

Integrated Electrochemical Systems for Scalable CO₂ Conversion to Chemical Feedstocks



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Project Members :

The University of Tokyo, Osaka University, RIKEN, UBE Corporation, Shimizu Corporation, Chiyoda Corporation, Furukawa Electric Co. Ltd., Maxell, Ltd.

1. Project Overview

2. Progress and Results

3. Summary and Future Challenges

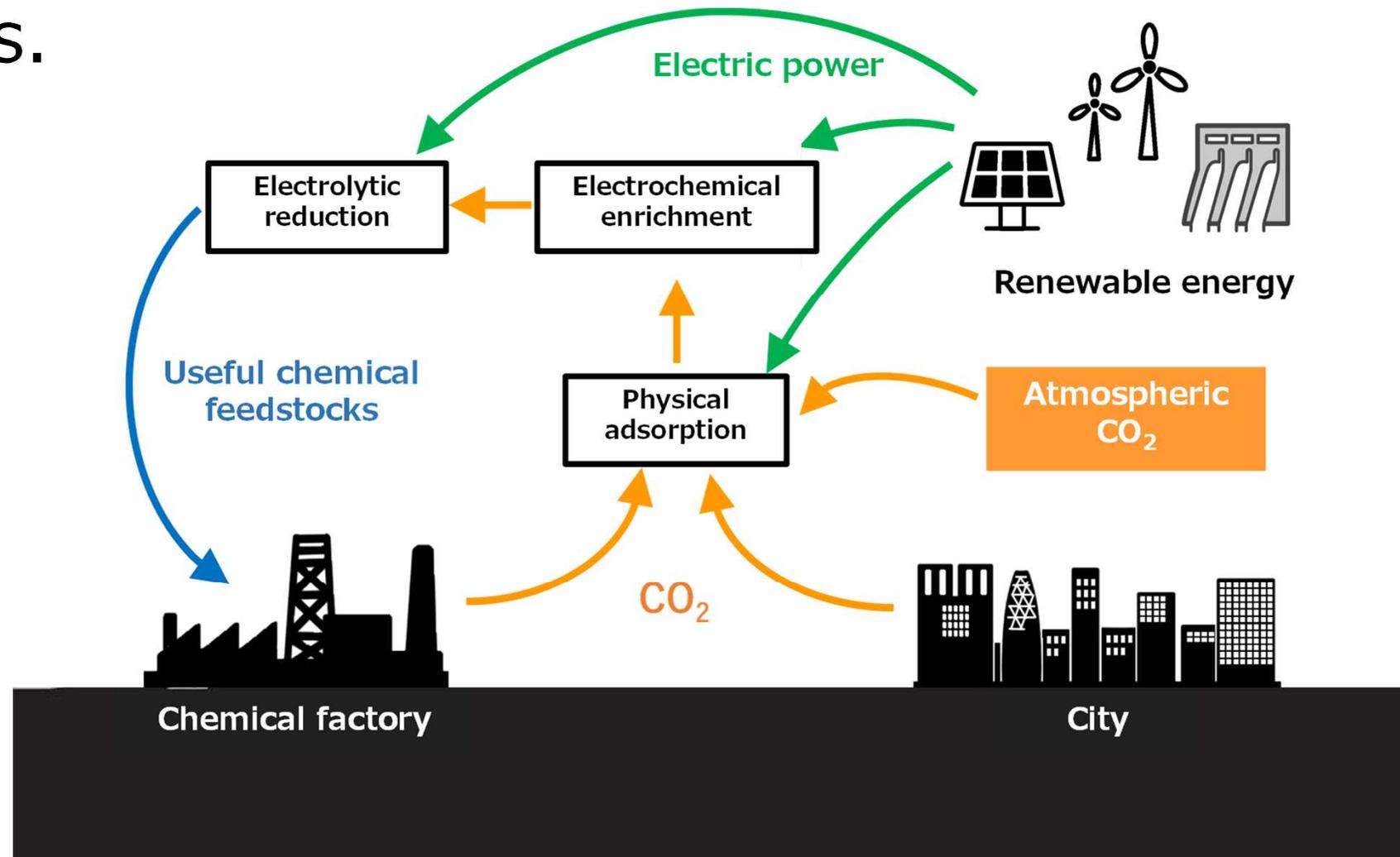
1. Project Overview

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(Goals in FY 2029)

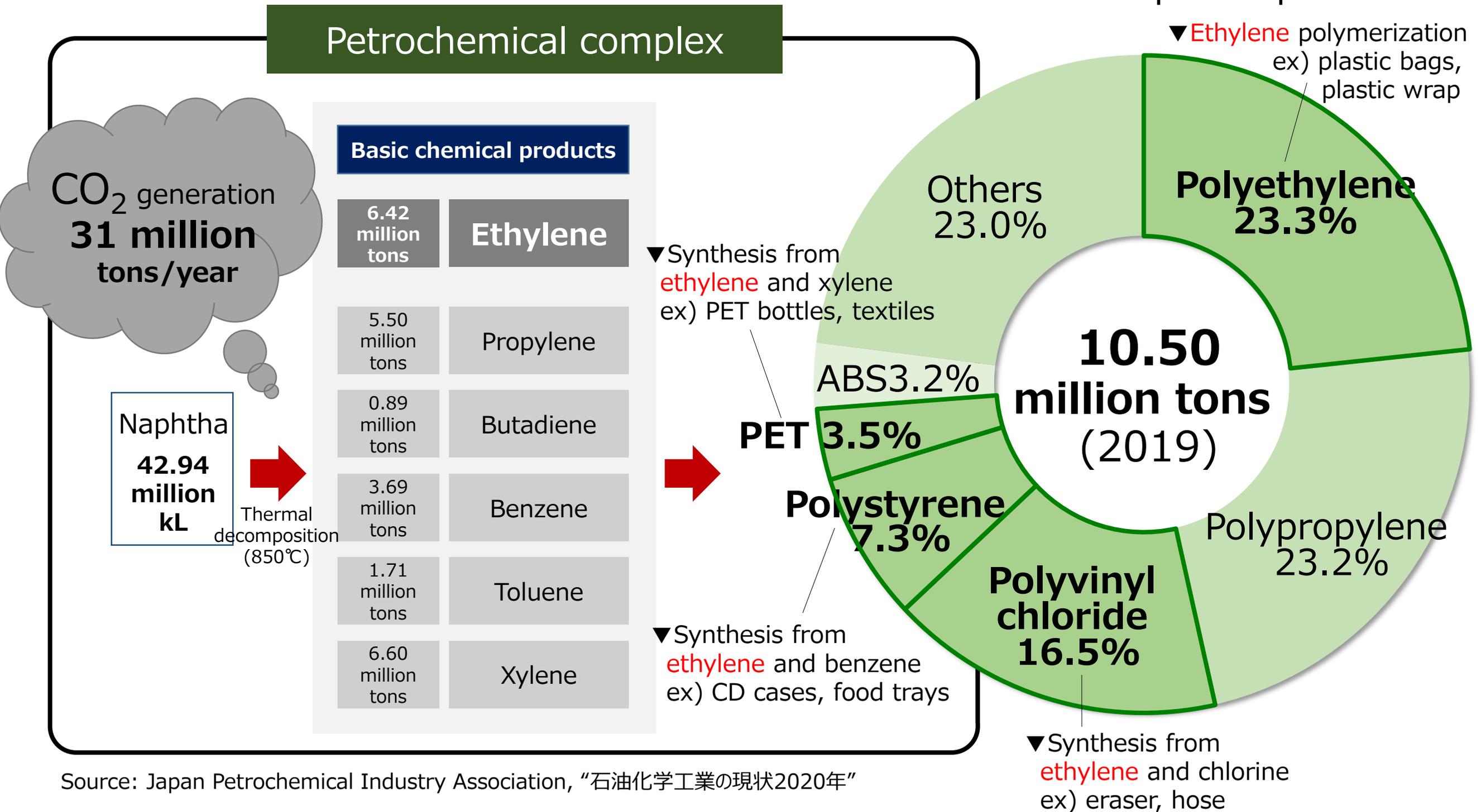
Development of a system to convert atmospheric CO₂ into useful chemical feedstocks based on electrochemical processes.



Achievement of carbon cycle based on electricity which is a platform of future energy system
~ Toward 100 million ton/year reduction of CO₂ emissions @ 2050 ~

Products : Ethylene

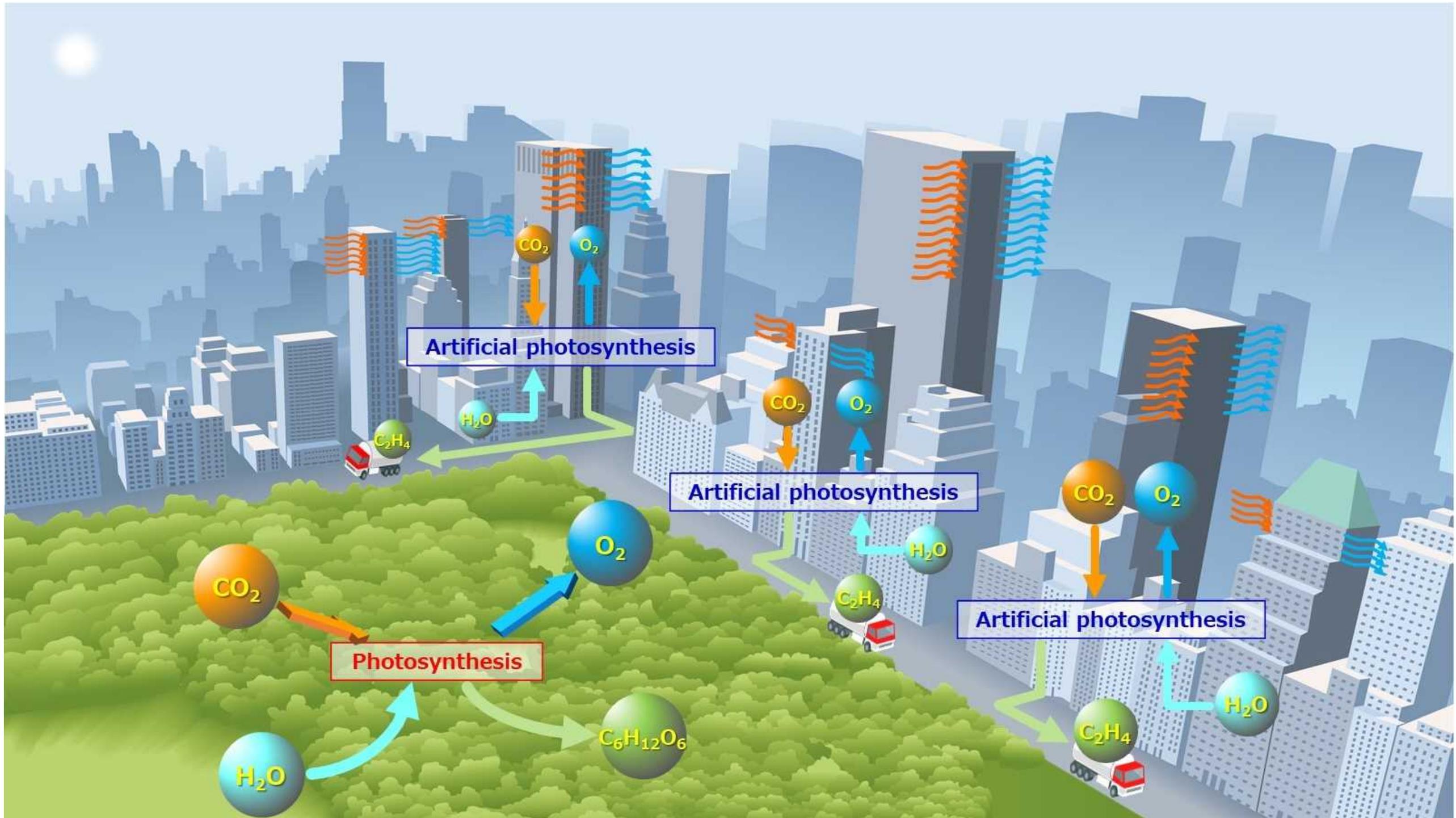
Ethylene is used in about half of all plastic products.



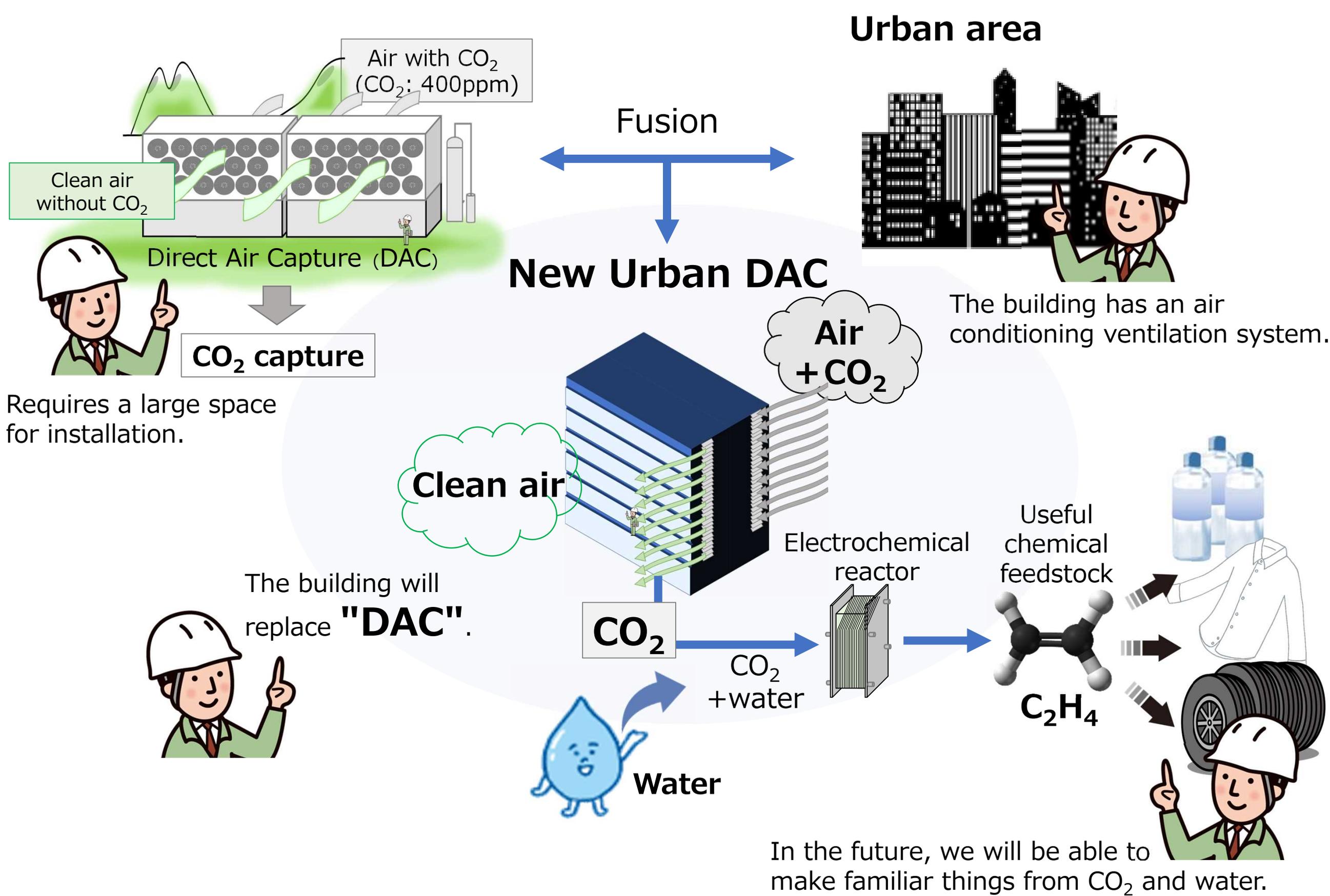
Source: Japan Petrochemical Industry Association, "石油化学工業の現状2020年"

Source: Plastic Waste Management Institute, "プラスチックリサイクルの基礎知識2020"
<http://www.pwmi.or.jp/pdf/panf1.pdf>

