

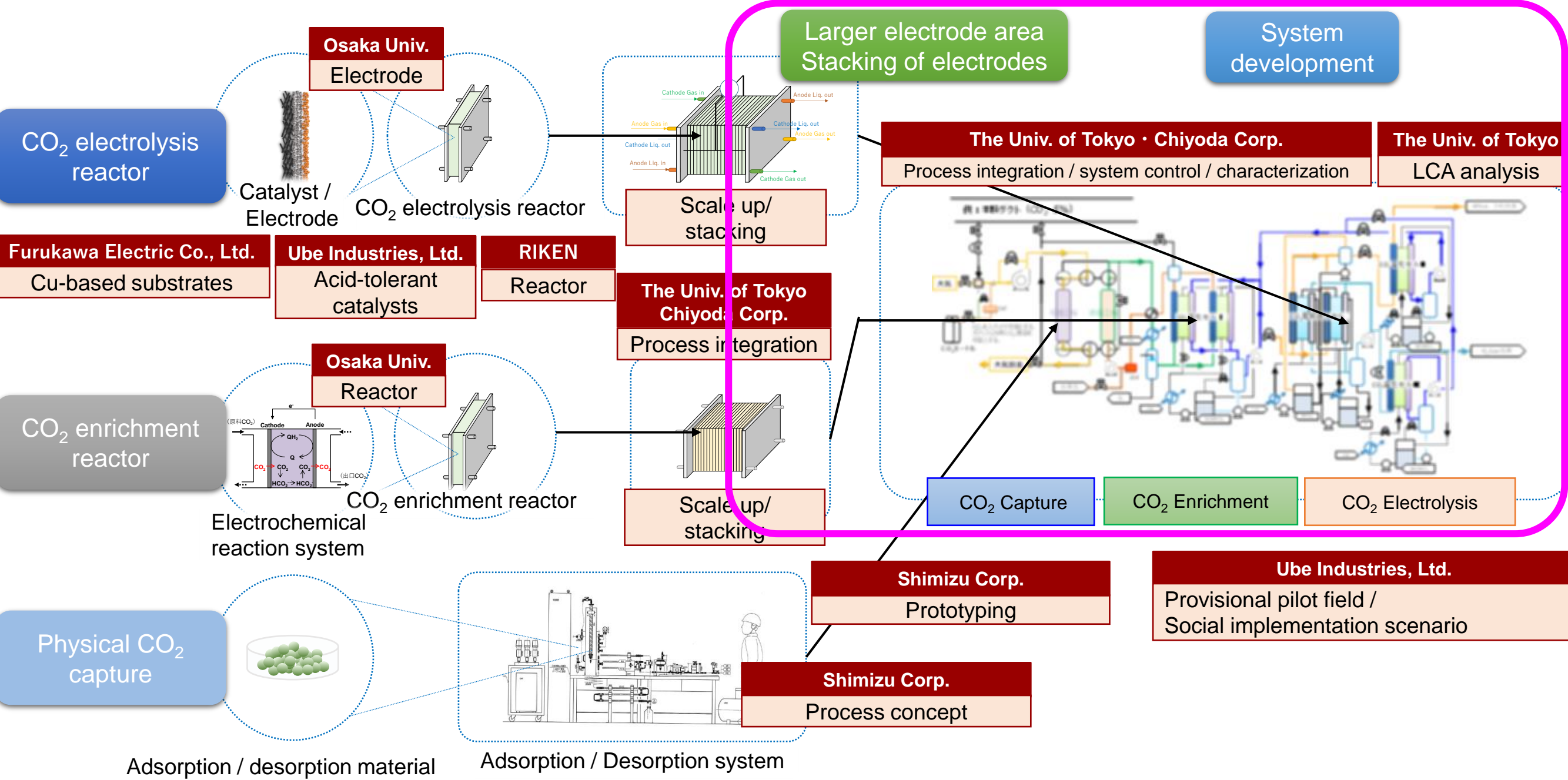
Integrated Electrochemical Systems for Scalable CO₂ Conversion to Chemical Feedstocks

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Implementing organizations : The University of Tokyo, Osaka University, Institute of Physical and Chemical Research (RIKEN), Ube Industries, Ltd., Shimizu Corporation, Chiyoda Corporation, Furukawa Electric Co., Ltd.

