

PJ: Development of Multi-lock Biopolymers Degradable in Ocean from Non-food Biomasses

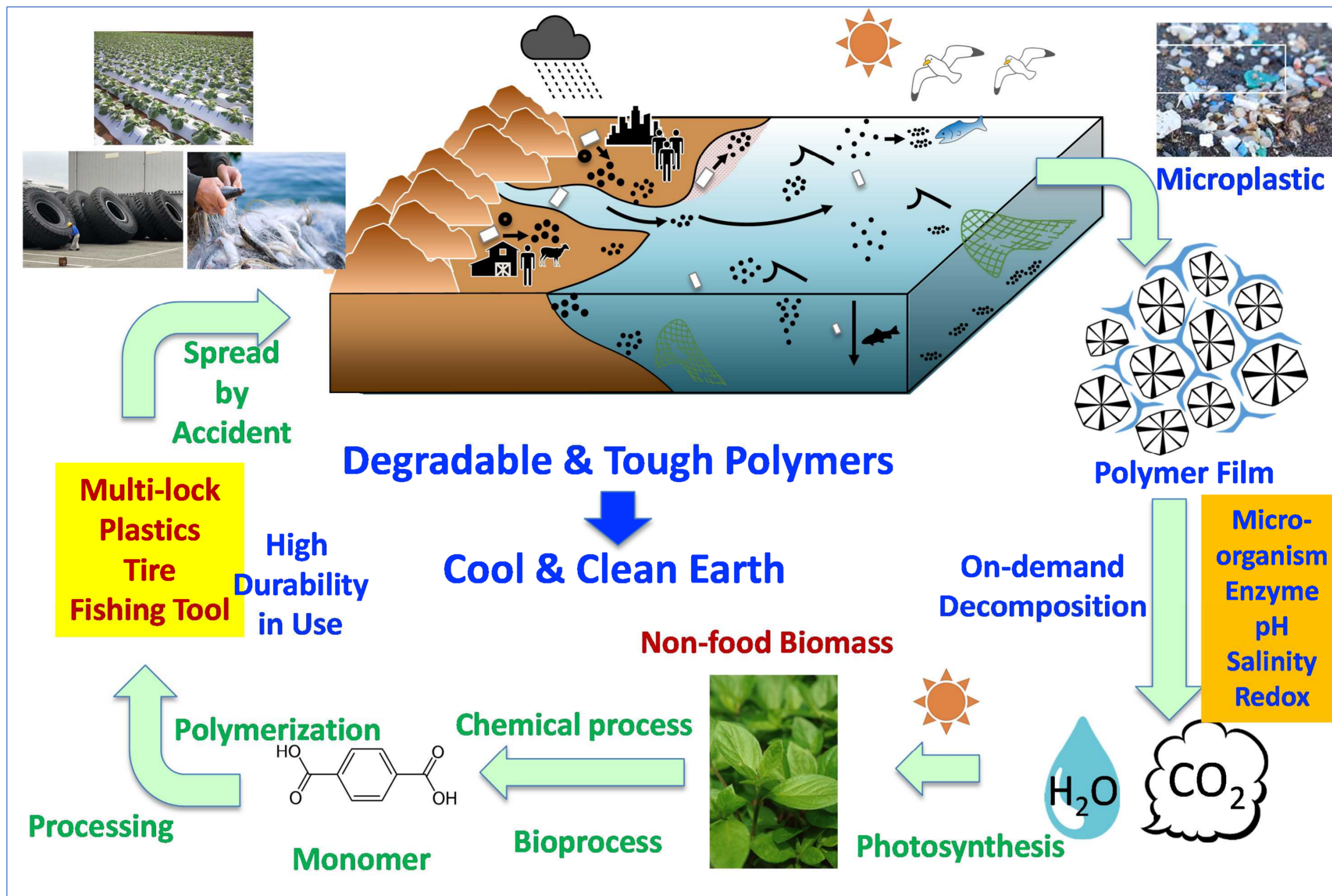
Organization: University of Tokyo

Contact: Kohzo Ito (kohzo@edu.k.u-tokyo.ac), Naoki Kato (naokikato@g.ecc.u-tokyo.ac.jp)



For seawater decomposition of **difficult-to-collect plastics, tire wear debris, and fishing gear**, we have introduced a multi-lock mechanism that achieves both degradability and durability, and realizes on-demand disassembly. In addition, by using non-edible biomass as a raw material, CO2 reduction can be achieved at the same time.

OVERVIEW



Organization

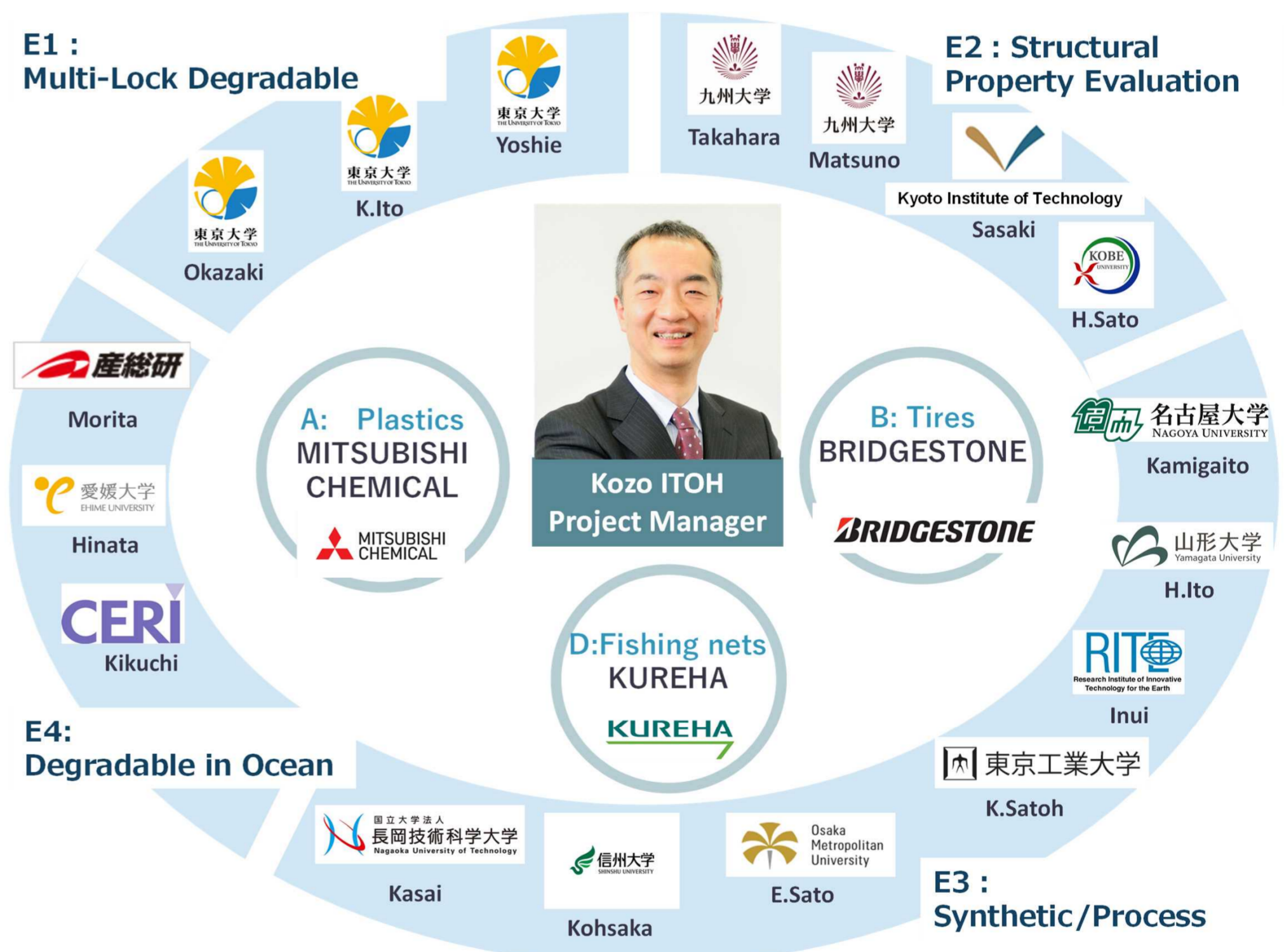
NEDO
Materials Technology and Nanotechnology Department
Bioeconomy Promotion Division