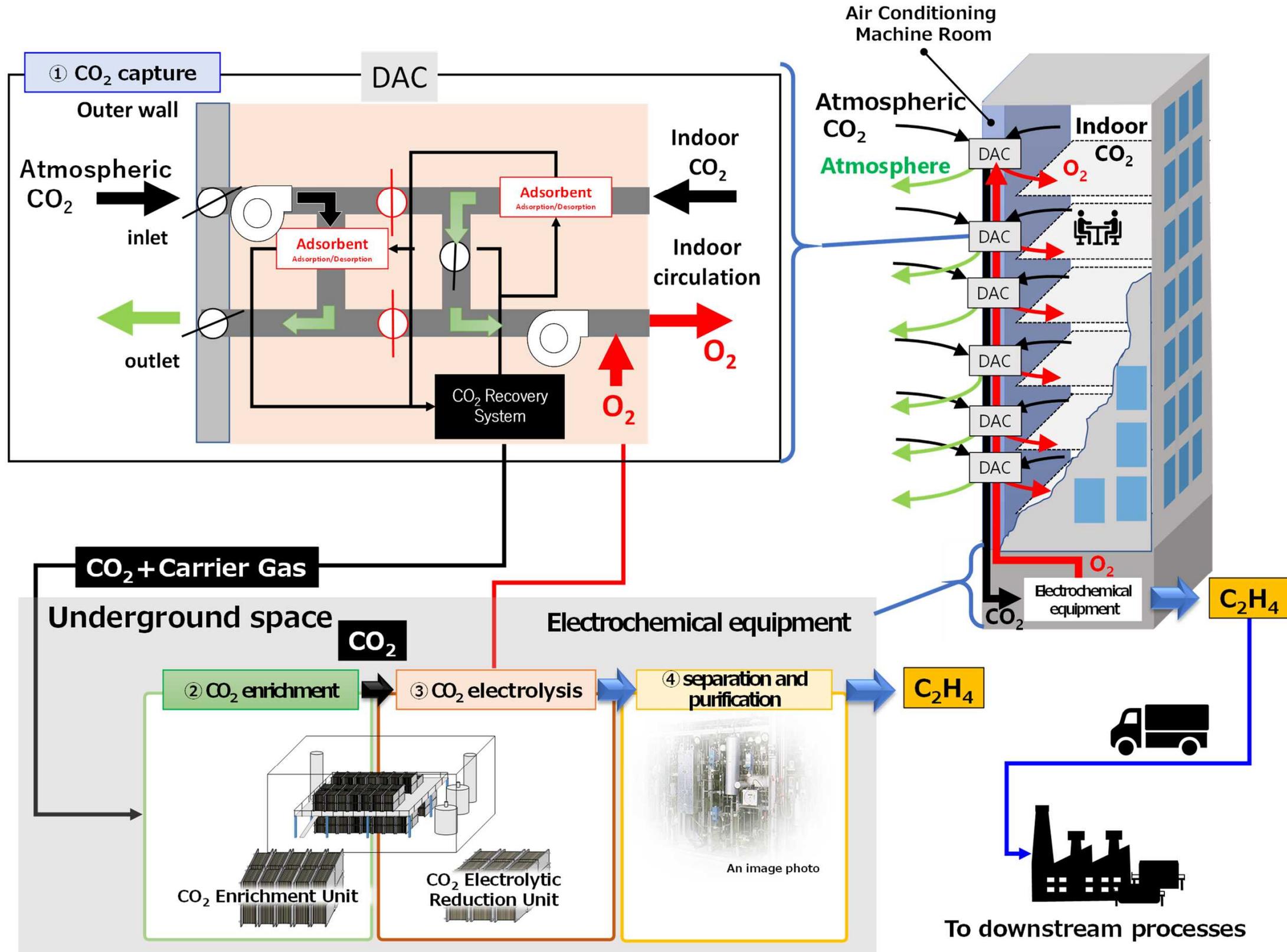
Urban DAC-U System (Artificial Photosynthesis)



Buildings Function as "Air purifiers for a city"

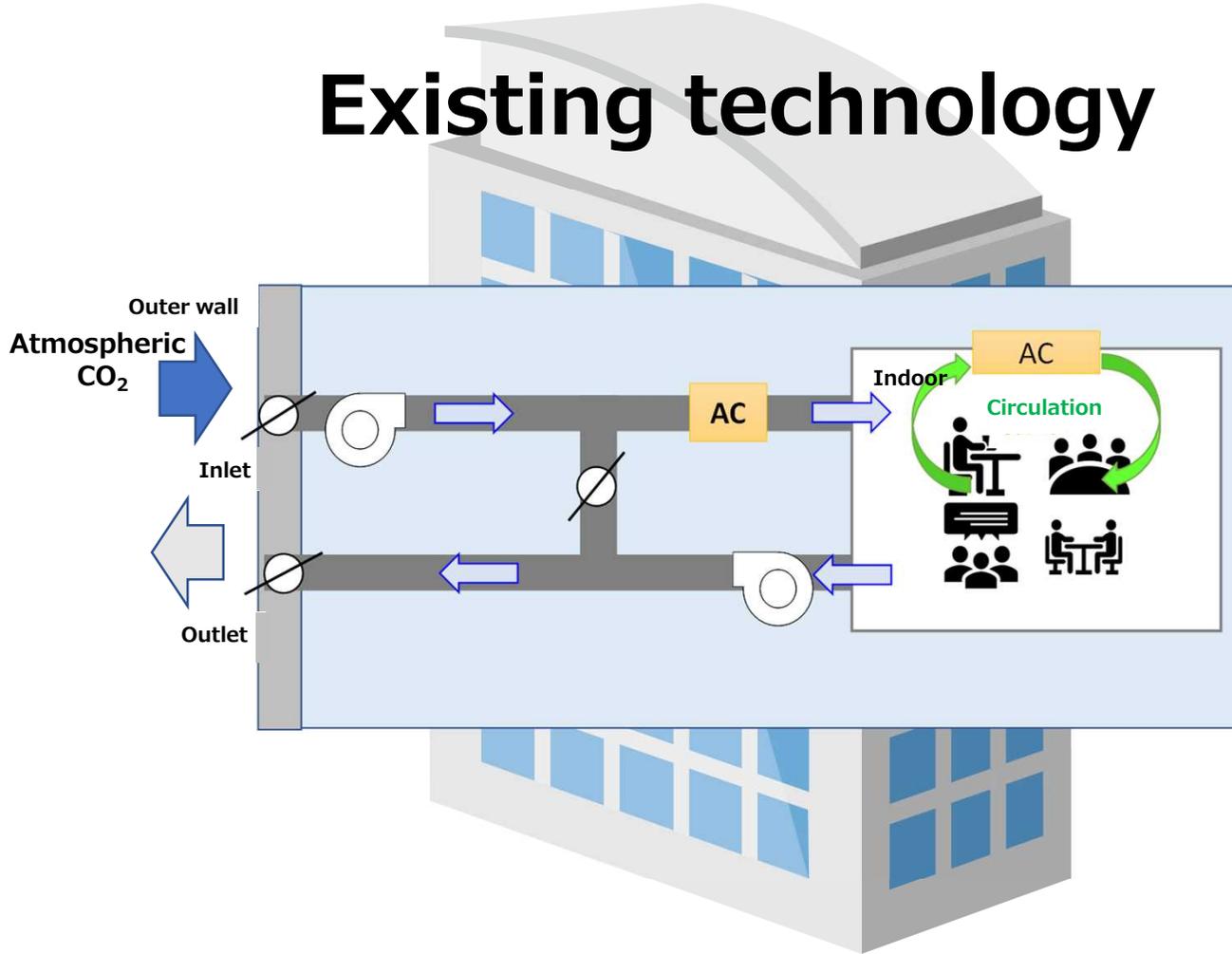


CO₂ and O₂ circulation

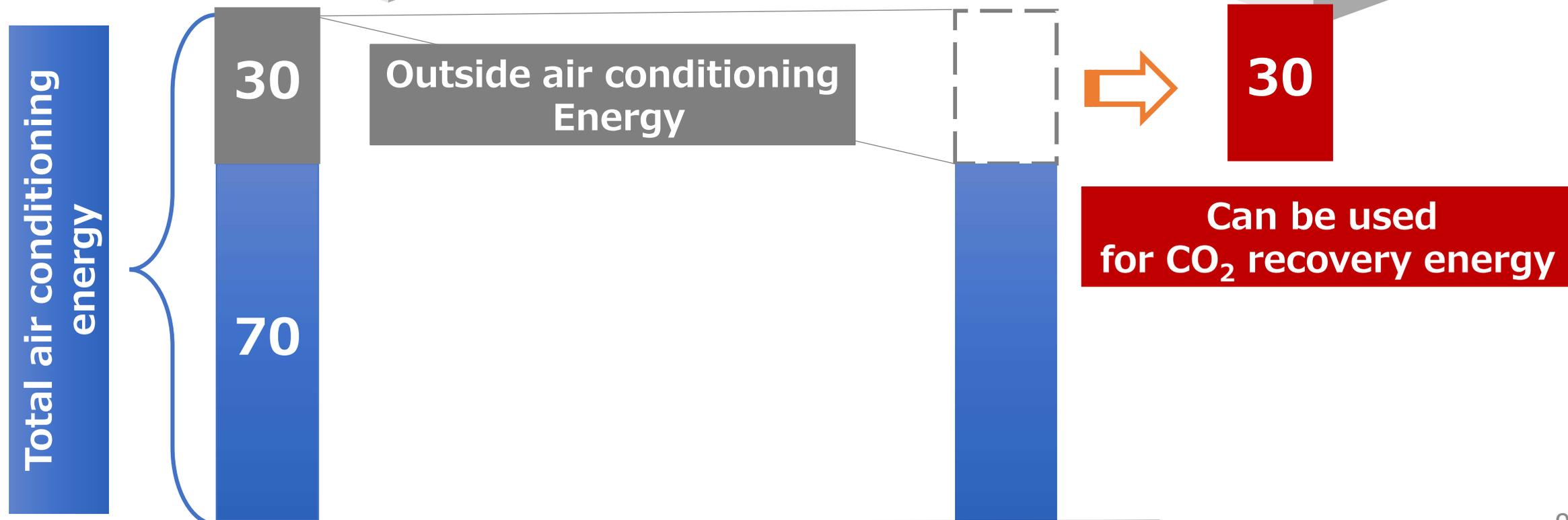
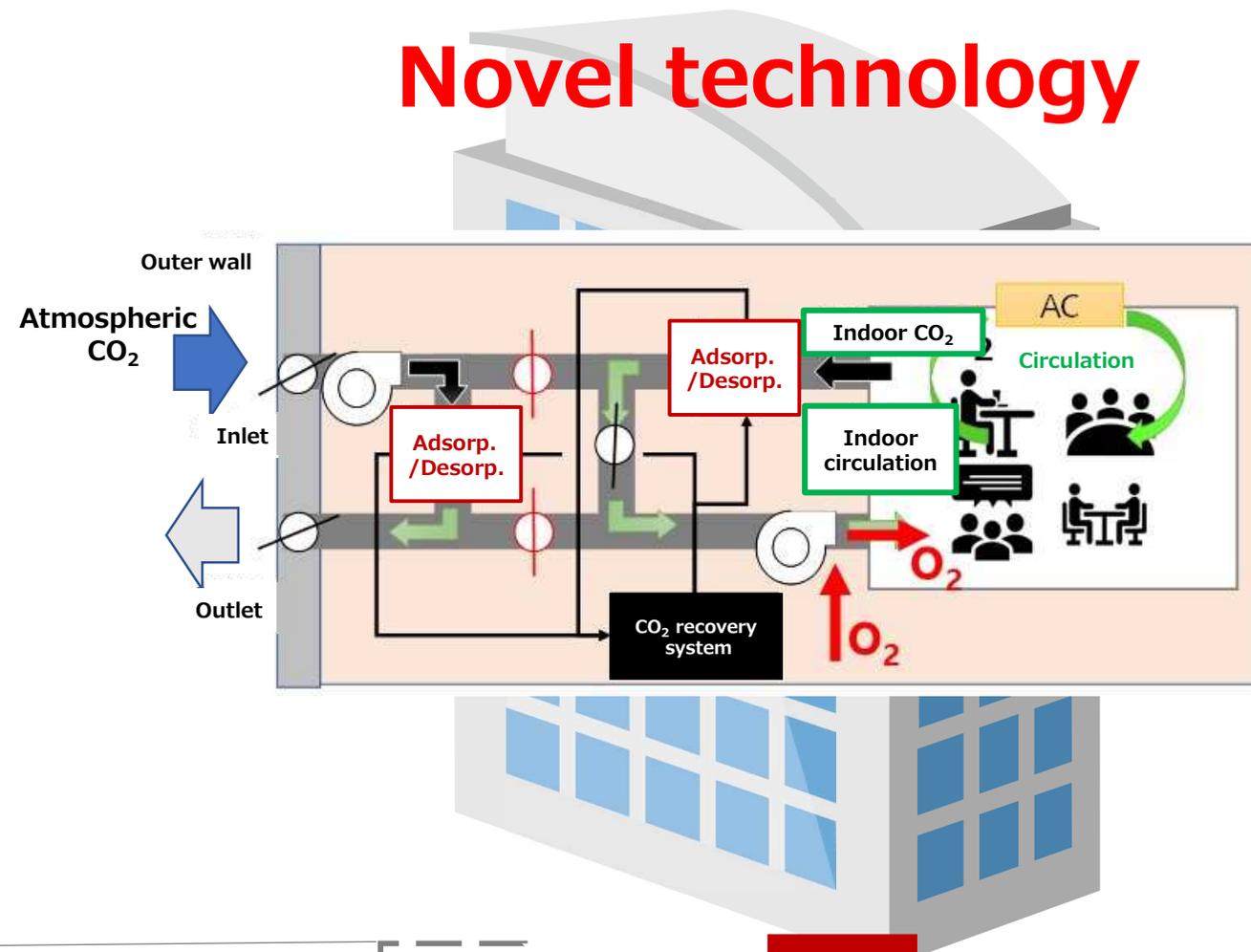


Energy Savings in Air Conditioning and Ventilation Systems

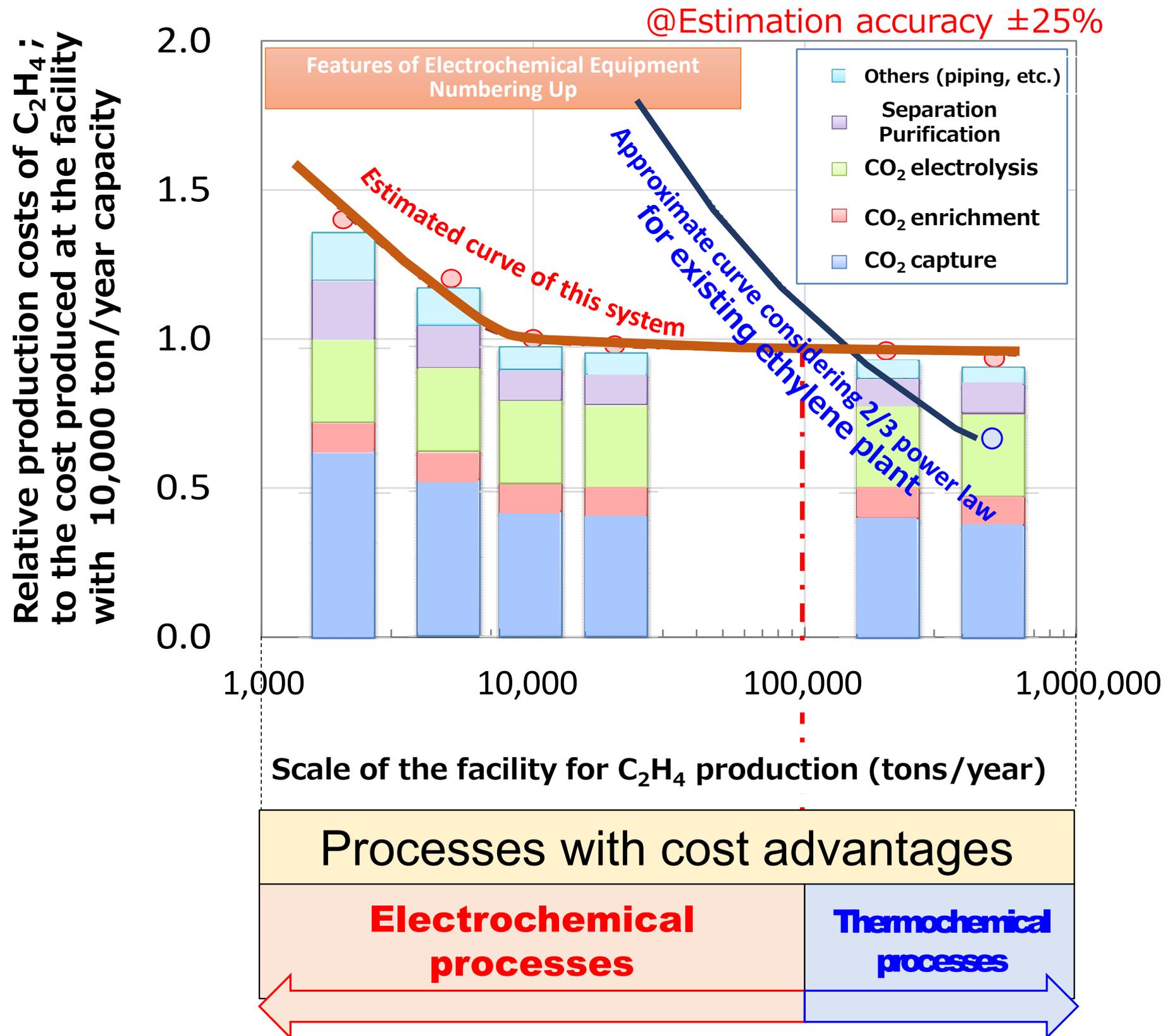
Existing technology



Novel technology



Estimated relative production cost of 1 ton ethylene

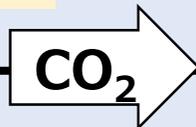
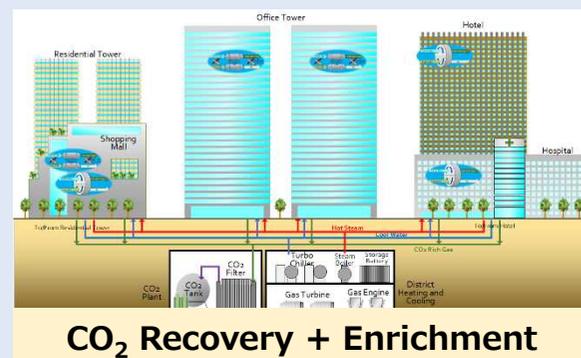


