

# Innovative Circular Technologies for Harmful Nitrogen Compounds/ To Solve Planetary Boundary Issues

Project outline

Theme 3 R&D items and achievement

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PM : Dr. Dr. KAWAMOTO Tohru , National Institute of Advanced Industrial Science and Technology (AIST)

Implementing organizations : National Institute of Advanced Industrial Science and Technology (AIST),

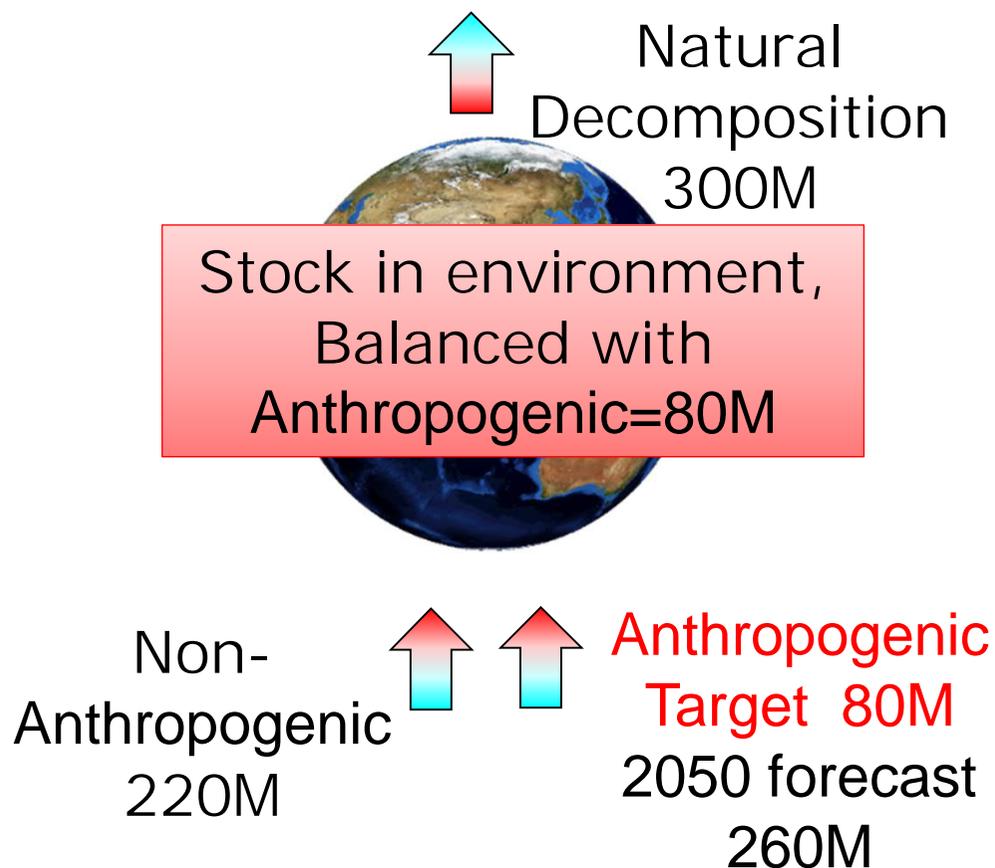
The University of Tokyo, Waseda University,