ITO PJ
Project Manager
Kohzo Ito
Project Office

Entrusted organization : 1
Reentrusted organization : 6

Company : 3
Academia : 14
(17Gr)

Entrusted organization	Reentrusted organization	Leader					
2 Mitsubishi Chemical	3 Bridgestone	5 Kureha	Asushi Kusuno Satoshi Hamatani Takashi Masaki				
1 The University of Tokyo	Kyoto Institute of Technology	Kobe University	Osaka City University	Shinshu University	Nagaoka University of Technology	CERI	Kohzo Ito Susumu Okazaki Naoko Yoshie Sono Sasaki Harumi Sato Eriko Sato Yshiro Kohsaka Disuke Kasai Tkako Kikuchi Jun Takahara Masami Kamigaito Hiroshi Ito Hisao Matsuno Masayuki Inui Hiroshi Morita Hirofumi Hinata Kotaro Sato
6 Kyushu University	7 Nagoya University	8 Yamagata University	9 RITE	10 AIST	11 Ehime University	12 Tokyo Institute of Technology	(敬称略)

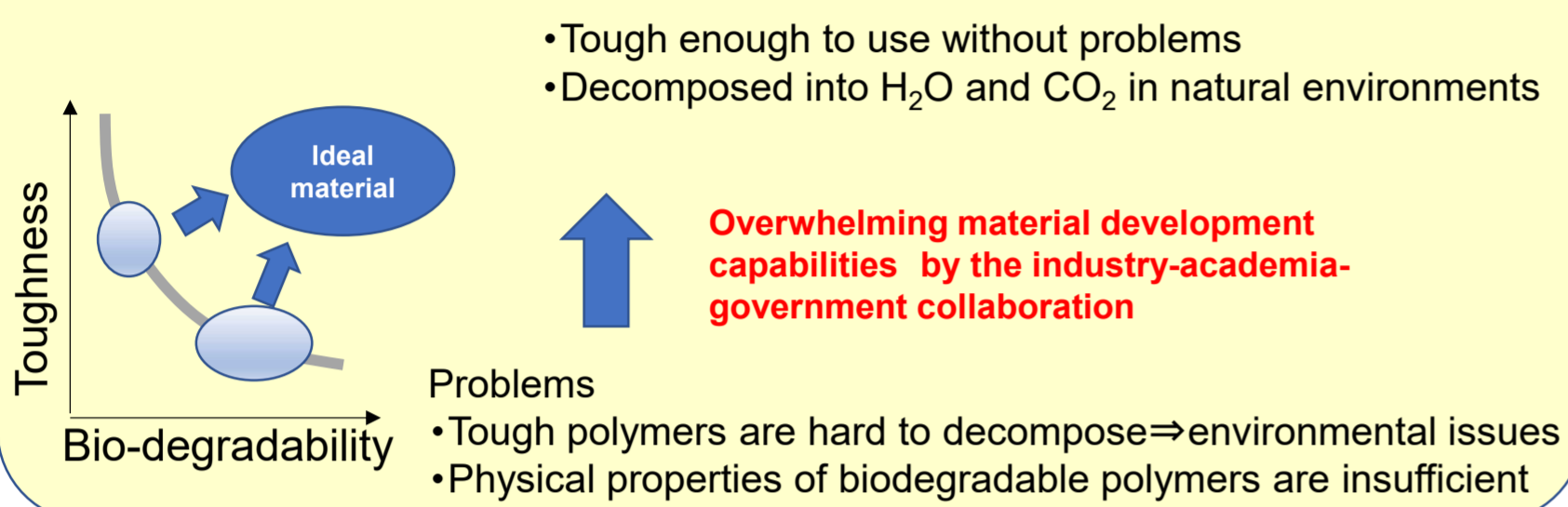


Common Issues		Goals	Members
E1+ E3	Multi-lock degradation mechanism (switch function)	Develop a multi-locked degradation mechanism for model resins and elastomers, utilizing copolymers, dynamic cross-linking, catalysts, and enzymes, which can be degraded on demand by multiple stimuli expected in the marine environment.	Univ Tokyo, Nagoya Univ, RITE, Tokyo Tech, AIST, Osaka City Univ, Shinshu Univ, Nagaoka Univ Tech
E2	Elucidation of environmental degradation mechanisms, including marine	Elucidate the degradation mechanisms of model resins and elastomers in natural environments, including the ocean.	Kyushu Univ, Kyoto Inst Tech, Kobe Univ, AIST, CERI
E3-1	Development of polymers from inedible biomass	Monomers from inedible biomass will be synthesized using enzymes and organic synthesis, as well as polymerization methods.	Nagoya Univ, RITE, Tokyo Tech, Shinshu Univ
E3-2	Improved durability and toughness of environmentally degradable polymers	The use of molding and processing techniques, dynamic cross-linking, copolymers, and supramolecules will be investigated to improve the durability and toughness of environmentally degradable polymers, including marine, as well as to study self-healing properties.	Yamagata Univ, Kyushu Univ, Univ Tokyo, Nagoya Univ, AIST
E4	Assessment of environmental degradability, including marine	Analyze the dynamics of plastic trash, fiber waste, fishing nets, and tire wear powder in the ocean, evaluate their degradation in the ocean, and study the development of a fast degradation evaluation method.	Ehime Univ, CERI
E5	Marine biodegradability and safety of oligomers	Synthesize oligomers equivalent to polymers developed by each company and evaluate marine degradability and safety	Kyushu Univ, Nagoya Univ, Tokyo Tech, Shinshu Univ, CERI
E6	Development of polymers made from seaweed for CO2 fixation	Synthesize marine biodegradable plastic using seaweed (Macroalgae) with excellent CO2 fixation performance provided by ARPA-E as a raw material, and evaluate its marine biodegradability and mechanical properties (joint research with ARPA-E)	Univ Tokyo, Nagoya Univ, RITE, Tokyo Tech, Osaka City Univ, Shinshu Univ, Ehime Univ, CERI, Yamagata Univ

1. Objectives and concept

【Objectives】 The purpose is to develop a bioplastic that **incorporates a multi-locking mechanism** in aliphatic polyesters produced from inedible resources and that **quickly biodegrades in seawater** after being unlocked by multiple external stimuli. We also aim to toughen biodegradable plastics while maintaining good biodegradability by introducing dynamic cross-linking or supramolecules and optimizing of higher-order structures. In this work, we will investigate the introduction of multi-locking mechanism and toughening of polybutylene succinate(PBS).

