

Development of Recovery and Removal Techniques of Dilute Reactive Nitrogen to Realize Nitrogen Circulating Society



■ P M Toru Wakihara

The University of Tokyo, Professor

■ PJ participating institutions

The University of Tokyo

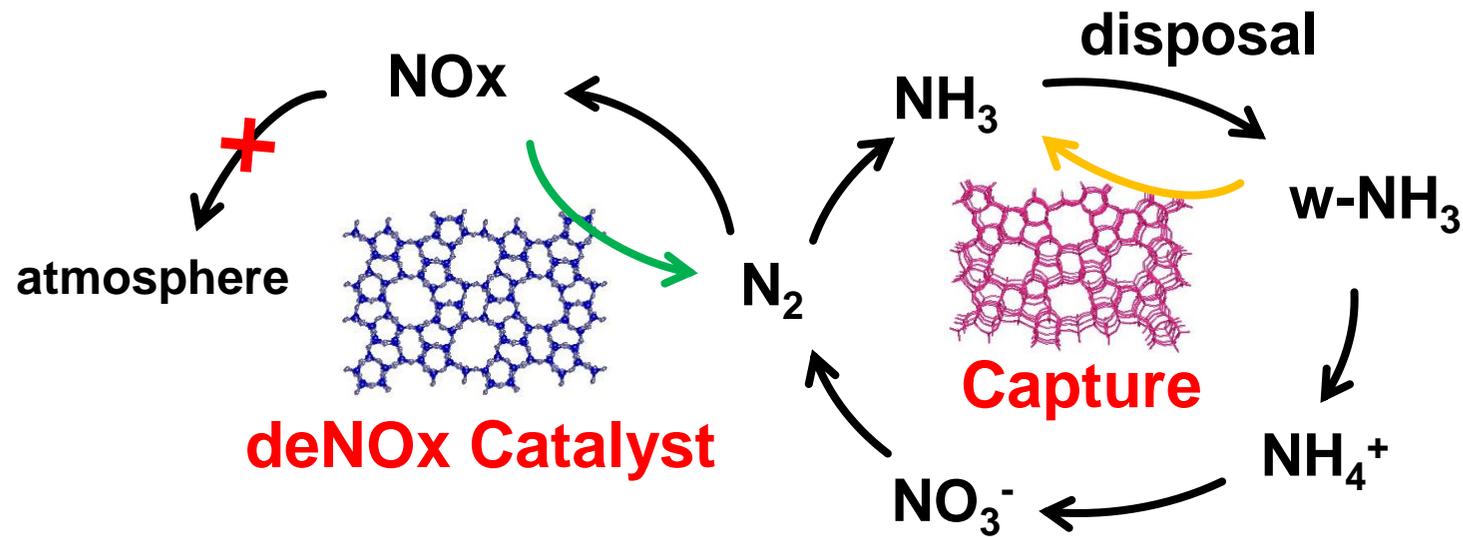
The National Institute of Advanced Industrial Science and Technology

Japan Fine Ceramics Center

Mitsubishi Chemical Corporation

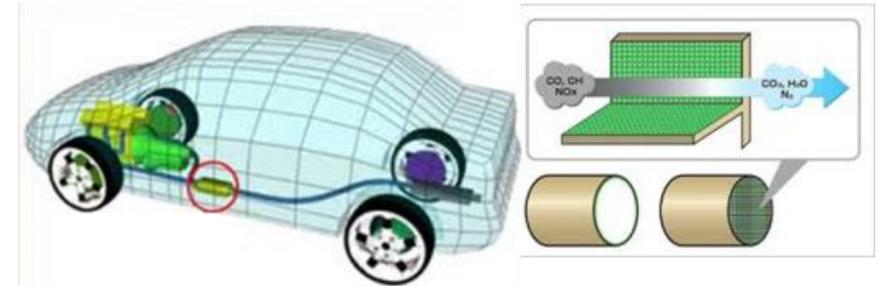


Project Overview



For building a nitrogen recycling society, development of **denitrification** and **ammonia recovery** technology is an urgent issue

Exhaust Gas (NOx)



Industrial Wastewater (w-NH₃)



- ◆ Although the transition to electric vehicles has been proposed for the realization of a carbon-neutral society, in Europe, reluctant to fully transition to electric vehicles.
- ◆ Considering the introduction of e-fuel, an internal combustion engine (especially for truck transportation) is essential.
- ◆ Truck-mounted catalyst does not need to be replaced even after running 1 million km → Cost reductions, wage increases, etc. are expected
- ◆ From the viewpoint of the nitrogen cycle, Realization of breaking away from the present treatment system wasting energy (industrial waste liquid, livestock farm, sewage treatment plant)
- ◆ Cost reduction by reducing manufacturing cost of urea for fertilizer by reusing recovered NH₃

Final Aim

- **Demonstration of NH₃ recover from wastewater at pilot facilities**
- **Pilot scale test using zeolite for high durability NOx purification**
- **Demonstration of NOx purification without NH₃**



Organization of the Project (at the early stage)

Institutions

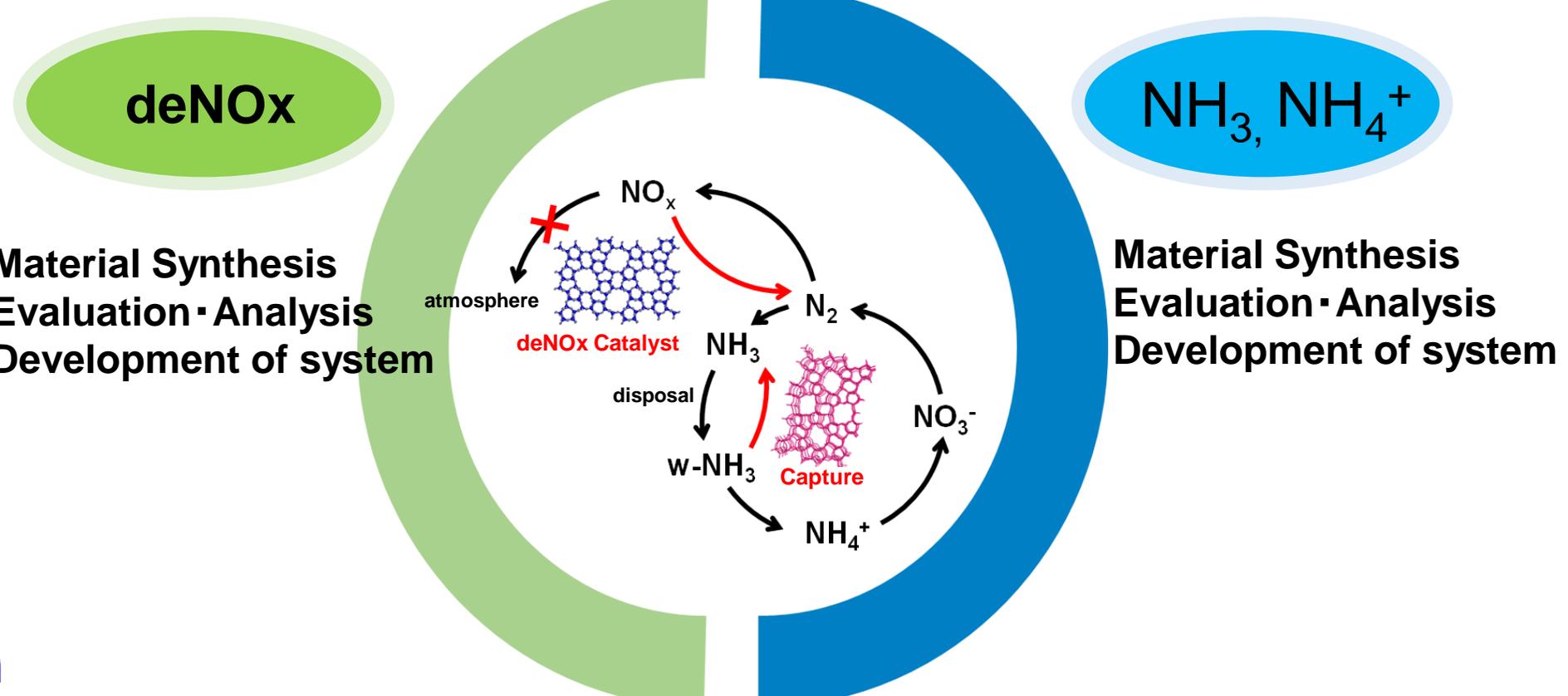
The Univ. of Tokyo

MCC

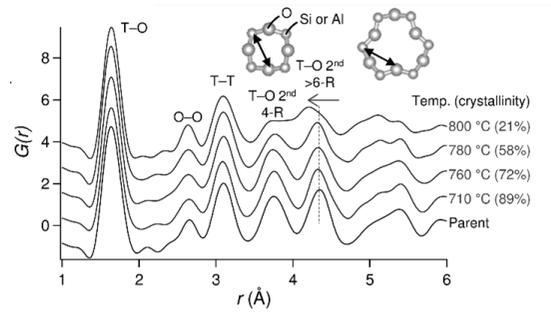
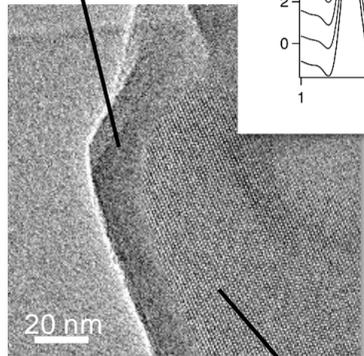
AIST

JFCC

Research Direction



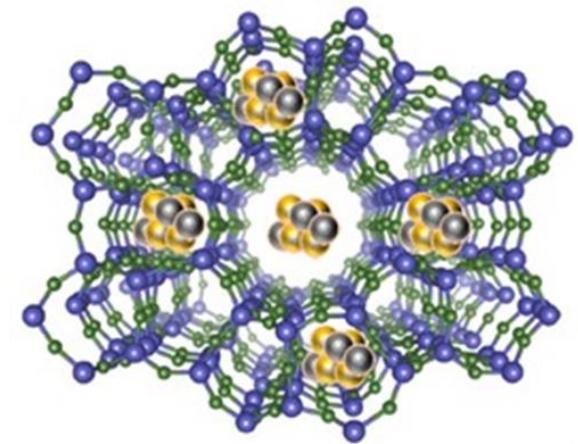
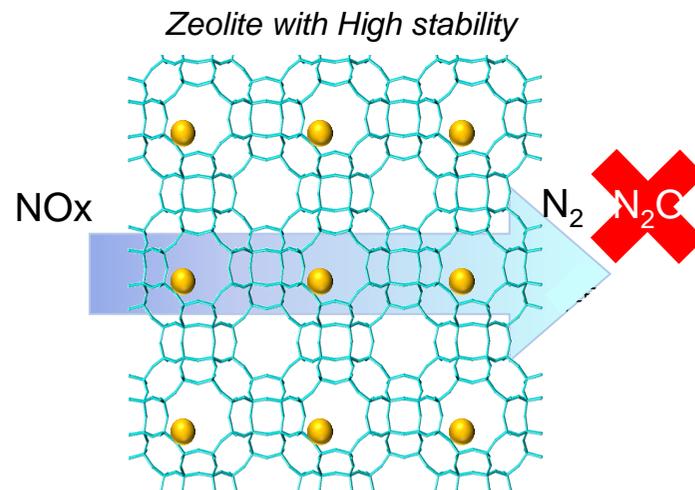
Amorphous Phase



Deactivation mechanism (at atomic scale)

Development of excellent catalyst

Social implementation



Metal Nanocluster Zeolites

New Active Sites

Focus on only zeolites



Organization and core technology (deNOx)

The Univ. of Tokyo

- ✓ Synthesis
- ✓ New System
- ✓ LCA evaluation

JFCC
AIST

- ✓ Analysis
- ✓ Evaluation

University of Alicante (Spain)

- ✓ Excellent catalyst

Budding

- Photocatalyst
- Plasma reaction
- Defect healing
- Dealumination

Core

- Fast Synthesis
- Flow synthesis
- Defect healing
- Dealumination

MCC

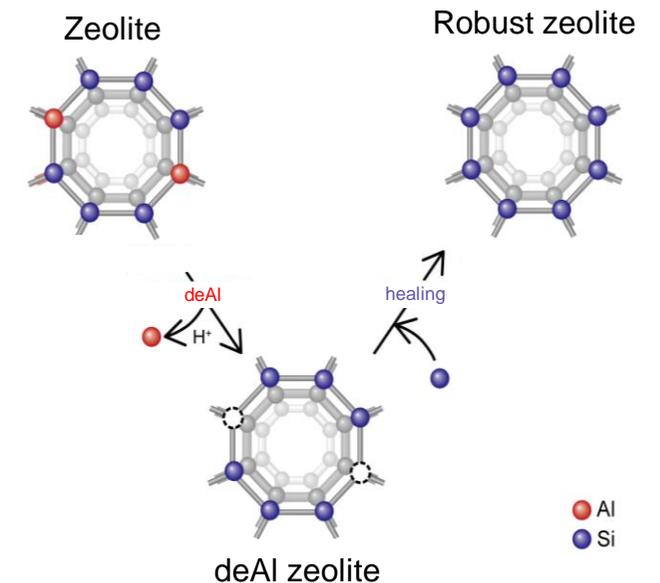
Company A
Company B

- ✓ Scale up
- ✓ Production – Sale

Company C

- ✓ Molding

- ✓ Fast synthesis of Excellent zeolite catalyst
→ High conversion · Low N₂O evolution!





