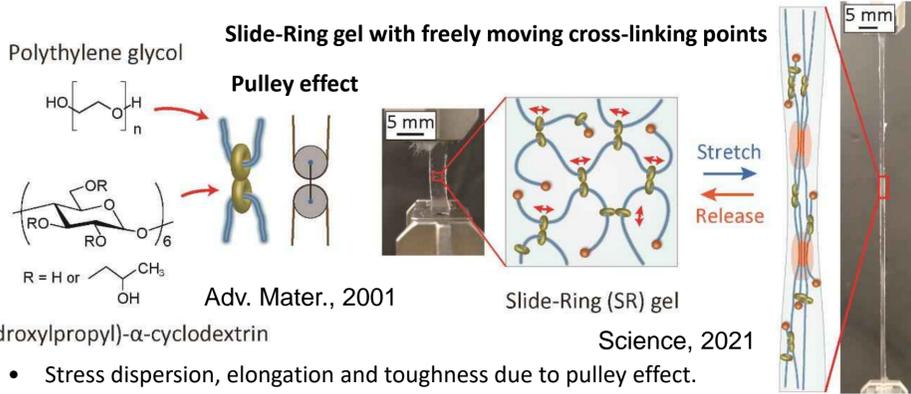
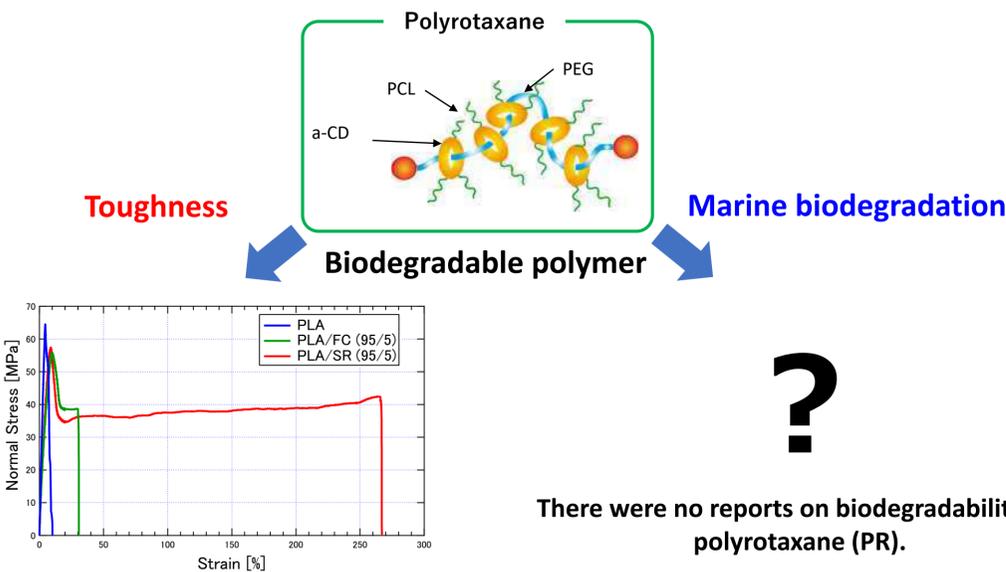


Background



- Stress dispersion, elongation and toughness due to pulley effect.
- Effective when added in small amounts to other polymers.
- In low coverage rate Slide-Ring gels, the elongation-induced crystallization provides both additional toughness and immediate recovery.
- Composed of cyclodextrins and PEG, showing high biosafety and compatibility.

Combination of biodegradability and toughness



There were no reports on biodegradability for polyrotaxane (PR).

Durability/toughness of PLA due to pulley effect

Assignment and Research Results

Research Objectives:  
To achieve both toughness and biodegradability through the use of polyrotaxane (PR)

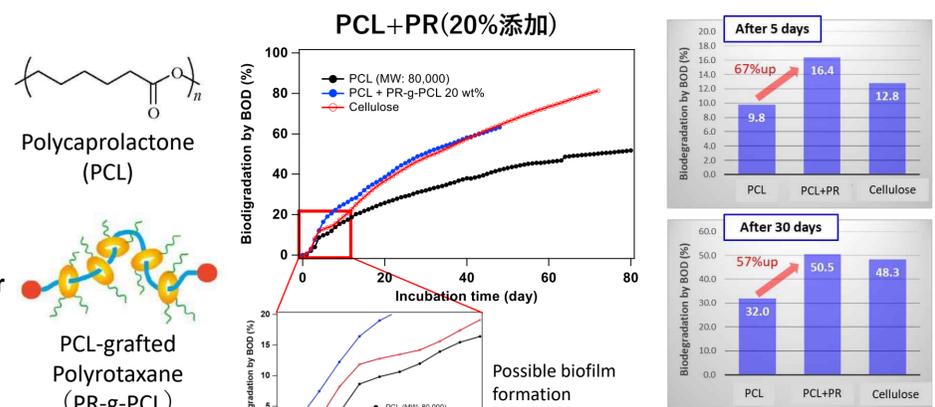
Effect of adding PR  
• Combination of toughness and biodegradability of resins  
• Combines toughness and biodegradability through ester exchange reaction  
• Achieved toughening and biodegradability of ester-exchange vitrimers

Multi-lock function  
• Synthesis of end-stimulated degradative PR polymer chain  
• Controlled structural parameters such as coverage ratio, axial molecular weight, and graft density

Social integration  
• One-pot synthesis of PR to reduce production cost and control coverage rate  
• Reduced production cost by PR extraction and purification method  
• Raw materials such as components, solvents, etc., for cost reduction.

Polyrotaxane Seawater Biodegradability

Seawater sampling location: Miyagawa beach, Miura city, Kanagawa  
Preparation of extracted seawater  
Sediment (100 g)  
Seawater (600 mL)  
Ultrasonication 10 s  
Extracted seawater  
CERI Dr. T. Kikuchi's protocol  
Degradation bacteria concentration was enhanced by approximately 10 times  
Degradation speed and stability also improved



Toughened polyester resin design

PCL  
MW : 80 kg/mol  
Catalyst : Dibutyltin dilaurate(DBTDL)  
PCL  
MW : 1 kg/mol  
S-S curves  
The transesterification between PR and PCL increased the elongation at break by more than 5 times.

Highly biodegradable formulation of polyester

Seawater sampling location: Miyagawa beach, Miura city, Kanagawa  
Sampling date: May 11, 2022  
BOD test  
Biodegradation by BOD (%)  
Incubation time (day)  
After 30 days  
The transesterification between PR and polyester has greatly improved seawater biodegradability.