【Concept】 Moonshot program led by the Cabinet Office
Achieve both high toughness and high biodegradability



3. Academia/Mitsubishi Chemical research content

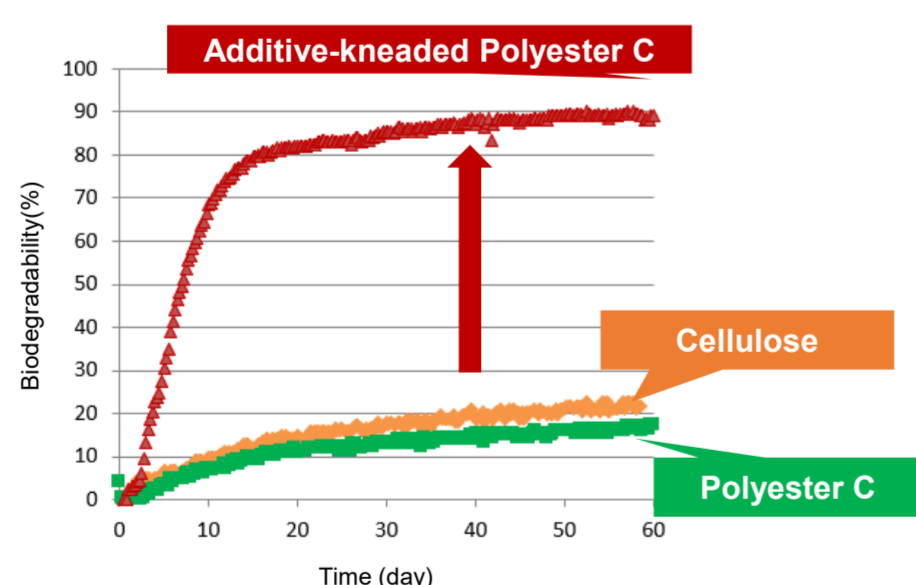
• Schedule for 2023-2024

- ① Biodegradability control (clarification of biodegradation mechanism, introduction of trigger mechanism)
- ② Improvement of tear strength (clarification and improvement of tear)

PIC	Study Items
Tokyo Univ. Prof. Ito	Improving toughness and promoting biodegradation (application using PR)
Yamagata Univ. Prof. Ito	Improvement of toughness (clarification of tearing mechanism and kneading technology)
Yamagata Univ. Prof. Matsuno	Analysis of enzymatic degradation behavior in water (AFM)
Kyoto institute of technology Prof. Sasaki	Time-resolved X-ray scattering measurement during tearing process (SPring-8)
Kyushu Univ. Prof. Takahara	Structural changes and degradability due to photooxidative degradation (natural environment model)
Kobe Univ. Prof. Sato	Effect of degradability on higher-order structure and intermolecular interactions (terahertz, low frequency Raman spectroscopy)
Mitsubishi Chemical Corporation	Introduction of multi-lock mechanism (expression of switch function by introducing copolymer monomer, biodegradation accelerator, and degrading enzyme)

5. Biodegradation accelerator kneaded polyester

• Various additives were kneaded into Polyester C and biodegradability was evaluated.



- Significantly improved biodegradability by kneading additives
- Biodegradation promoted by the action of microorganisms in seawater and additives
- Achieved speed control

7. Summary and Future plans

● Results in FY2023

- Succeeded in tracking nano structural changes in the torn part during in situ PBS and PBSA film tearing process. (Kyoto Institute of Technology, Professor Sasaki)
- Detection of crystal structure change from α crystal to β crystal during the PBS tearing process. Examined PBS/PR/catalyst blend to improve tearing. (Yamagata University, Professor Ito)
- Confirmed that toughness and biodegradability were improved by adding PR and PPR nanosheets to PBSA. Study of introducing degrading enzymes onto the surface of PPR nanosheets. (Professor Ito, University of Tokyo)
- Confirmed repeated changes in crystal structure and hydrogen bond state of PBS copolymer film in extracted seawater. (Kobe University, Professor Sato)
- Observe $\alpha \rightarrow \beta$ crystal transition due to photooxidative decomposition of PBS and PBSA. Photodecomposition proceeds preferentially in the amorphous region. Biodegradability is promoted by UV irradiation. (Kyushu University, Professor Takahara)
- Surface structure changes due to PBSA hydrolase were evaluated by in situ AFM observation. Clear differences in biodegradation rates were confirmed due to differences in crystal orientation. (Yamagata University, Professor Matsuno)
- Confirmed the high seawater biodegradability of PBS copolymer throughout the year. Weight reduction confirmed in field test of additive-kneaded copolymerized polyester. (Mitsubishi Chemical)

● Future plans

- ① Degradability control (introduction of point control)
- ② Improvement of tear strength

2. Targets

• FY2022 Intermediate Target: **Proof of the multi-locking concept**

- Degradation rate is more than 3 times higher for multiple external stimuli than for a single external stimulus.

• FY2024 Intermediate Target : **Achieve both high toughness and multi-locking mechanisms**

- Degradation rate is more than 10 times higher for multiple external stimuli than for a single external stimulus.
- 5 times higher tear strength than existing aliphatic polyesters