Tokyo University of Agriculture and Technology, Kobe University,

Osaka University, Yamaguchi University, Kyowa, Hakko Bio Co., Ltd.,

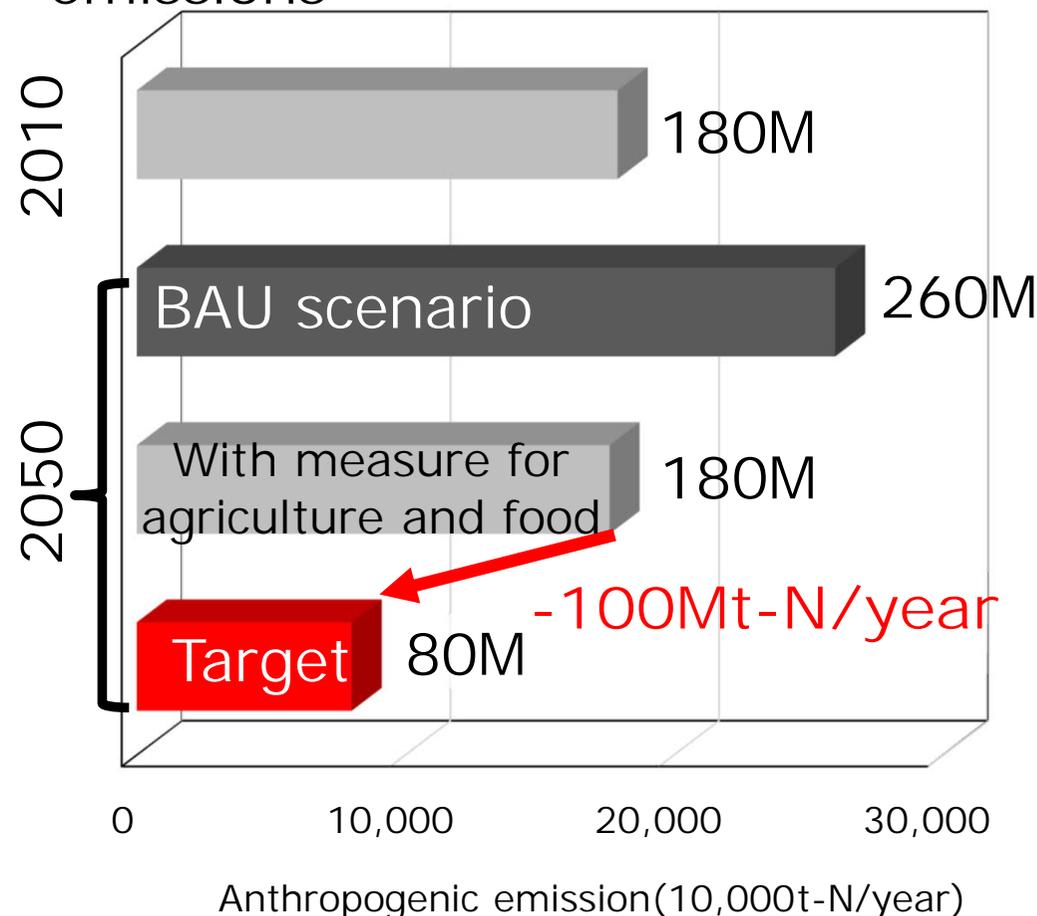
ASTOM Corporation, Toyobo Co., Ltd., FUSO Corporation, Ube Industries, Ltd,

(a) Target of Nitrogen compounds emission (t-N/year)