Multi-lock degradation by UV irradiation

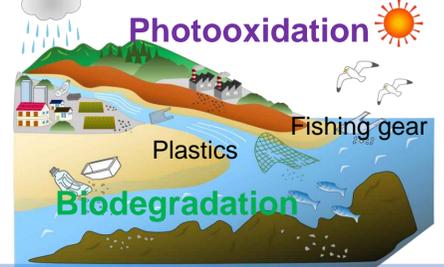
Polymer chain  
Seawater immersion  
Sunlight  
Biodegradation  
Degradation point  
Cross-linked chain  
Polyrotaxane cross-linking for toughening under use  
Photodegradation of the ends degrades the polyrotaxane structure (Point-control)  
Components of polyrotaxane attract degradation bacteria as food and accelerate matrix biodegradation (Speed-control)

Toughening and point control of polyurethanes

Irradiation conditions :  
• UV wavelength: 325 nm  
• Irradiation intensity : 650 mW/cm<sup>2</sup>  
• Irradiation time: 30 min  
Tensile test  
Measuring conditions :  
• 10 mm/min  
• Room temperature  
• 1 kN load cell  
S-S curves  
Stress (MPa)  
Strain (%)  
PU  
PU after UV irradiation  
PU\_TD-PR  
PU\_TD-PR after UV irradiation  
• TD-PR increases elongation at break by about 30 times.  
• UV irradiation of PU with TD-PR reduced elongation at break by about 1/5  
→ Suggests degradation of PR by UV irradiation

Samples	Young's modulus (GPa)	Strain at Break (%)
PU	0.87	4.0
PU_after UV irradiation	0.85	4.1
PU_TD-PR	0.64	117.4
PU_TD-PRPU_after UV irradiation	0.65	21.6

Waste Plastics in Environment

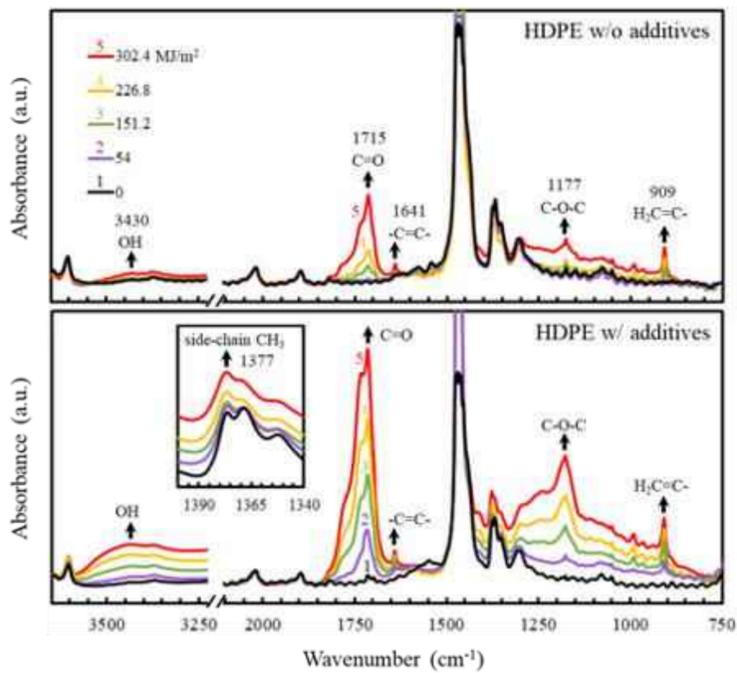


**Introduction:** Microplastics (MPs) has become global environment problems due to the environmental degradation of waste plastic products. In order to face this challenge, it is important and necessary to study the degradation behaviors of plastics in environment. Also, development of biopolymer which can be degraded by microorganisms will be a promising way to solve the MPs problems in environment. In this study, the environmental degradation behaviors of plastics were studied by simulating photooxidation and biodegradation behaviors of polymers in laboratory using weathering test chamber and extracted seawater.

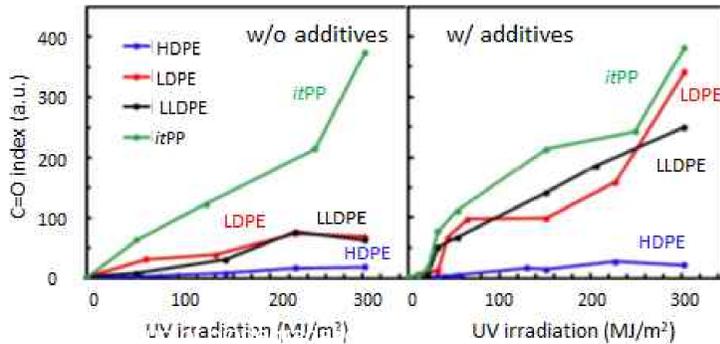
1. Photo-oxidation and biodegradation behaviors of polyolefins containing oxo-biodegradable additives

The effects of the commercially available pro-oxidants of the oxo-biodegradable type (P-Life Japan Inc.) on the photo-oxidation and biodegradation of polyolefin (HDPE, LDPE, LLDPE, and *it*PP) films were investigated.

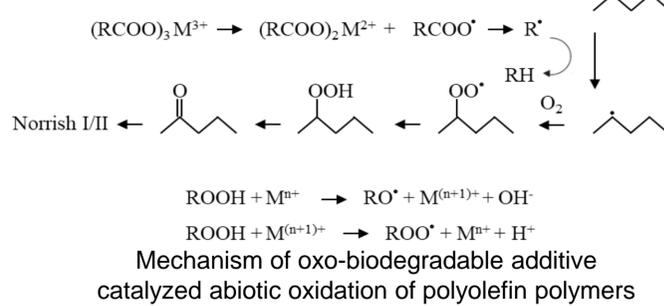
Photo-oxidation



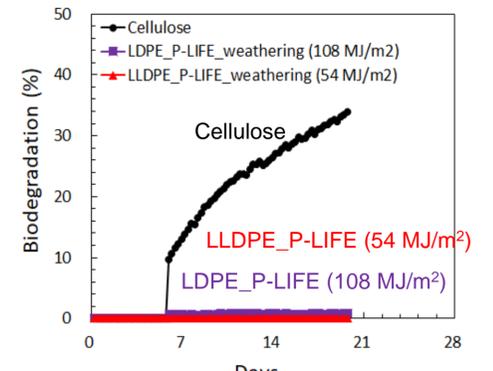
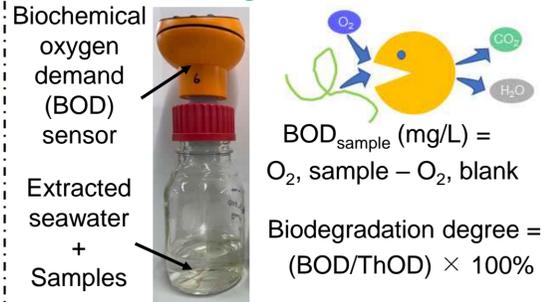
FT-IR spectra of *it*PP w/o and w/ the oxo-biodegradable additives



Variations of the C=O Index of HDPE, LDPE, LLDPE, and *it*PP w/o and w/ oxo-biodegradable additives with UV exposure (polymer chain)