• FY2027 Intermediate Target : **Demonstration of the Bench-scale production**

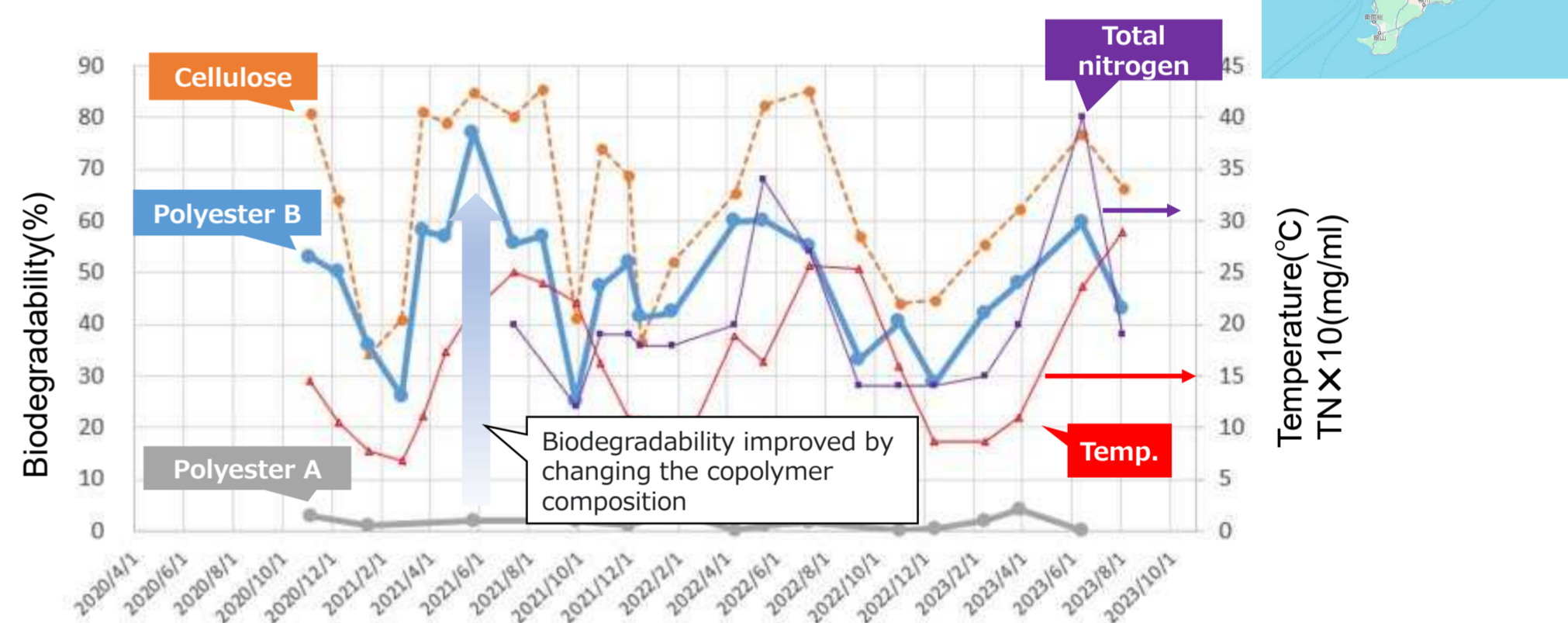
- Can be manufactured in scales of 20 kg or more

• FY2029 **Final Target**: Achieve the followings with scaled-up products

- Marine biodegradation after unlocked :40% biodegradability in sea water (25°C) in 30 days.
- **Tear strength**: More than 10 times that of existing biopolymers
- Polymer production on a scale larger than bench scale

4. Seasonal changes in biodegradation

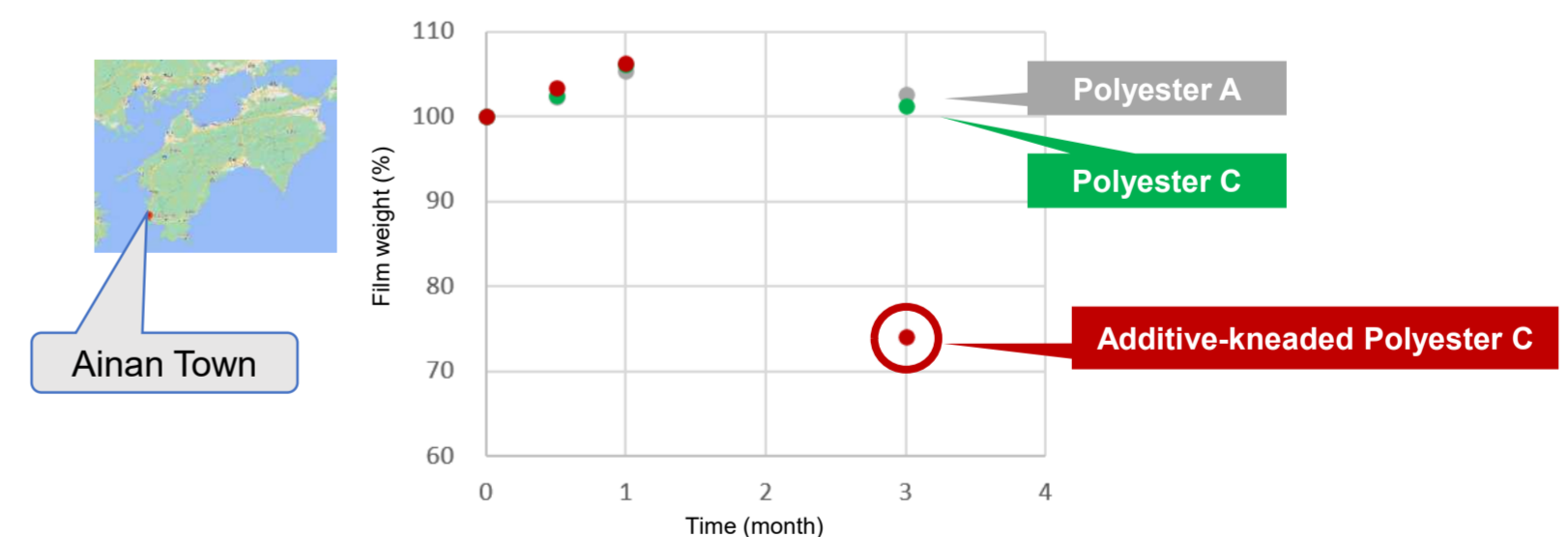
• Evaluation of seasonal changes in biodegradability of cellulose and polyesters A and B using seawater sampled at a fixed point (Tokyo Bay)



- Seawater biodegradability has been significantly improved by changing the copolymer composition (polyester A \rightarrow B)
- Polyester B, like cellulose, maintained relatively high biodegradability throughout the year.
- The biodegradability of polyester B and cellulose is correlated with the total nitrogen content in seawater.

6. Field test

• Participated in a field test conducted by Ito Project in Ainan Town, Ehime Prefecture
 • Evaluation of weight change of various polyester films of 3cm x 3cm x 200 μ m