Scalable Social Implementation for Various Scenes

Transfer material



Carbon dioxide



Electrolytic reduction

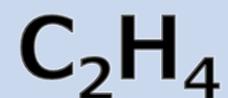
Ethylen production \Rightarrow Chemical plant

Ethylene

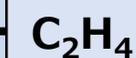
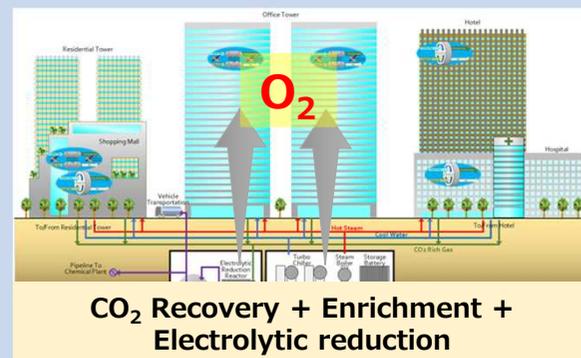
Butadiene

Ethylene glycol

Electrolysis units are located away from urban areas



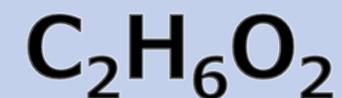
Ethylene



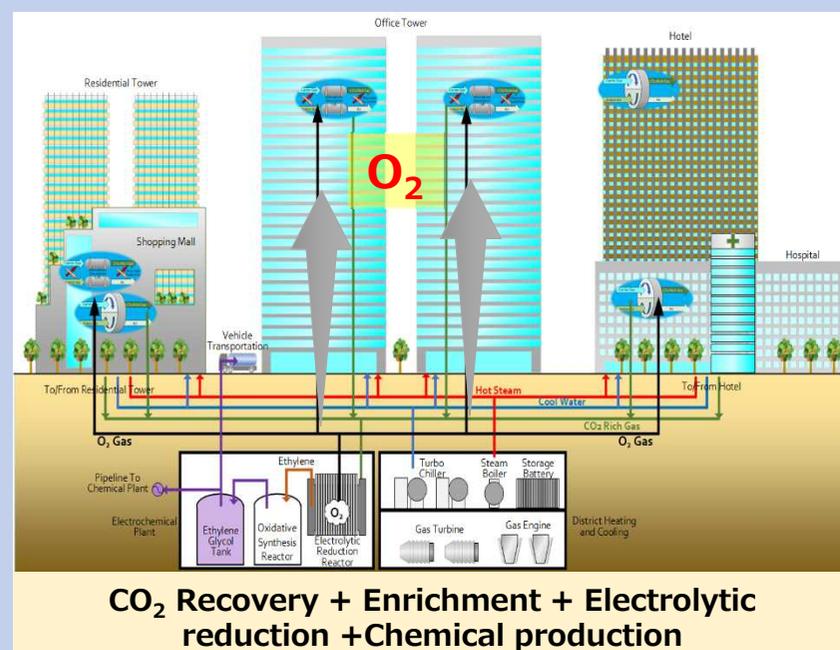
Chemical plant

Butadiene

Ethylene glycol



Ethylene glycol

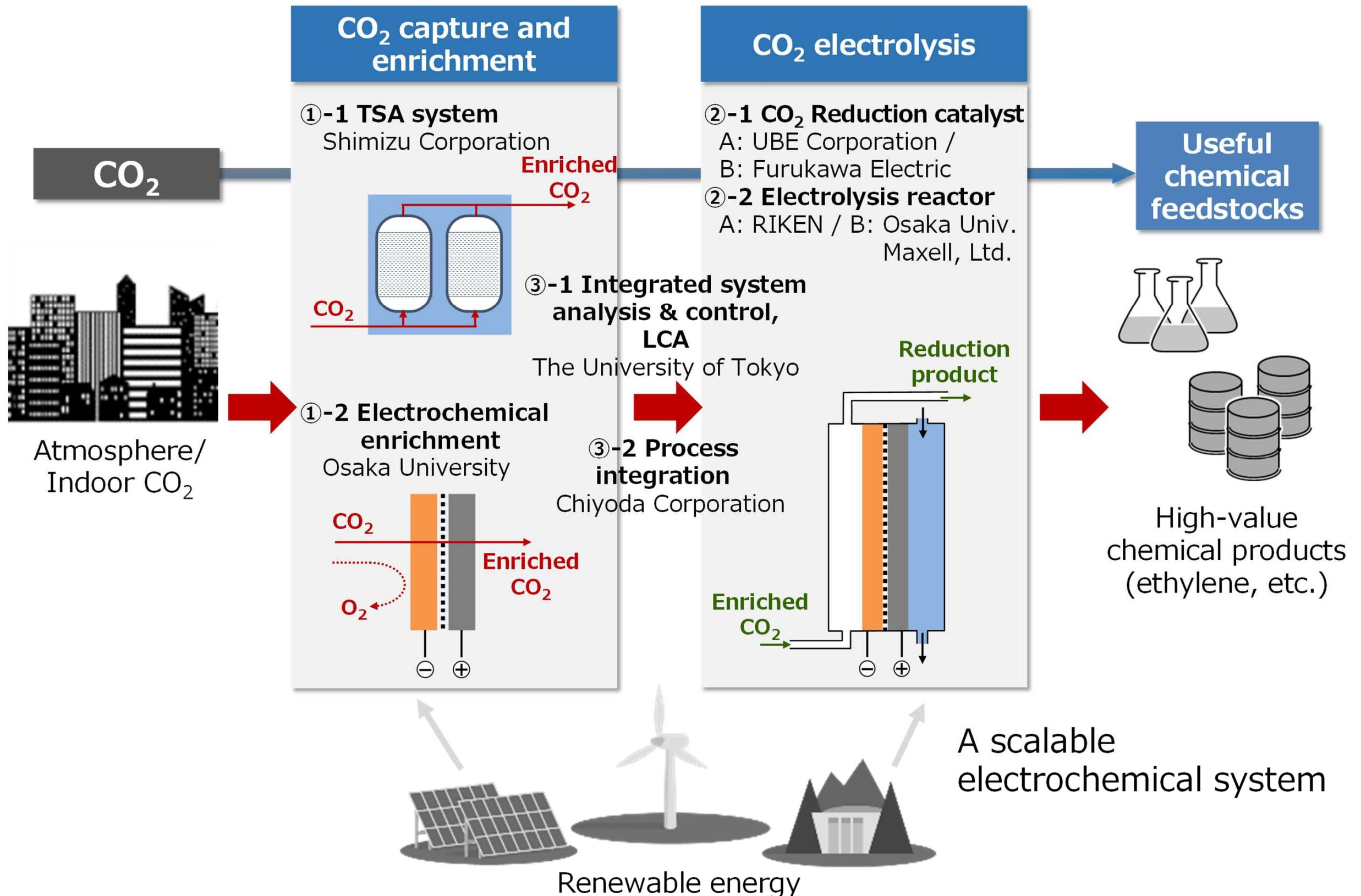


EG

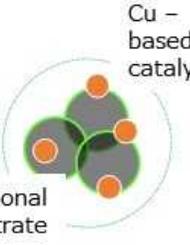
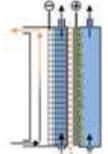
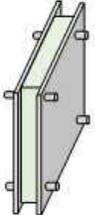
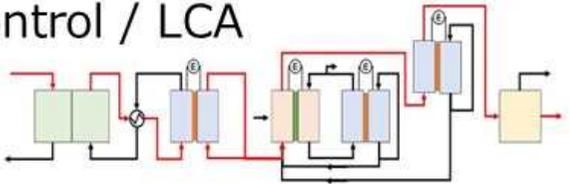
Antifreeze, Solvents, Cosmetics, Polyester raw materials, etc.

Related industrial plants

Work Packages of the Project



R&D Items and Players

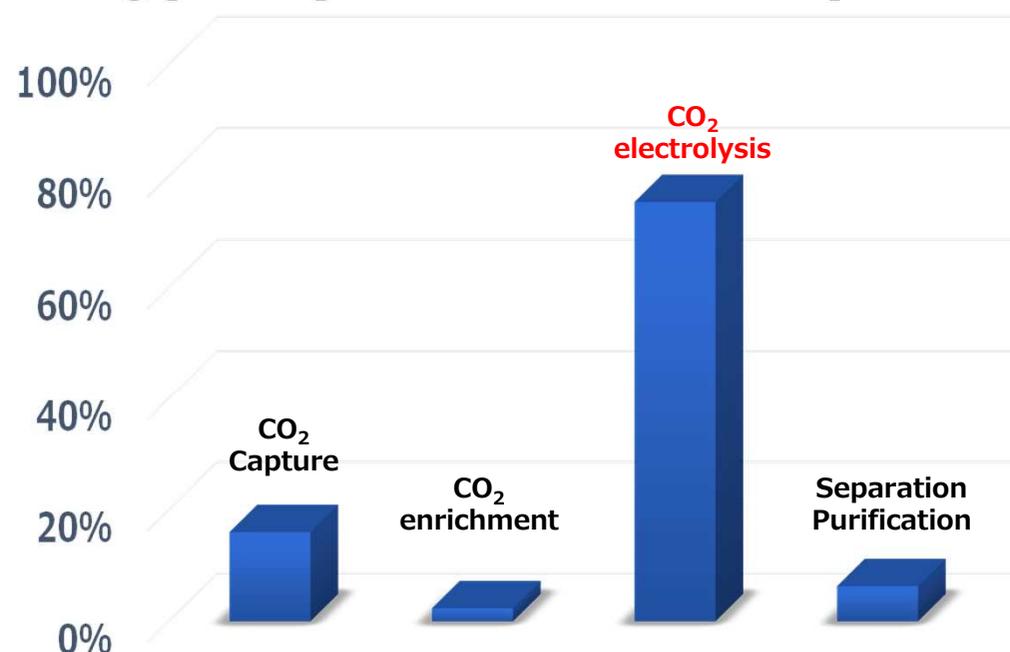
R&D items		Player			
CO ₂ capture and enrichment	CO ₂ capture by TSA method 	SC	Collaborative member		
	Electrochemical CO ₂ enrichment 	OSU	Collaborative member		
CO ₂ electrolysis	Reactor member	Catalyst 	Substrates	OSU UTK	UBE
		Cu-based materials	FKW		
		Gas-Diffusion Electrode (GDE) 	UBE, FKW, Maxell		
		MEA-based reactor 	Membrane	RIKEN	Collaborative member
			Reactor		
Stack 					
System integration	Reaction process development / Process integration Integrated system analysis & control / LCA 	UTK	CYD		

*UTK: The University of Tokyo, OSU: Osaka University,
 RIKEN: Institute of Physical and Chemical Research,
 UBE: UBE Corporation, SC: Shimizu Corporation,
 CYD: Chiyoda Corporation, FKW: Furukawa Electric Co., Ltd, Maxell : Maxell, Ltd.

	2022	2024	2029
CO ₂ emission* (t-CO ₂ /t-C ₂ H ₄)	+1.0 ~ +1.5 at device level	+0.5 ~ +1.0 at laboratory scale 1,000 hours	< -0.5 at pilot plant scale 5,000 hours
CO ₂ emission during operation	-0.5 ~ 0.0 (5.0~4.5 V, FE= 55 ~65%)	-1.0 ~ -0.5 (4.5~3.8 V, FE= 55 ~80%)	< -2.0 (3 V, FE= 80%)
CO ₂ emission upon equipment manufacturing	+1.5	+1.5	+1.5

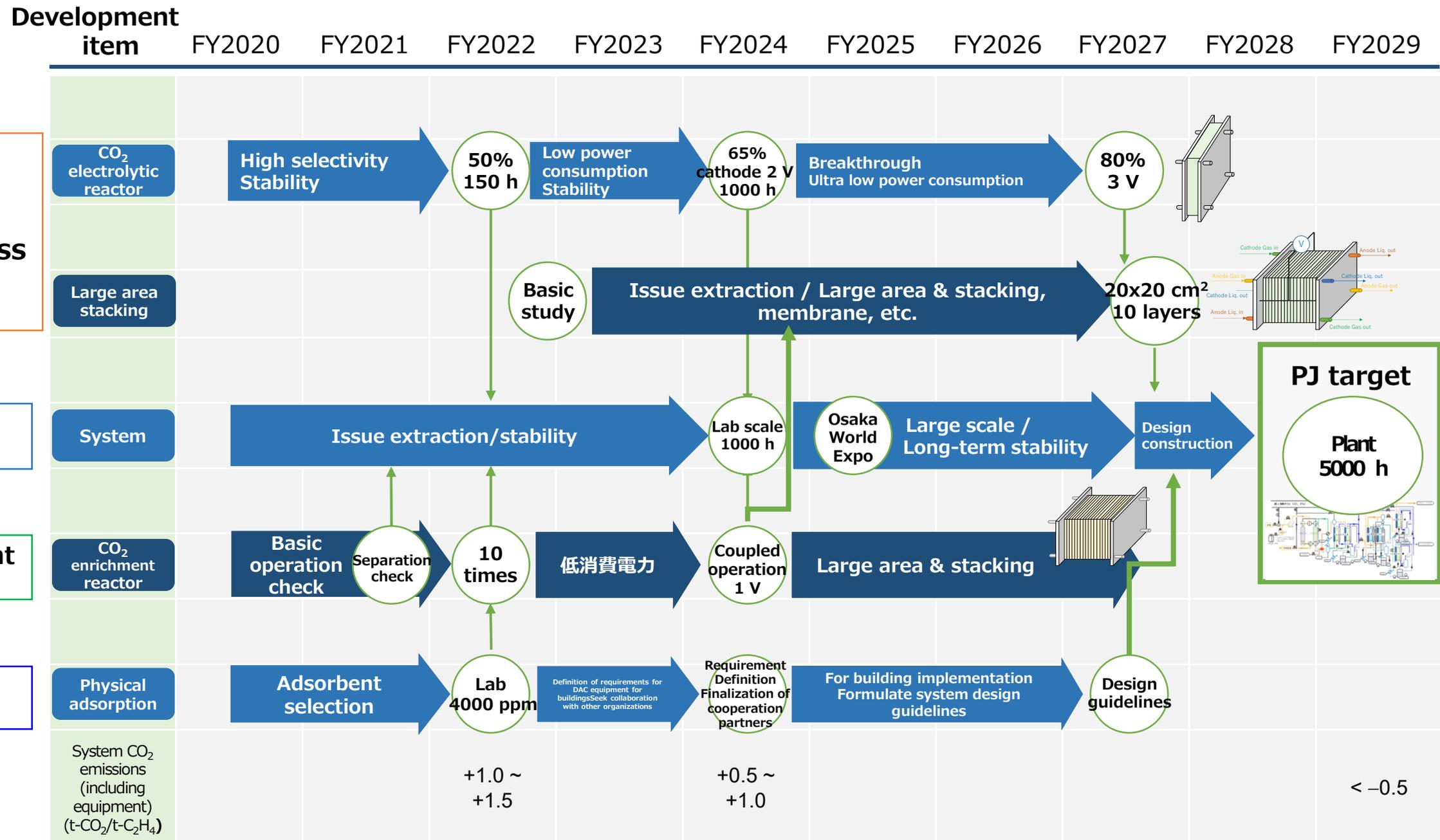
※CO₂ emission of the entire system from atmospheric CO₂ capture to ethylene production (including emission upon manufacturing of equipment)

Energy required for each process



- Rough estimation with ±20% accuracy
- Assuming solar power generation of 30g-CO₂/kWh as CO₂ emission from electricity
- Includes CO₂ emission from equipment manufacturing

