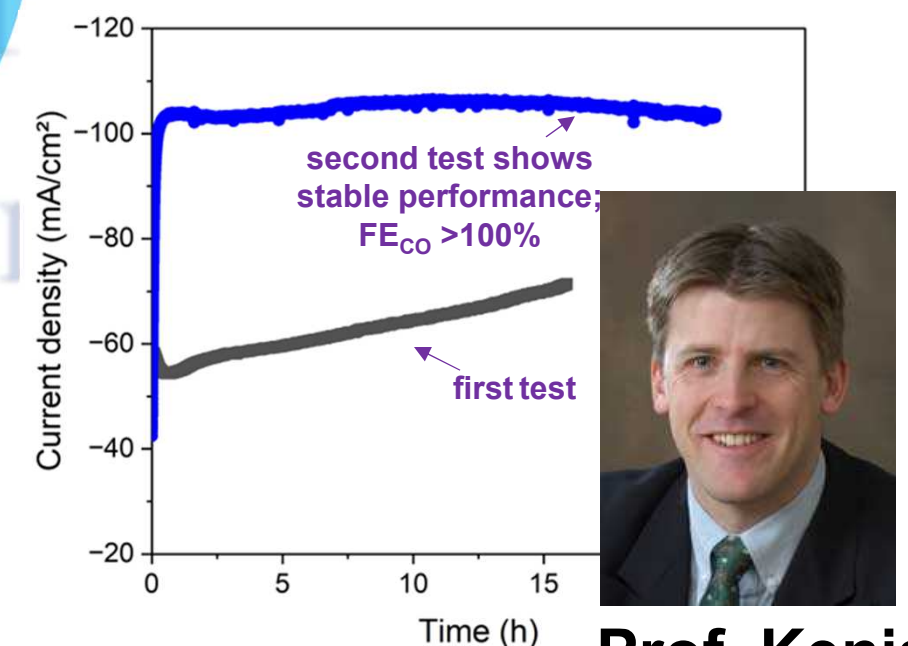
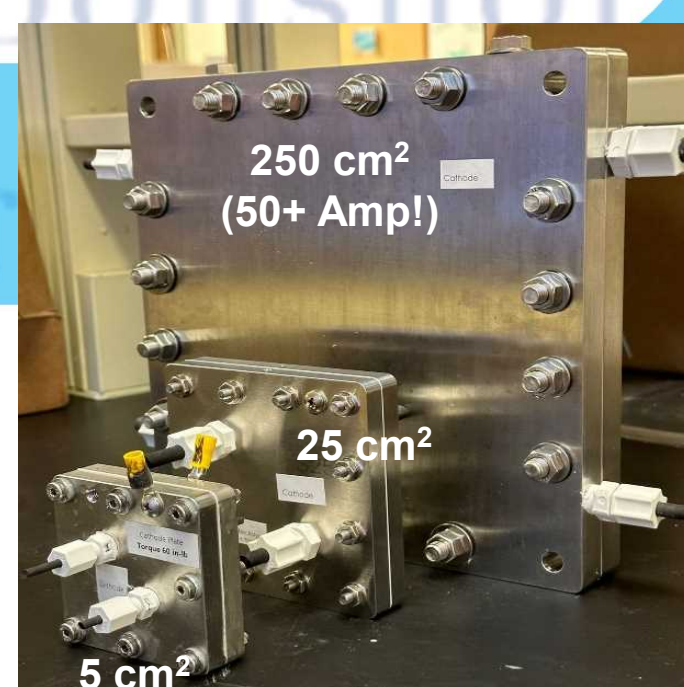
CO<sub>2</sub> conv. rate : 45.4 mol day<sup>-1</sup> (CO<sub>2</sub>)  
CH<sub>4</sub> prod. rate : 1.29 mol day<sup>-1</sup> (CH<sub>4</sub>)

(Wh) = P (W) × t (h)より  
1<sup>st</sup> step : 0.6 kWh  
2<sup>nd</sup> step : 1.0 kWh  
→ 1.6 kWh

10円/kWh → 16円 (eCO<sub>2</sub>R)