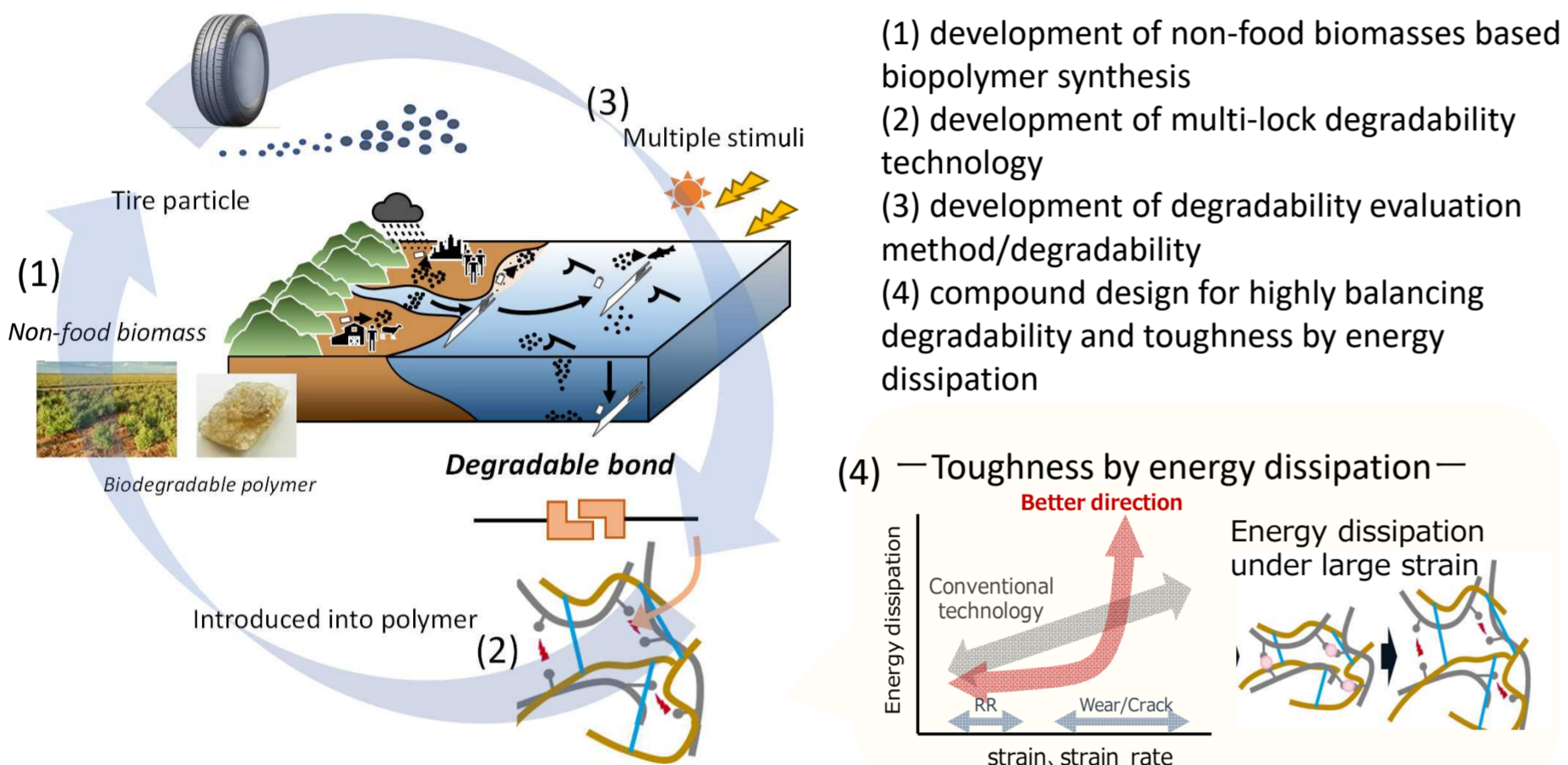


- Significant weight loss after 3 months of additive-mixed polyester C installed on the surface of seawater
- Showed point-controlled decomposition behavior

Research outline of this project

In recent years, there has been growing concern about the influence of tire wear particle on the environment. While its substantial contribution to the environment is still debatable, technological development is desired from a view of environmental pollution/circulation of resources. In this study, we aim to solve these issues by developing non-food biomasses based multi-lock tough polymer which can be decomposed by multiple stimuli. Combined with the toughness technology by energy dissipation proposed in IMPACT project (2014-2019), the developed tough polymer is applied to tire tread, and it demonstrates toughness by effective energy dissipation in use and quickly decomposes by multiple stimuli (microorganism, light, heat, oxygen, etc.) after use in the state of wear particle.

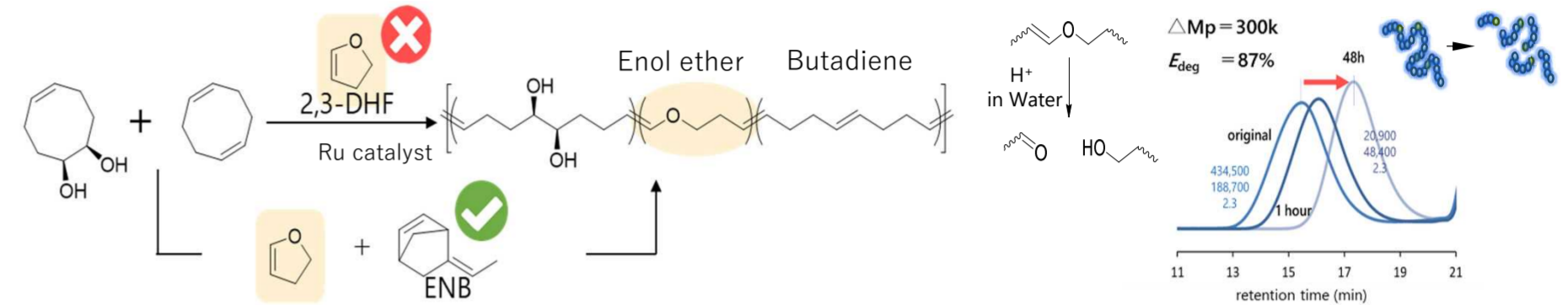
—Biodegradation cycle of tire wear particle—



Biodegradation technology by introducing degradable unit

To develop diene rubber copolymer which can be decomposed by multiple stimuli cooperating with academia