Biodegradation



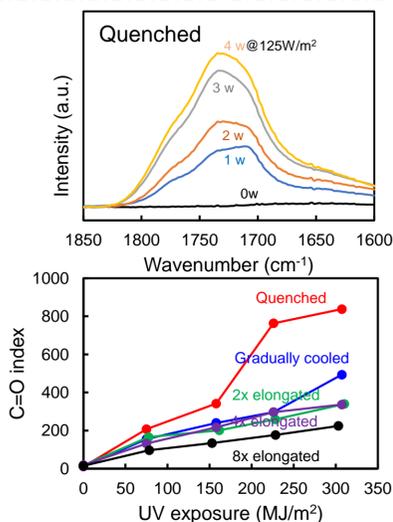
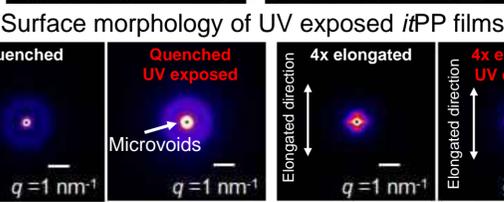
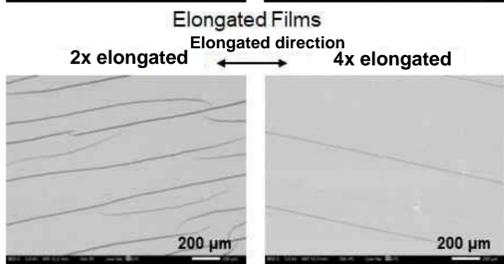
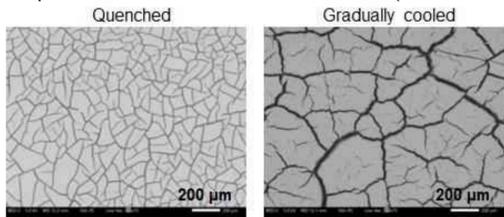
Biodegradation of LDPE & LLDPE with oxo-biodegradable additive

The carbonyl index calculated for the neat and additive-containing samples suggested that the additives promoted the photo-oxidative degradation. The overall degradation rate of the additive-containing polymers was in the order of *it*PP > LDPE = LLDPE >> HDPE. The biodegradation of additive-containing polyolefins was not proceeded in extracted seawater.

2. Photo-oxidation behaviors of elongated itPP films

Most of practical *it*PP products are oriented *it*PP. In this study, the photooxidative degradation of elongated *it*PP was examined.

UV exposure test @ 125W/m² for 4 weeks (302.4 MJ/m²)



IR carbonyl region and C=O index in *it*PP films

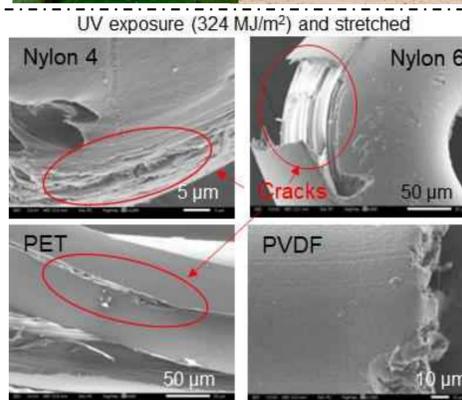
Microvoids formed in quenched *it*PP films after UV exposure

The oxidation susceptibility of *it*PP was in the order of quenched > gradually cooled > 2x elongated > 4x elongated > 8x elongated. Crystalline phase has strong resistance against photooxidation

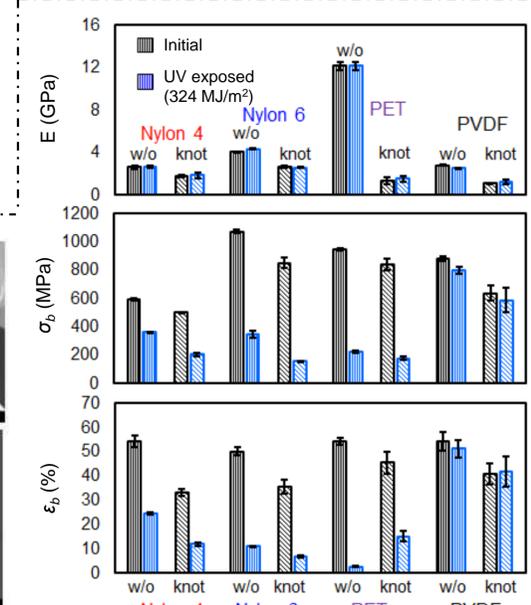
3. Environmental degradation of fishing lines

Ghost fishing caused by lost fishing gear has strongly affected the marine organisms. In this study, the environmental degradation behaviors of nylon 4, nylon6, PET and PVDF fishing lines were investigated.

Ghost fishing in Ocean



Surface morphology of fishing lines after UV exposure

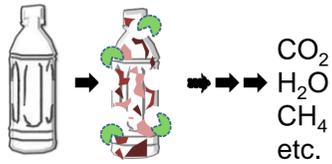


Comparison of mechanical properties of fishing lines before and after UV exposure

PET, nylon 6 fishing lines were degraded after photooxidation while PVDF was stable against UV exposure. Also, biodegradability of nylon 4 fishing lines were confirmed in extracted seawater.

## Background

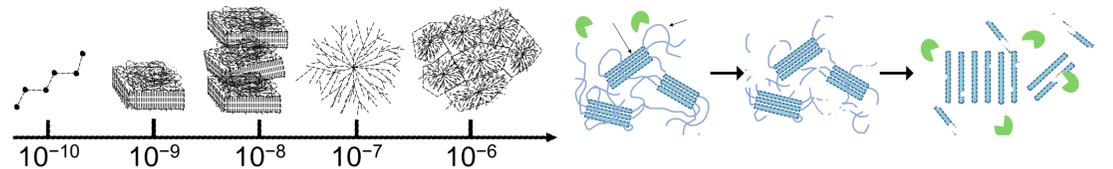
### Biodegradable polymers



- Agricultural mulch films
- Fishing nets
- PET bottles



### Hierarchical structures and degradation process



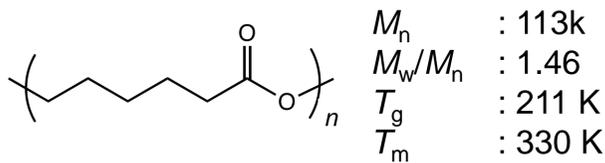
A detailed understanding of the relationship between hierarchical aggregation structures and degradation behaviors of polymers is required.

## Objective

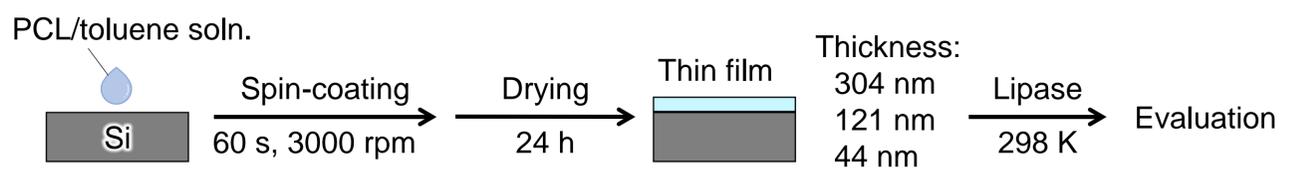
To reveal the relation between the crystalline lamellar structures of polymers and their biodegradability.