Organization and core technology (NH₃ capture)

The Univ. of Tokyo

- ✓ Synthesis
- ✓ New System
- ✓ LCA evaluation

JFCC

- ✓ Analysis
- ✓ Evaluation

Nazarbayev University (Kazakhstan)

- ✓ NH₃ removal using natural zeolite and amorphous aluminosilicate

Company A
Company B
Company C

- ✓ Production
- ✓ Sale

Company D
Company E
Company F

- ✓ Recycle
- ✓ Urea Production

Company G

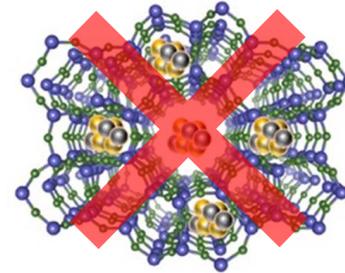
- ✓ Equipment design

Company H

- ✓ Implementation

Company I

- ✓ Molding



Synthetic zeolite

Budding

- Amorphous aluminosilicate
- Waste material

Core

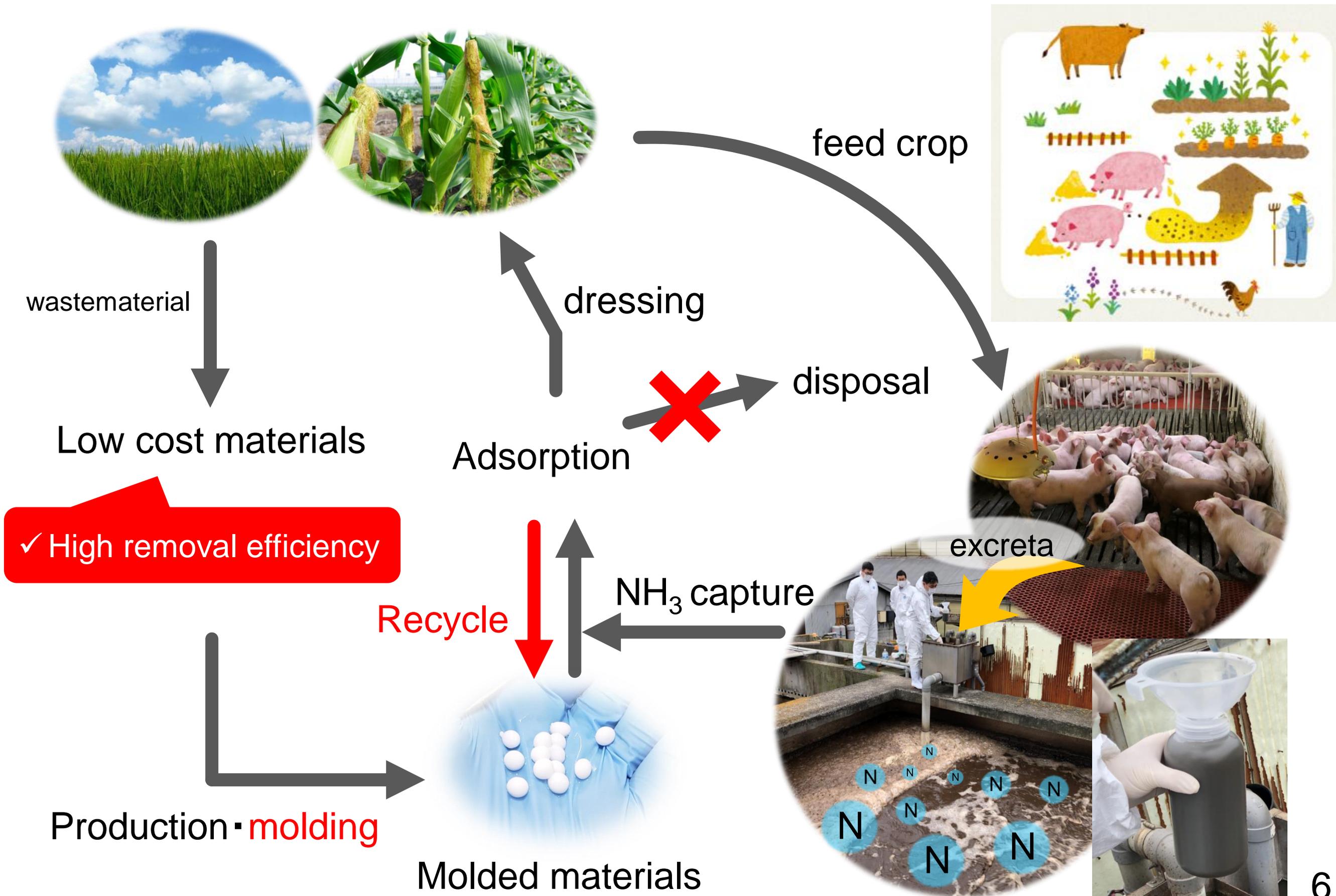
- Amorphous aluminosilicate
- Waste material
- Natural zeolite



- ✓ Low-cost • facile process
 - ✓ High removal efficiency
- **Regional nitrogen circulation system**



Regional nitrogen circulation system





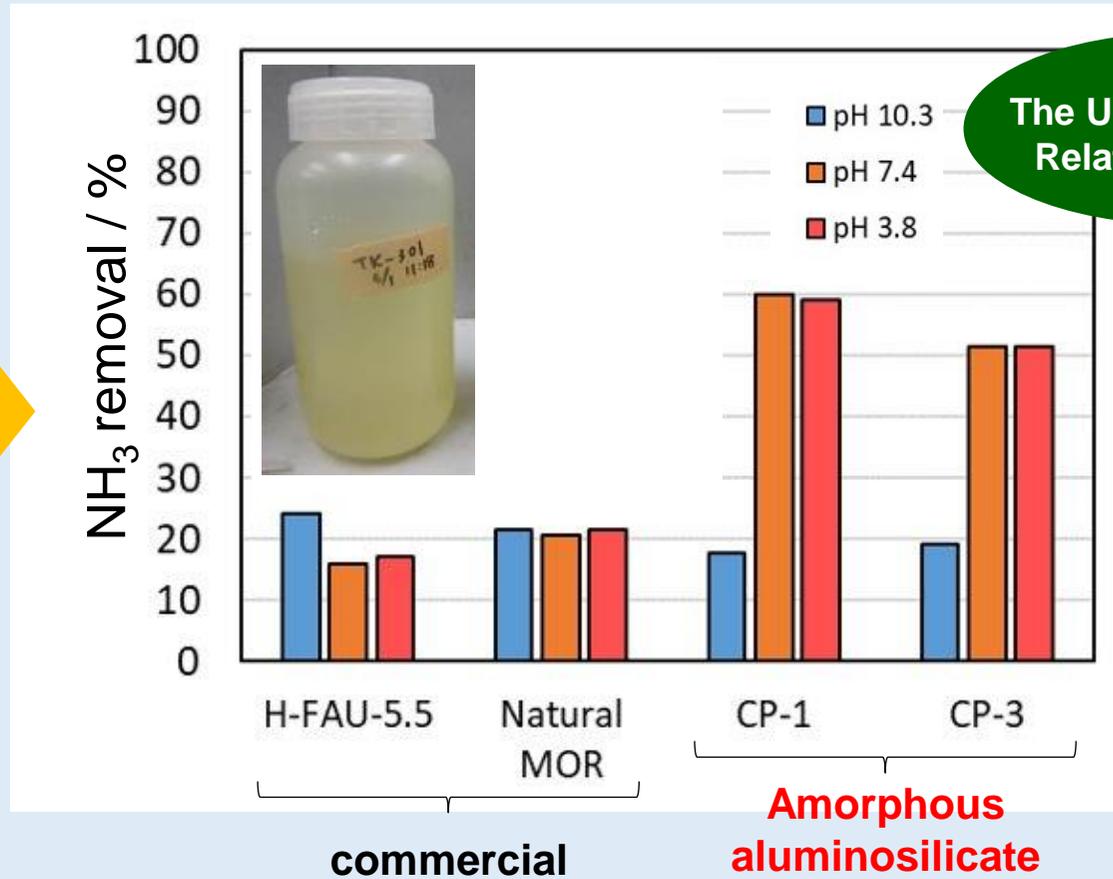
NH₃ capture

Recovery and recycling from industrial waste liquids

Related Company
Company A-C



exhaust



The Univ. of Tokyo
Related Company

- ✓ Highly concentrated (1000-2000 ppm NH₄⁺)
- ✓ Ammonia odor spreads to the floor in factory

Many similar companies

As fuel

200 ton / year

Related Company

Compost to the market

Related Company

Distilled concentrated NH₃ water

✓ Aim: NH₃ removal over 50% from industrial wastewater

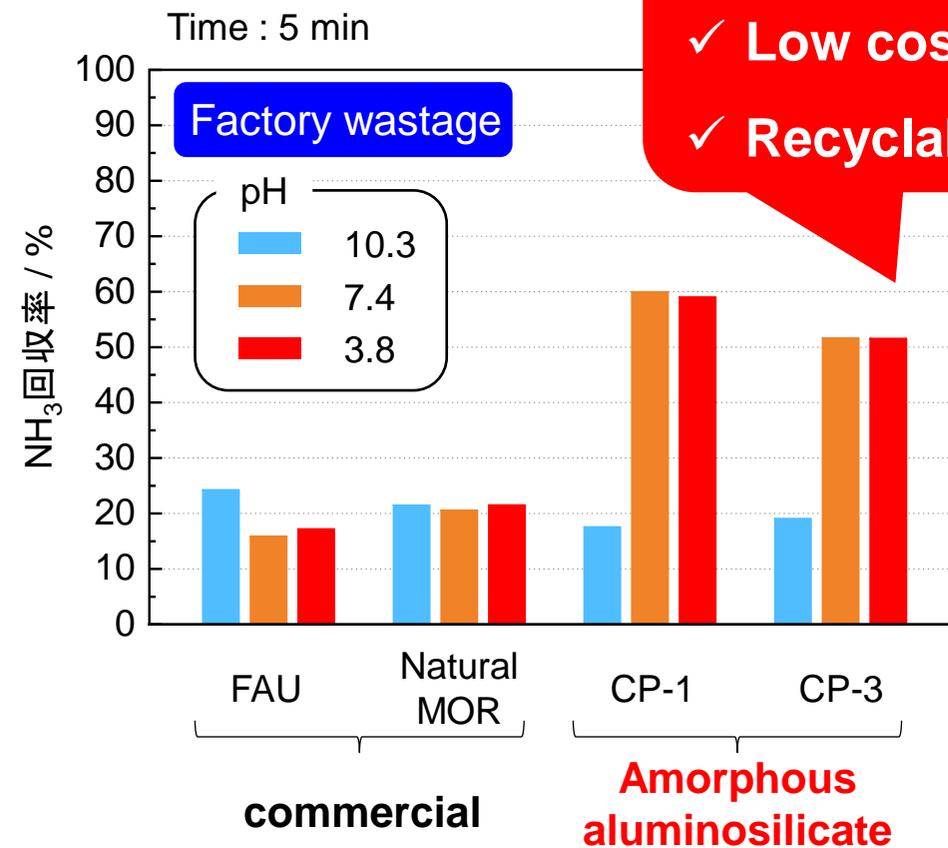
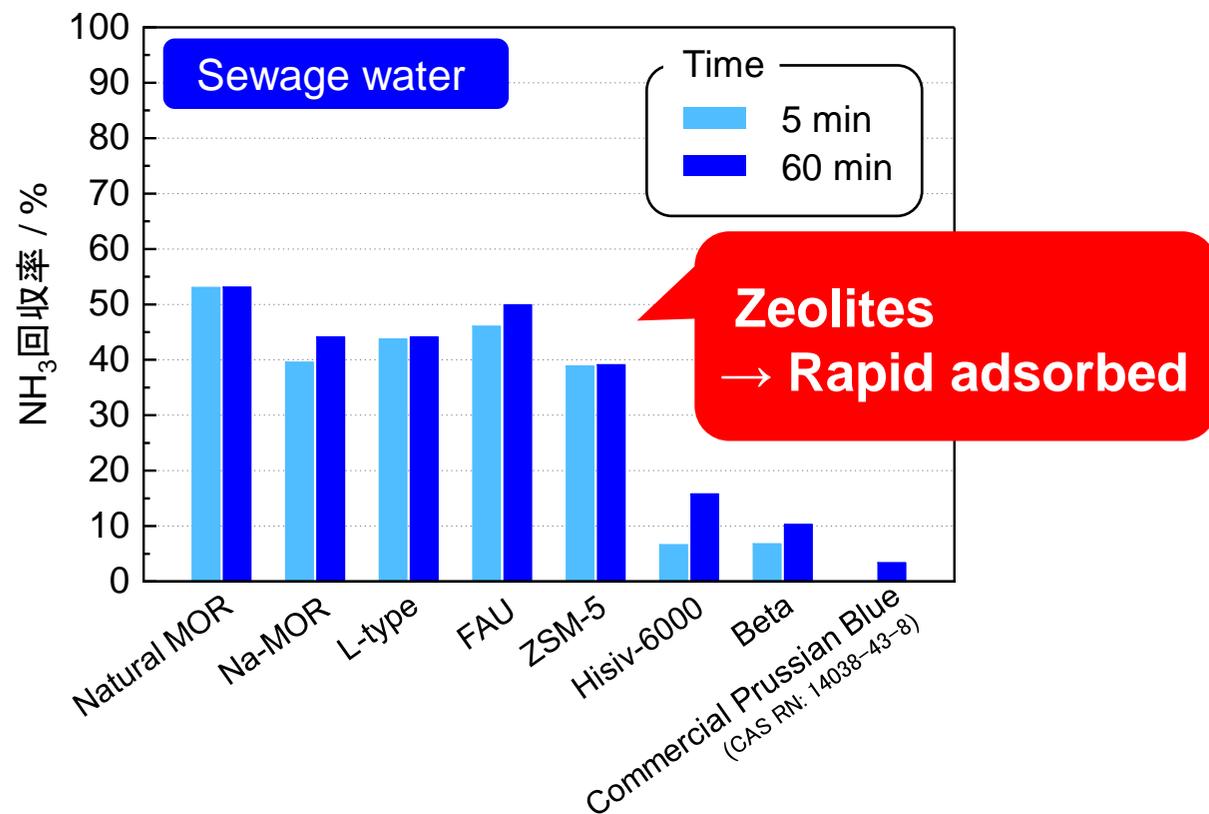


Selection of Materials

List of industrial wastewater

	Sample	NH ₄ ⁺ concentration / mM
Sewage water	Position A	1.7~2.3
	Position B	1.6~1.9
	Activated sludge stripper	75
Swine wastewater	-	110
Factory wastage	Company A	70
	Company B	12