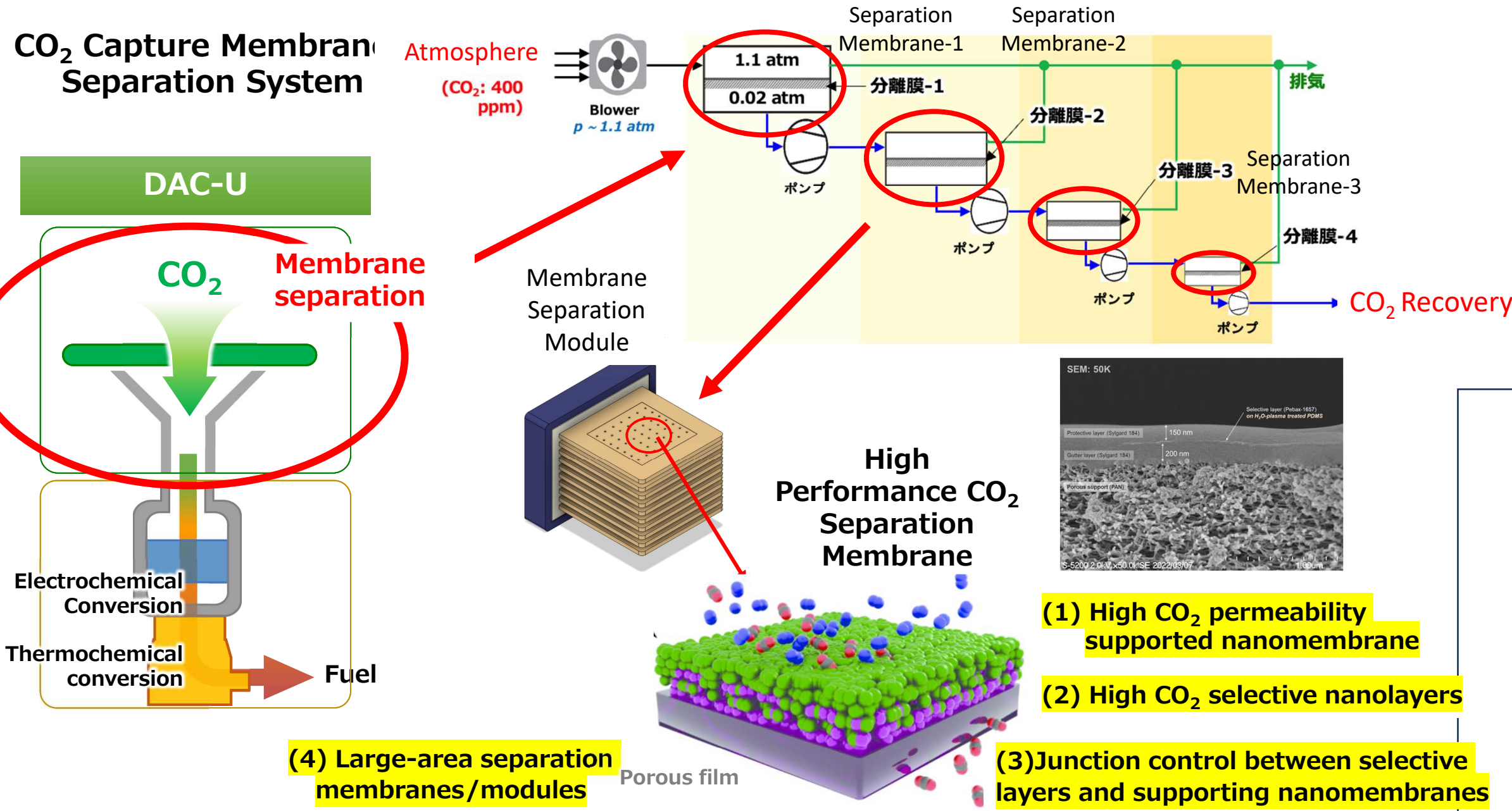
If electricity bill were reduced to 1/10....