## Method

### Poly( $\epsilon$ -caprolactone) (PCL)



### Thin film preparation & enzymatic degradation



### Aggregation structures

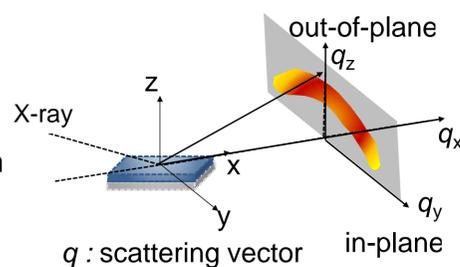
GIWAXD measurements

SPring8 BL03XU (2022A7218)

Sample to camera distance : 294 mm

Wavelength : 0.1 nm

Incident angle : 0.13°



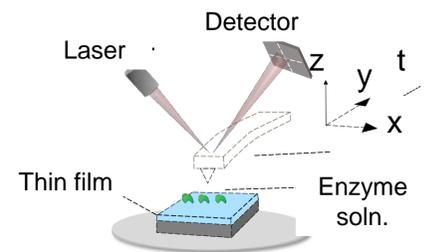
### Surface morphologies

AFM observations

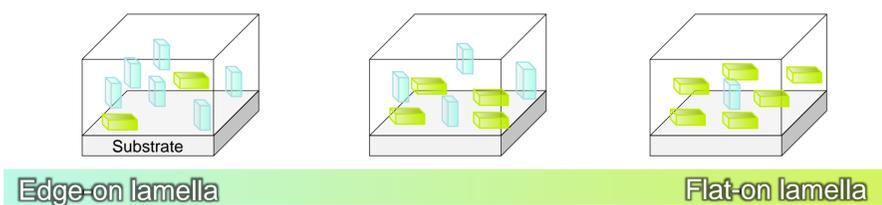
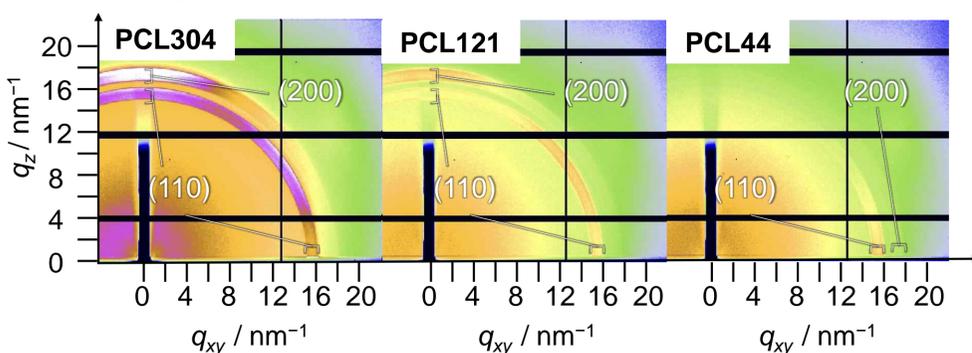
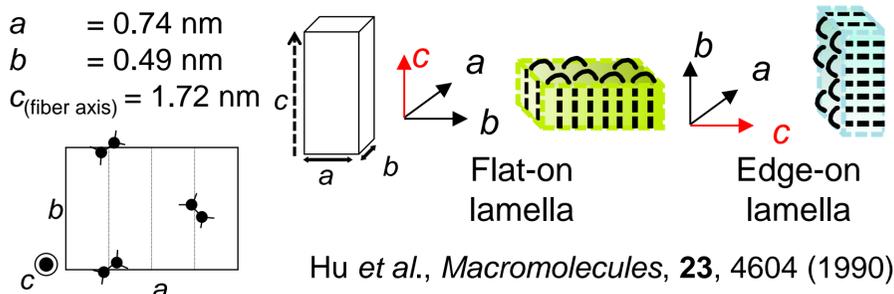
Cypher ES (OXFORD)

Mode : AC mode

Temperatures : 298 K

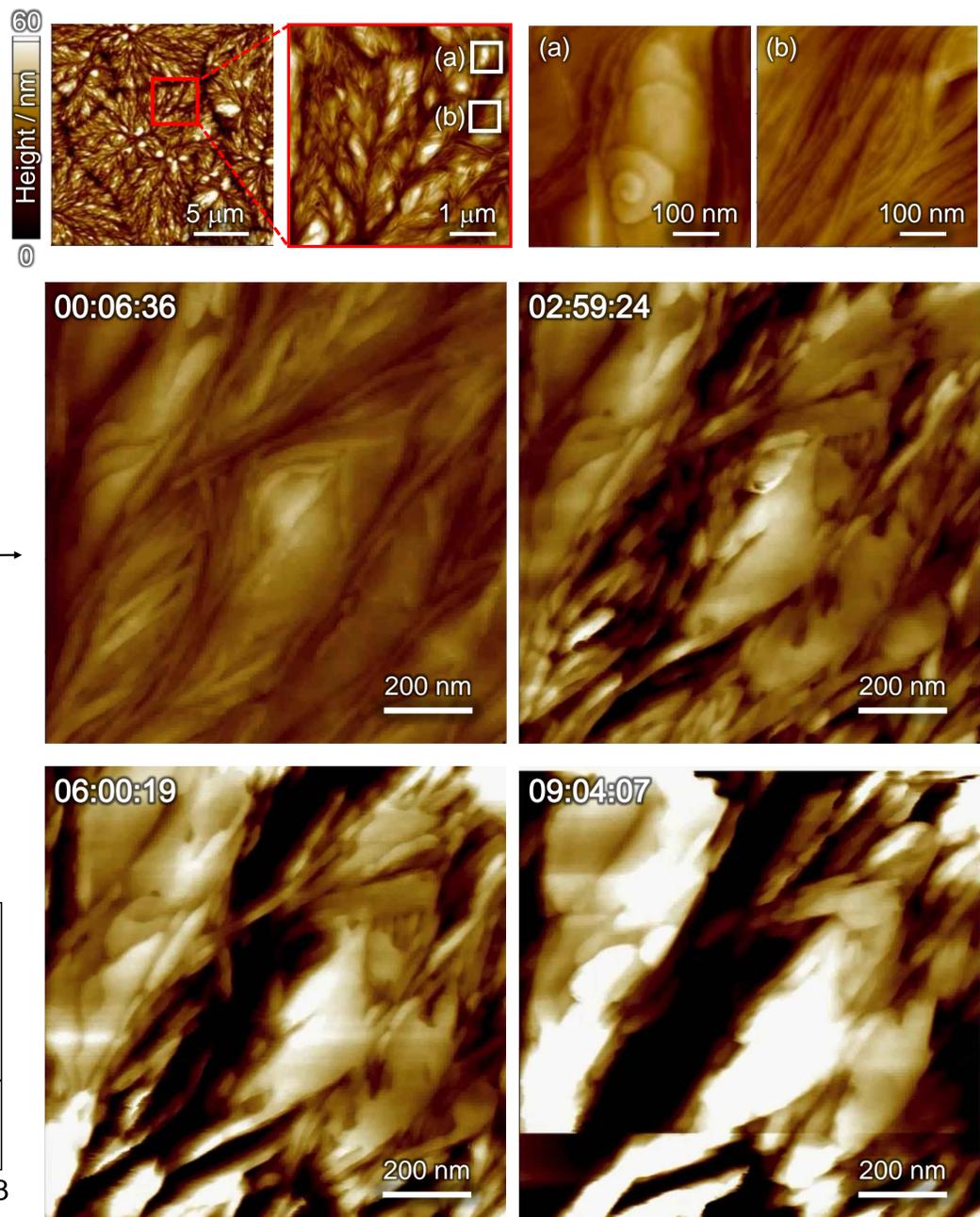
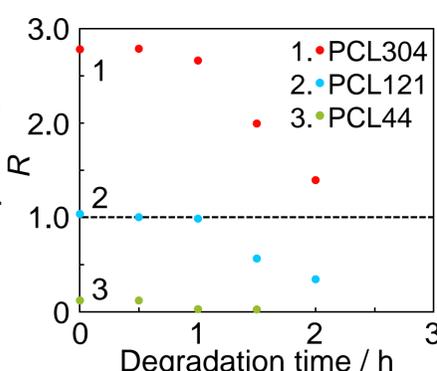