NH₃ removal efficiency

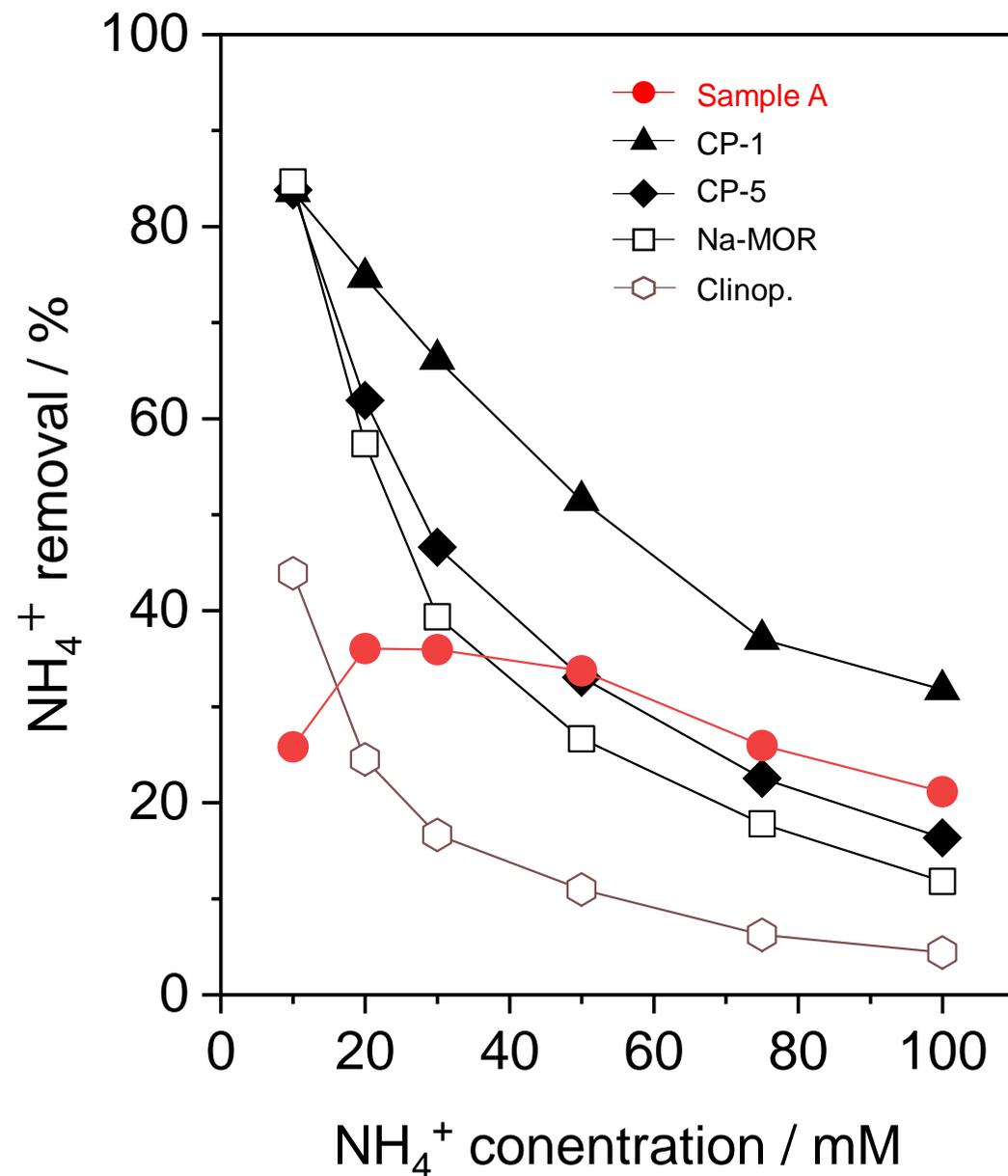


- ✓ High efficiency !!
- ✓ Low cost (CP-3) !
- ✓ Recyclable

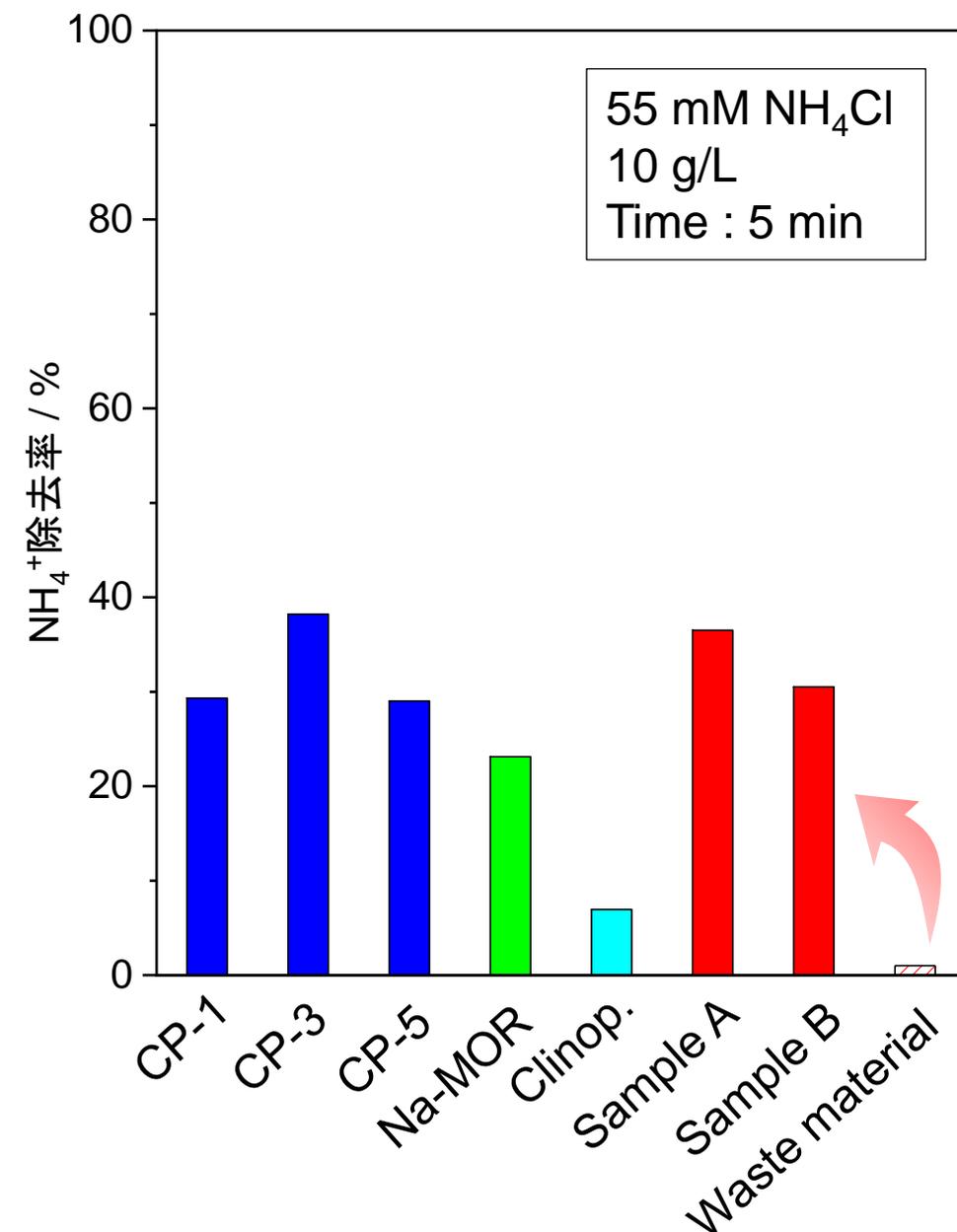


NH₃ removal efficiency

Effect of NH₄⁺ concentration



Various adsorbents

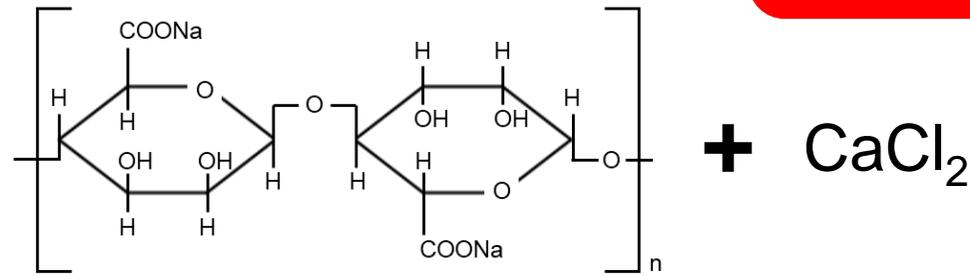


✓ High NH₃ removal (similar to CP samples)



Molding - Continuous flow system

Molding method



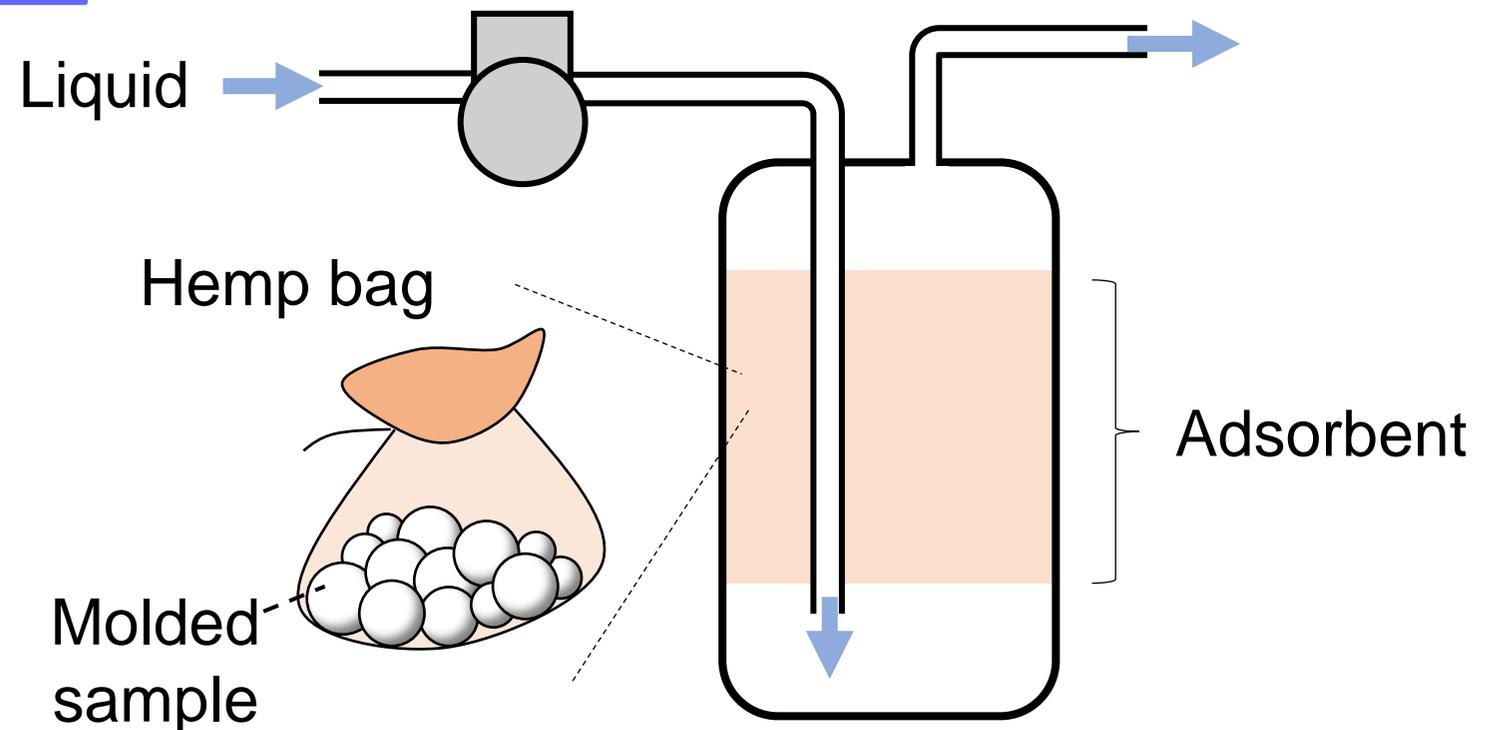
✓ Low temperature
→ Low cost !



Continuous Flow System



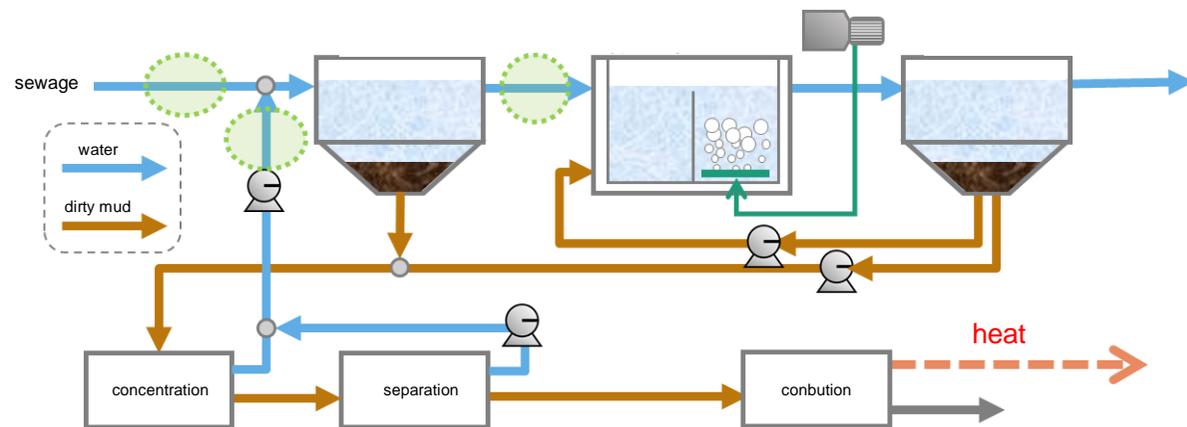
✓ able to adsorb NH_3



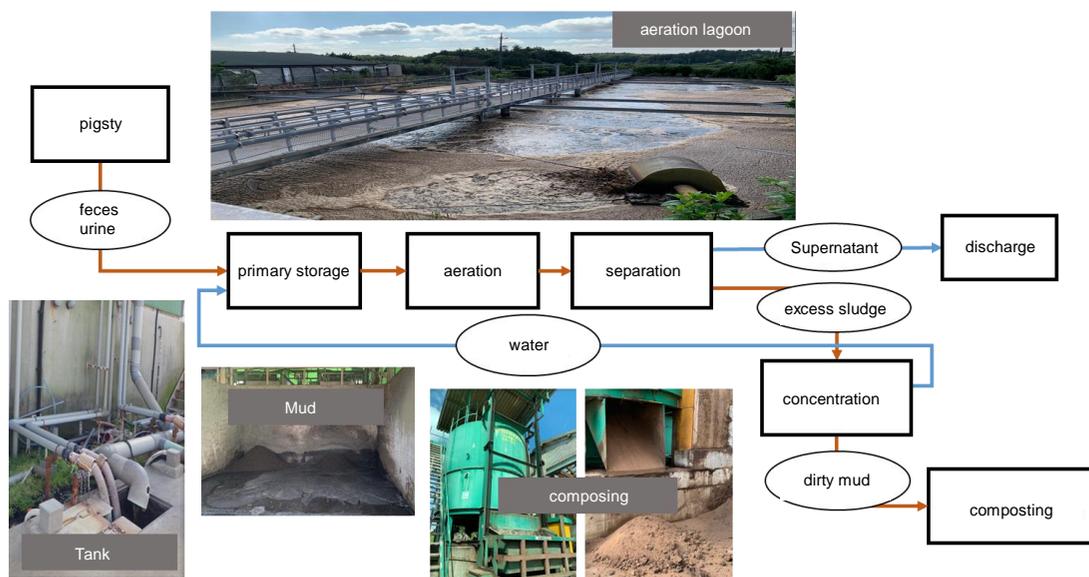


LCA (NH₃ capture)

Sewage treatment



Livestock Wastewater Treatment



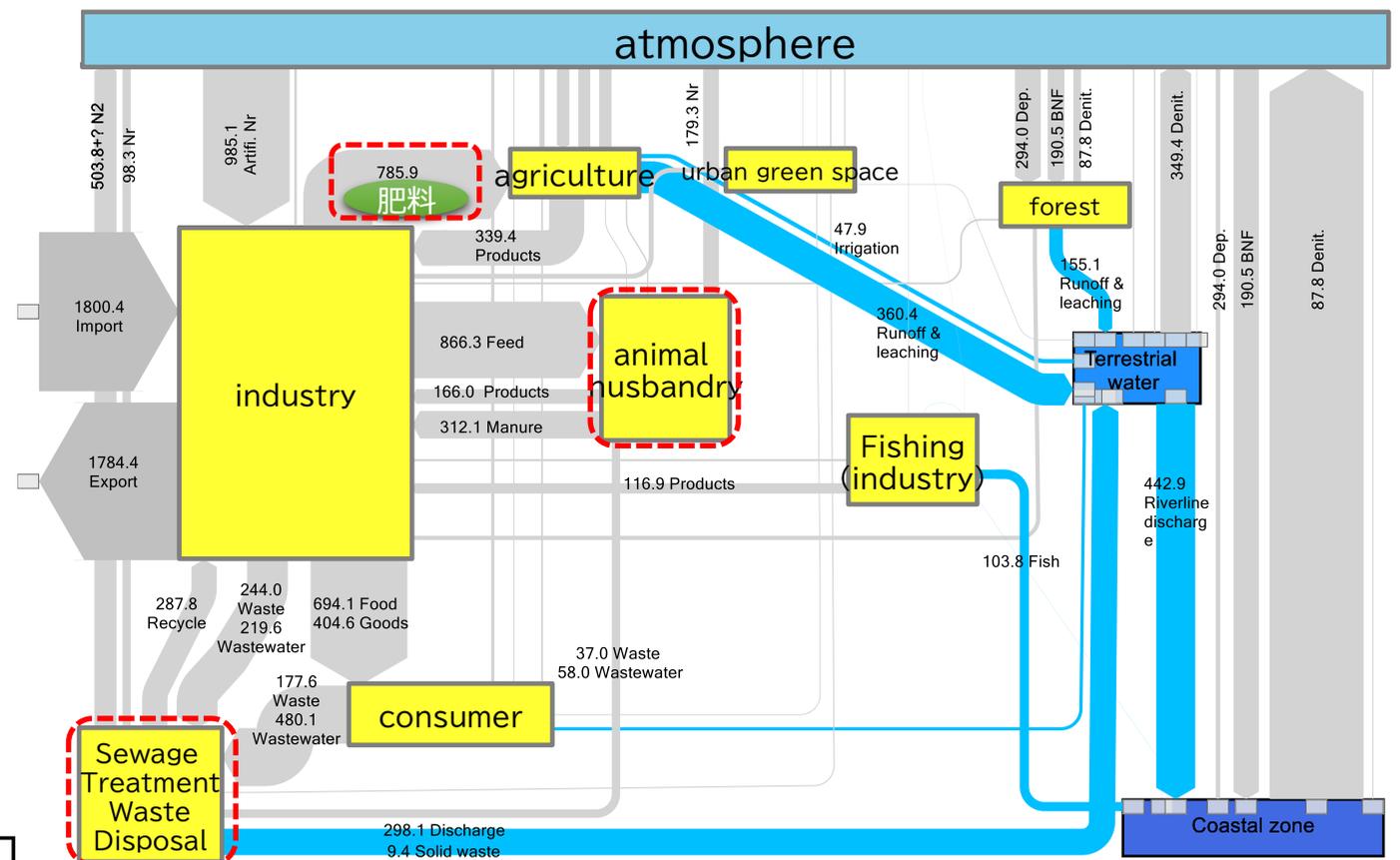
The University of Tokyo Materials

Team: Wakihara, Iyogi Urban

Engineering Team: Katayama, Hashimoto, Tobino

LCA Team: Kanematsu

Nitrogen flow in Japan



Based on Hayashi *et al.*, (2021)