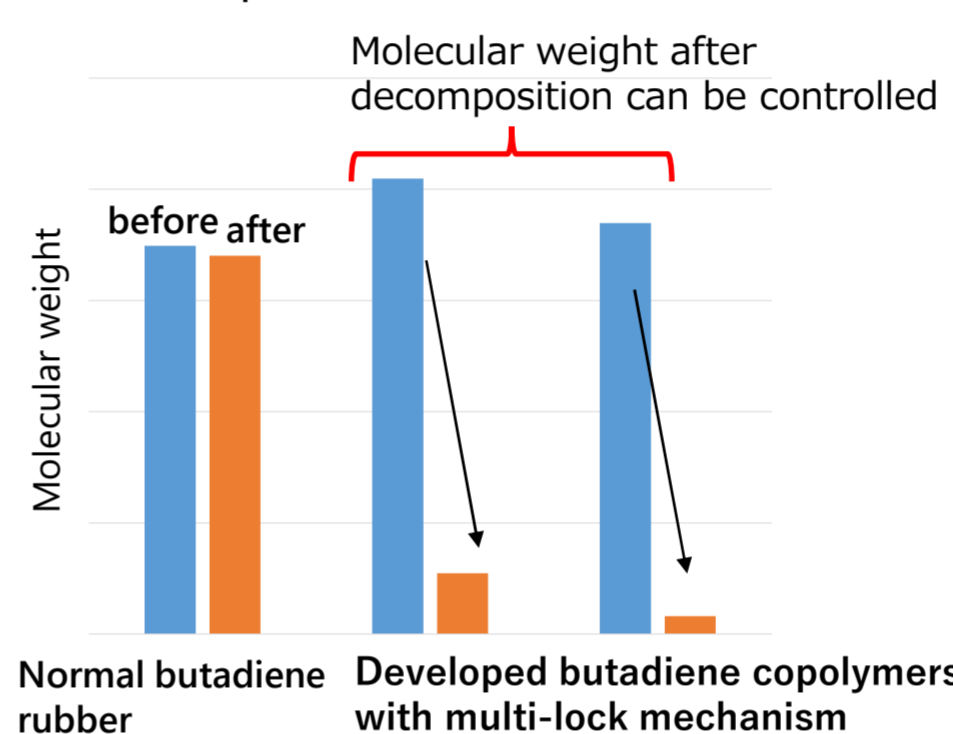
-Polymer Design-



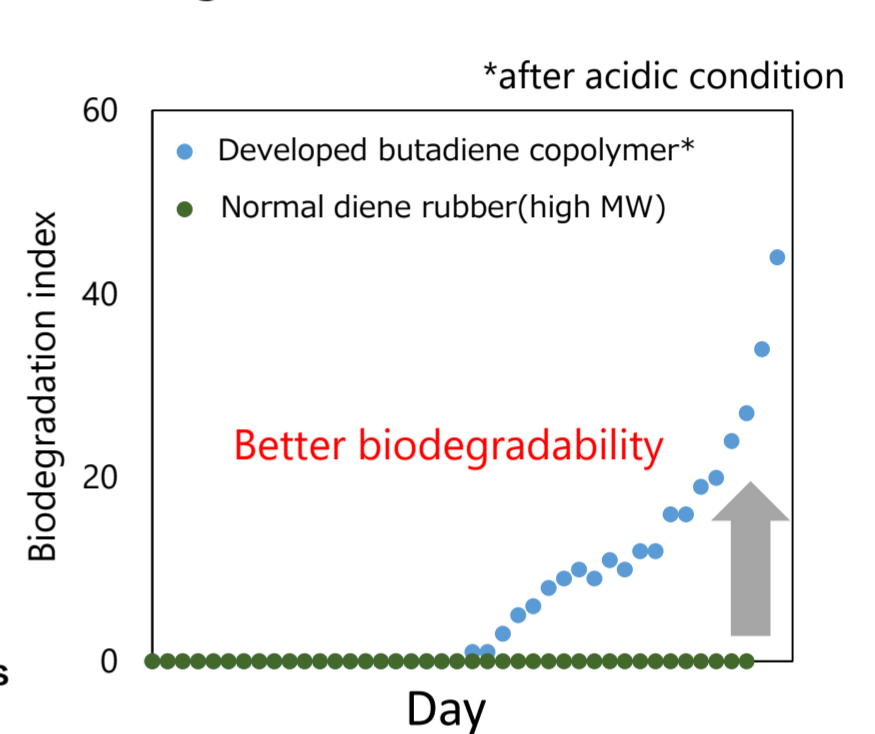
Discovered synthetic route for introducing decomposition unit

⇒ Butadiene copolymer with multi-lock mechanism was developed

—lab decomposition test result —



—Marine biodegradation test result —



Fast decomposition in acidic condition (molecular weight reduced to 1/10)

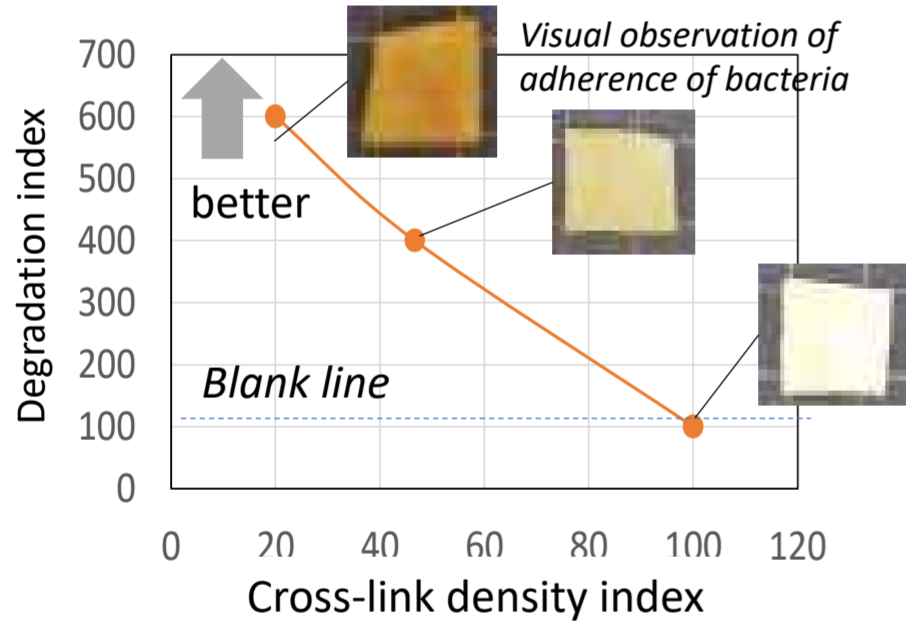
Improved biodegradability due to lower molecular weight

Diene copolymer with multi-lock system was developed. POC, that improves biodegradability by more than 10 times, was achieved.

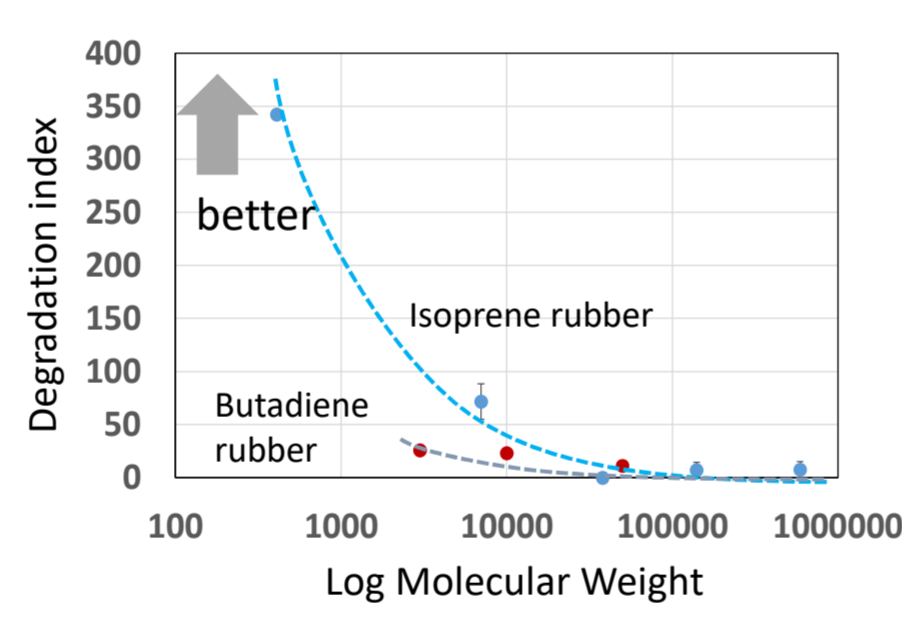
Multi-lock degradable system

★Two locks need to be unlocked for Rubber biodegradation !