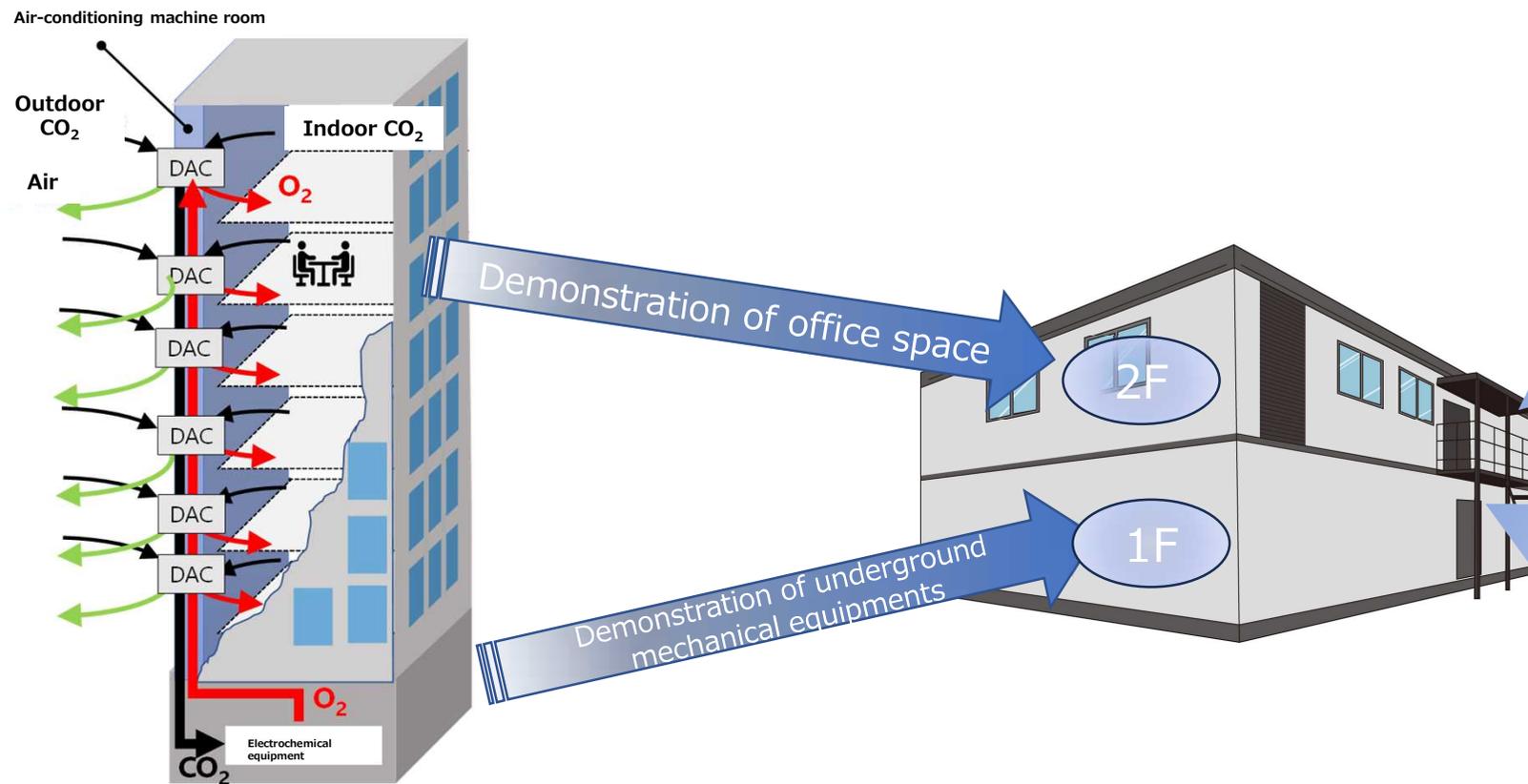
Development Schedule



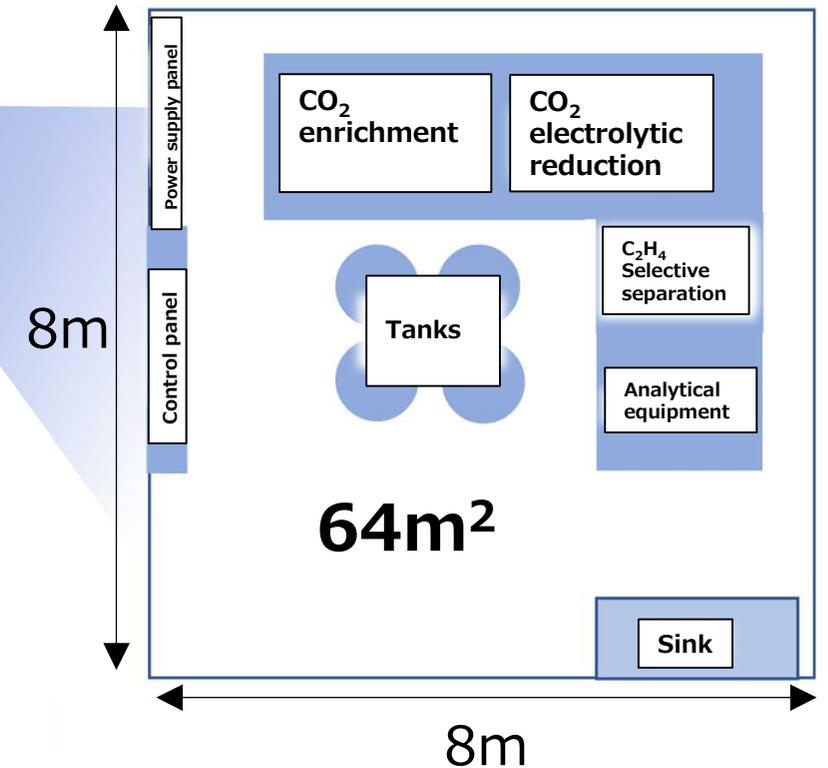
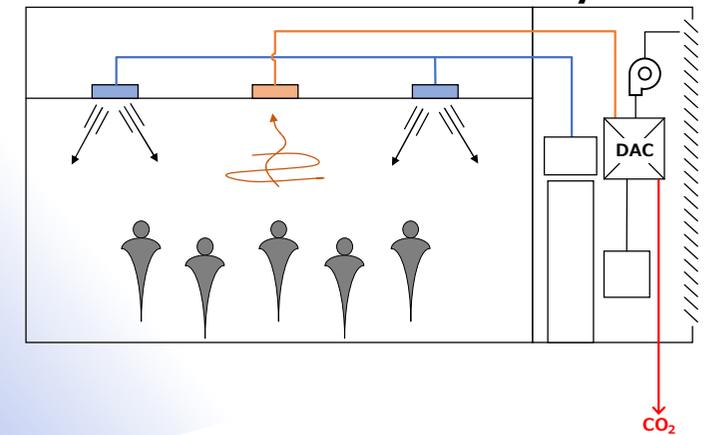
Pilot Implementation Image

- CO₂ throughput : ~10 kg/day
- Ethylene production : ~ 3 kg/day

Image of building implementation



DAC air conditioning and ventilation system



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CO₂ Capture and Enrichment : Basic Performance Achieved

□ CO₂ capture :

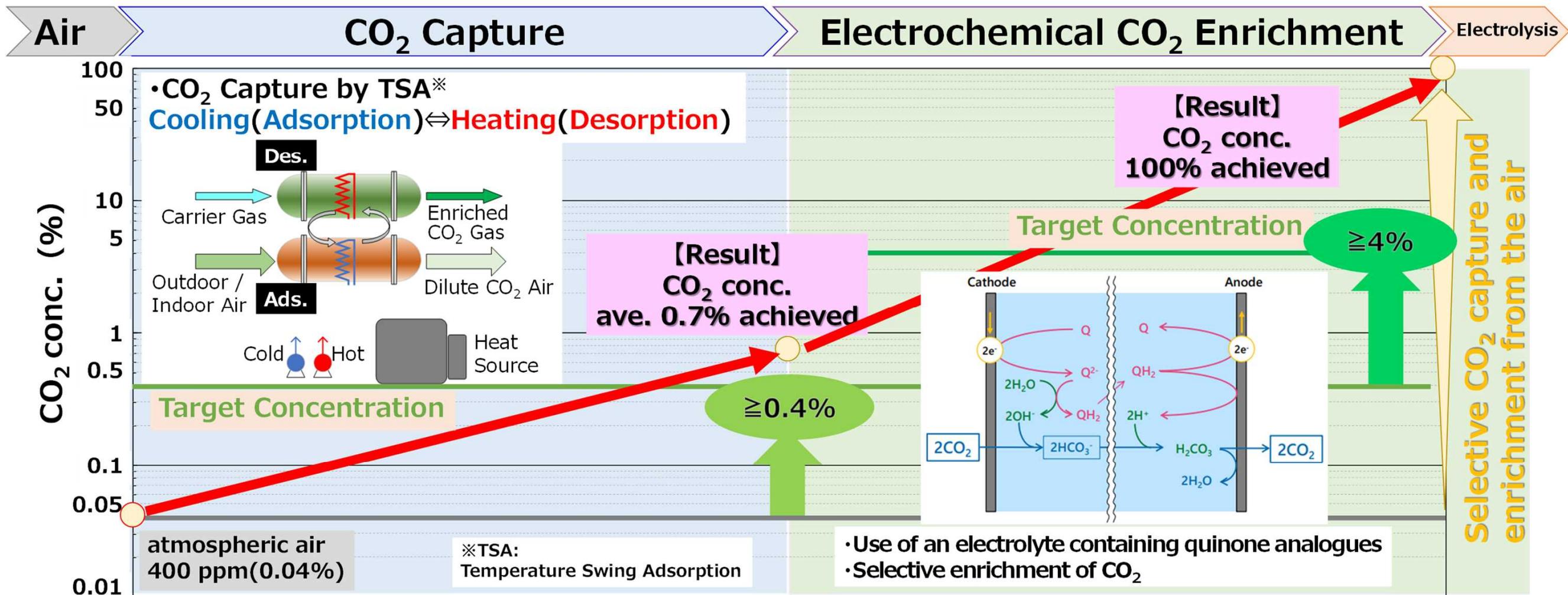
Concentration 0.04% → ave. 0.7%

Target for FY2022 (0.4%) achieved

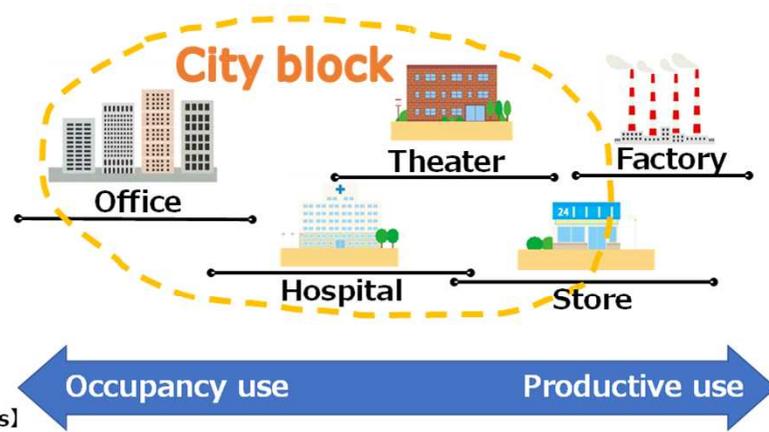
□ CO₂ enrichment

Concentration 0.7% → 100%

Target for FY2022 (10 times enrichment) achieved



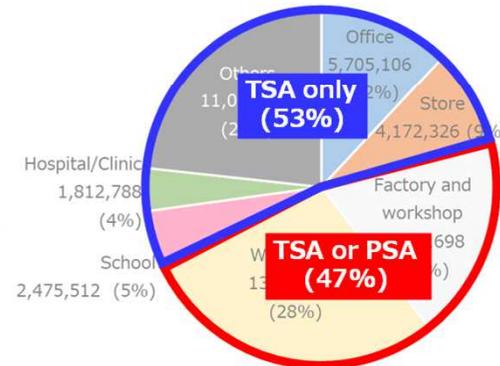
□ Concept of DAC requirement definition for buildings



※1) Japanese market for air conditioning systems and partners' positioning (AHU maker)



※2) Percentage of applied DAC mechanism in Japanese construction start floor area



2022 Annual construction start floor area (m²)
Ref.) MLIT: Statistical survey of construction starts 2022

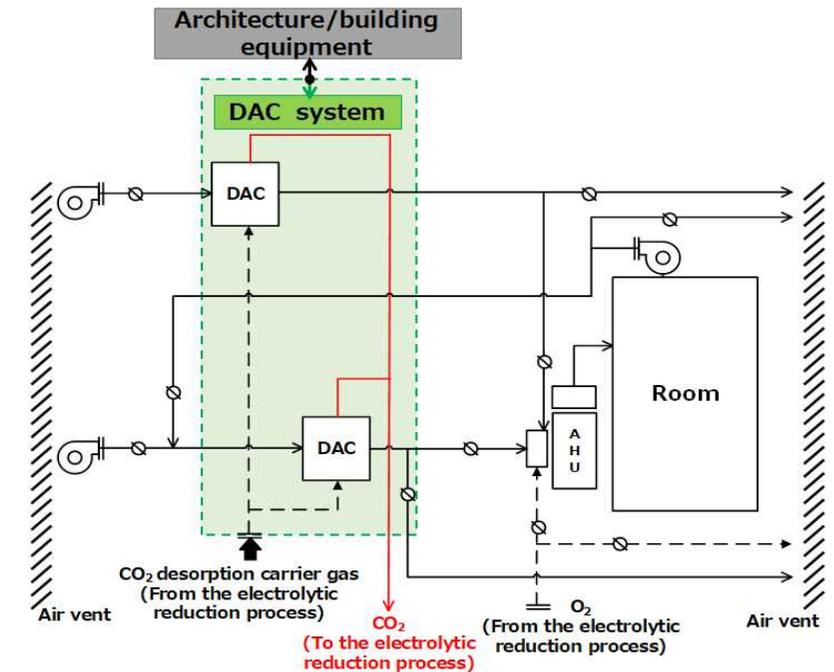


Diagram with the boundary

[Defining DAC requirements]

① Indoor Air Quality: Act on Maintenance of Sanitation in Buildings (Occupancy use) / Industrial Safety and Health Act (Productive use)

② Noise Level (dBA) - 40 — 45 — 50 — 55 — 60 — 65 — 70 — 75 -

[Corresponding specifications]

DAC solid adsorbent: Physics (Measures against moisture on the primary side need to be considered) / Chemical

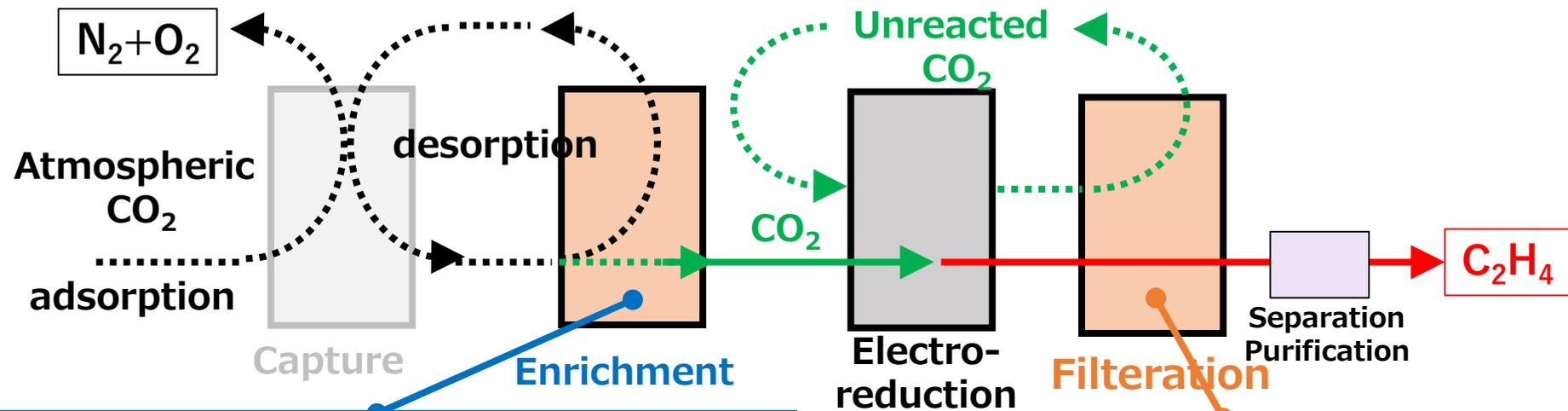
DAC mechanism: TSA (Packaged air conditioners (PAC) maker, Air handling unit (AHU) maker) / PSA (Engineering companies, PSA air dryer maker)

Prof. Kodama, Kanazawa University, Kodama PJ

	Office			Theater	Store	Hospital	Factory
	Complex	Large	Below medium				
CO ₂ conc.	≤ 1,000 ppm					≤ 1,000 ppm	≤ 5,000 ppm
Noise /Vibration	NC45/VAL≤			NC45/VAL ≤		NC45/VAL≤	NC45/VAL≤
Space	4~6% of floor area						
Temp. /Humi.	17~28°C/40~70%R.H.						
Air quality	HCHO:100 µg/m ³ , TVOC:400 µg/m ³ or less						
Odor	-						Offensive Odor Control Law
Energy type	Electricity					Electricity /heavy oil/gas	Electricity /heavy oil/gas
Special facilities	District heating and cooling					Private power generation	※ cogeneration
Adsorbent	Solid adsorbent					Solid adsorbent	Solid adsorbent
DAC mechanism	TSA			TSA/PSA	PSA /TSA	TSA/PSA	PSA/TSA
Partner candidate							