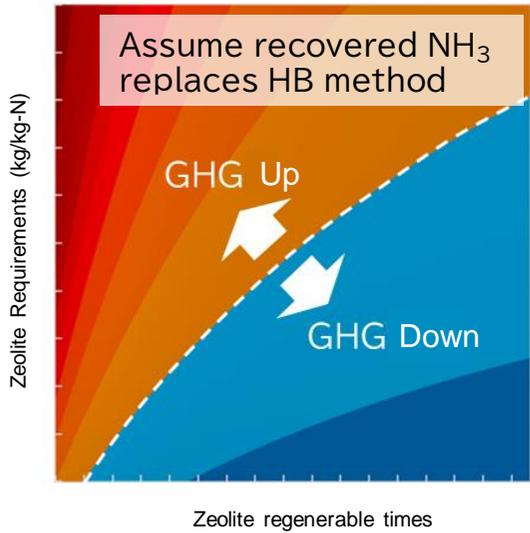
To transform social systems and processes by moving away from wastewater treatment, which is energy-intensive and dumps resources



LCA and nitrogen flow

Study of LCA framework for exploring material performance conditions

Change in GHG emissions from sewage treatment process



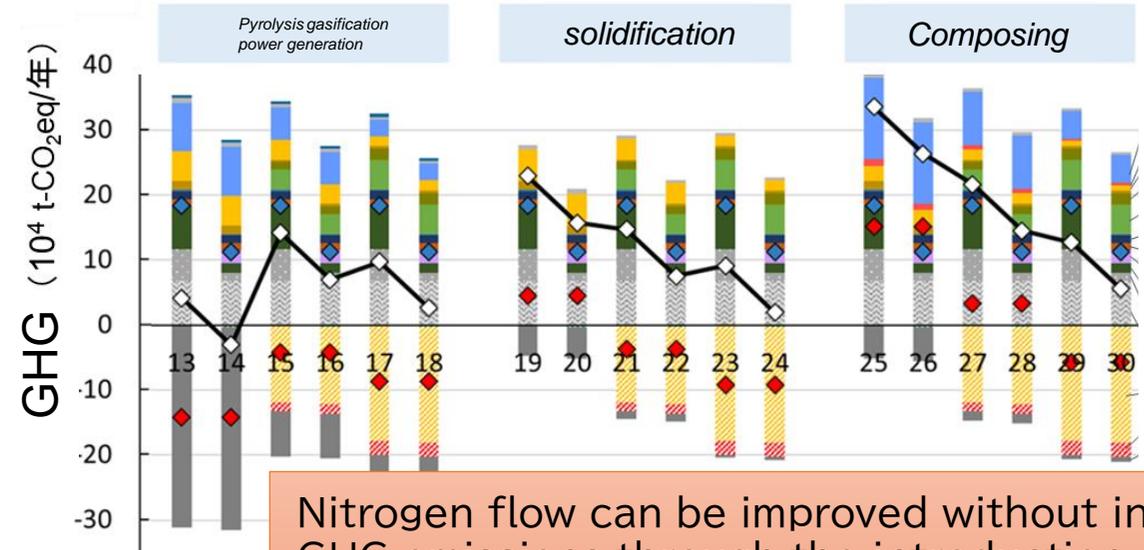
- As an initial LCA, GHG emission intensity during production of Type A zeolite for which inventory data is available is applied.
- Under these conditions, the required amount of zeolite ($\hat{=}$ adsorption efficiency) and the number of times the zeolite can be regenerated are identified as having high GHG sensitivity

→ Feedback to the experiment

LCA to guide technology development

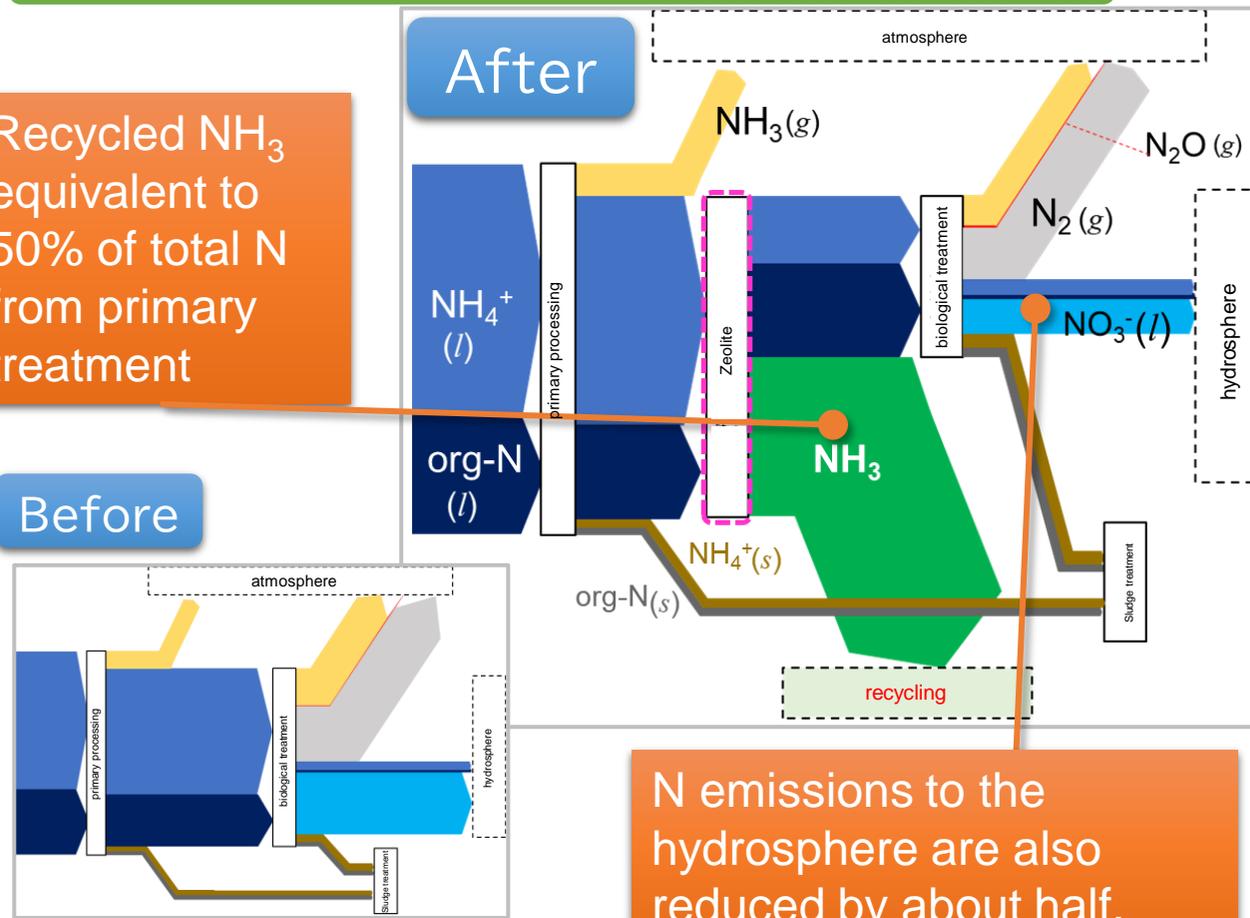
LCA for real processes

Some excerpts from the evaluation results of 42 cases (↓ even numbers are the results when zeolite was introduced)



Nitrogen flow can be improved without increasing GHG emissions through the introduction of this technology.

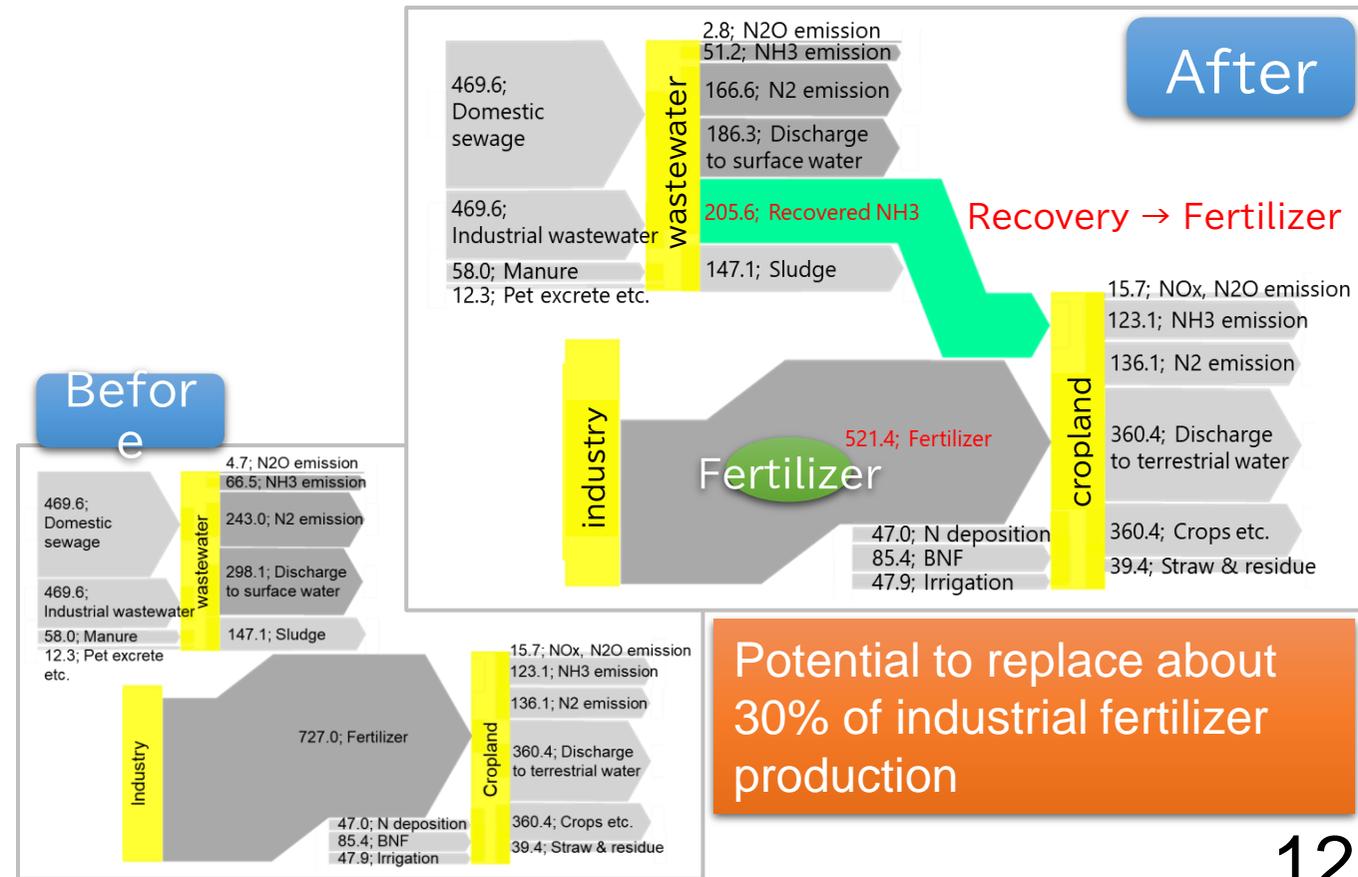
Nitrogen flow changes during sewage treatment process (site units)



Recycled NH₃ equivalent to 50% of total N from primary treatment

N emissions to the hydrosphere are also reduced by about half.