Large area, Stack



Prof. Kenis (UIUC)



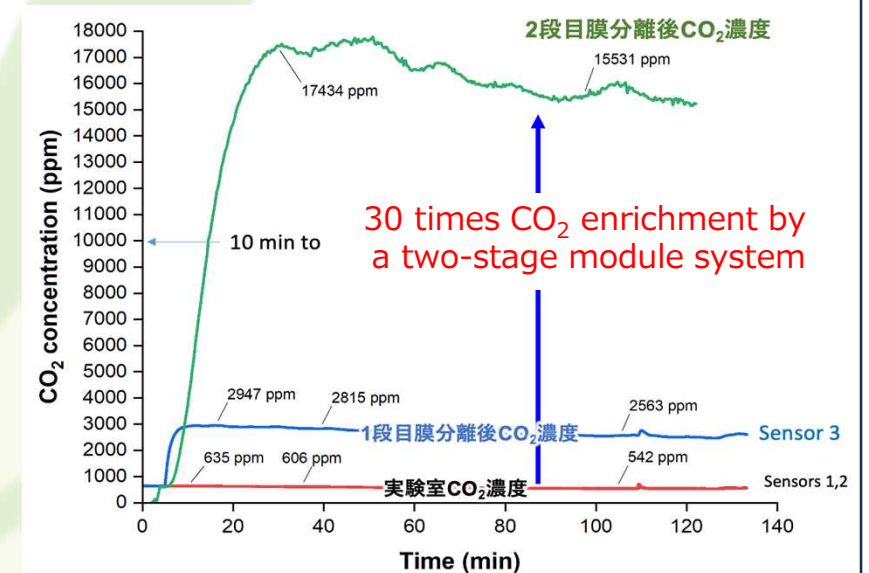
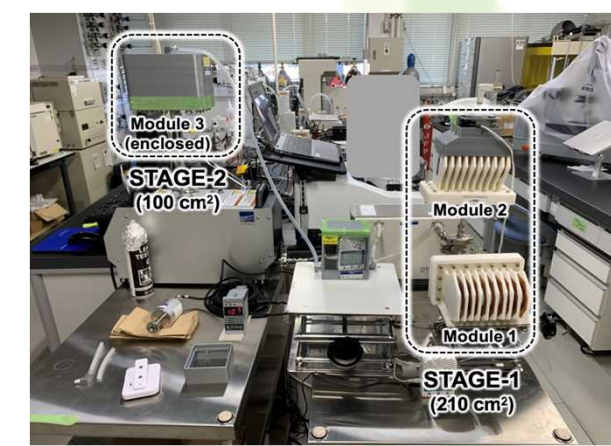