## Results & Discussion



### Degradation time dependence of edge-on lamella to flat-on lamella ( $R$ )

$$R = \frac{I_{\text{out of plane (110)}} + I_{\text{out of plane (200)}}}{I_{\text{in-plane (110)}} + I_{\text{in-plane (200)}}$$



## Conclusions

Biodegradability of polymeric materials can be controlled depending on the orientation of crystal lamellae.

(Acknowledgement: JPNP18016)

## 1. Introduction

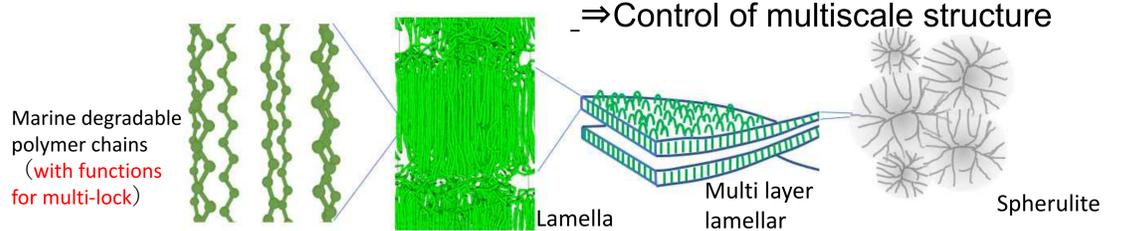
### Our technology

- Multi-scale simulation technology for polymer materials from atomic to mesoscopic scales.
- Material analysis using informatics technology

### Objective until end of FY2022

- Model study of degradation process of polymer crystals** : Major parts of marine plastic is the crystalline polymer, and we must consider the function degrading the crystal of polymer by unlocking. In this study, we perform the model simulation of the degradation of polymeric crystalline body by heat to analyze the degradation mechanism. We develop the precise analysis method to distinguish the crystalline and melt parts using machine learning, and using its tool we clarify the degradation process in detail.
- Model study of degradation of multi-lock polymer** : Yoshie et al developed the dynamic bond elastomer having the functions of toughness in use and degradation in marine. In this study, we made the model of dynamic bond elastomer to clarify the functions of both toughness and degradation. In the near future, we will design its material having those functions in high level.

### Problems in development of polymer materials



## 2. Model study of degradation process of polymer crystals

### 2.1 development of analysis method of local structure

#### Problems:

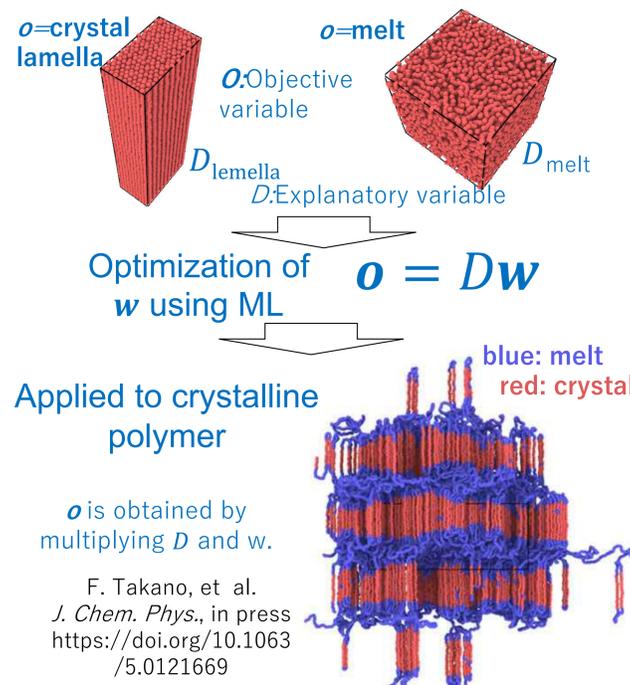
- Too many candidates of order parameter
- Arbitrariness to choose order parameter

⇒ We develop the software to identify the local structure automatically using machine learning technique. Its name is Molecular Assembly structure Learning package for Identification of Order parameters (MALIO)

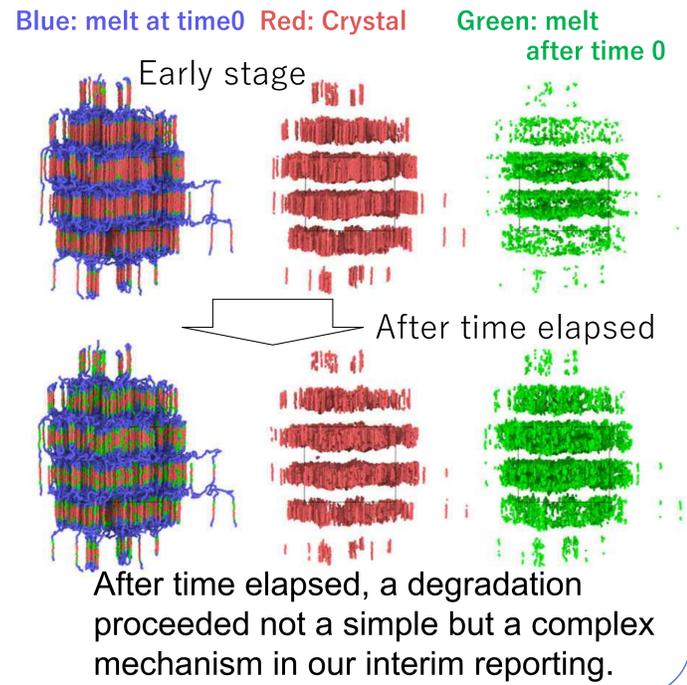
MALIO can identify local structures 40 times faster than the previous our software which is developed in another NEDO project.