Flow change in agricultural use of recovered NH₃ (national potential)



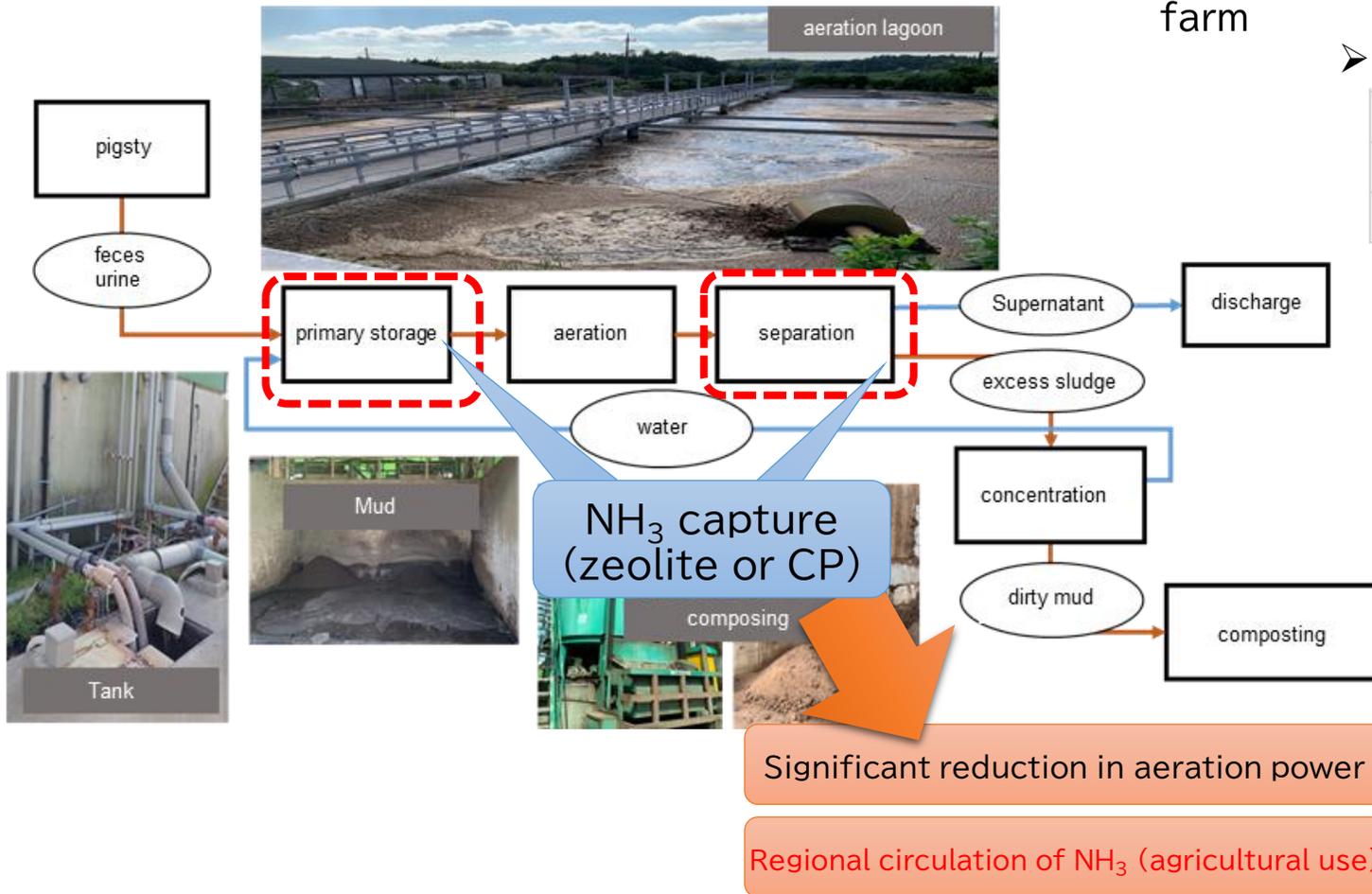
Potential to replace about 30% of industrial fertilizer production



Expand application to livestock wastewater

Modeling of pig farm effluent treatment processes

→ Field survey and interviews were conducted at a pig farm where a lagoon wastewater treatment system is being introduced. Conducted on-site survey and interviews at a pig farm



➤ number of pig farms

Mother	count	2000
Child	count	20000
Total	count	22000

➤ Emissions

excrement	kg/day/匹	1.9
urine	kg/day/匹	3.5
Total	kg/day/匹	5.4
BOD	mg/L	24000
SS	mg/L	80000
TN	mg/L	6800
TP	mg/L	2700

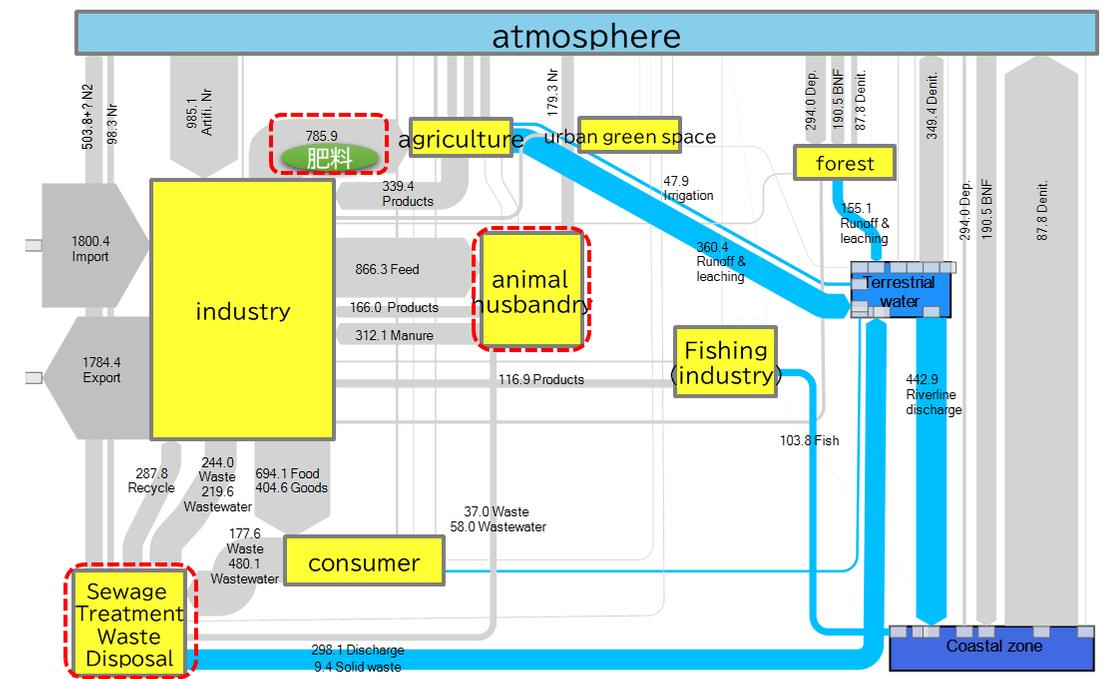
Equivalent to the emissions of a city of tens of thousands of people at one site

More room for further purification compared to sewage treatment
→ large reduction potential per m³

Modeling is underway to analyze the current situation and evaluate the effectiveness of technology implementation

Future development

- Evaluation of flow when zeolite/CP is applied (per pig farm)
- Macroscopic flow impact assessment when recycling is implemented



To quantify the effects of N recovery from sewage + livestock wastewater at the national level

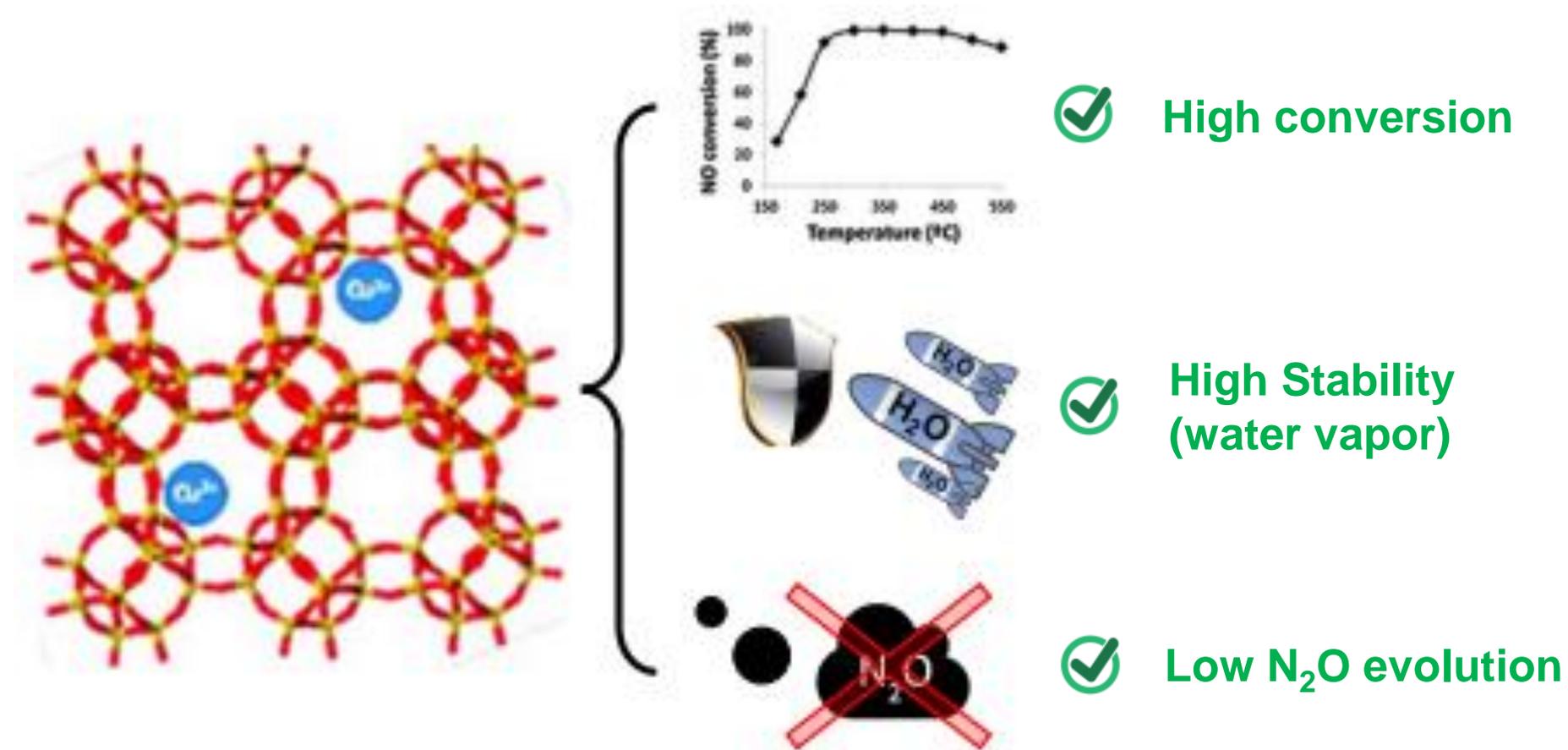


NH₃-SCR (for automobile)

◆ Conventional Cu-exchanged zeolite

Low stability → Decrease of NO_x conversion
N₂O production

Desired properties for zeolite catalyst



broken catalyst in NH₃-SCR (by urea water)