### CO<sub>2</sub> Capture Membrane R&D Unit

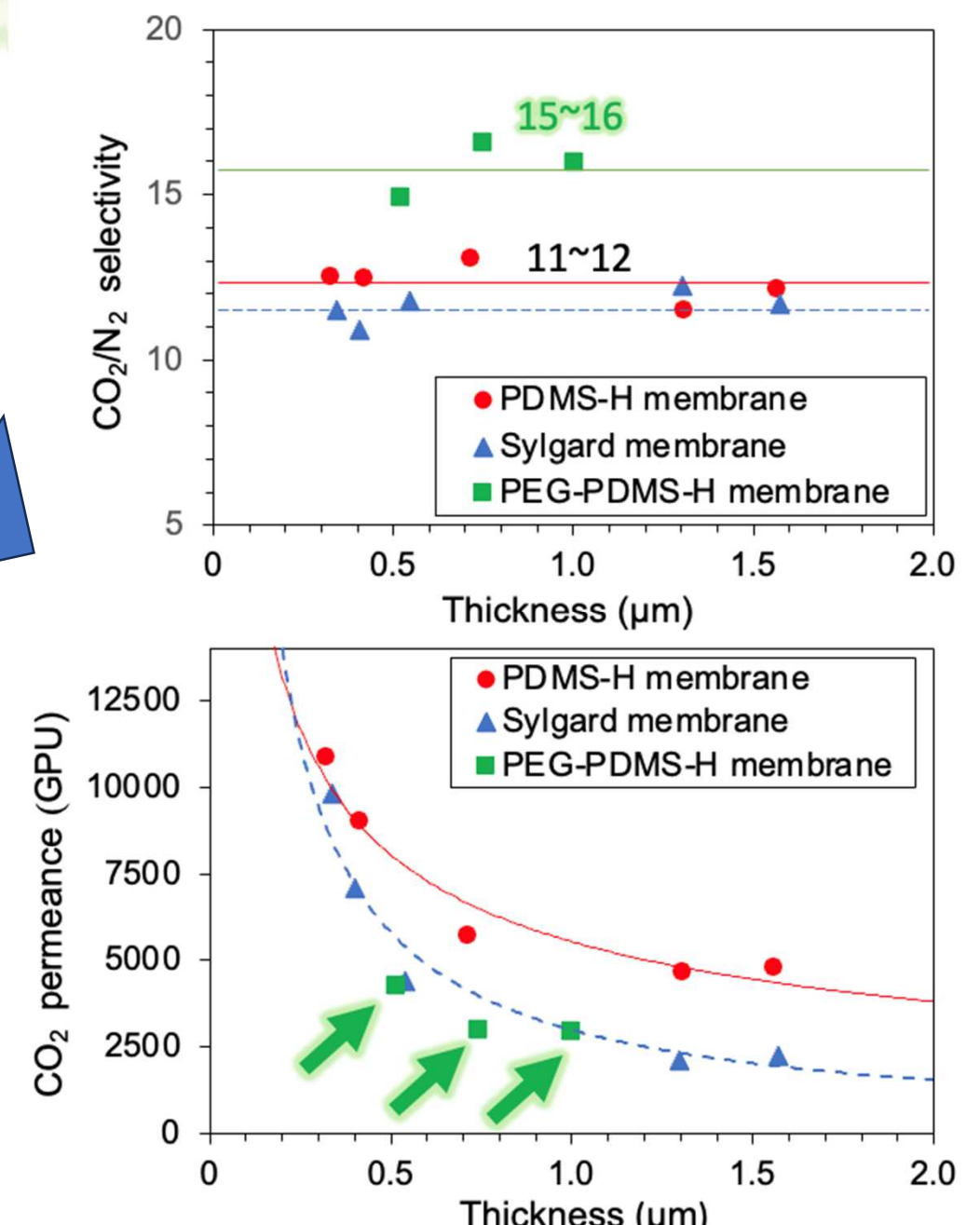