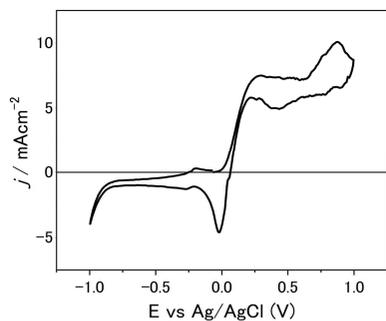
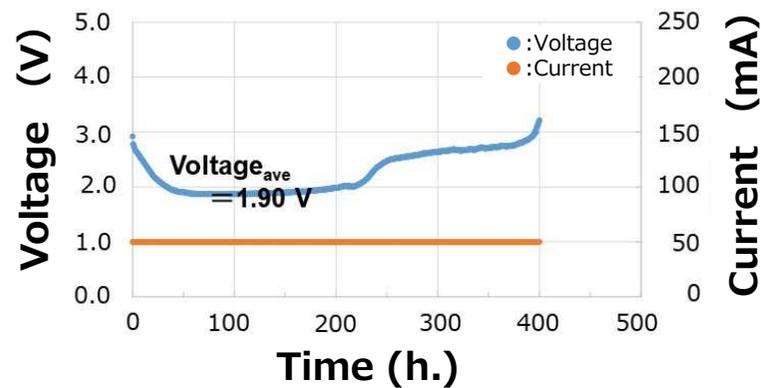
Requirements definition table

CO₂ Enrichment : Stability and Scale-up

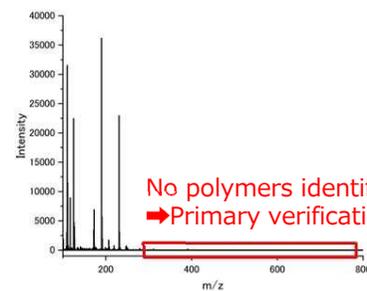


Extract Issues

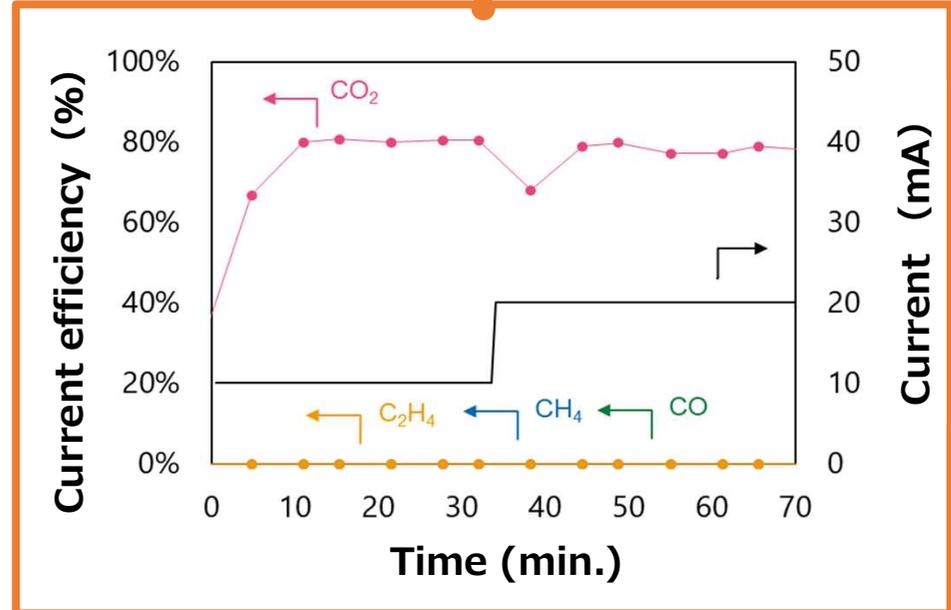
- Asymmetric Electrochemical Properties in the High pH Region (Near the Electrode)



Asymmetric electrochemical properties



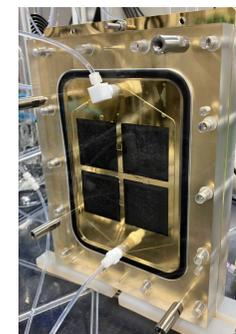
MS spectrum results of post-reaction solutions



- Selective separation of CO₂ from the mixture of CO₂/C₂H₄



2.5 cm Square



10 cm Square

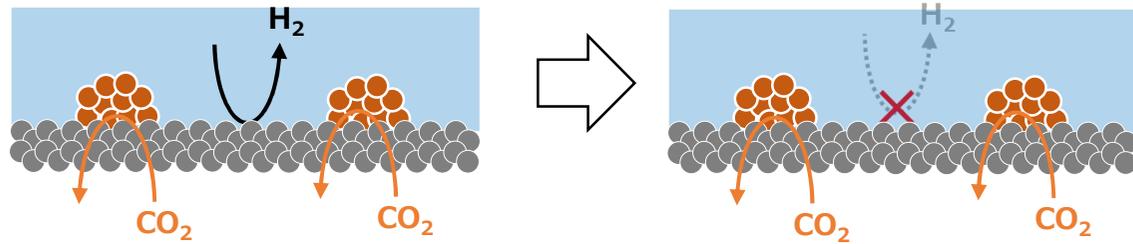
Improved reactor and larger area

CO₂ Electrolytic Reduction : Cathode Electrode Design Guidelines

Measures to improve current efficiency

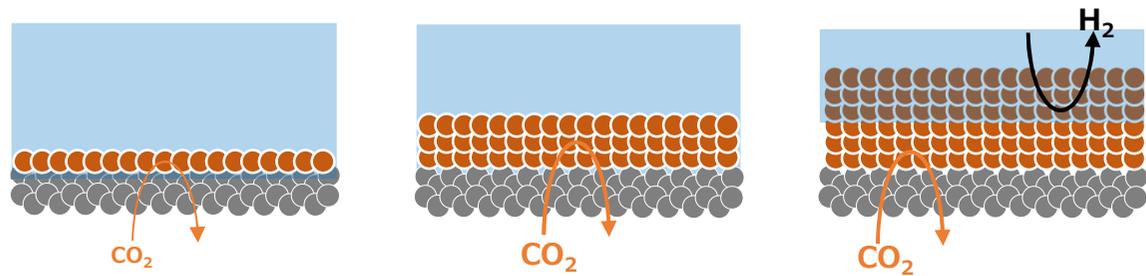
vs. Hydrogen

① Thorough elimination of macro hydrogen generation sites



② Appropriate catalyst structure control according to target current value

Example: Appropriate catalyst loading according to target current value

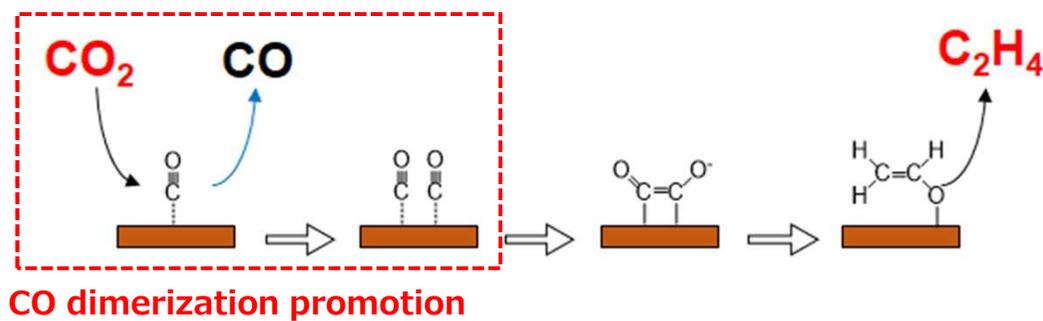


Current density : **small**
Current efficiency : **Large**

Current density : **Large**
Current efficiency : **Large**

Current density : **Large**
Current efficiency : **small**

vs. Other CO₂ reduction products



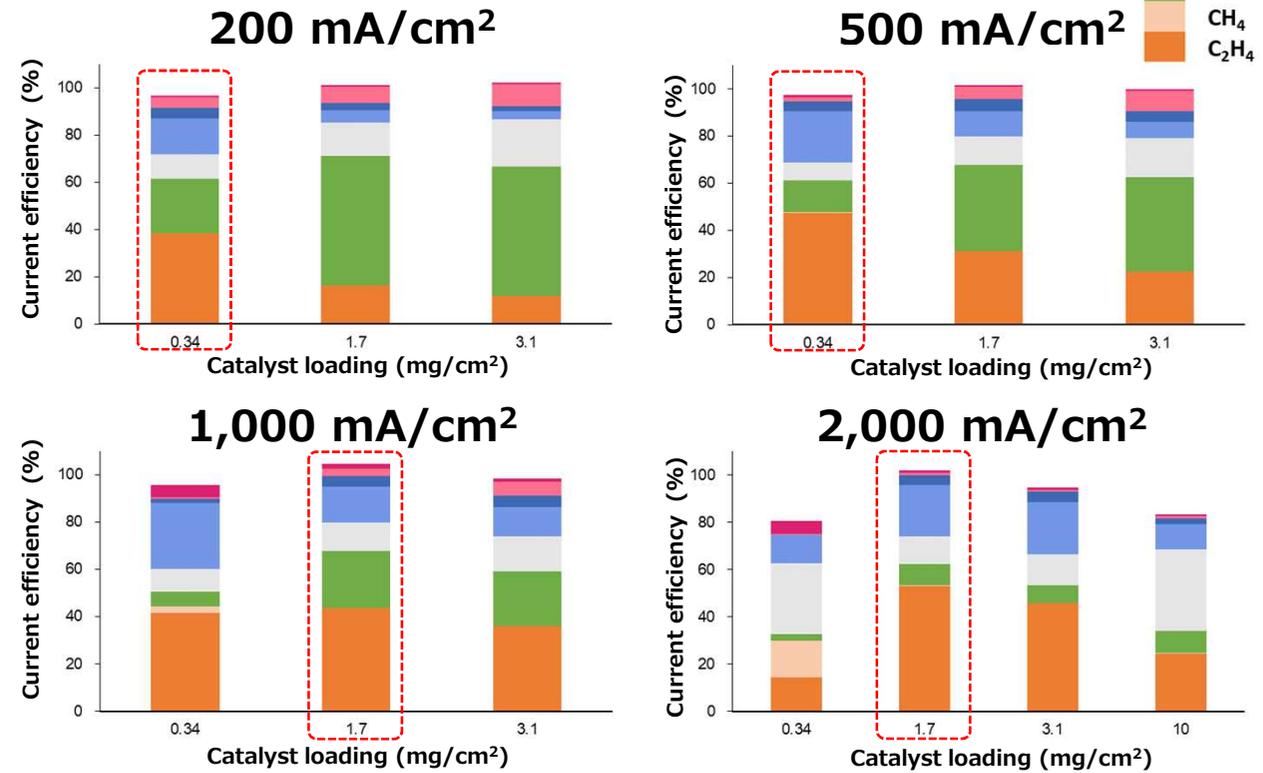
Counter measures

Improvement of local partial pressure of CO by increasing catalyst/electrode porosity
Improvement of CO local partial pressure by increasing current density

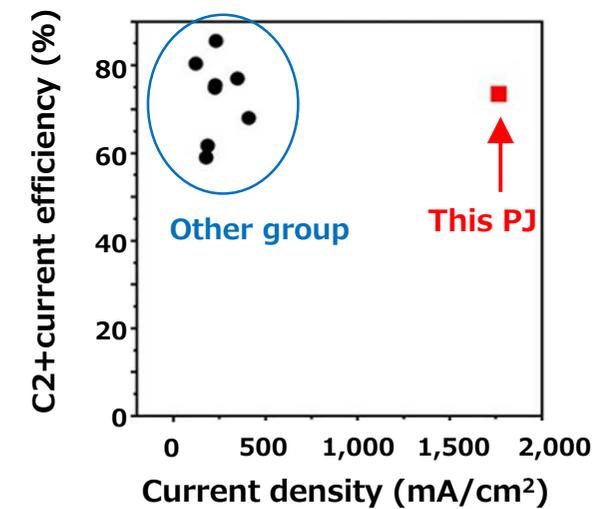
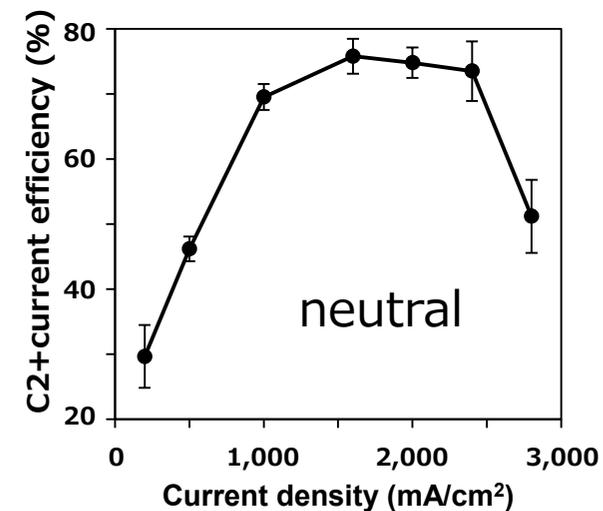
Establish optimal electrode design guidelines

Enables adjustment to target current density

CH₃COOH
HCOOH
C₃H₇OH
C₂H₅OH
H₂
CO
CH₄
C₂H₄

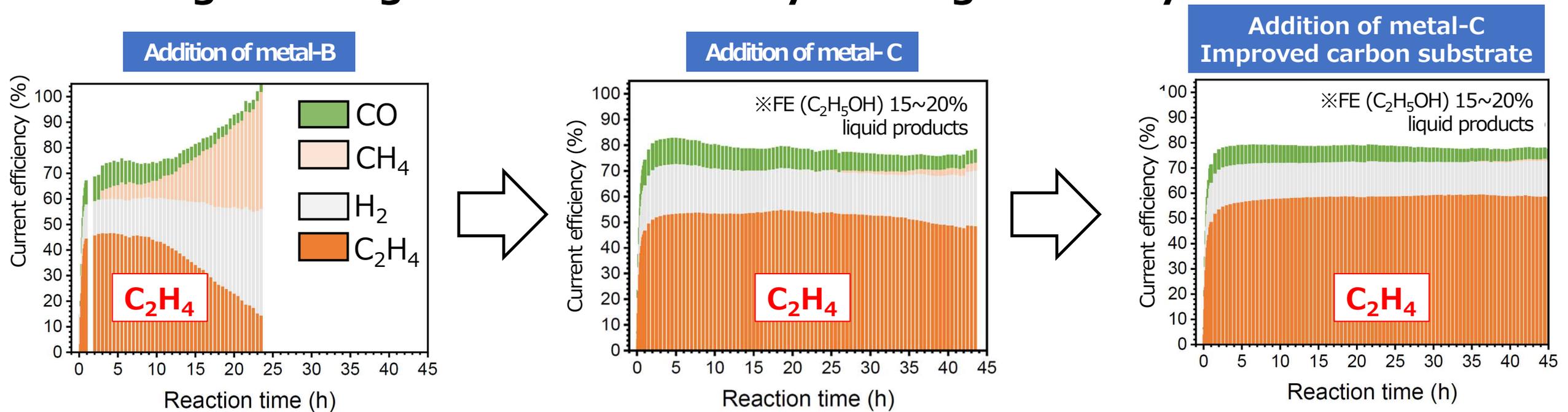


Compatibility of current efficiency with current density



CO₂ Electrolytic Reduction: Cathode Electrode Catalyst Development

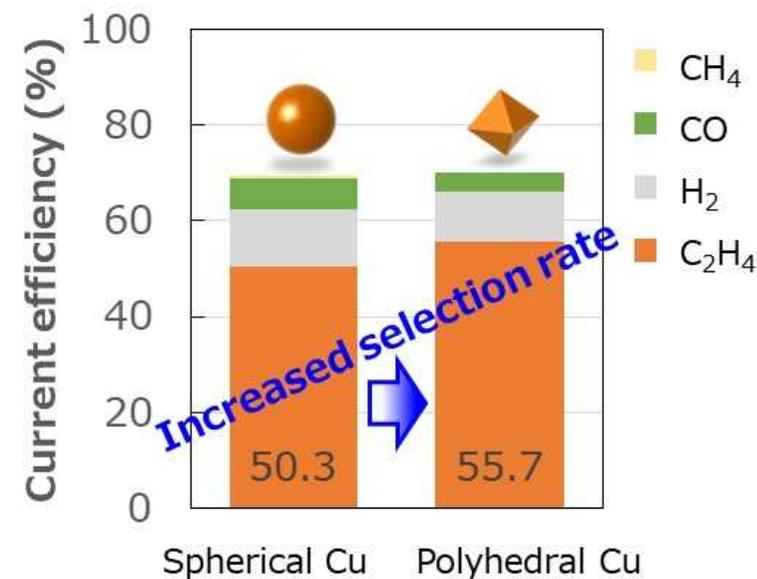
Attaining both high current efficiency and high stability



Efforts to achieve a high current efficiency

- Ethylene current efficiency of 56% with polyhedral Cu
- Factors are being identified and design guidelines are being developed

	Spherical Cu	Polyhedral Cu
Particle image		
Production method	Sintering	Wet reduction
State of purchase	Powder	Dispersion



CO₂ Electrolytic Reduction: Cathode Electrode Study for Industrialization

□ Start of electrode prototyping through institutional collaboration / Preliminary report of results (Prototype 1)

UBE
【 MPL/GDL production 】

Water repellent treatment, etc.