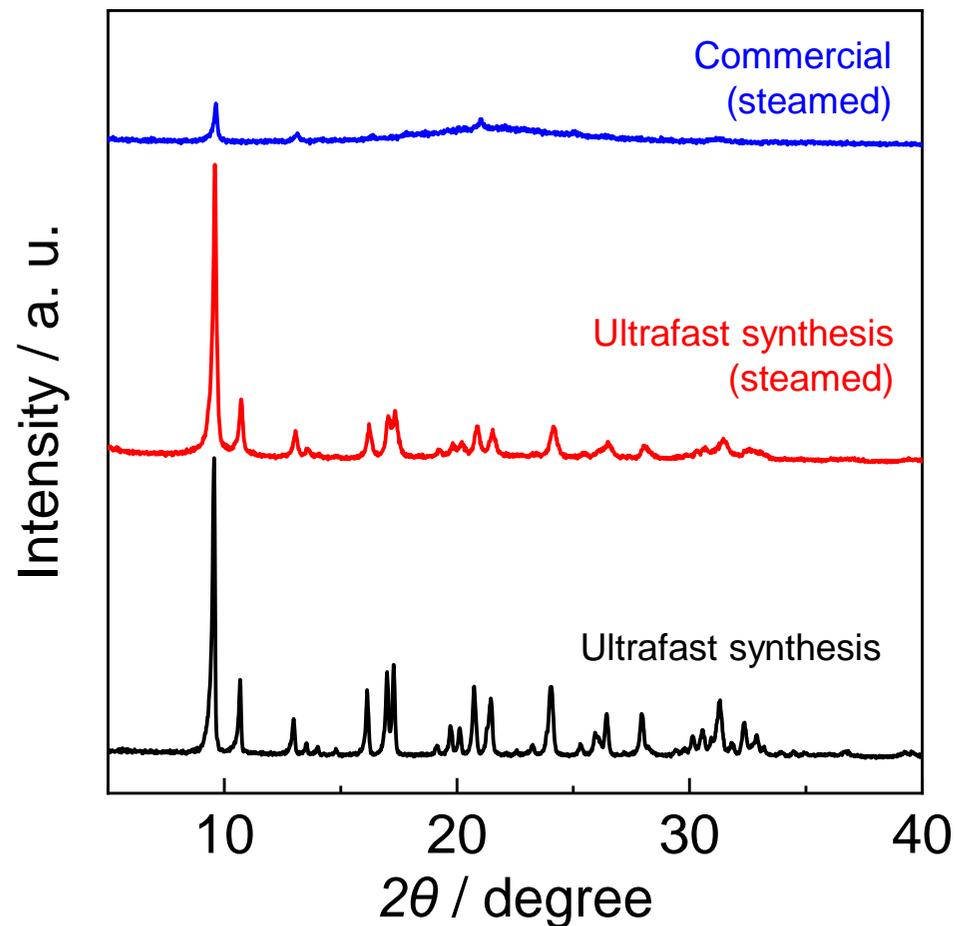


Hydrothermal stability of zeolites catalyst

Stability against water vapor

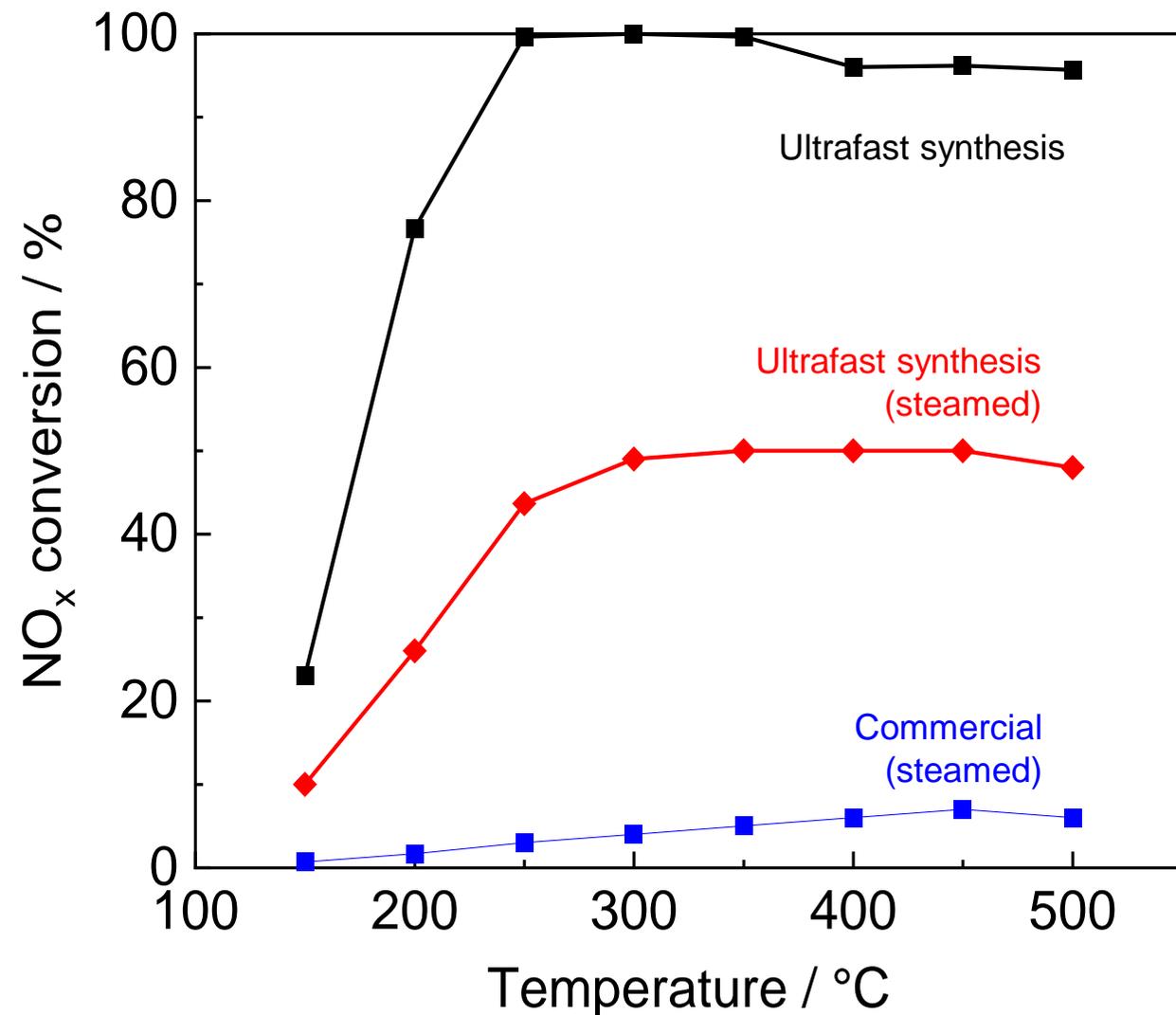
Condition: H₂O-10vol%
900°C 1 h

X-ray Diffraction



NH₃-SCR

NO 300 ppm, NH₃ 300 ppm, 5% O₂
Flow rate 100 cm³/min

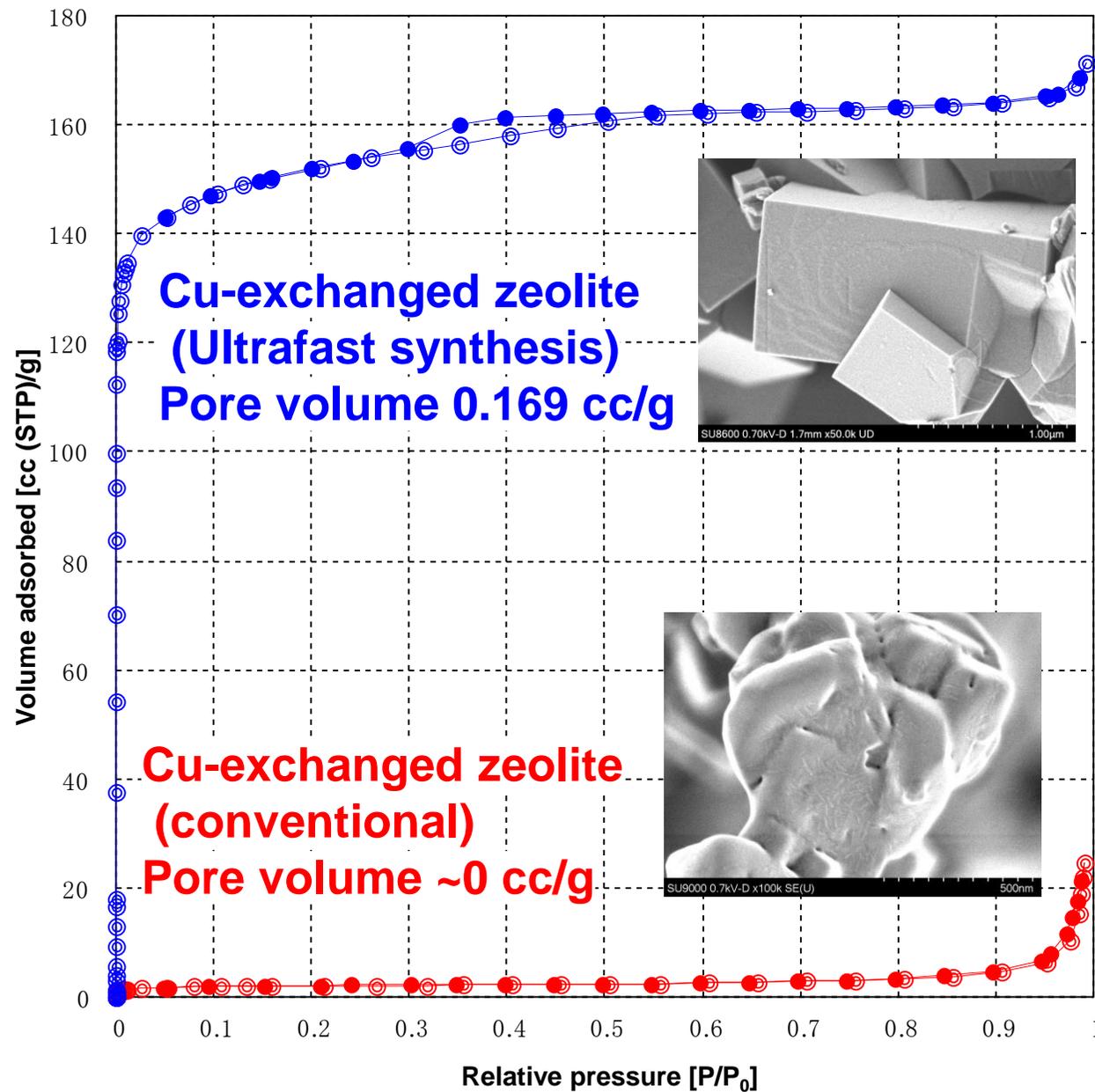


- ✓ Retain the original crystalline structure after steaming (2022 KPI achieved)
- ✓ Better activity than commercial catalyst

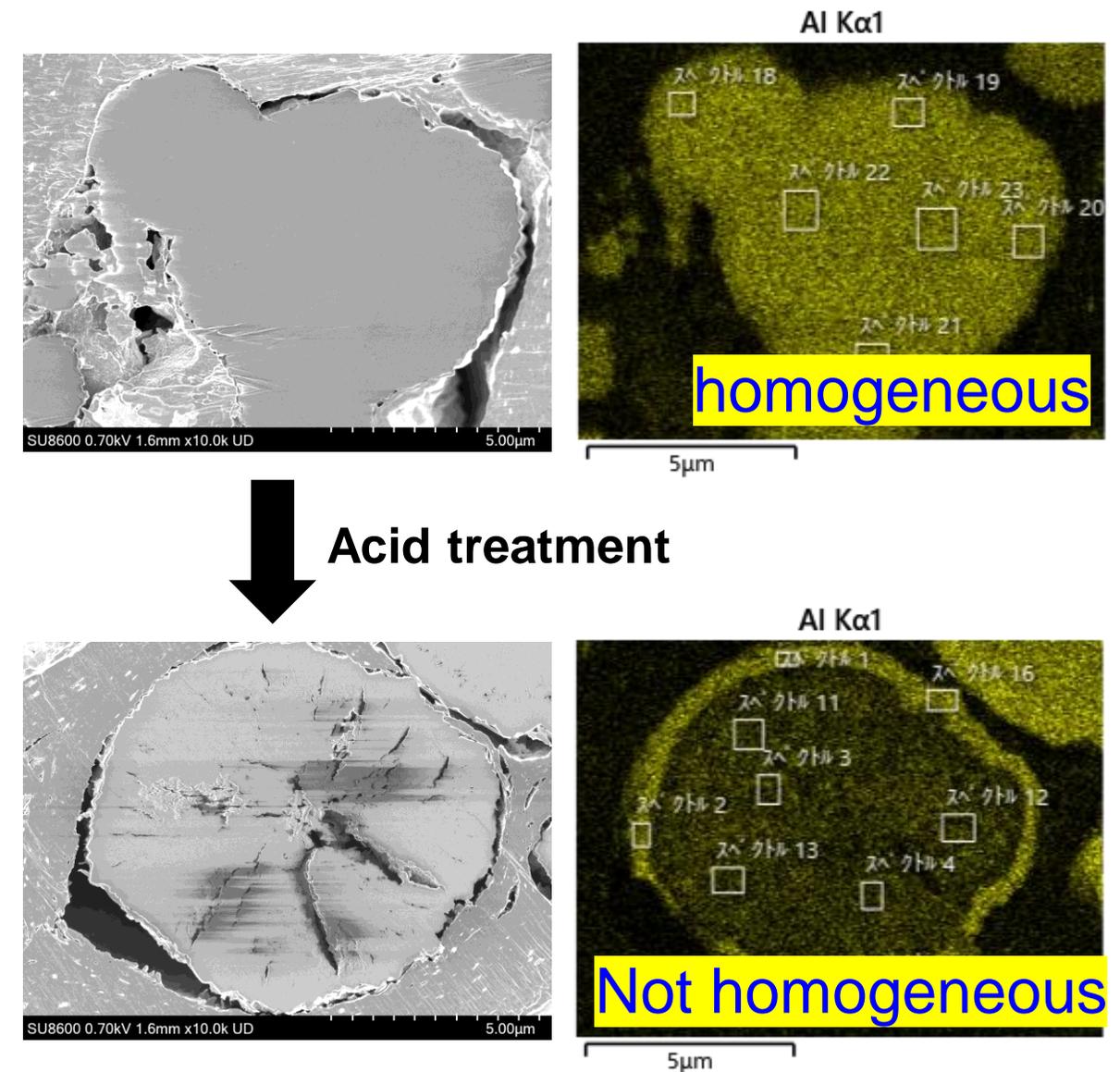


Evaluation of degradation of zeolite catalyst

Ar adsorption-desorption



Elemental Mapping



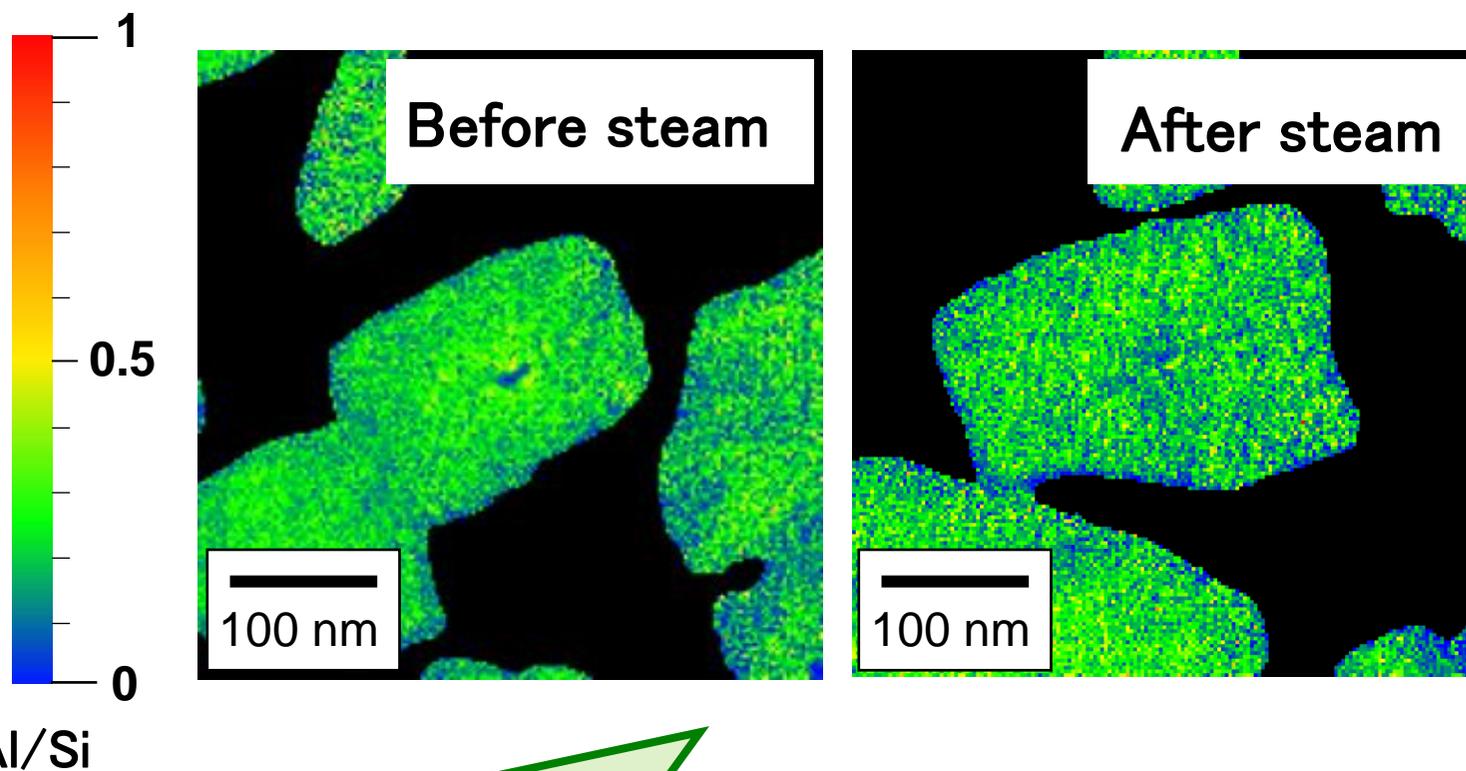
Developed the analysis technique of the degradation and dealumination of zeolite catalyst



STEM-EDS mapping

Establishment of a method for analyzing Si/Al ratio distribution with a spatial resolution of **2.4nm~** for zeolites sensitive to electron beam.

Ex) Cu-exchanged zeolite



Pixel: 2.4nm × 2.4nm

*From Qualitative to Quantitative
Microdomain Analysis*

**Quantification of spatial
distribution**

Quantify visual information

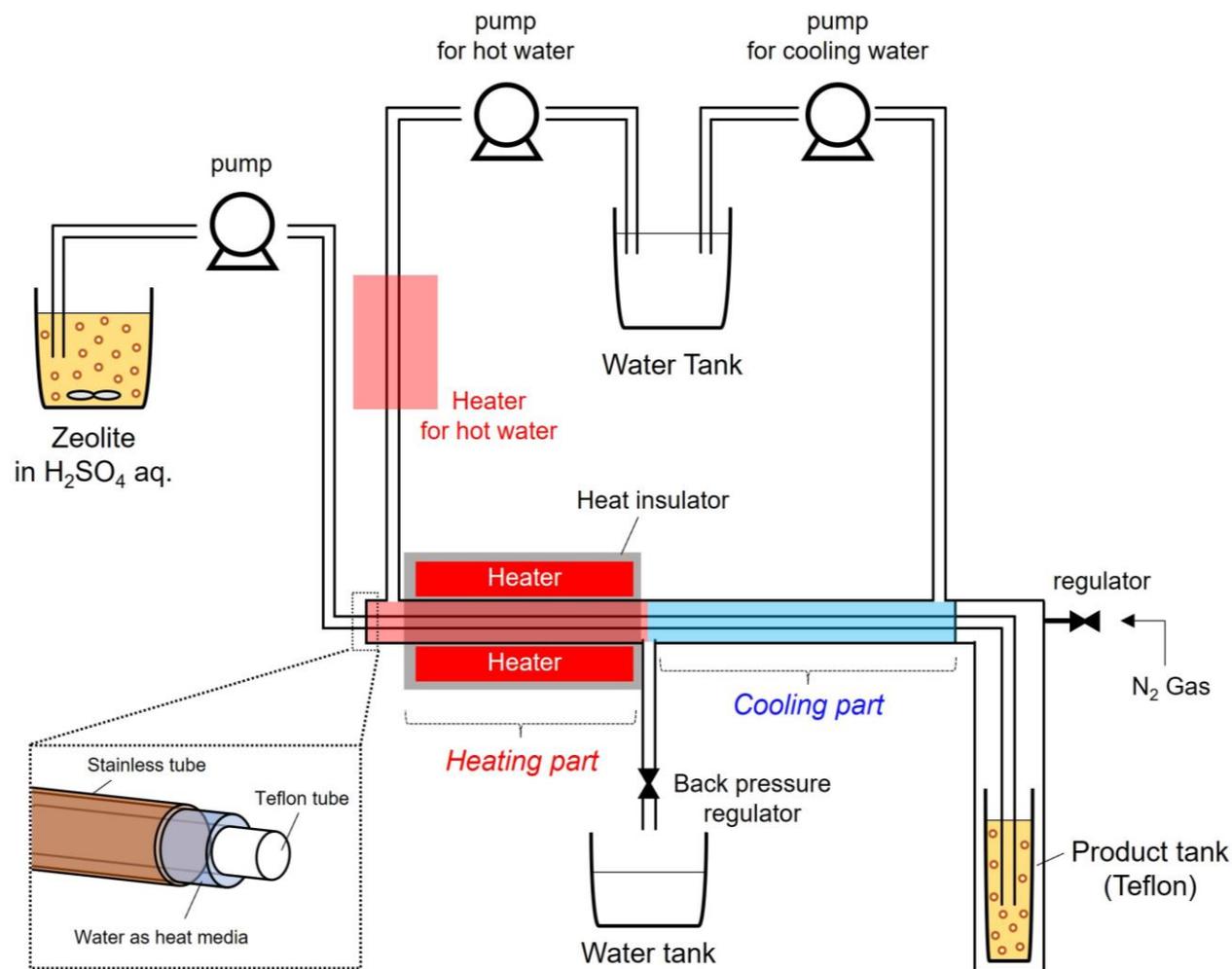
→ characterized and analyzed for differences in properties due to their chemical composition.