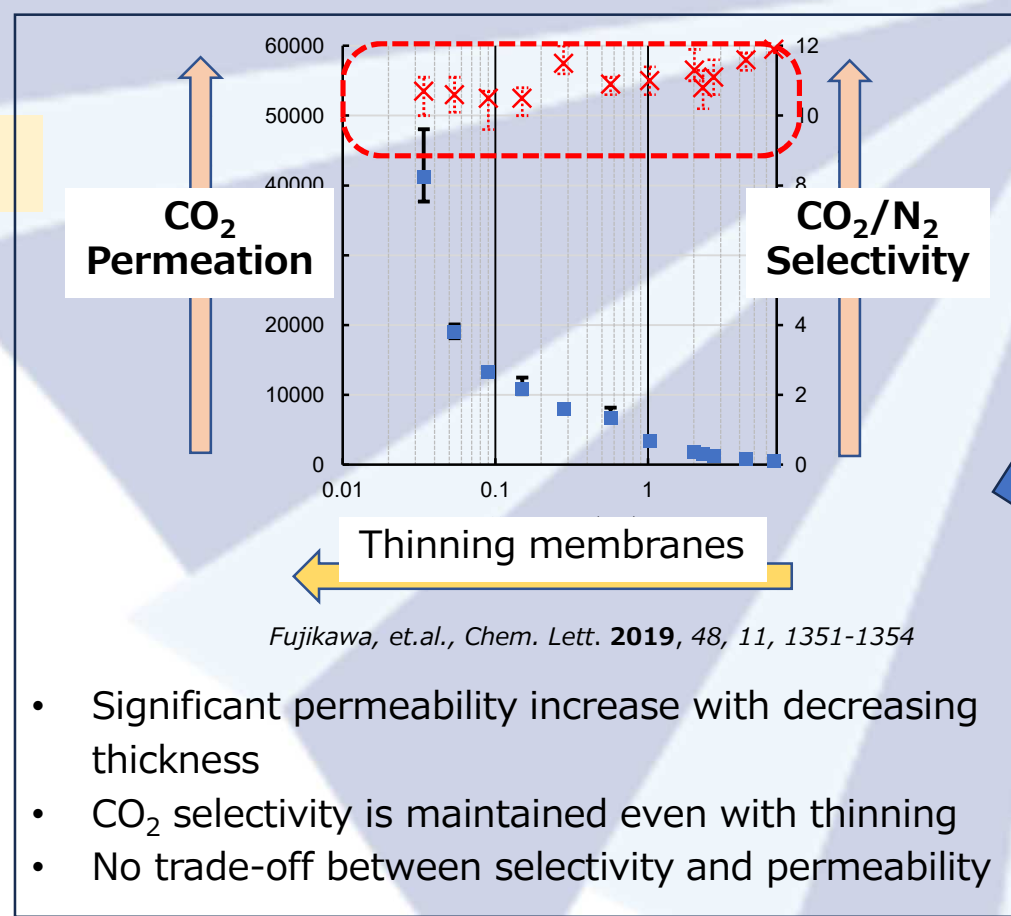
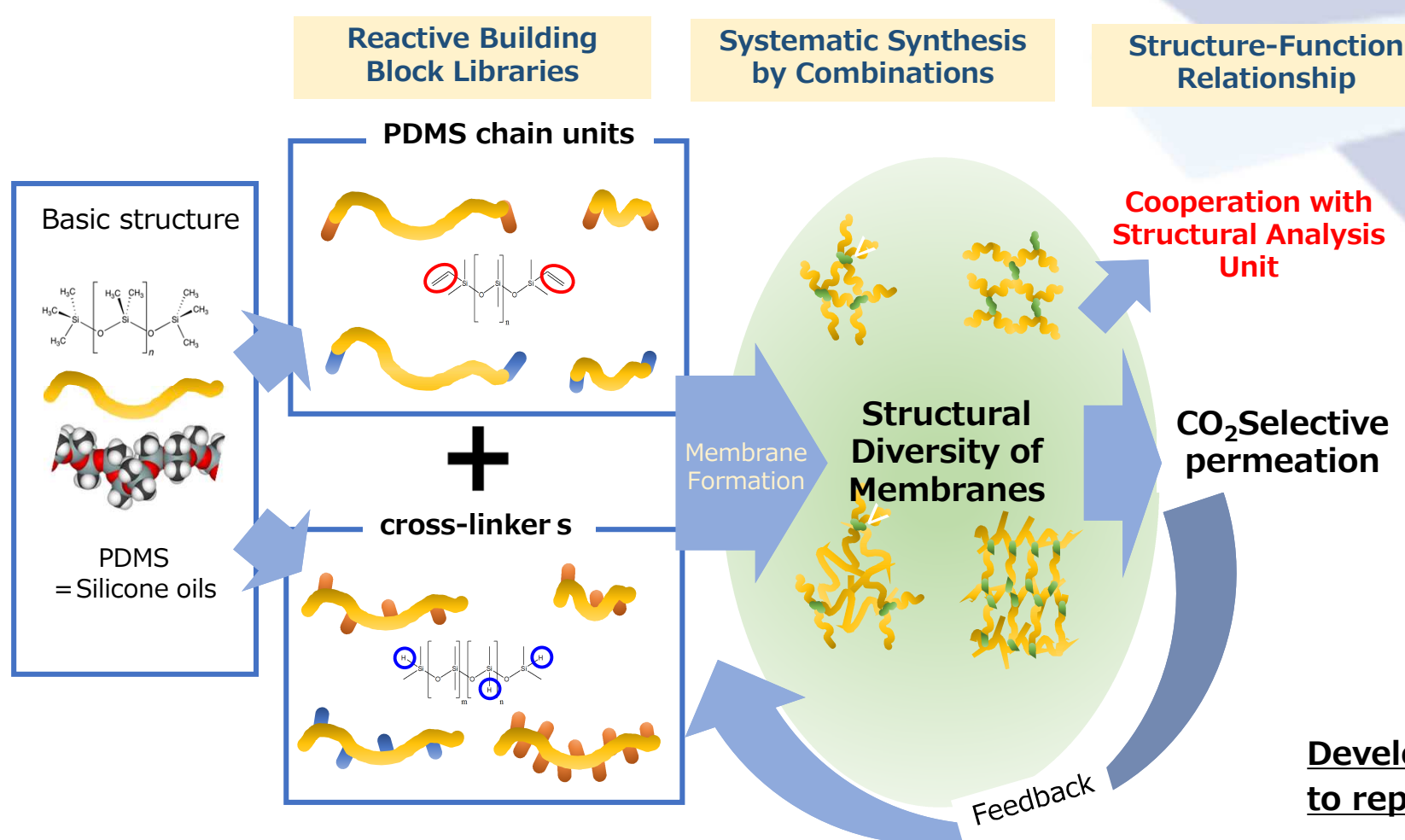