Substrate area 6cm square

Research Center for Advanced Science and Technology
The University of Tokyo

FURUKAWA ELECTRIC
【 Catalyst Development & Preparation 】

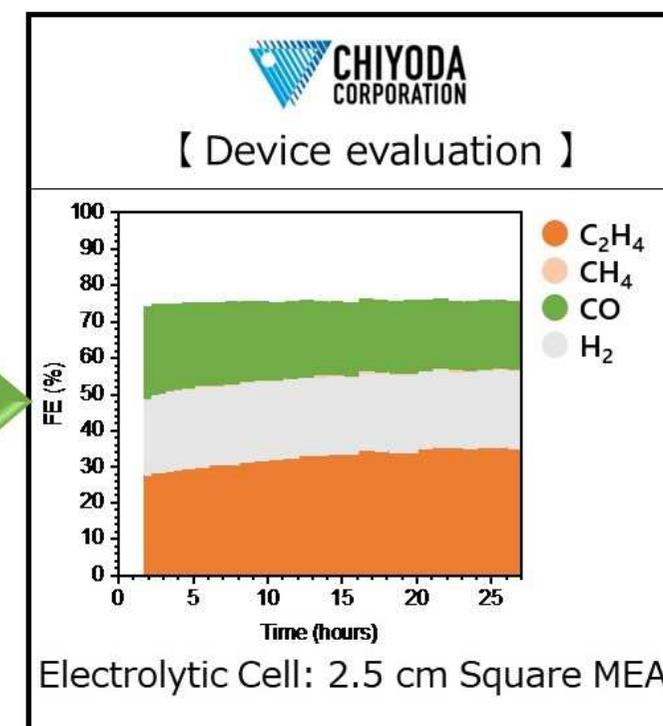
Cu-Al alloy catalyst preparation

Image	SEM
 ●: Cu ●: Al solid solution alloy	
Preparation Method	Thermal plasma method

maxell
Within, the Future
【 Catalyst coating /GDE fabrication 】

Coating treatment, etc.

Image of catalyst coating treatment on substrate



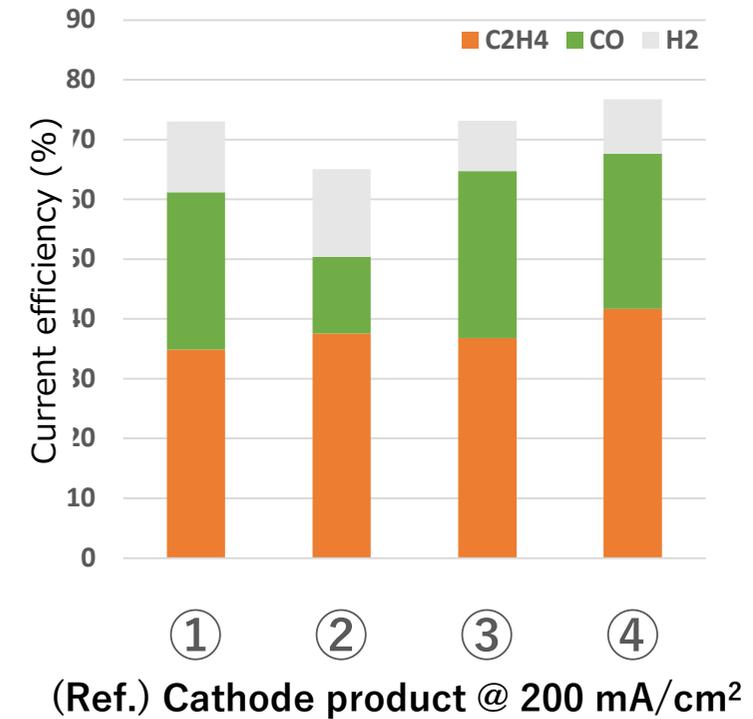
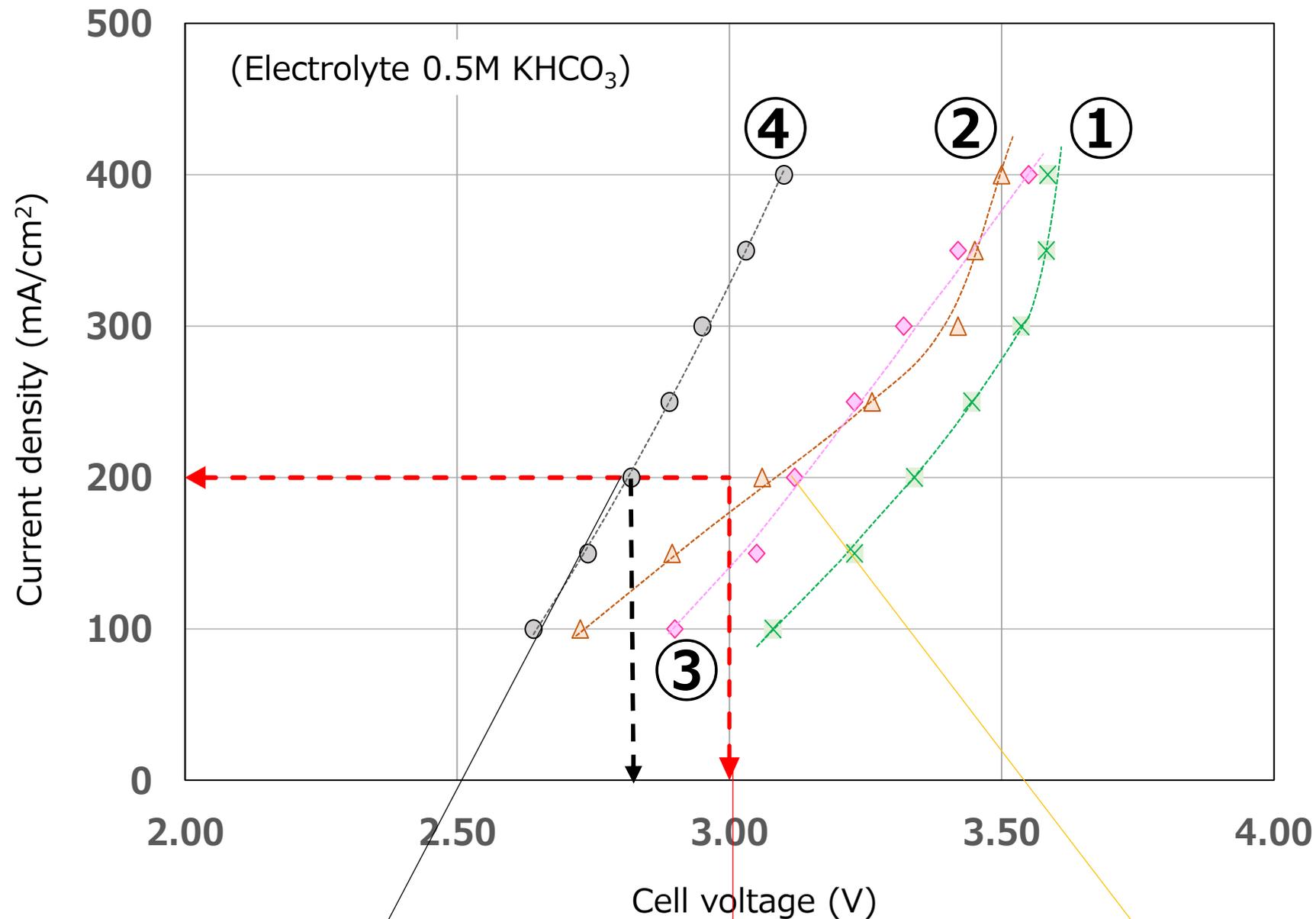
Evaluation results of the first prototype of the catalyst addition electrode



【Future】
Aiming to implement roll-to-roll coating prototypes

CO₂ Electrolytic Reduction: Anode Electrode Development

□ Cell voltage of 2.82V @ 200mA/cm² was confirmed
by optimizing the anode electrode structures



	structure
①	Commercial
②	Catalyst α
③	Catalyst β
④	Catalyst γ

④ Cell voltage:
2.82 V
@200 mA/cm²

Target :
3.0V
@200mA/cm²

② Cell voltage
3.06 V
@200 mA/cm²

CO₂ Electrolytic reduction : Reactor Performance Improved

□ World's highest level of high current density

- C₂+ FE 80% @ 2,000 mA/cm²

□ Higher current efficiency / lower operating voltage

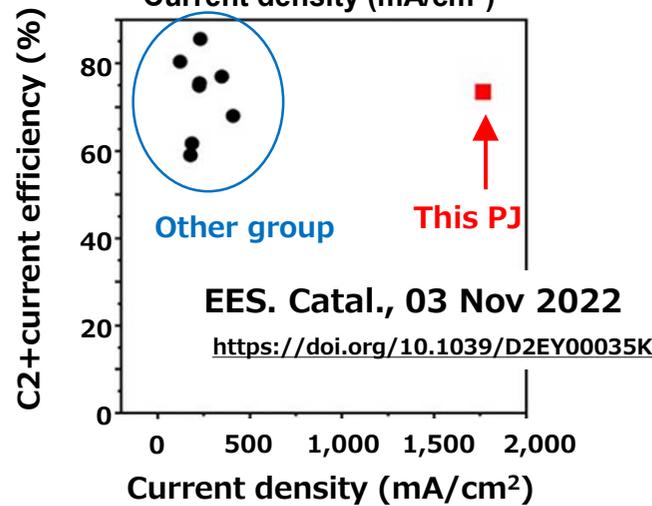
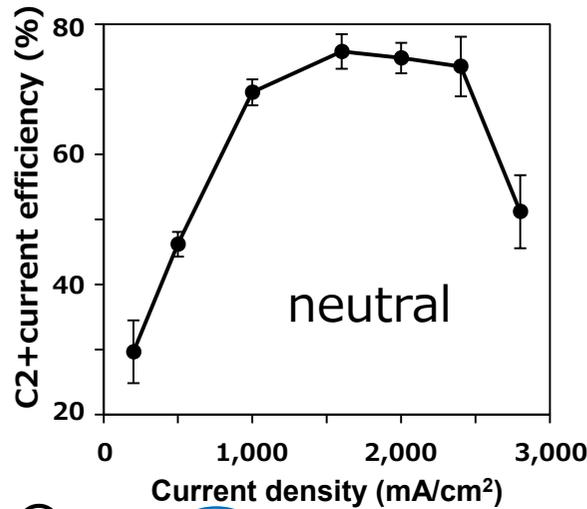
- C₂H₄ FE 60% @ cell voltage of 2.8 V

World's top

FY2022 Target : 100 mA/cm²

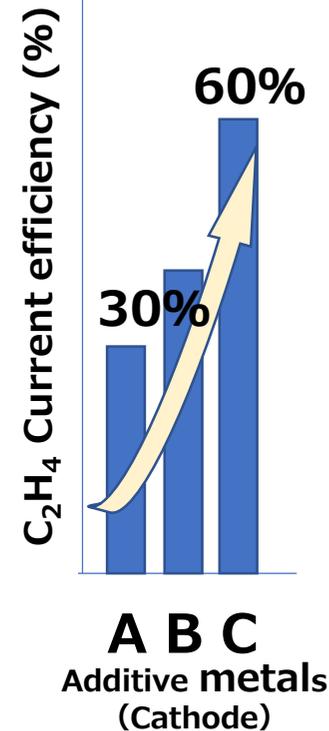
FY2024 KPI
CO₂ emissions (t-CO₂/t-C₂H₄)
+0.5~+1.0

- Current density: 2,000 mA/cm²
- C₂+ Current efficiency 80%

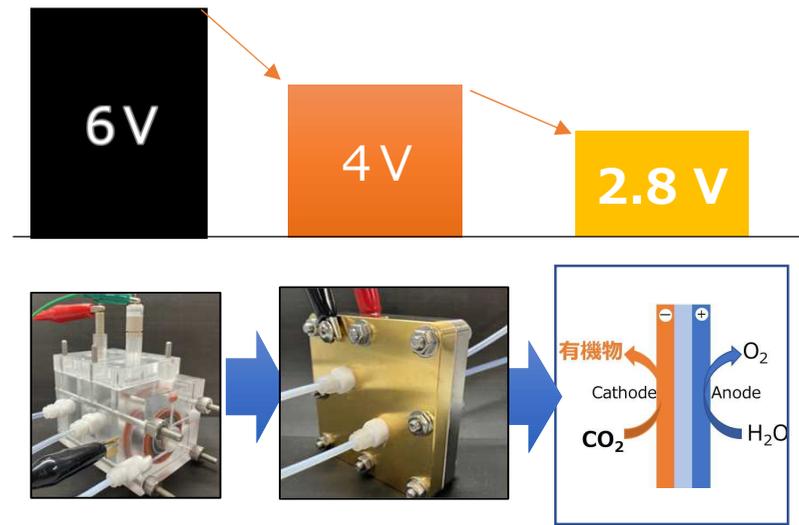


High current efficiency at high current density

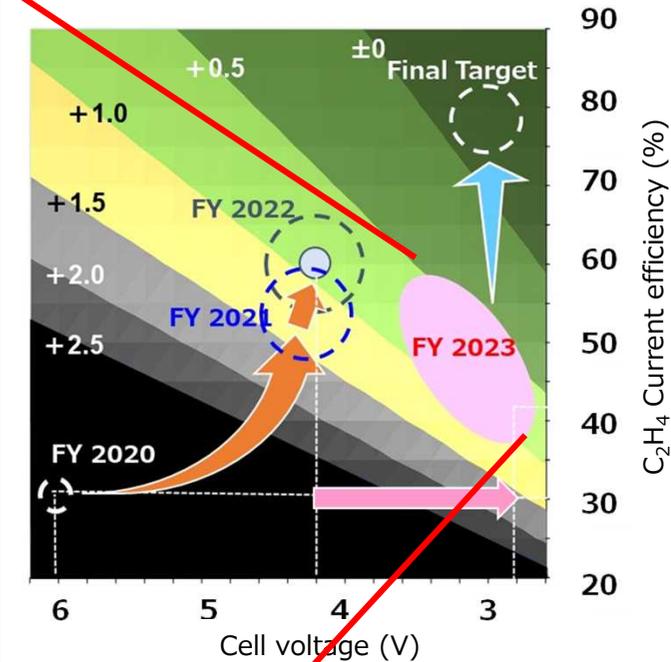
- Current efficiency 60%
- Cell voltage of 2.8 V