K. Z. Takahashi, Phys. Chem. Chem. Phys. (in press)  
DOI: 10.1039/D2CP03696G

### 2.2 precise analysis of local structure



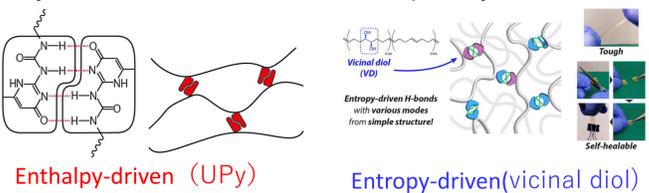
### 2.3 Analysis of degradation process



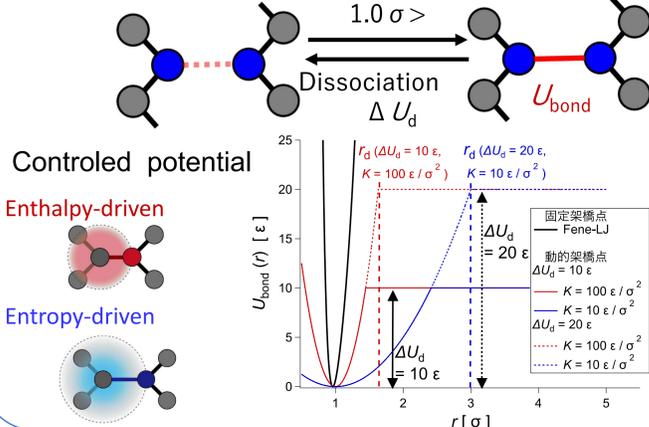
## 3. Model study of degradation of multi-lock polymer

### 3.1 Modeling of dynamic bond elastomer

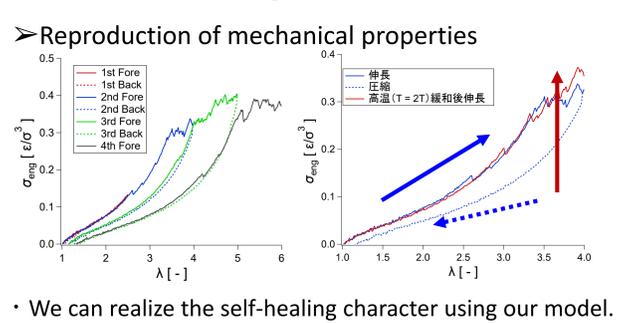
- Dynamic bond elastomer developed by Yoshie et al



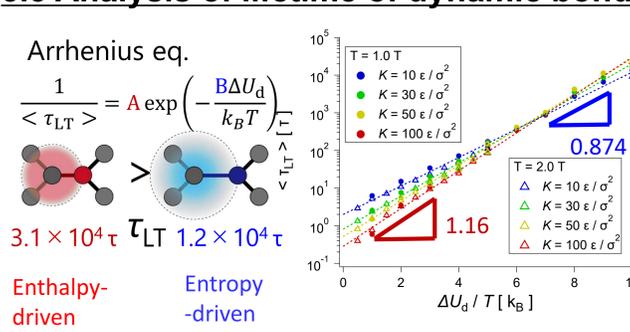
- Model of dynamic bond



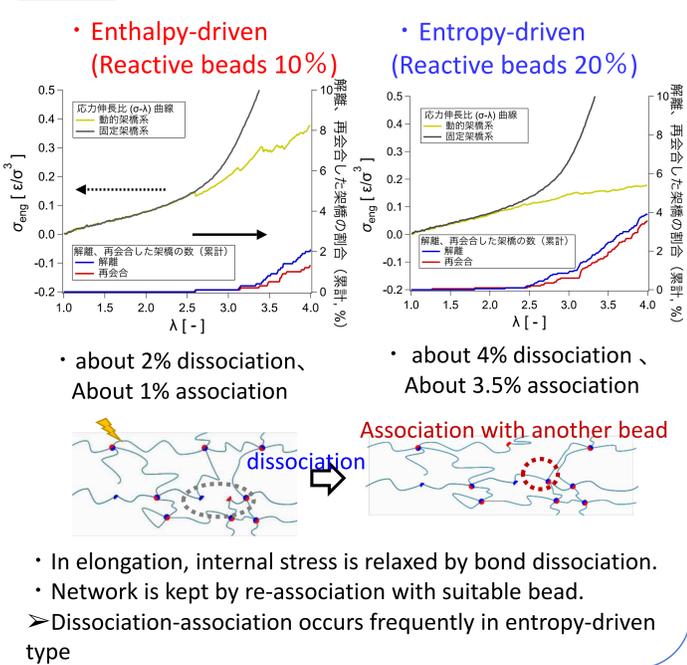
### 3.2 Reproducing experimental results



### 3.3 Analysis of lifetime of dynamic bond



### 3.4 Dissociation-association in each model



## 4. Summary and Future work

### Model study of degradation process of polymer crystals

- Development of MALIO which can identify the local structure.
- Application study of MALIO to the problem of dynamic structure in degradation.

### Future work

- MALIO is applied to the degradations of other polymers.
- ⇒ We will consider proposal to accelerate degradations due to unlocking effect

### Model study of degradation of multi-lock polymer

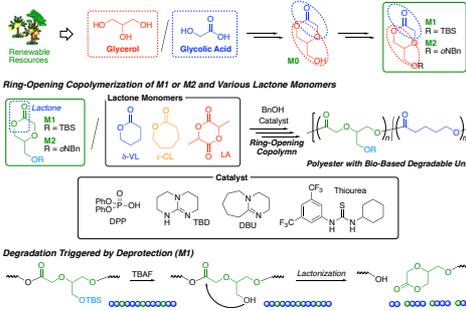
- Development of models for both entropy-driven and enthalpy-driven types.
- Detail analysis of dissociation-association process which cannot be observed in experiments.

### Future work

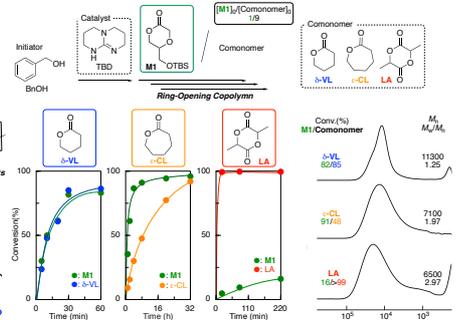
- Development of updated model and design of suitable elastomer for marine degradation.
- ⇒ Our simulation is applied to the collaborative study with company team.