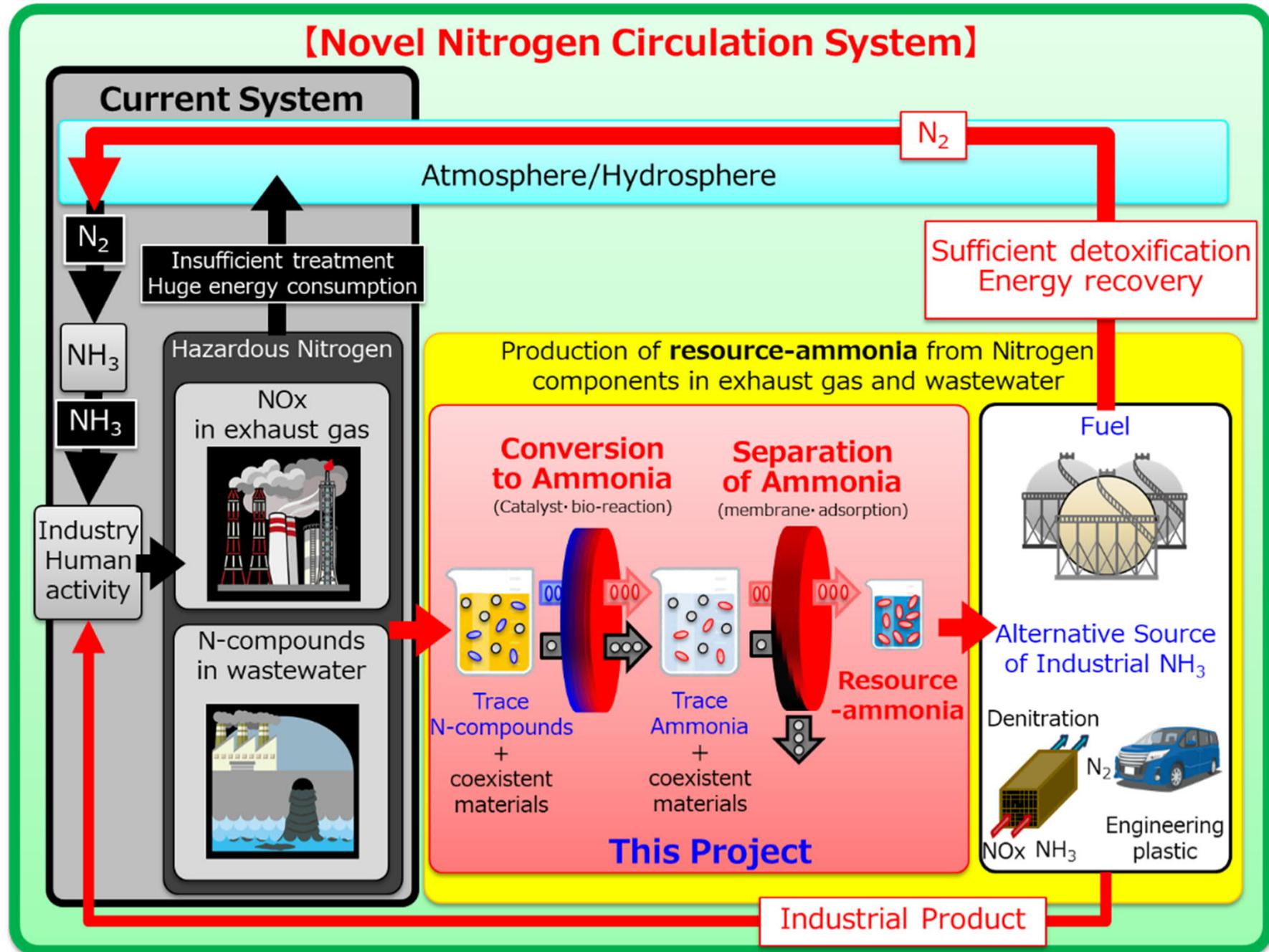


*Galloway, Biogeochem. (2004),  
Fowler, Philos. Trans. R. Soc. B Biol. Sci. (2013)  
de Vries, Curr. Op. Env. Sus. (2013)*