Project organization and goals



- Goals**
- Development of an integrated system that electrochemically converts CO₂ captured from an atmospheric air to valuable chemical substances
 - Conducting a life cycle assessment on a pilot-scale plant to evaluate the effectiveness as a measure against global warming

- **R&D items**

 - Analysis and control of the integrated system

 - Life-cycle assessment (LCA)

- **Major results so far**

- 1) Development of on-line analysis system for CO₂ electrolysis

 - ➔ Product analysis over days with a time resolution of minutes

- 2) Precise potential measurement of a gas-diffusion electrode (GDE)

 - ➔ Precise evaluation of the overpotential for CO₂ electrolysis

- 3) LCA for CO₂ footprint on the product C₂H₄

 - ➔ Clarification of the blanching point for a carbon-negative process

□ FY 24

Theoretical design of the operation of DAC, electrochemical CO₂ enrichment, and CO₂ electroreduction in series connection is completed.

Verify the feasibility of a lab-scale system consisting of an electrochemical CO₂ enrichment device and a CO₂ electroreduction device connected in series.

(Target continuous operation time: 1,000 hours, current density: 200 mA/cm², a current utilization efficiency : 50%)

LCA evaluation considering a wide range of process conditions.

□ FY 27

Prospects for 5,000 hours of continuous operation on a laboratory scale at a current density of 200 mA/cm² and a current utilization efficiency of 80% for the product will be determined. Also, obtain the necessary specifications for the pilot design.

LCA evaluation based on the pilot plant design is completed.

1) Development of on-line analysis system for CO₂ electrolysis



- Current: up to 2 A
- Electrolyte flow: 0~100 mL/min
- CO₂ flow: 0~100 mL/min
- Reaction temperature: 0~80°C

Gas product analysis:

- 2 minutes/analysis
- 1 ppm~100%

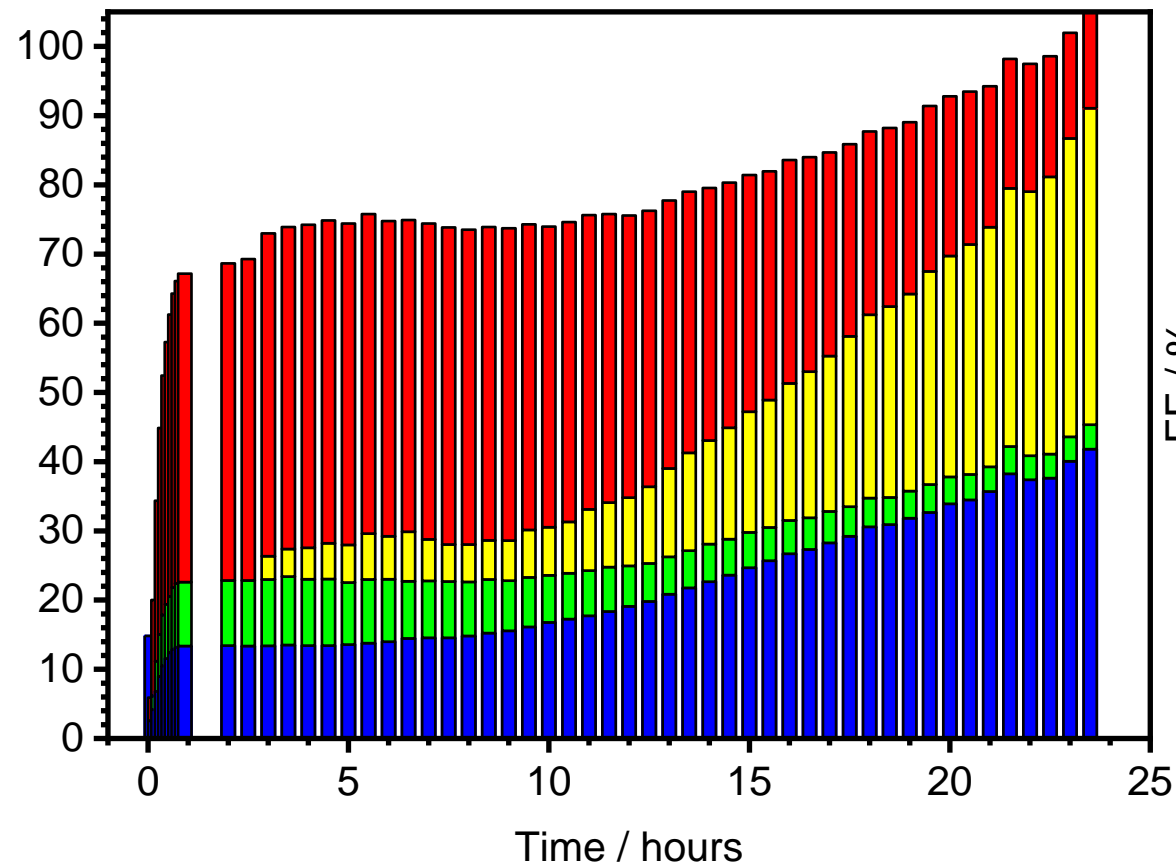


Agilent 490 Micro GC

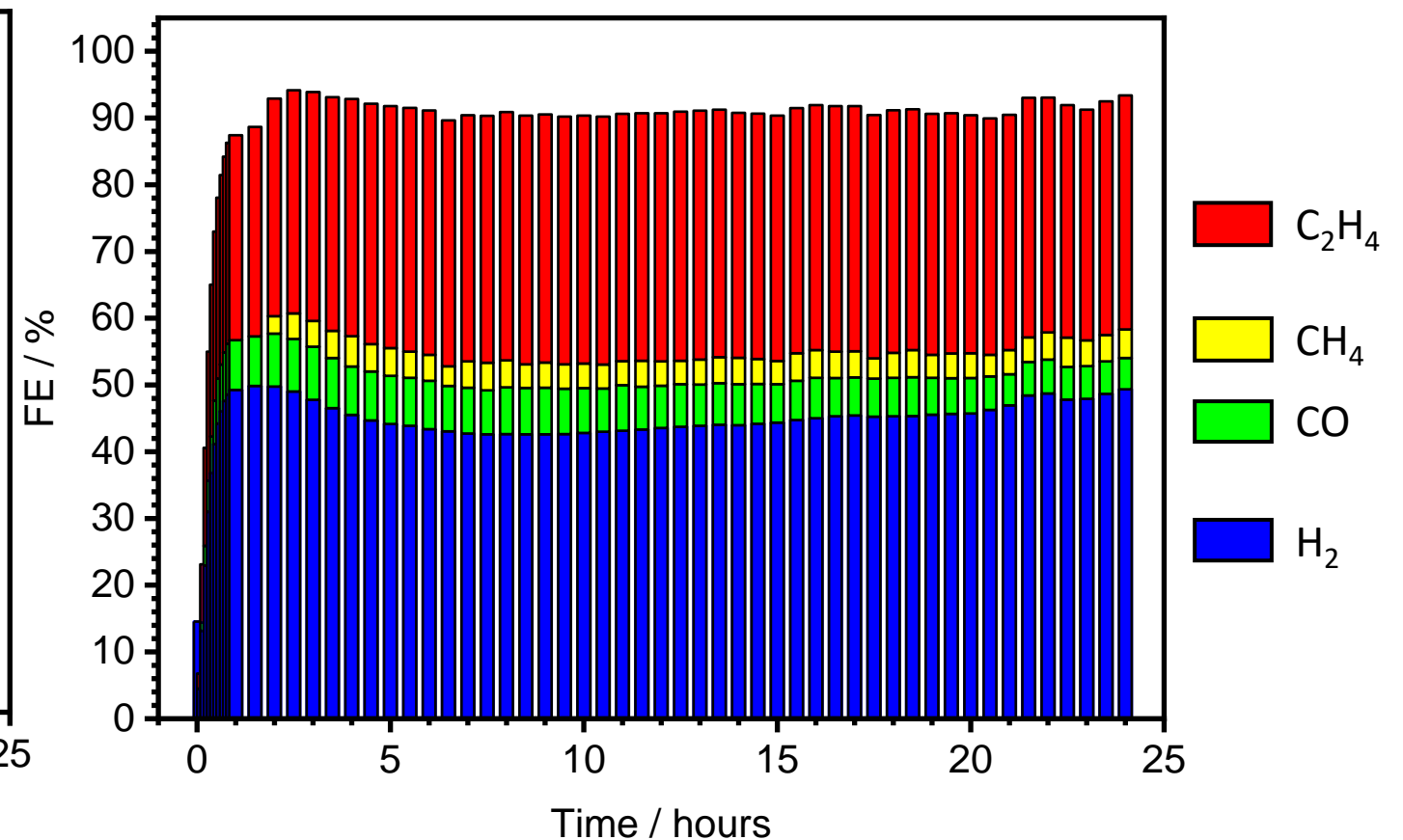
- 2 channels
Ch1: Ar, MS5A + Buckflush
Ch2: He, Pora Plot Q
Detectors: MicroTCD