Selectivity improved



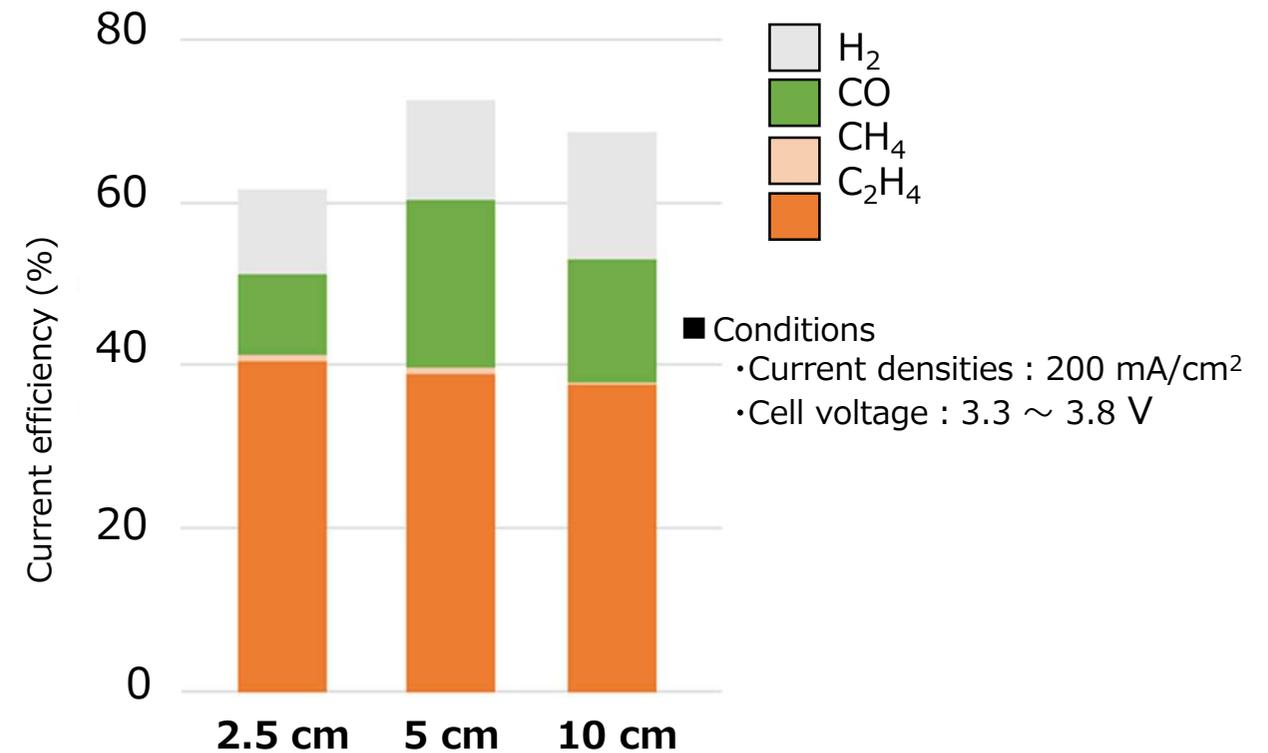
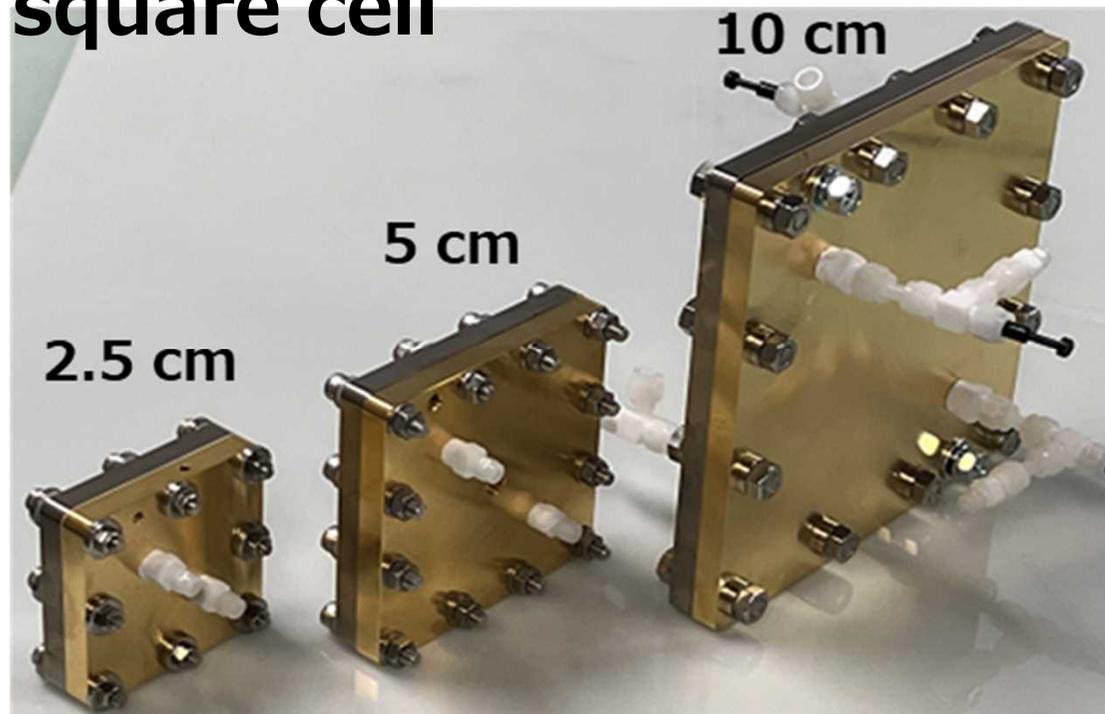
Cell voltages reduced



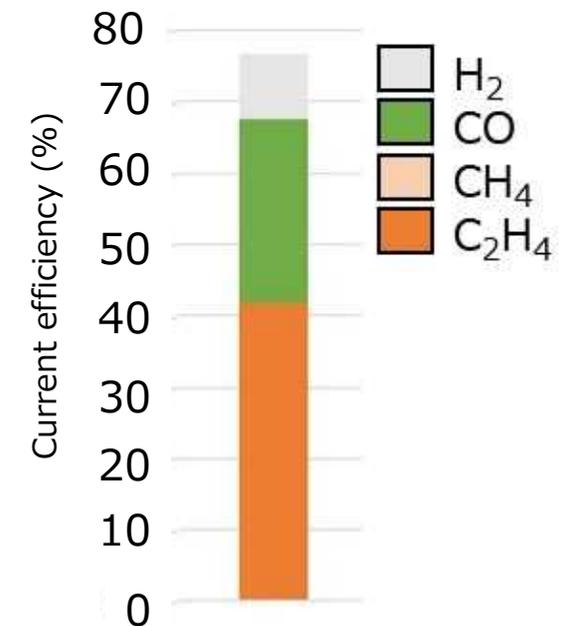
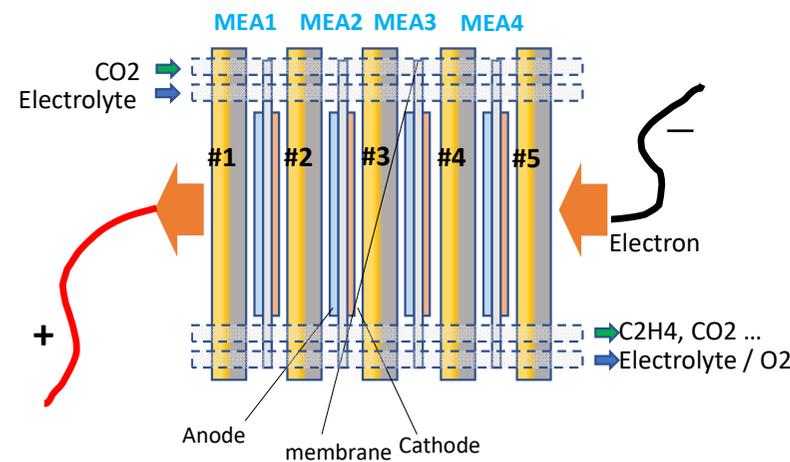
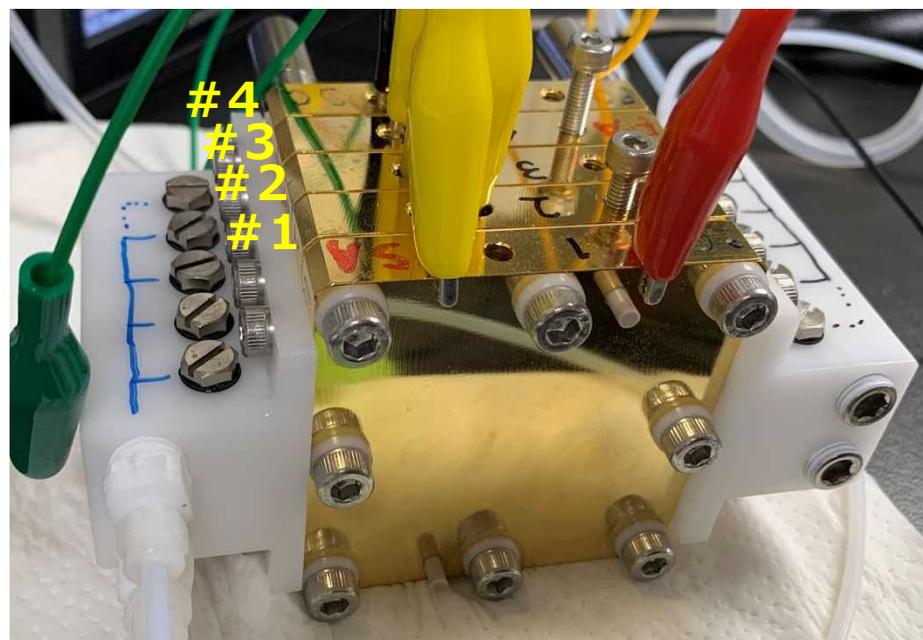
Overall system CO₂ emissions
(t-CO₂/t-C₂H₄)

CO₂ Electrolytic Reduction: Larger Area and Stacked Reactors

□ 10 cm square cell achieves performance equivalent to 2.5 cm square cell



□ Ethylene formation was confirmed in a 2.5 cm square x 4 stack



□ Development of a simulation model was initiated

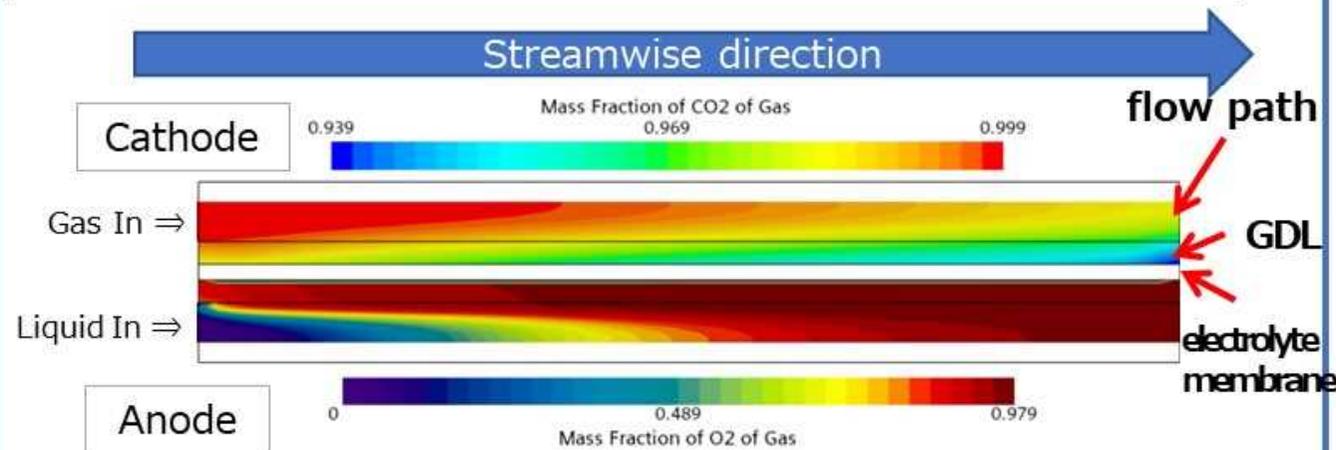
Electrolytic reaction model

[Review]

- Development of reaction process model between cathode and anode
- Consideration and reflection of CO₂ crossover

○ Example of response analysis in the basic study model

Distribution of CO₂ and O₂ concentration between Anode and Cathode

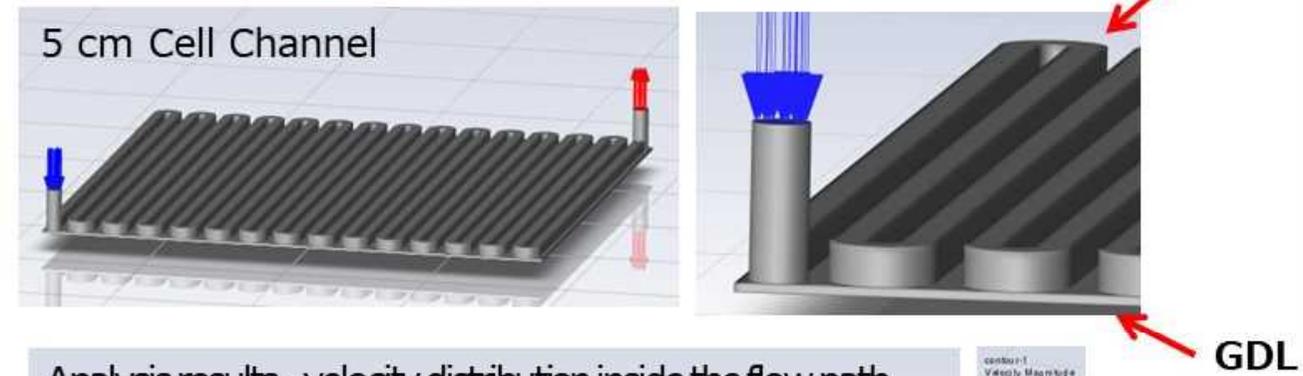


Reproduce the reduction of CO₂ concentration (reduction reaction) as you go downstream in Cathode.
 → Analysis of the effect of flow on reaction efficiency.

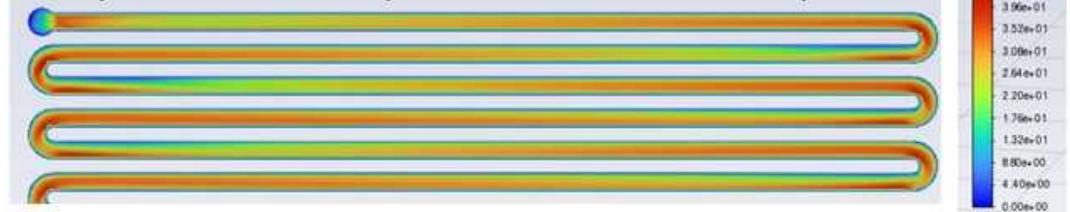
Fluid model

[Review]

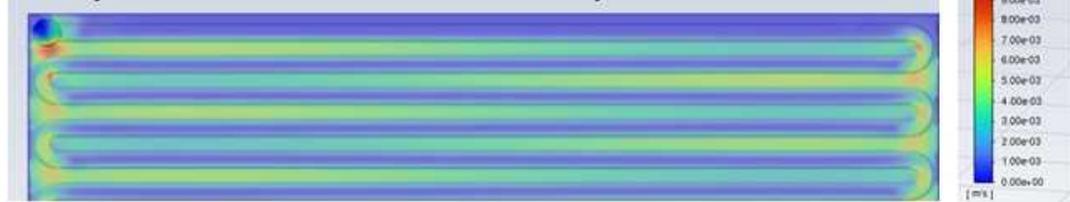
- Flow analysis in cell channel and diffusion layer film (GDL)
- Design of channel for large area



Analysis results - velocity distribution inside the flow path



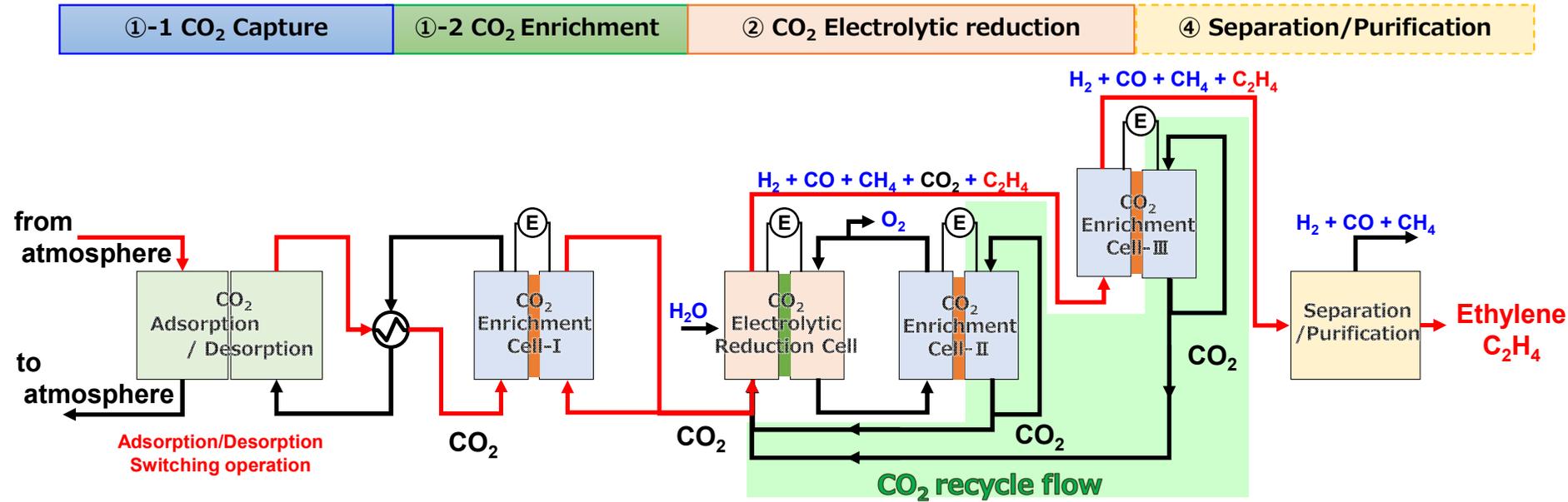
Analysis Results - GDL Internal Velocity Distribution



Optimal flow path design for large-area cells based on flow paths and flow conditions inside the GDL.