Ring-Opening Polymerization of Novel Lactones with Protected Hydroxy Group Derived from Biomass and Deprotection-Induced Polymer Degradation

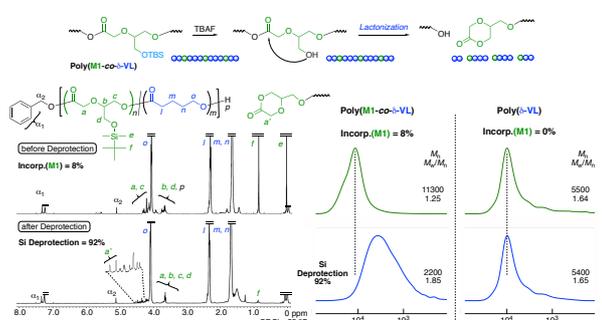
Introduction



Ring-Opening Copolymerization of Silyl-Protected Lactone

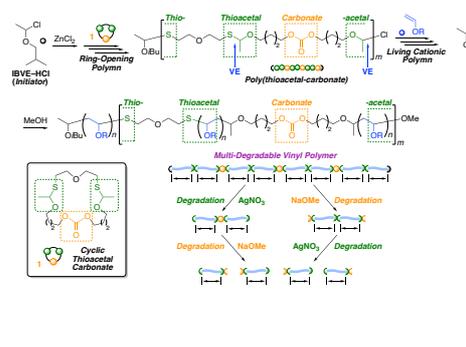


Deprotection-Induced Polymer Degradation

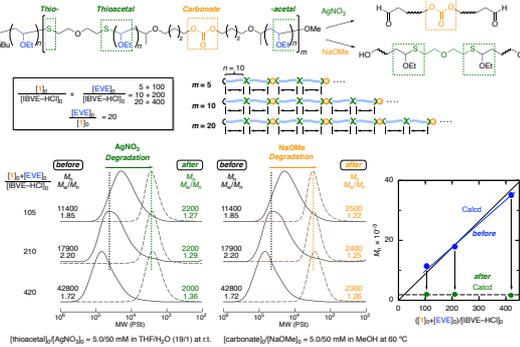


Synthesis of Multi-Degradable Vinyl Polymers by Living Cationic Polymerization of Cyclic Thioacetals with Cleavable Bonds

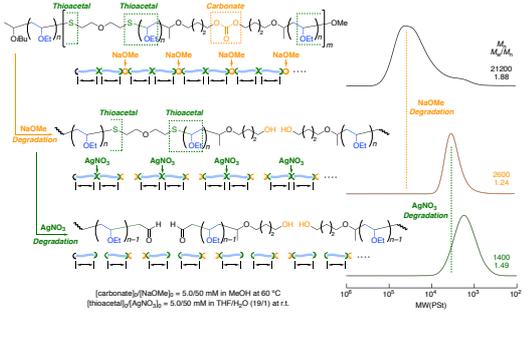
Introduction



Control of Molecular Weights before and after Degradation

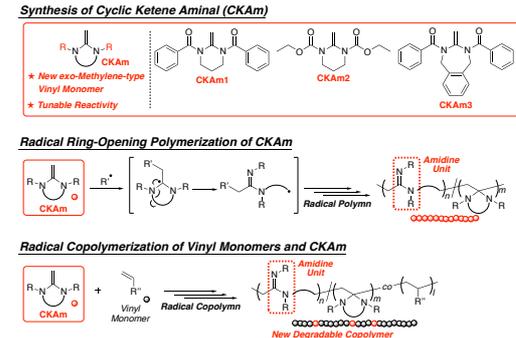


Multi-Degradation Based on Carbonate and Thioacetal Bonds

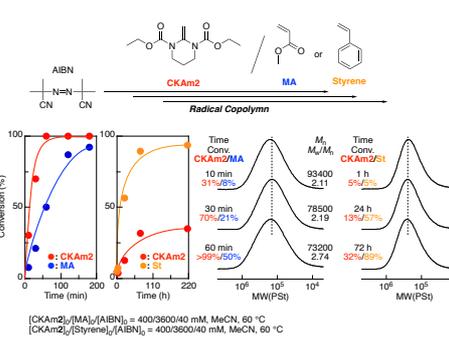


Synthesis and Radical Polymerization of Cyclic Ketene Aminals for Degradable Vinyl Polymers

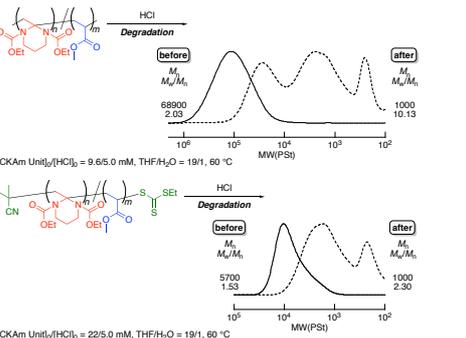
Introduction



Radical Copolymerization of Cyclic Ketene Amino (CKAm)

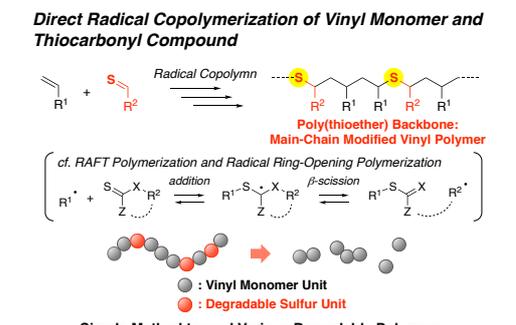


Degradation of Copolymers of Acrylate and CKAm

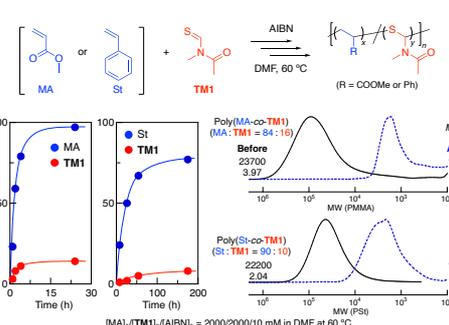


Direct Radical Polymerization of Thiocarbonyl Group for Degradable Vinyl Polymers