Ex) CO₂ reduction using electrodeposited Cu₂O electrode

1 cycle electrodeposition



12 cycles electrodeposition



Deposition amount:

Small → High initial FE_{C₂H₄}, poor durability

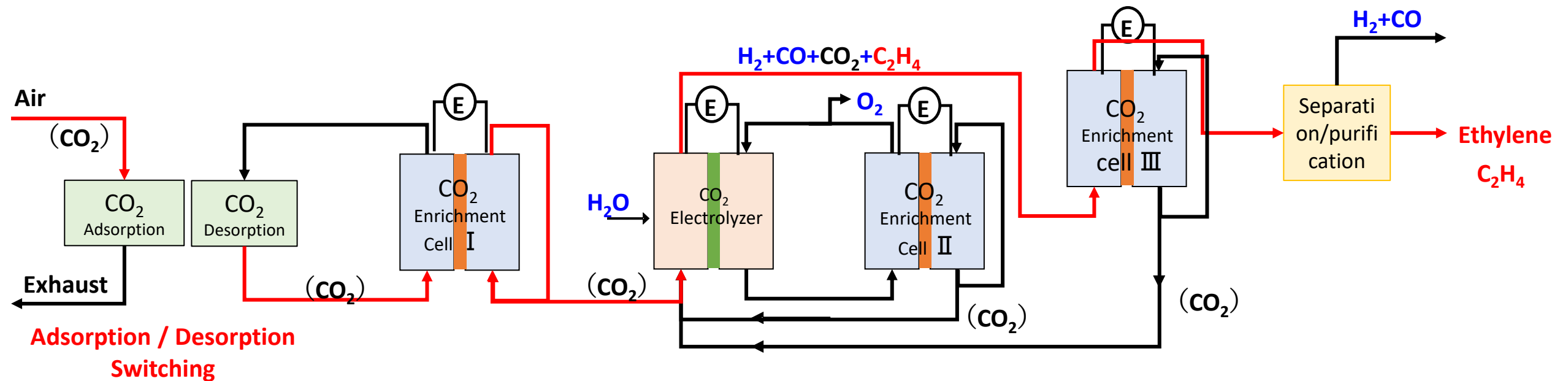
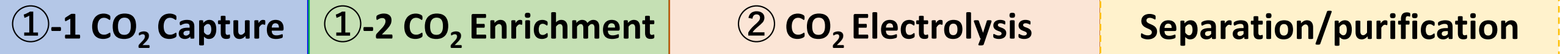
Large → Low initial FE_{C₂H₄}, better durability

Development of electrodes for both initial performance and durability.