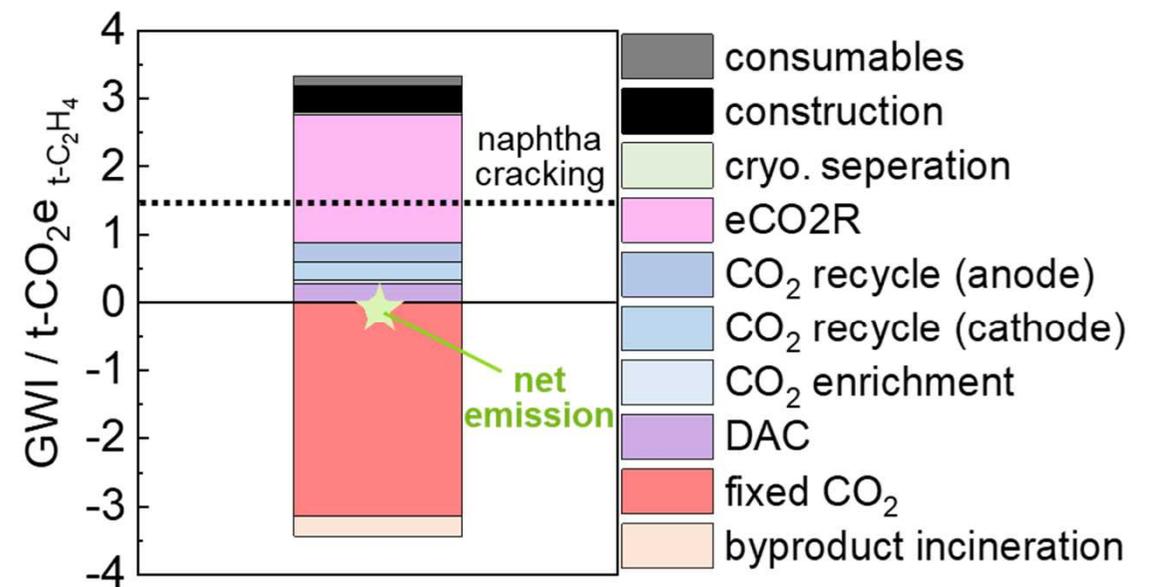
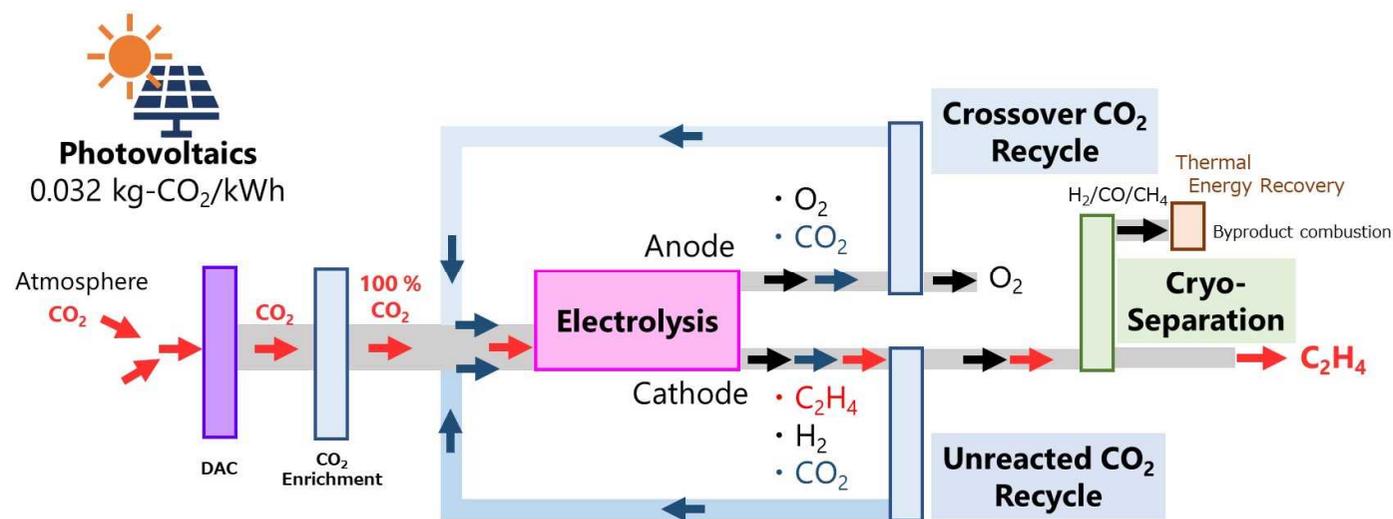
Integrated Systems: Conceptual Design and LCA

Integrated system design: introduction of CO₂ recycling flow using enrichment cells

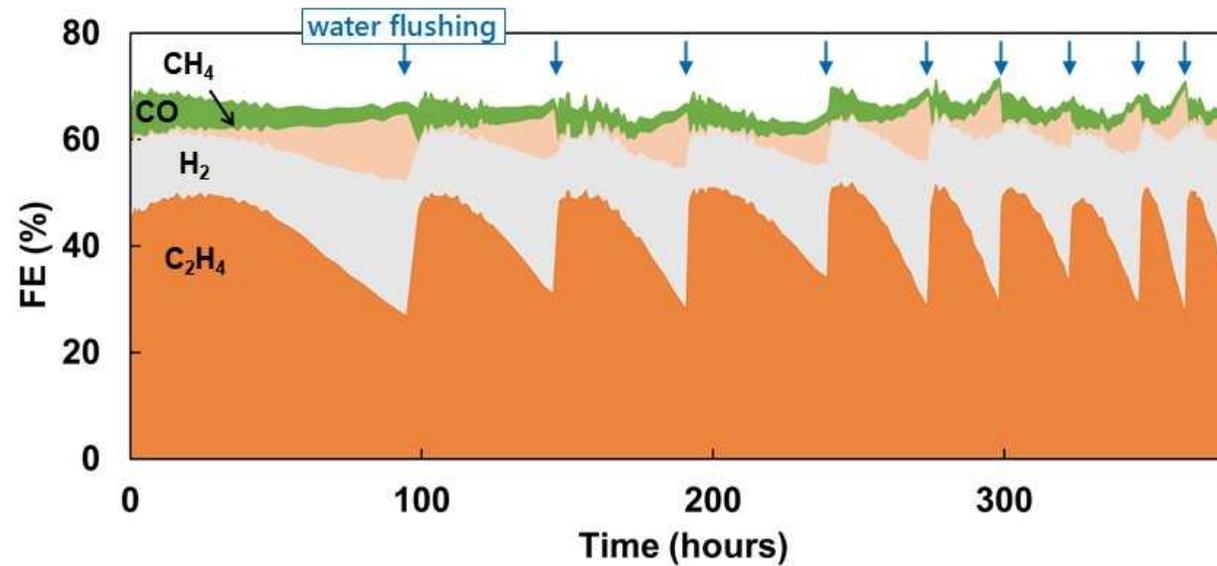


LCA evaluation of CO₂ recycling

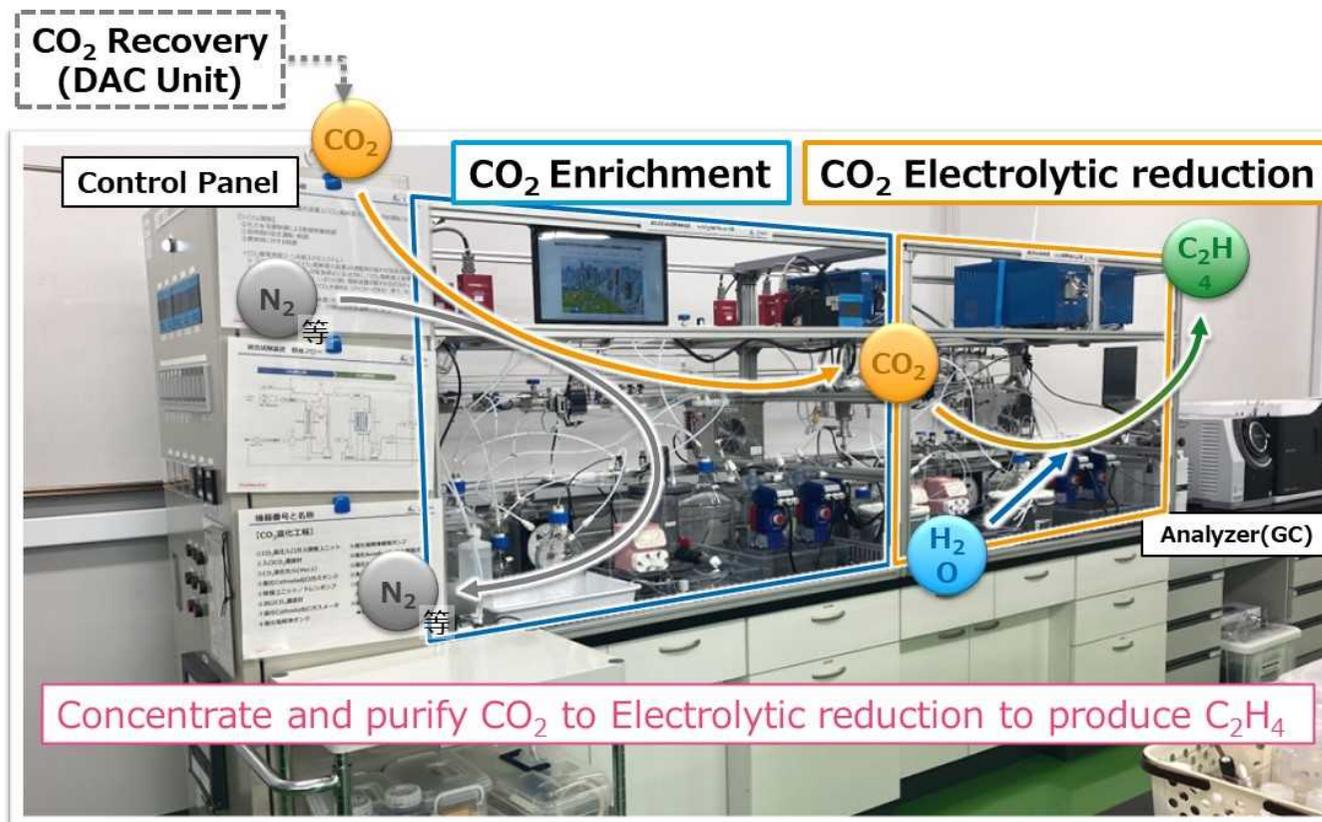
Study LCA for a system that reuses heat from combustion of by-products as heat for CO₂ desorption in DAC



- Study on system control for continuous operation
 - ➔ Confirmed the impact of operation control



- Started evaluation of "CO₂ enrichment + CO₂ electrolysis" coupled operation at lab. scale



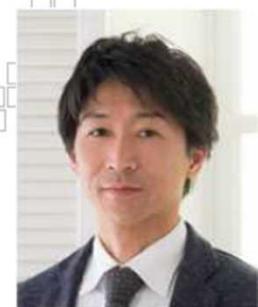
International Collaborations



The Director and Prof. Sugiyama



Prof. Joel Ager



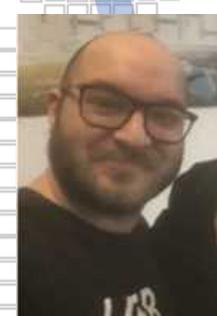
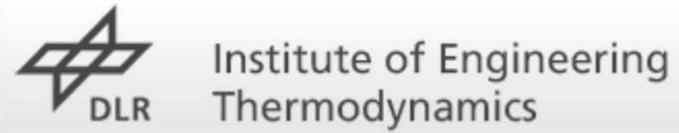
Prof. Nakanishi



Dr. Fujii



Information Exchange Meetings with European Embassies and Others
 /Embassy of Norway
 /Trade and Investment Promotion Agency of the German State of NRW, etc.
 /Consulate of Switzerland in Osaka



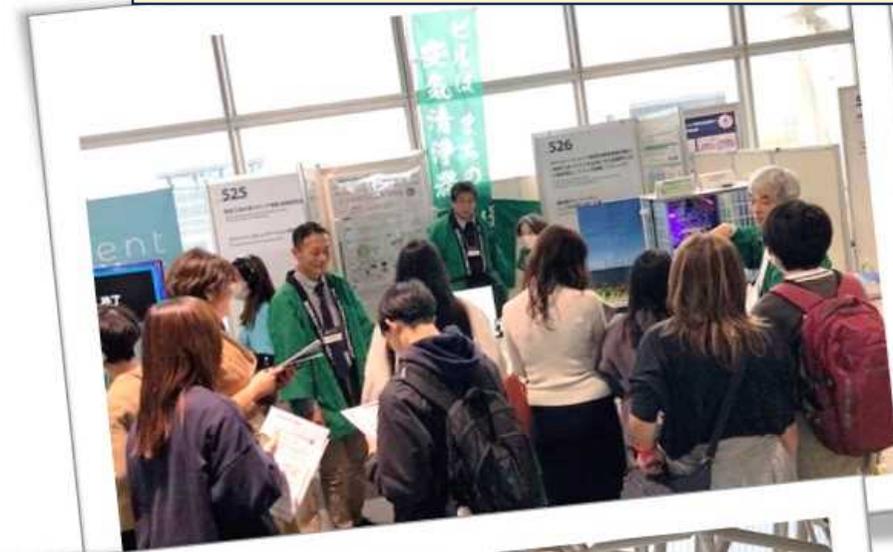
Dr. Seyed Schwan Hosseiny

DLR Institute of Engineering Thermodynamics: Dr. Seyed Schwan Hosseiny
 Study abroad [Osaka Univ. : Prof. Nakanishi]

Lawrence Berkeley National Laboratory, Materials Science & Engineering : Prof. Joel Ager
 Collaborative research [RIKEN : Dr. Fujii]
 Study abroad [Osaka Univ. : Prof. Nakanishi]

Exhibited at SCIENCE AGORA 2023 organized by JST

Nov. 18 - 19, 2023, at Telecom Center
Number of visitors to the project booth: over 400
Exchange of business cards: about 15 companies



1. Project Overview

2. Progress and Results

3. Summary and Future Challenges

Summary and Future Challenges

□ Summary

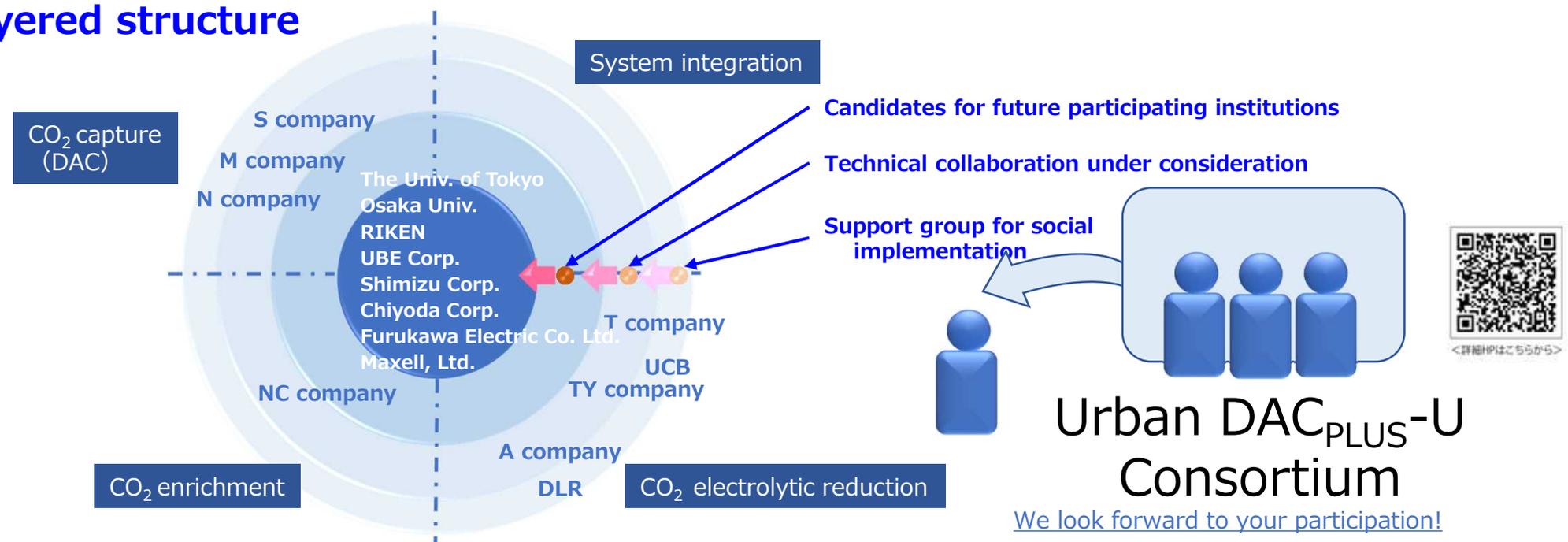
Theme	Major Results	Future Works
CO ₂ Capture and Enrichment	<ul style="list-style-type: none"> Clarified the concept (requirements) of the implementation model for buildings Successful enrichment of atmospheric CO₂ from 400 ppm to 100% (pure CO₂) 	<ul style="list-style-type: none"> Design and manufacturing of prototypes Low drive voltage and long-term stable operation
CO ₂ Electrolysis	<ul style="list-style-type: none"> FE to ethylene 60%, 2.8 V operating potential between 2 poles achieved Efforts to achieve large area / 10cm square cell evaluation and institutional collaboration 	<ul style="list-style-type: none"> Development of electrodes that simultaneously satisfy current efficiency, current density, and stability
System Integration LCA	<ul style="list-style-type: none"> Conceptual system design from atmospheric CO₂ capture to ethylene production and LCA for CO₂ emission 	<ul style="list-style-type: none"> Continuous process benchmark of "CO₂ Enrichment + Electrolysis." Improvement of LCA accuracy

□ Targets

Fiscal Year	Scale	CO ₂ throughput (kg/y)	CO ₂ emission reduction target (per ton of ethylene produced)
2024	Laboratory	25	+0.5 ~ +1.0 ton
2027	Bench	250	± 0.0 ~ +0.5 ton
2029	Pilot	3,300	< -0.5 ton (Carbon negative and 5,000 hours of continuous operation achieved)

□ Efforts toward social implementation

Establish a four-layered structure



END