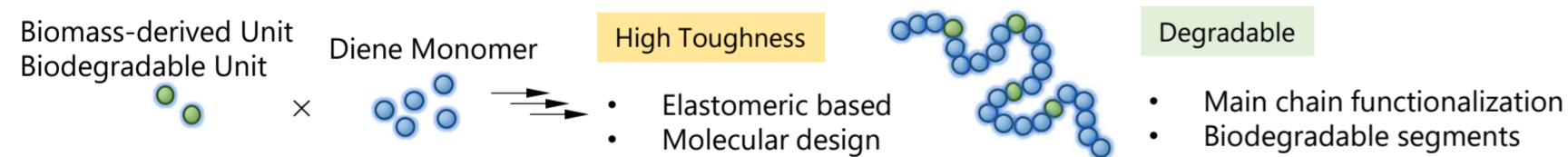
—Degradation test result of rubber with different cross-link density—



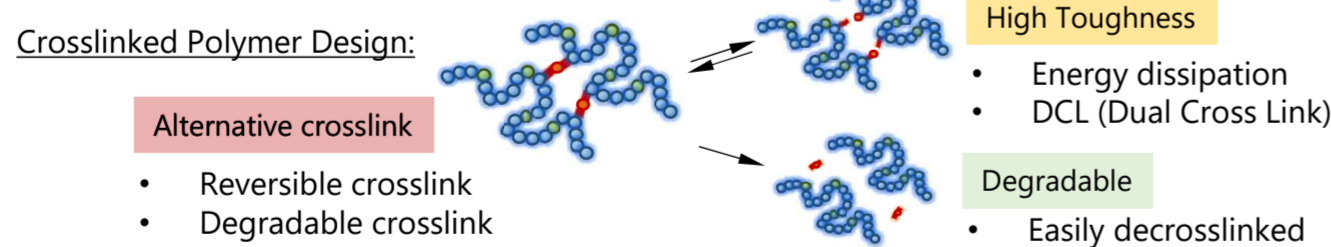
—Marine biodegradation test result of rubber with different molecular weight —



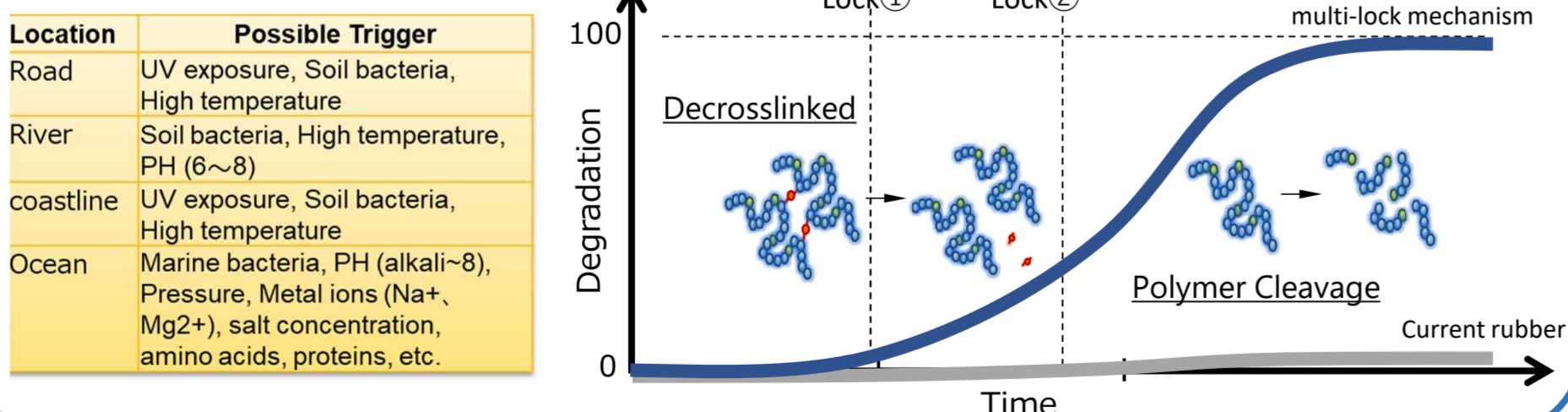
—Development of biodegradable polymer by introducing degradable unit—



—Development of tough and biodegradable polymer by introducing reversible cross-link—



—Biodegradable scheme by multi-lock mechanism

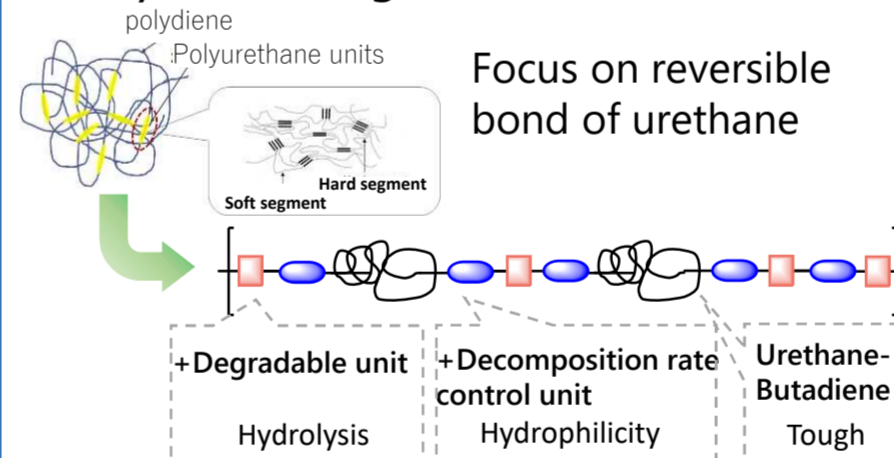


Technology for highly balancing biodegradation and toughness by introducing multi-lock system

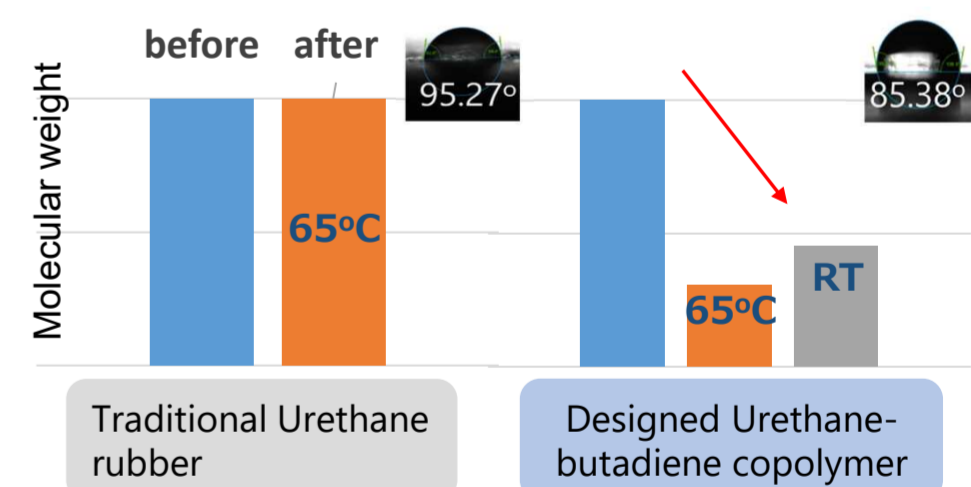
With extending reversible bond technology that strengthen rubber by effective energy dissipation, we newly designed degradable cross link system cooperating with academia.