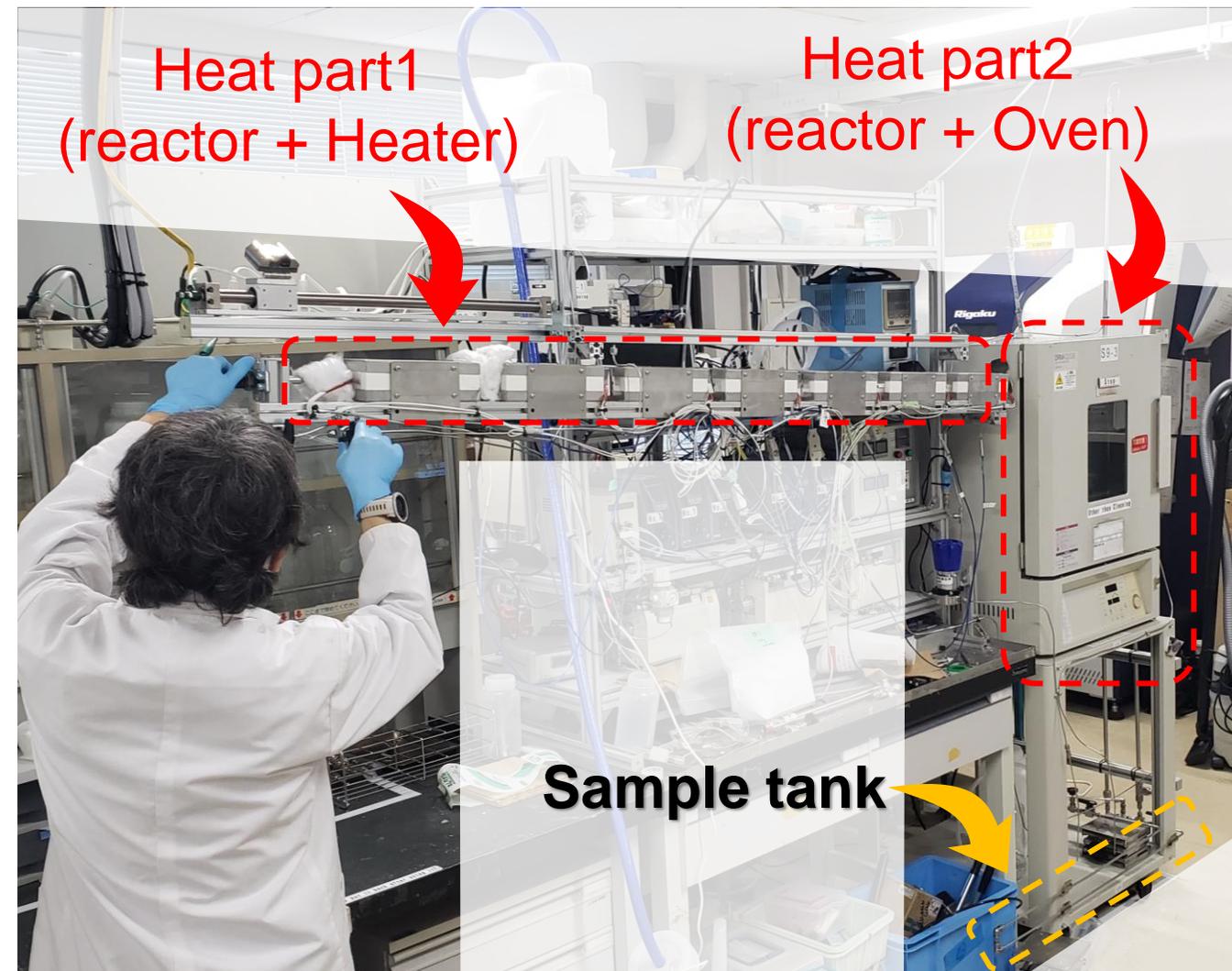
Continuous flow system for synthesis and post-treatment

Flow reactor

◆ deAl of zeolites in flow system



◆ Machine for zeolite synthesis



Publication; ultrafast deAl of Beta in continuous flow reactor

A. Minami, M. Takemoto, Y. Yonezawa, Z. Liu, Y. Yanaba, A. Chokkalingam, K. Iyoki, T. Sano, T. Okubo, T. Wakihara
Advanced Powder Technology, 33, 103702 (2022).

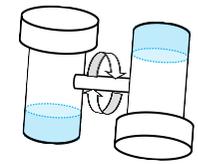
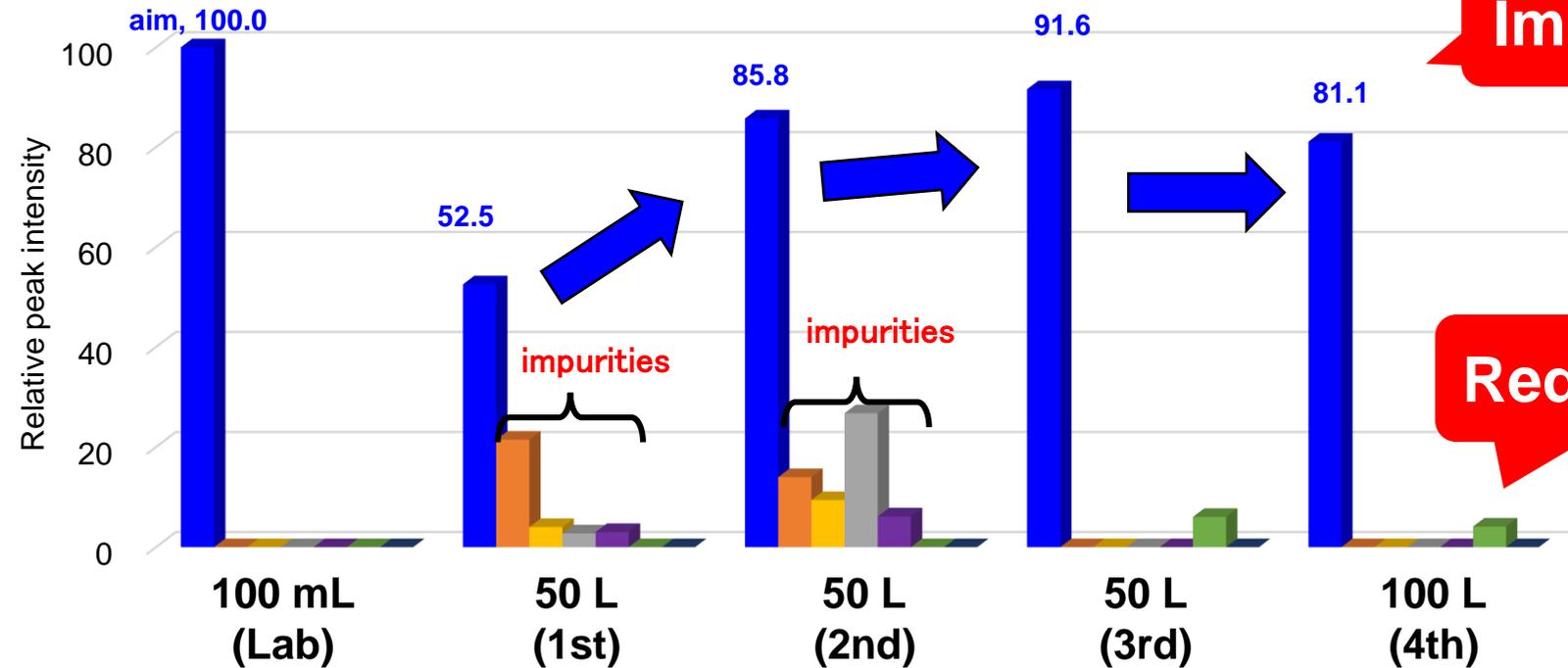


Started the design of continuous flow reactor for zeolite synthesis

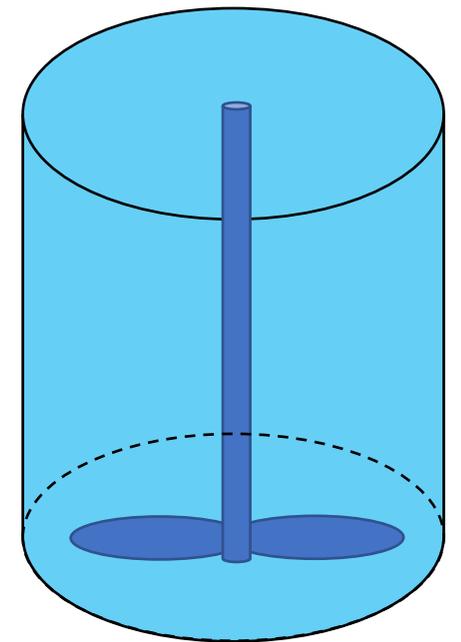


Large Scale Synthesis of zeolites

Optimization



Lab scale

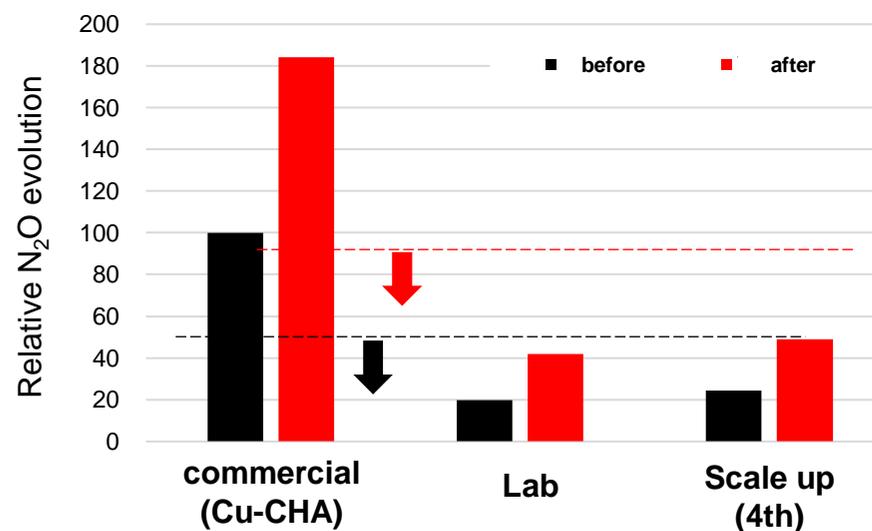
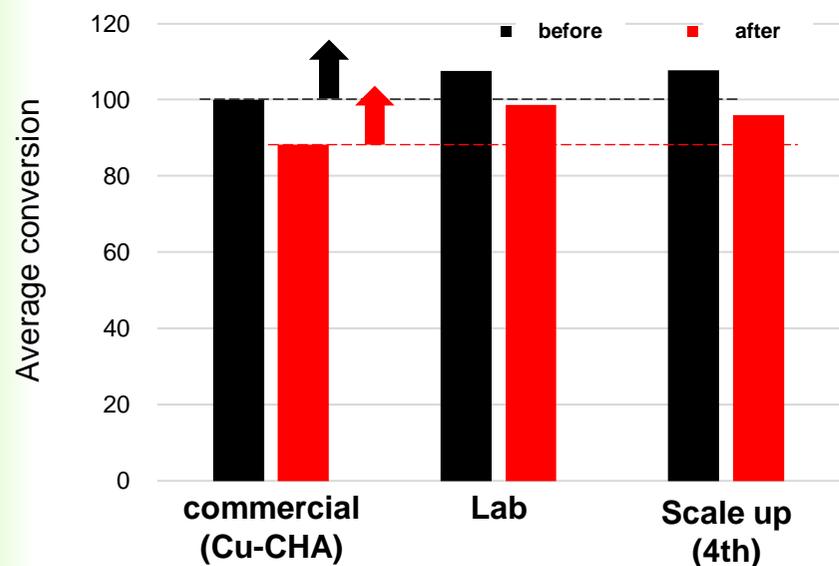


Prototype (image)

NH₃-SCR(scale up + effect of steaming)

■ Average NO_x conv. (150-500°C)

■ Average N₂O evolution (150-500°C)

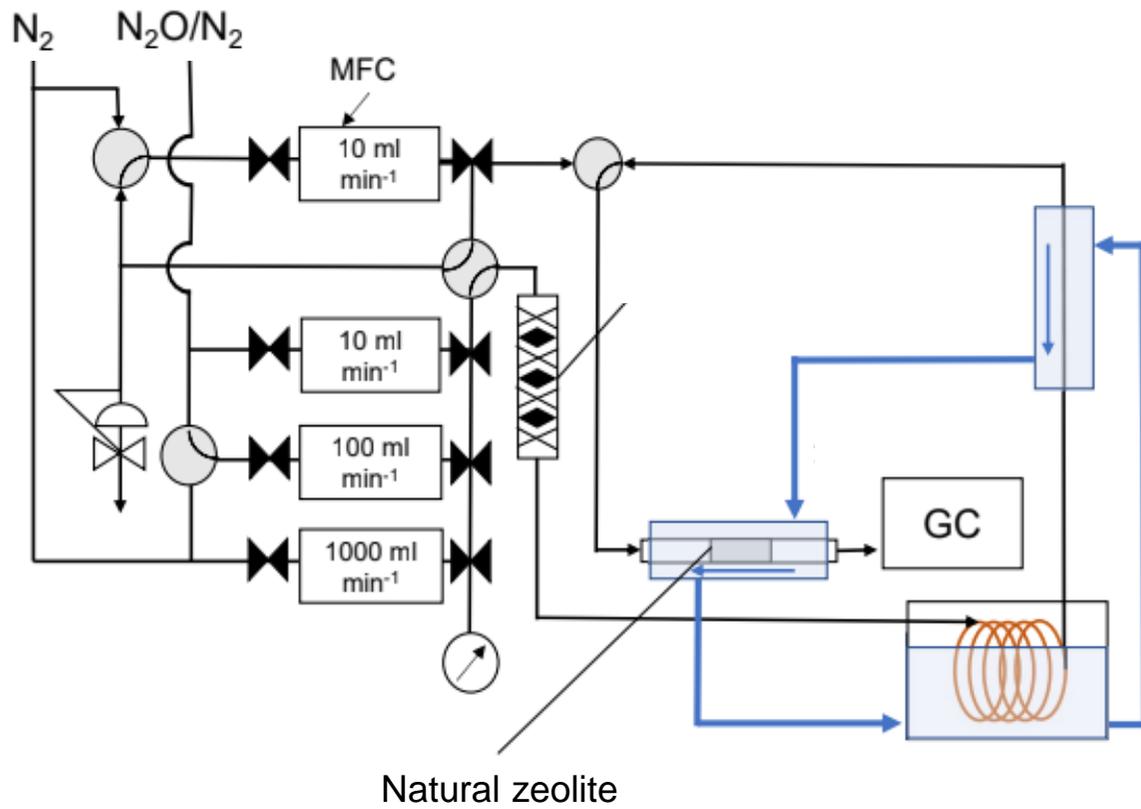


Low N₂O evolution → Future plan: large scale (2m³)