(b) Current status and target for anthropogenic nitrogen compound emissions

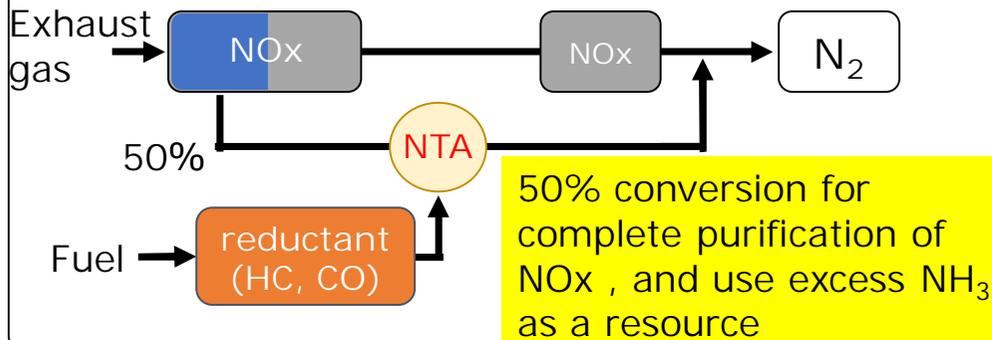


**Additional 100Mt-N/year reduction is necessary**



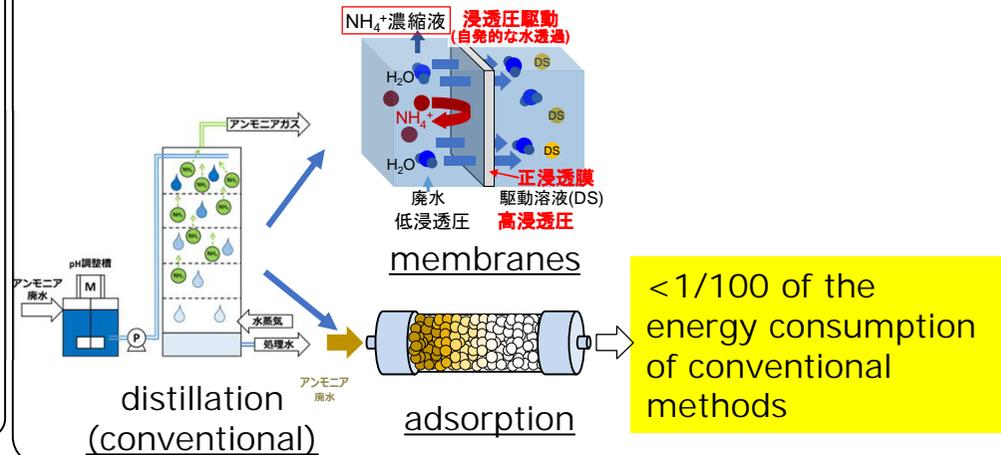
## 【1. NOx to ammonia】

- ① concentration of NOx·NH<sub>3</sub> in gas phase
- ② catalyst for NOx→NH<sub>3</sub> at low temperature
- ③ catalyst even with O<sub>2</sub> coexistence



## 【2-2. ammonia recovery in water】

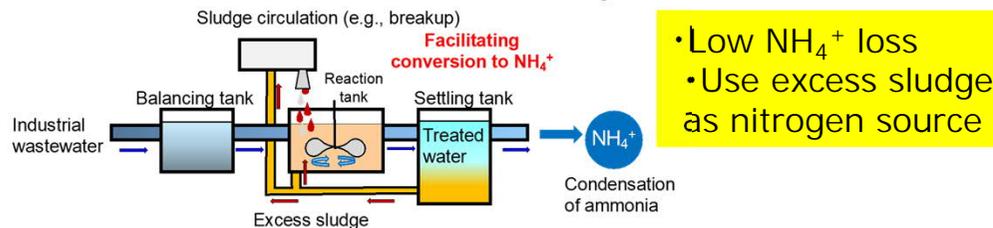
- (1) new membrane/adsorption separation
- (2) Construction of ultra energy-saving separation and concentration process



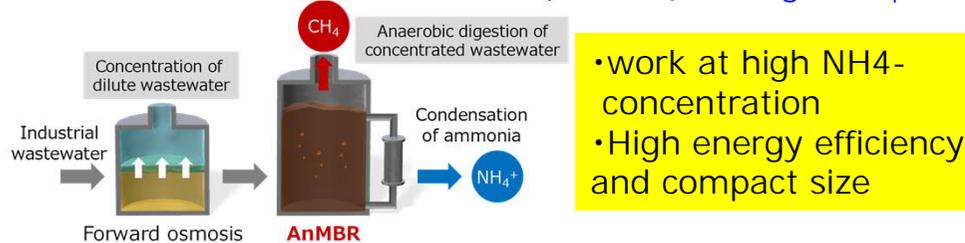
## 【2-1. conversion to NH<sub>3</sub> in wastewater】

Efficient NH<sub>4</sub><sup>+</sup> conversion bioprocess for various conditions