Shigenori Fujikawa (Kyushu University, PM)  
 Masashi Kunitake (Kumamoto University, UL)  
 Tomoyasu Hirai (Osaka Institute of Technology)  
 Yoshiro Kaneko (Kagoshima University)  
 Shinichiro Noro (Hokkaido University)  
 Toyoki Kunitake (NanoMembrane Technologies, Inc.)

### Demonstration of a two-stage separation prototype system



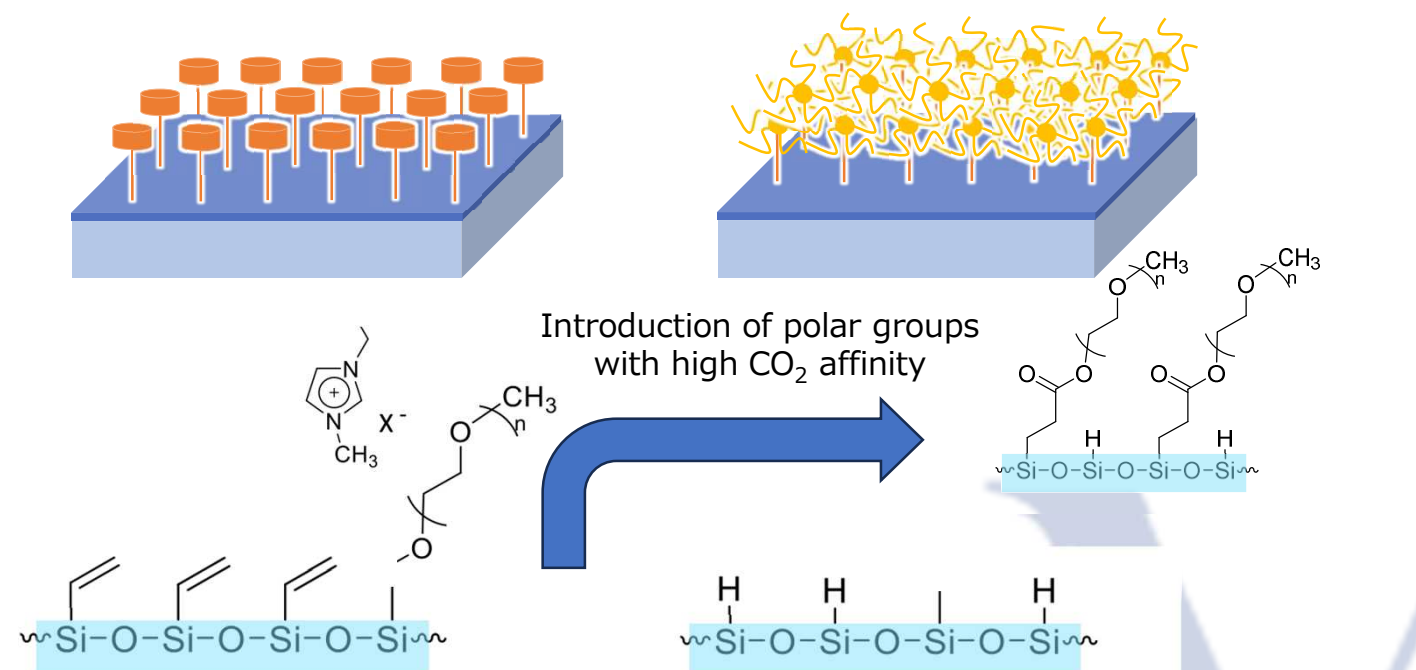
### Supported Membranes with High CO<sub>2</sub> Permeability