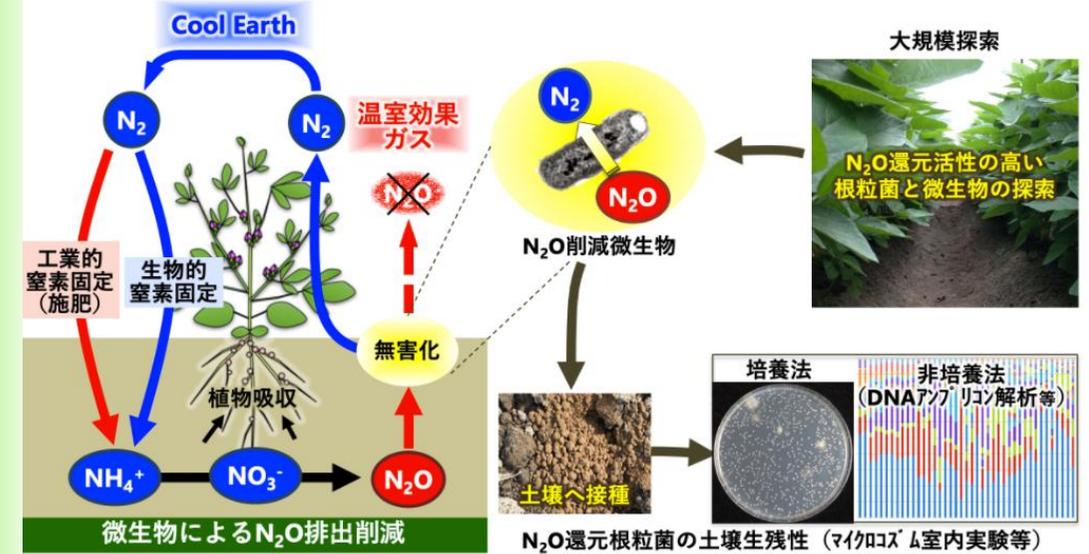


Adsorption and decomposition of dilute N₂O

N₂O capture by natural zeolite



N₂O degradation by rhizobium (Prof. Minamisawa, Tohoku Univ.)



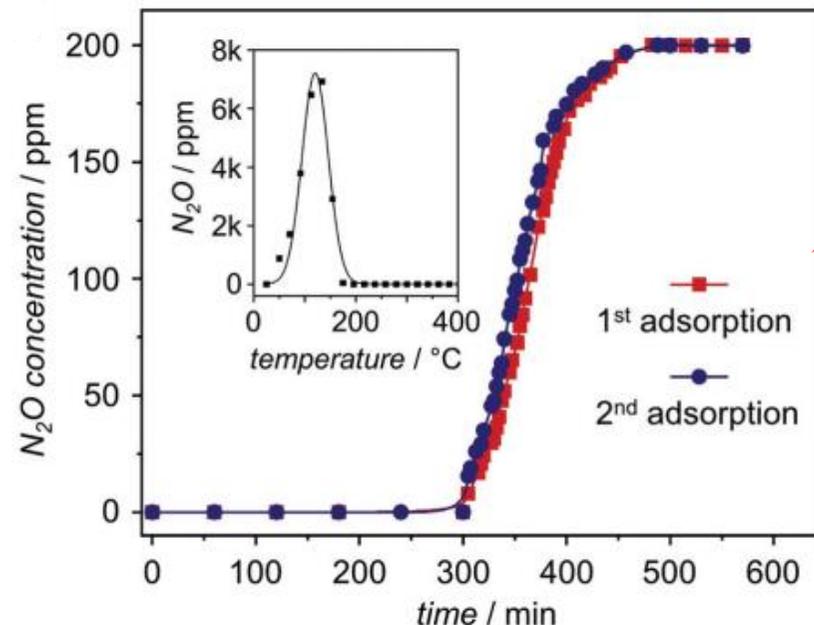
Cited from dSOIL

□ problem : concentration of N₂O (>100 ppm)

Equipment



N₂O adsorption-desorption test



✓ Concentration of N₂O !!

✓ Increase the amount of adsorbed N₂O by temperature control

✓ Hybrid with rhizobium

✓ Discussion with JACI (> 80 company)

✓ RFI (submitted)



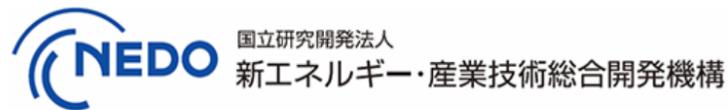
Publication · Conference · Seminar

Publication (14 papers + 2 submitted papers)

- ◆ T. Yoshioka, K. Iyoki, Y. Hotta, Y. Kamimura, H. Yamada, Q. Han, T. Kato, C. A. J. Fisher, Z. Liu, R. Ohnishi, Y. Yanaba, K. Ohara, Y. Sasaki, A. Endo, T. Takewaki, T. Sano, T. Okubo, T. Wakihara **Science Advances**, 8, (2022).

Press Release

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ニュース

イベント

実施者募集(公募)

事業紹介

刊行物・資料

調達

NEDOについて



ホーム > ニュース > ニュースリリース一覧 > 新しく小細孔ゼオライトの組成チューニング法を開発し、耐久性向上を実現

新しく小細孔ゼオライトの組成チューニング法を開発し、耐久性向上を実現

—環境問題の解決へ向け、窒素酸化物を浄化する触媒応用に期待—

2022年6月23日

NEDO (国立研究開発法人新エネルギー・産業技術総合開発機構)
国立大学法人東京大学

Information toward the public

- ◆ JACI (Japan Association for Chemical Innovation) : discuss about N₂O capture
- ◆ seminar for middle and high-school students (at the Univ. of Tokyo, 6/5)
- ◆ Science Agora 2022
- ◆ Home page launched

Conference (>25)

Seminar (>8)



- ◆ R. Simancas, A. Chokkalingam, S. P. Elangovan, Z. Liu, T. Sano, K. Iyoki, T. Wakihara, T. Okubo
Chemical Science, 12, 7677-7695 (2021)
- ◆ C.-T. Chen, K. Iyoki, P. Hu, H. Yamada, K. Ohara, S. Sukenaga, M. Ando, H. Shibata, T. Okubo, T. Wakihara
Journal of the American Chemical Society, 143, 10986-10997 (2021)
- ◆ Y. Sada, A. Chokkalingam, K. Iyoki, M. Yoshioka, T. Ishikawa, Y. Naraki, Y. Yanaba, H. Yamada, K. Ohara, T. Sano, T. Okubo, Z. Liu, T. Wakihara
RSC Advances, 11, 23082-23089 (2021)
- ◆ R. Sato, Z. Liu, C. Peng, C. Tan, P. Hu, J. Zhu, M. Takemura, Y. Yonezawa, H. Yamada, A. Endo, J. García-Martínez, T. Okubo, T. Wakihara
Chemistry of Materials, 33, 8440-8446 (2021)
- ◆ Z. Liu, A. Chokkalingam, S. Miyagi, M. Yoshioka, T. Ishikawa, H. Yamada, K. Ohara, N. Tsunoji, Y. Naraki, T. Sano, T. Okubo, T. Wakihara
Physical Chemistry Chemical Physics, 24, 4136-4146 (2022)
- ◆ K. Iyoki, T. Onishi, M. Ando, S. Sukenaga, H. Shibata T. Okubo, T. Wakihara
Journal of the Ceramic Society of Japan, 31, 187-194 (2022)
- ◆ P. Hu, K. Iyoki, H. Fujinuma, J. Yu, S. Yu, C. Anand, Y. Yanaba, T. Okubo, T. Wakihara
Microporous and Mesoporous Materials, 330, 111583, (2022).
- ◆ R. Simancas, M. Takemura, Y. Yonezawa, S. Sukenaga, M. Ando, H. Shibata, A. Chokkalingam, K. Iyoki, T. Okubo, T. Wakihara
Nanomaterials, 12, 396 (2022).
- ◆ T. Yoshioka, K. Iyoki, Y. Hotta, Y. Kamimura, H. Yamada, Q. Han, T. Kato, C. A. J. Fisher, Z. Liu, R. Ohnishi, Y. Yanaba, K. Ohara, Y. Sasaki, A. Endo, T. Takewaki, T. Sano, T. Okubo, T. Wakihara *Science Advances*, 8, (2022).
- ◆ M. Takemoto, K. Iyoki, Y. Otsuka, H. Onozuka, A. Chokkalingam, T. Yokoi, S. Tsutsuminai, T. Takewaki, T. Wakihara, T. Okubo
Materials Advances, 3, 5442 (2022).
- ◆ A. Minami, M. Takemoto, Y. Yonezawa, Z. Liu, Y. Yanaba, A. Chokkalingam, K. Iyoki, T. Sano, T. Okubo, T. Wakihara
Advanced Powder Technology, 33, 103702 (2022).
- ◆ J. Tomita, S. P. Elangovan, K. Itabashi, A. Chokkalingam, H. Fujinuma, Z. Hao, A. Kanno, K. Hayashi, K. Iyoki, T. Wakihara, T. Okubo
Advanced Powder Technology, 33, 103741 (2022).
- ◆ Y. Sada, S. Miyagi, K. Iyoki, M. Yoshioka, T. Ishikawa, Y. Naraki, T. Sano, T. Okubo, T. Wakihara
Microporous and Mesoporous Materials, 344, 112196 (2022).
- ◆ A. Minami, P. Hu, Y. Sada, H. Yamada, K. Ohara, Y. Yonezawa, Y. Sasaki, Y. Yanaba, M. Takemoto, Y. Yoshida, T. Okubo, T. Wakihara
Journal of the American Chemical Society, Accepted.



Main results at this time

NH₃ capture

- ✓ Low-cost, quick and simple adsorbent synthesis process
- ✓ Recovery of more than 50% NH₃/NH₄⁺ ions in actual wastewater
- ✓ Collaborating with participating companies, taking into consideration about reuse recovered NH₃.
- ✓ Demonstrated recyclability of developed product in NH₃ recovery
- ✓ Development of superior adsorbent material using waste raw materials
- ✓ Establishment of molding technology for developed products for social implementation
- ✓ Started studying the treatment of liquid waste by continuous distribution system

deNOx

- ✓ Development of zeolite with both high NOx conversion and low N₂O emissions
- ✓ Development of zeolite that maintains crystallinity after exposure to high temperature steam
- ✓ Establishment of zeolite production technology on a large scale
- ✓ Started synthesis of highly durable zeolite catalyst by continuous flow method

Evaluation of system

- ✓ LCA evaluation of adsorption and denitration systems using zeolite production, zeolite, etc.
- ✓ Visualization of nitrogen flow/extension of further evaluation targets
- ✓ Proof of reduction of emission N and resource saving by introduction of development agents



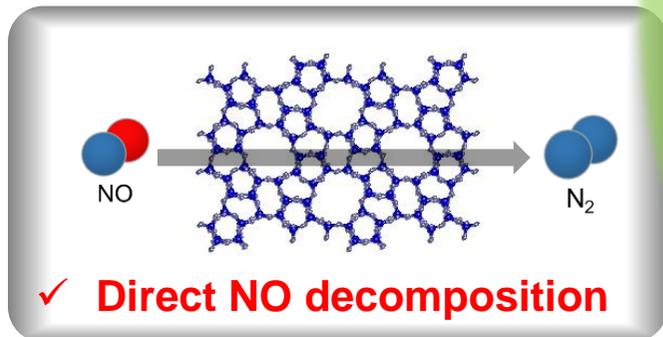
Future Outlook

✓ High stability

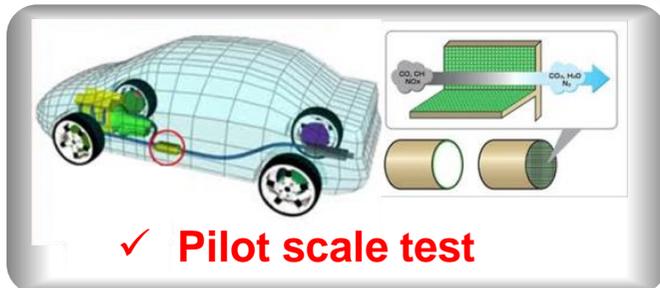
✓ Concentration of N₂O by natural zeolite

✓ Scale up synthesis

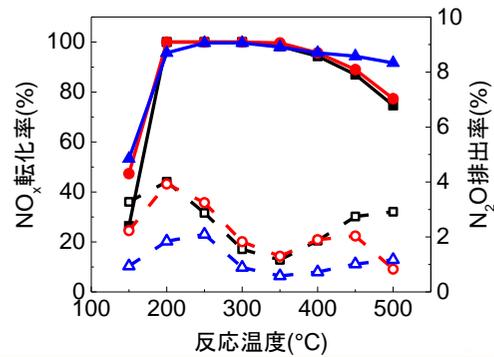
✓ Continuous flow synthesis



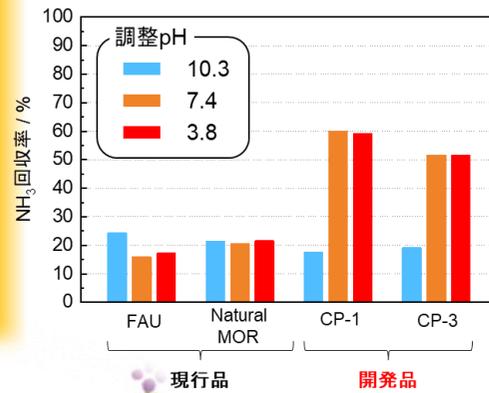
✓ Hybrid with natural zeolite and rhizobium



✓ High NO_x conv. • Low N₂O



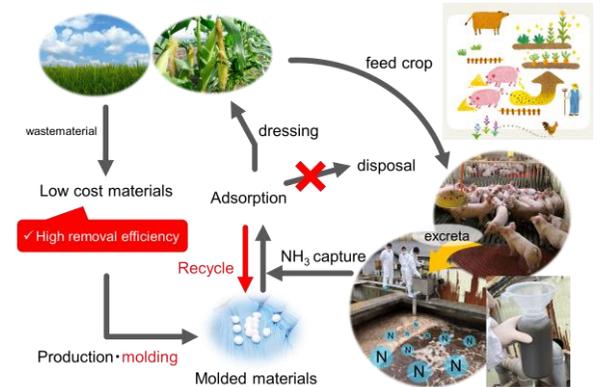
✓ Excellent adsorbent !!



✓ Recyclable

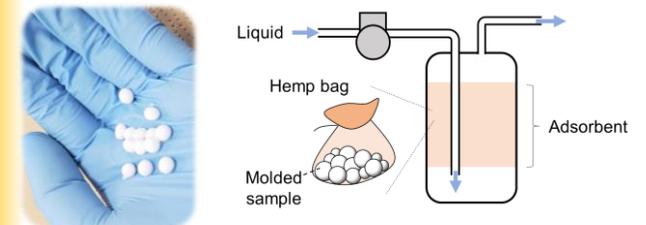
✓ From waste raw material

✓ Regional nitrogen cycle !!



✓ Molding method

✓ Flow system



✓ LCA assessment taking into account the increase in greenhouse gases and environmental pollutants

✓ Pilot scale test



Scale-up demonstration test for pilot test ahead of schedule