### ● microaerobic conversion to NH<sub>4</sub><sup>+</sup>

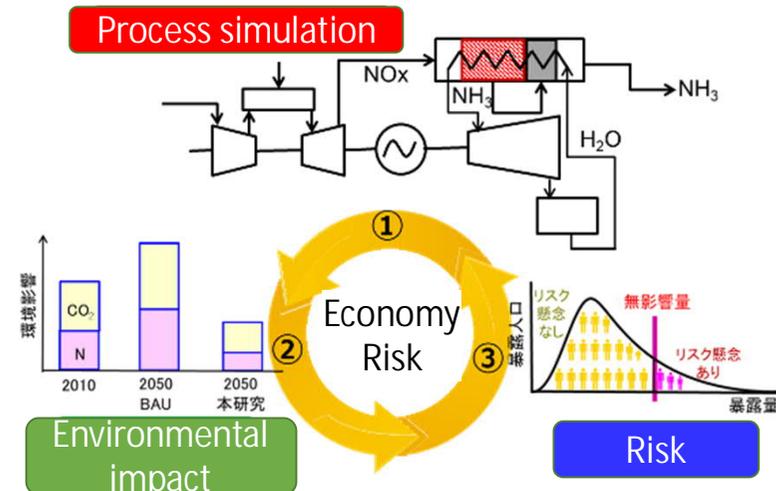


### ● anaerobic membrane bioreactor (AnMBR) for high-NH<sub>4</sub><sup>+</sup>



## 【Theme 3. Process and evaluation】

- 1) Process design of actual equipment and pilots
- 2) economic and environmental impact assessment.



Time	Theme	Speaker, Affiliation
10:00	Outline & Theme 3	Dr. T. Kawamoto, AIST(1)
10:06	Theme 1. NO <sub>x</sub> to Ammonia	Prof. M. Ogura, Univ. Tokyo
10:15		Prof. M. Iwamoto, Waseda Univ.
10:21		Dr. M. Tanaka, Ube Industries, Ltd.
10:24	Theme 2-1. conversion of nitrogen compounds to NH <sub>4</sub> <sup>+</sup>	Dr. T. Hori, AIST(2)
10:28		Prof. A. Terada, TUAT
10:34		Dr. R. Ohashi, Kyowa Hakko Bio Co., Ltd.
10:37		Prof. M. Ike, Osaka Univ.
10:46	Theme. 2-2. NH <sub>4</sub> <sup>+</sup> recycling by Separation and concentration	Prof. H. Matsuyama, Kobe Univ.
10:58		Mr. H. Sakurai, Toyobo Co., Ltd.
11:02		Prof. M. Higa, Yamaguchi Univ.
11:07		Mr. S. Doi, ASTOM corporation
11:10		Dr. T. Kawamoto, AIST(3)
11:17		Dr. K. Kogure, Fuso Corporation
11:20		Q&A

A more detailed overview will be given at 14:35 in Room 1.

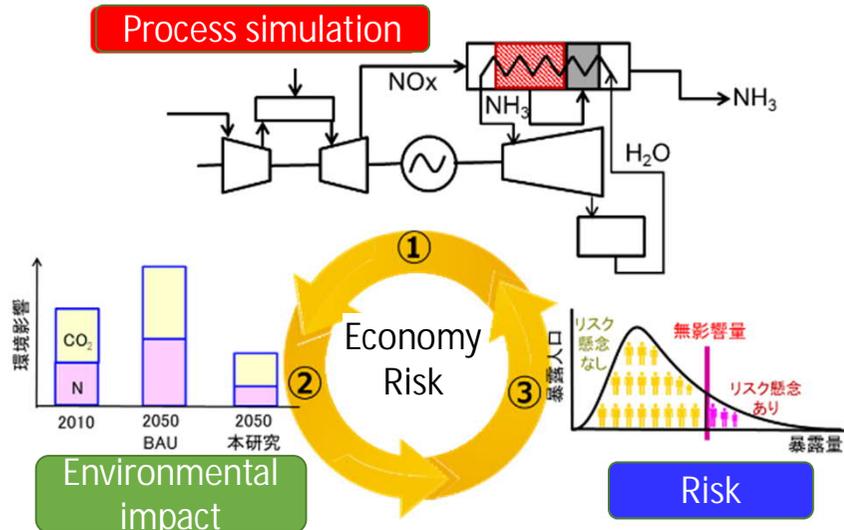
- Evaluation of the effects our technologies
- Clarification the issues in the practical application of nitrogen cycle technologies.