**Development of siloxane products for permeation membranes to replace structure-unknown commercial materials**

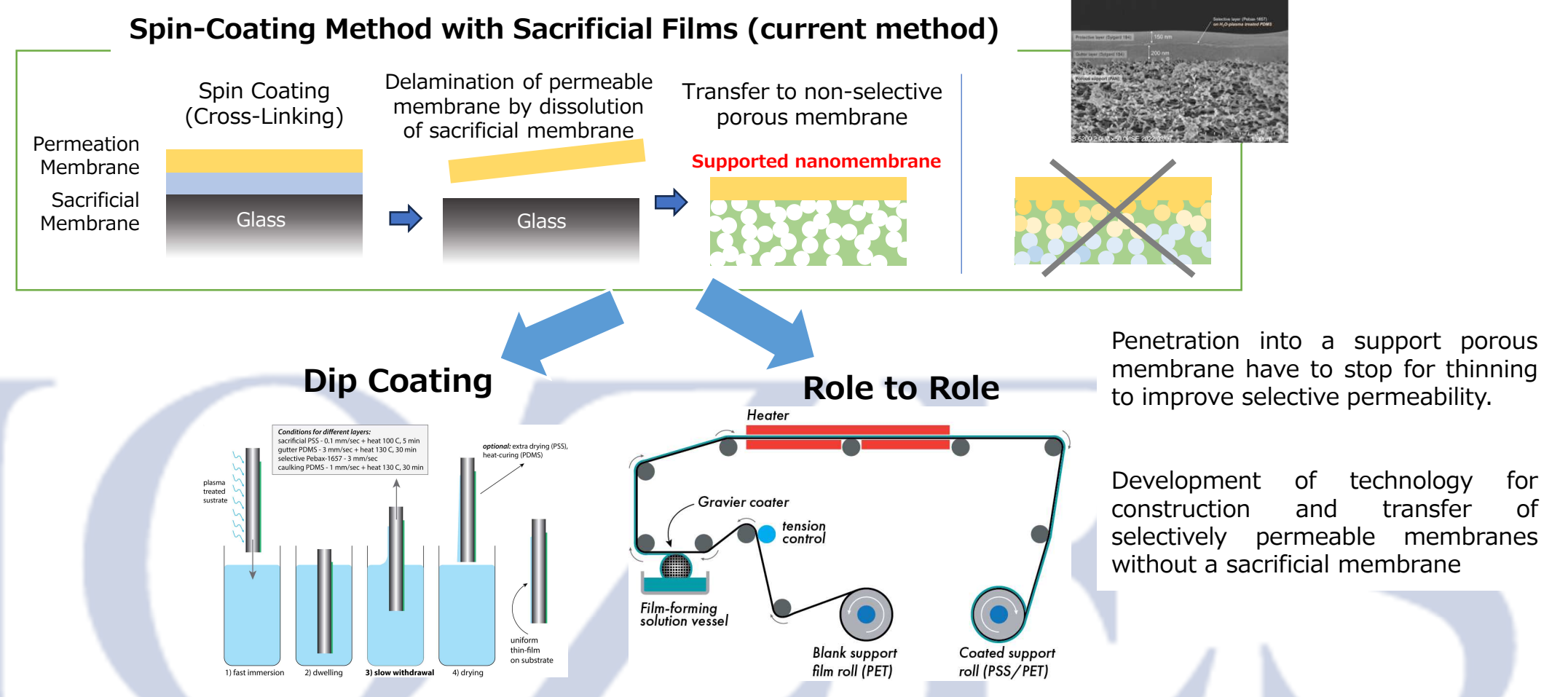
We have successfully developed several membrane material systems with CO<sub>2</sub> selective permeability equivalent to or superior to Sylgard, based on PDMS crosslinking reaction systems that can be hierarchically structured and have a well-defined chemical structure. In each system, we reconfirmed the improvement of permeability by thinning while maintaining the selective permeability.

### Exploration of Selective Layers, and Junction Control to Supporting Membranes



We have developed a technology for selectively introducing reactive units such as vinyl groups and hydrosilyl groups to PDMS film surfaces. This has made it possible to introduce various polar units to the non-polar PDMS film surface.