➤ Necessity for novel analysis and materials

3) Integrated system design

Collaborative work with Chiyada corporation

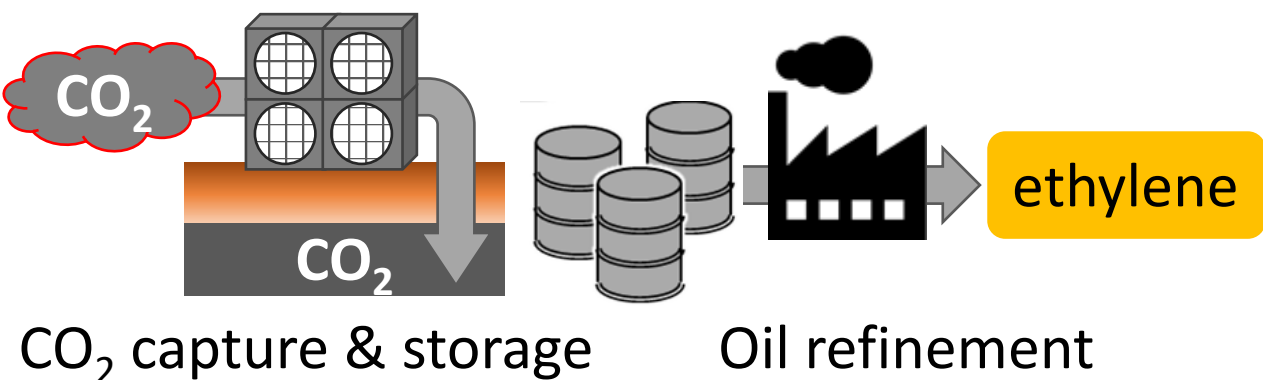


- ❑ Design of a system flow that integrates the elemental processes under development, from CO₂ capture to ethylene production.
- ❑ Material balance and heat balance are examined.
- ❑ Basic Study of LCA

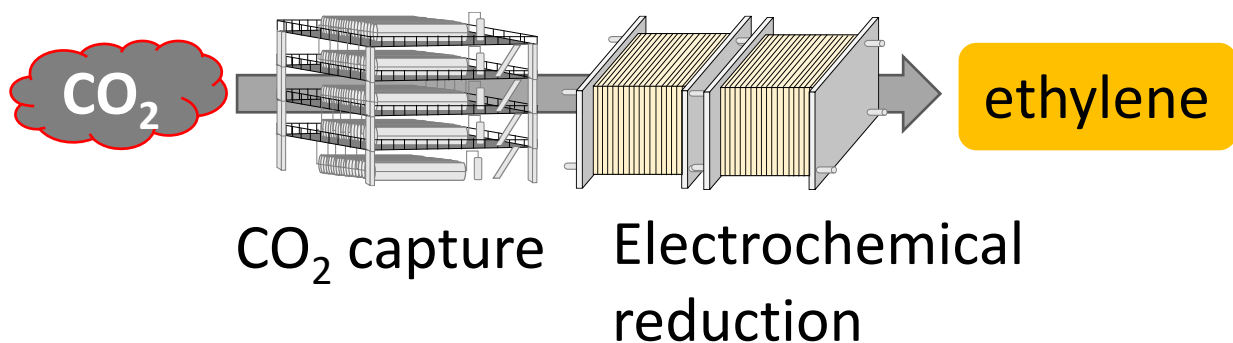
3) LCA

Integrated assessment of the environmental impact from production to disposal.

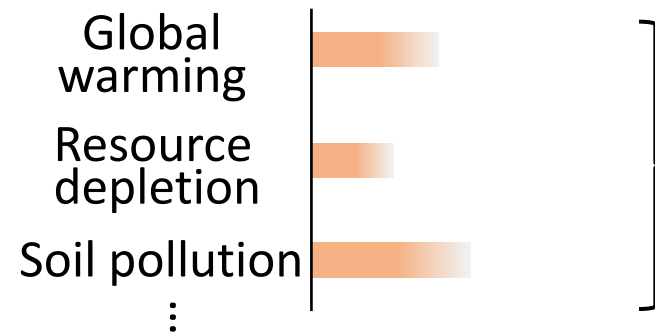
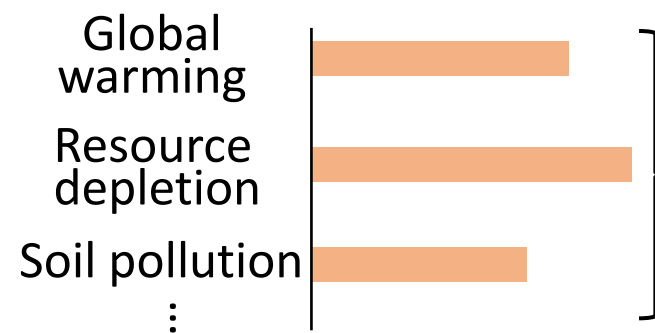
Existing technologies



This project



LCA



Compare environmental impact with existing technologies

Current results

- Confirmed that the electrochemical reduction process has the greatest impact of all processes.
- Clarify the operating conditions necessary to achieve carbon negativity (CO₂ emissions < CO₂ fixation)