Optimal combination of each elemental technology and constructed a practical system.

## 【Required technologies】

- 1) Management of R&D data and models
- 2) Estimation of various environmental impacts
- 3) Risk estimation of nitrogen compounds

### 1. Process design of practical equipment



2. Various cost evaluation (cost, CO<sub>2</sub>, N emission etc.)

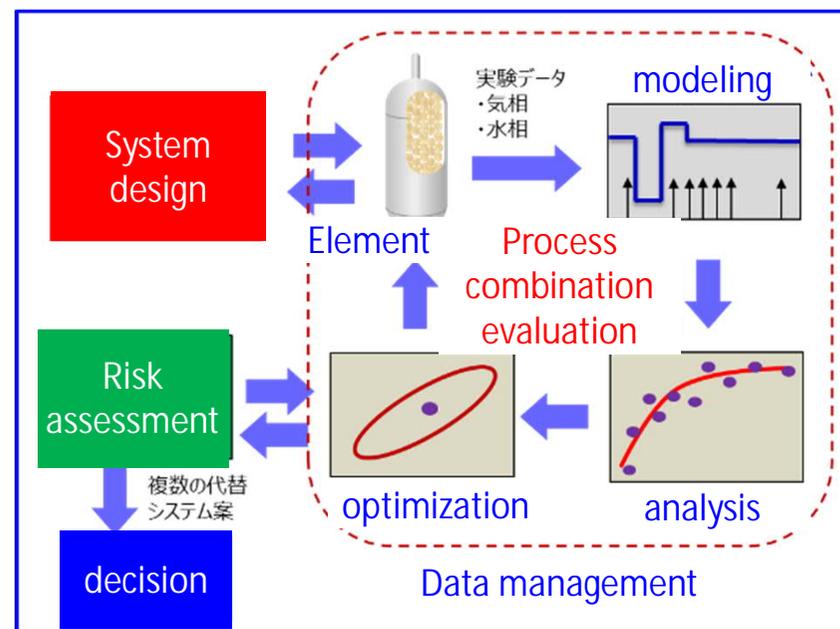
③Risk assessment of implementation of new technologies

【Method】 Construction of a frame for the comprehensive evaluation of the R&D

①Managing R&D data and models for nitrogen cycle technologies

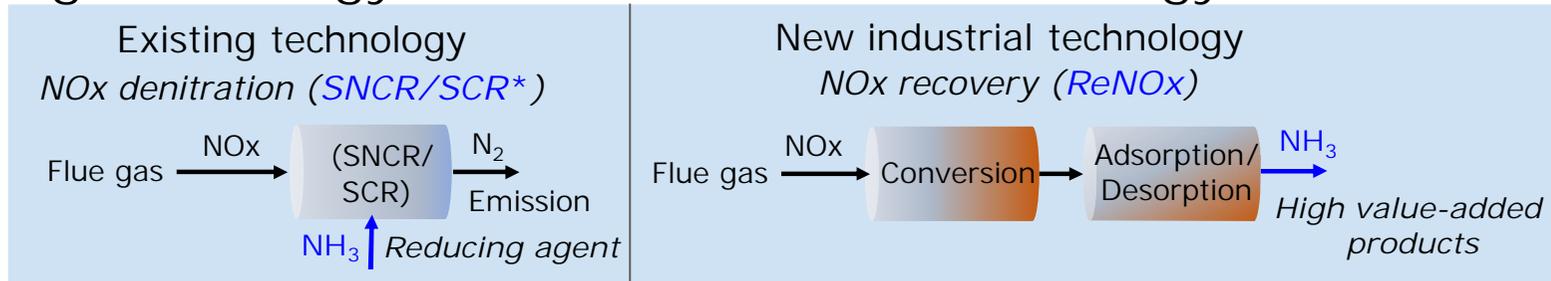


②Unique nitrogen inventory data, and development of a regional atmospheric chemical transport model applicable to local and urban scales



- Conduct life cycle assessment of NOx recovery, compare with existing NOx detoxification.
- The results indicated the effect that NOx recovery can reduce the environmental impacts.