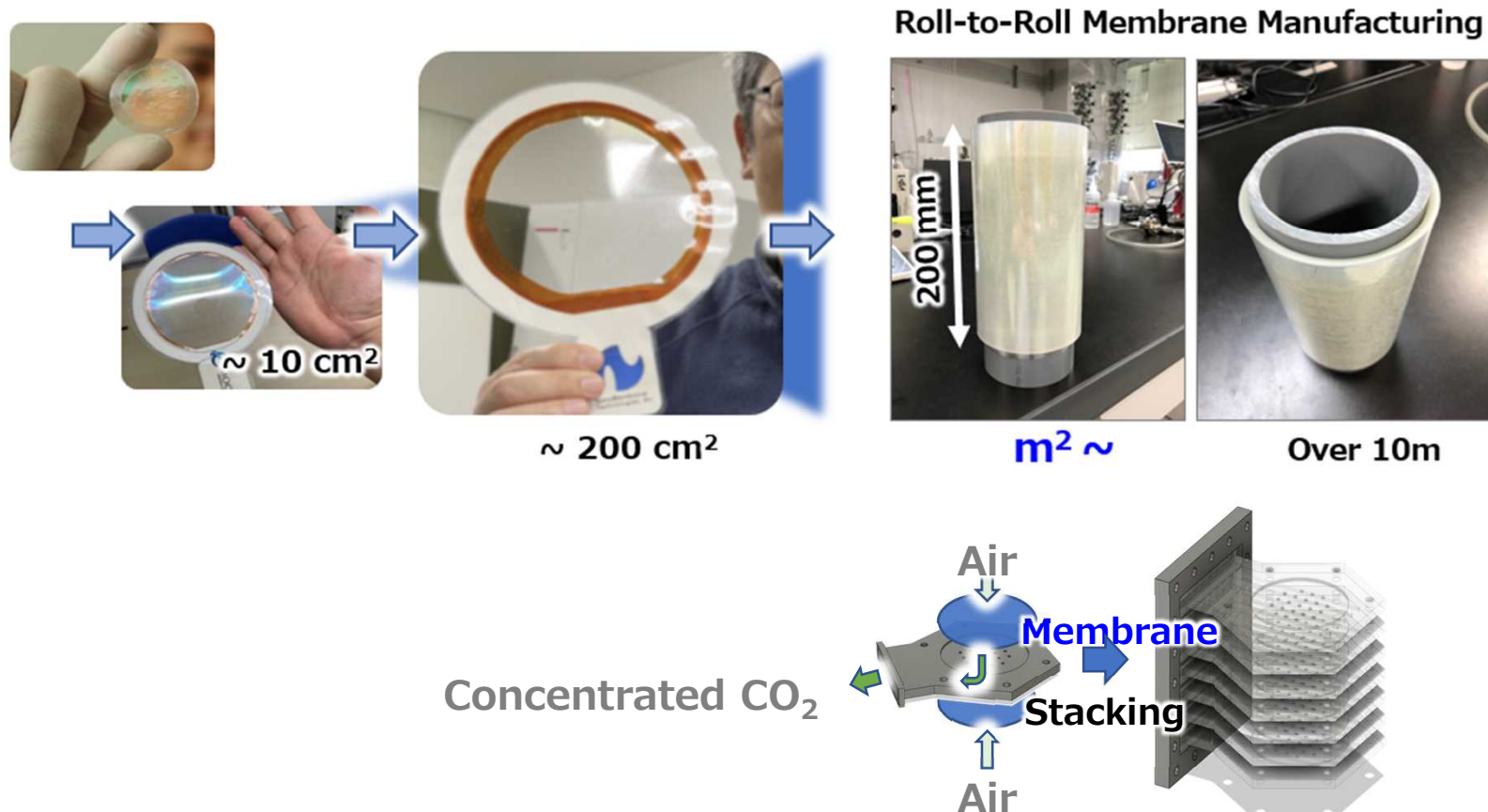
### Development of Manufacturing Technology to Industrialize a Large-Scale Thin Permeation Membrane Production



Penetration into a support porous membrane have to stop for thinning to improve selective permeability.

Development of technology for construction and transfer of selectively permeable membranes without a sacrificial membrane

### Large-Area/Modular Separation Membrane System



### Development of Membrane Materials that do not Require Chemical Cross-Linking