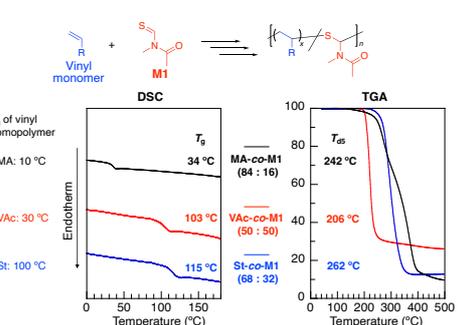
Introduction



Radical Copolymerization of Thioamide and Vinyl Monomers

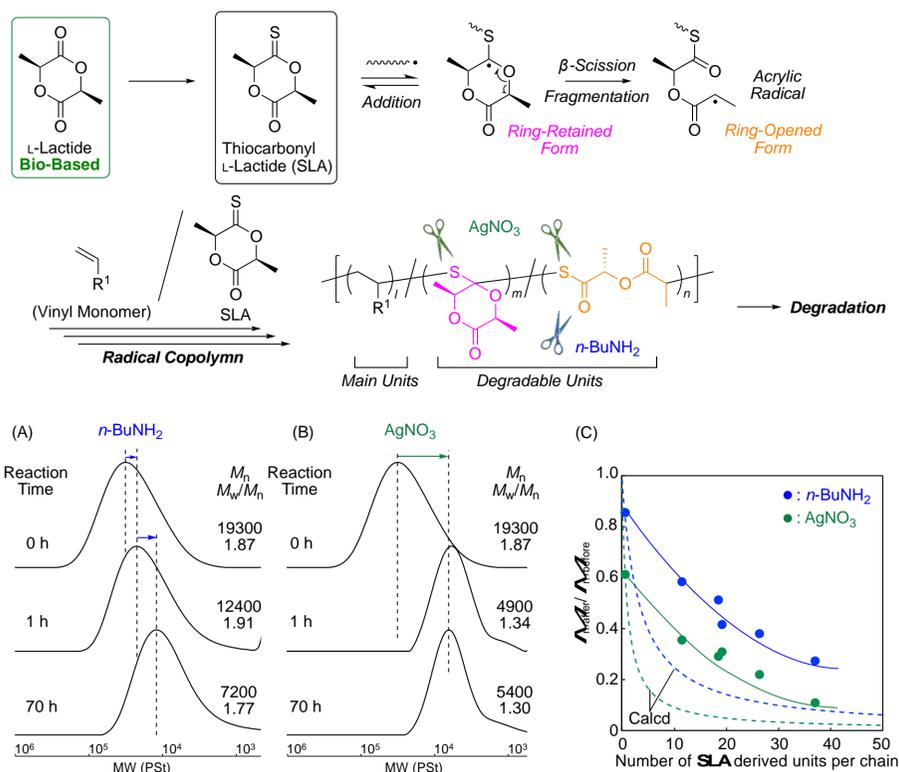


Thermal Properties of Vinyl Copolymers with Thioamide



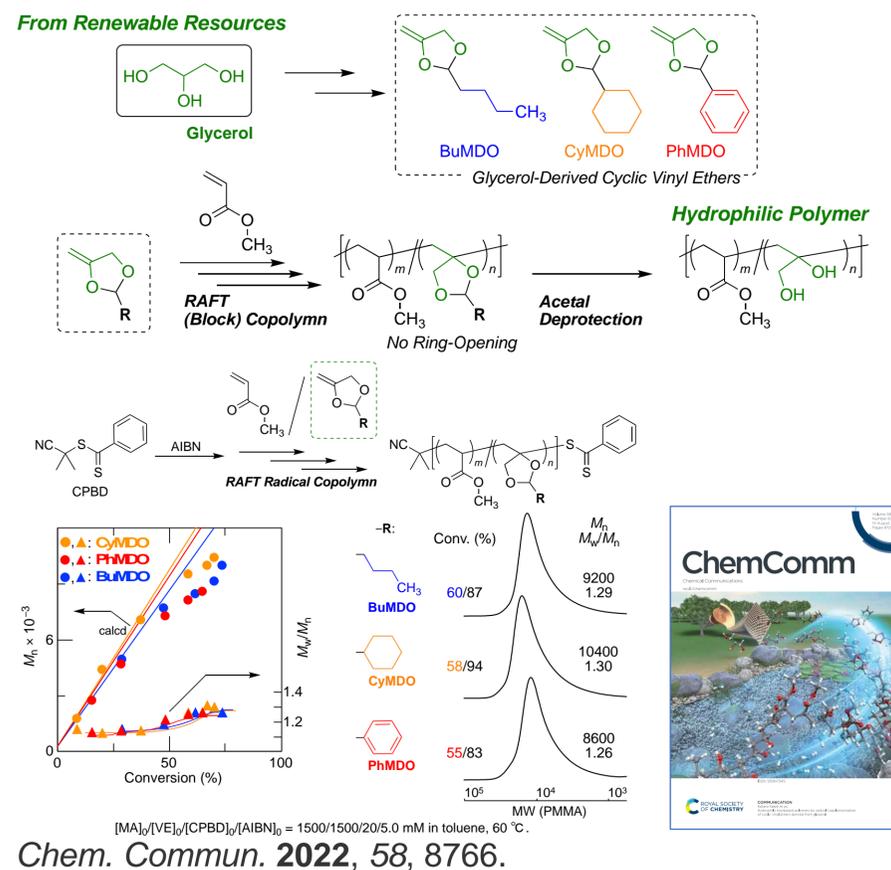
For developing multi-locked degradable polymers from non-edible biomass, we will develop a multi-lock technology by utilizing the technology of precision polymerization, which we had cultivated in the petroleum chemicals, to biomass-based and multi-locked degradable polymers. By the polymerization of non-edible biomass as a raw material, we propose the concept of a manufacturing method for multi-lock biopolymers that can be degraded in the ocean collaborating with industry.

### Addition-Fragmentation Ring-Opening Polymerization of Bio-Based Thiocarbonyl L-Lactide for Dual Degradable Vinyl Copolymers

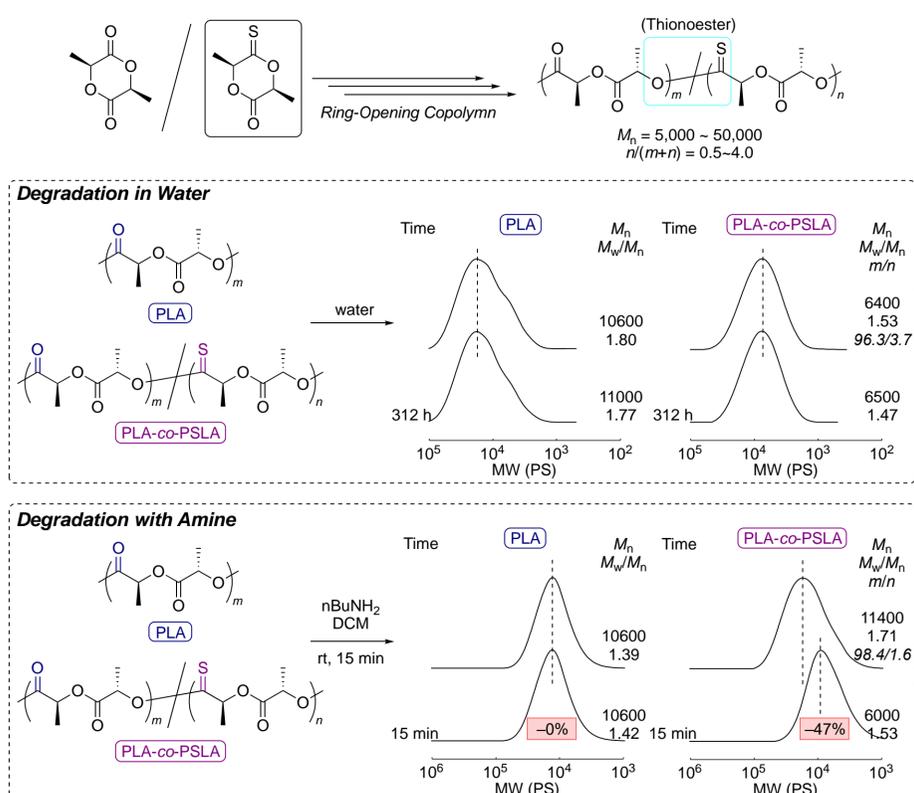


Japanese Patent Application 2021-131293  
*Macromol. Rapid. Commun.* **2022**, in press.

### Hydrophilic Bio-Based Polymers by Radical Copolymerization of Cyclic Vinyl Ethers Derived from Glycerol



### Improved Degradability of Poly(Lactic Acid) by Introducing Thionoester Linkages via Ring-Opening Copolymerization



Japanese Patent Application 2021-131293

### Interlocking Degradation of Vinyl Polymers via Main-Chain C-C bonds Scission by Introducing Pendant-Responsive Comonomers