## Existing Technology vs New Industrial Technology



\*SNCR : selective non-catalytic reduction  
SCR : selective catalytic reduction

## Life cycle assessment

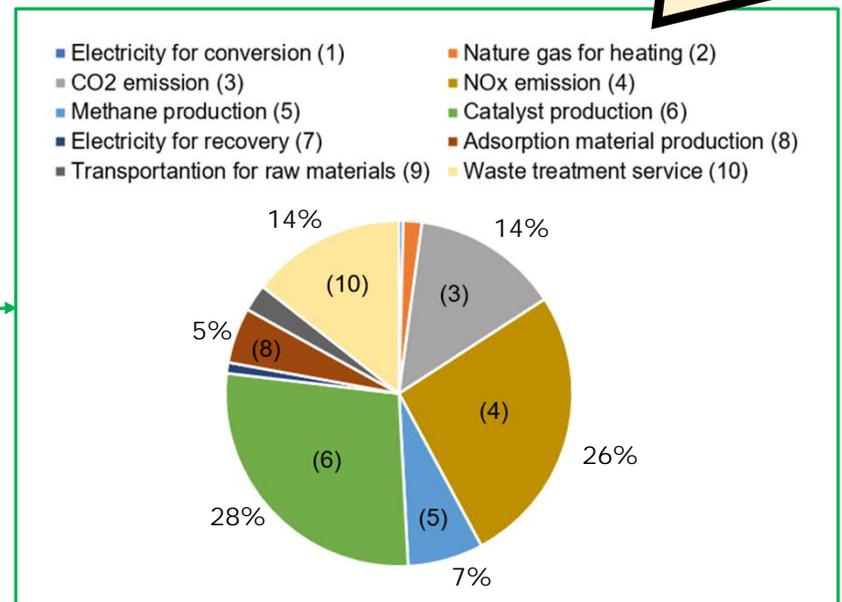
Damages (LIME2, 4 categories)

SNCR	3.64E-03	2.94E+04	1.82E-11	8.36E+01
SCR	1.39E-03	1.34E+04	3.92E-10	5.43E+01
ReNOx	2.29E-04	1.61E+03	9.85E-11	6.16E+00
	human health (DALY)	social assets (JPY)	biodiversity (ENES)	Primary production (kg DW eq.)

Integration (LIME2)

SNCR	8.58E+04
SCR	4.19E+04
ReNOx	7.16E+03
	eco-index (JPY)

Driving factor analysis for ReNOx



Catalyst production & NOx emission are critical for ReNOx.

The integration assessment revealed that the environmental impact of ReNOx is the minimum.

Maximum (Red)  
Middle (Yellow)  
Minimum (Green)

## 【Background】

- Release of nitrogen compounds into the environment is one of the biggest challenges in the planetary boundary → In terms of recoverability, it is a greater risk than climate change, etc. In order to solve this issue, it is necessary to reduce emissions from industry and daily life by 100 million tons.

## 【 Objective of the research and development 】

- Development of technology to reduce emissions of nitrogen compounds by 100 million tons in 2050.

## 【 Development items 】

- Technology to convert NO<sub>x</sub> in exhaust gas into ammonia for detoxification and conversion to resources
- Technology to convert nitrogen compounds in wastewater into ammonia and convert them into resources
- Evaluation of the effectiveness of the developed technology.

## 【 Results (Theme 3) 】

- Conduct life cycle assessment of NO<sub>x</sub> recovery, compare with existing NO<sub>x</sub> detoxification.
- The results indicated the effect that NO<sub>x</sub> recovery can reduce the environmental impacts.

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