★Introduced lock unit to diene rubber polymer chain to control tough and decomposition

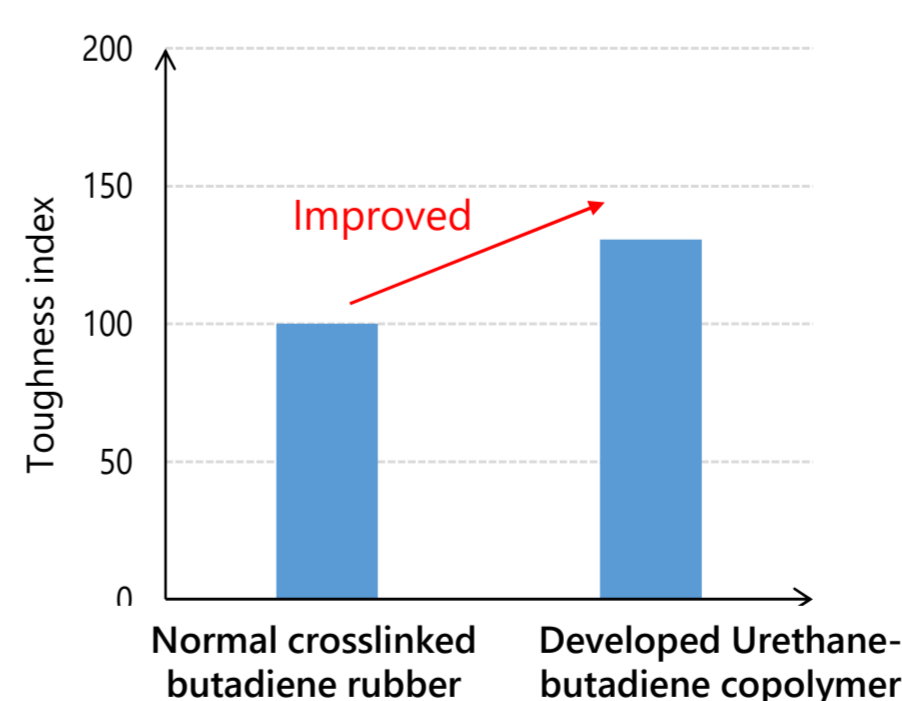
-Polymer Design-



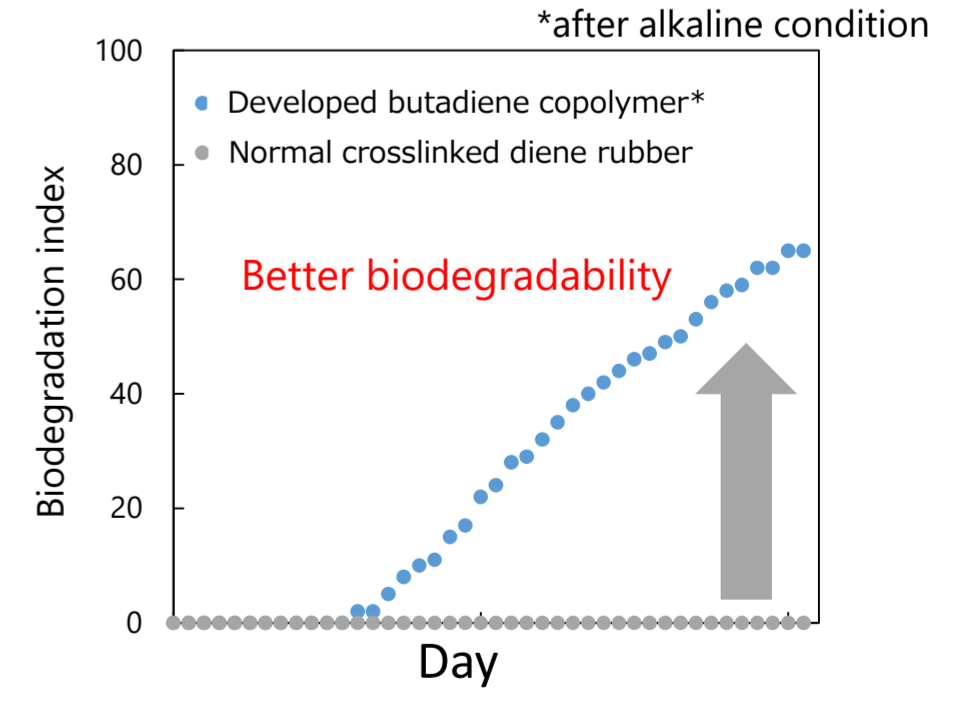
—lab decomposition test result —



—Physical property test result—



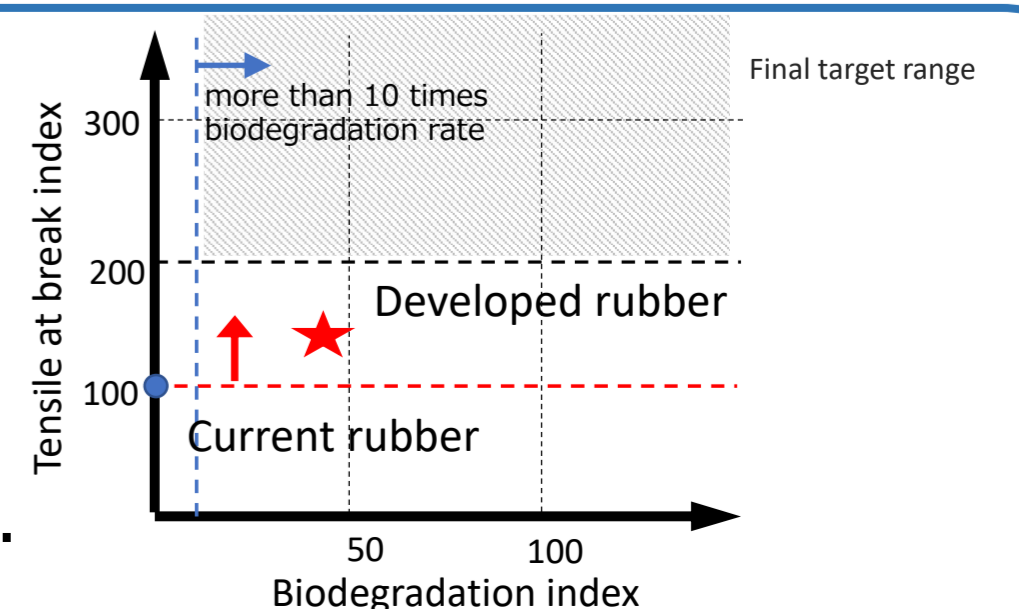
—Marine biodegradation test result —



Tough diene copolymer with multi-lock system was developed. More than 10 times higher degradability without sacrificing toughness was achieved.

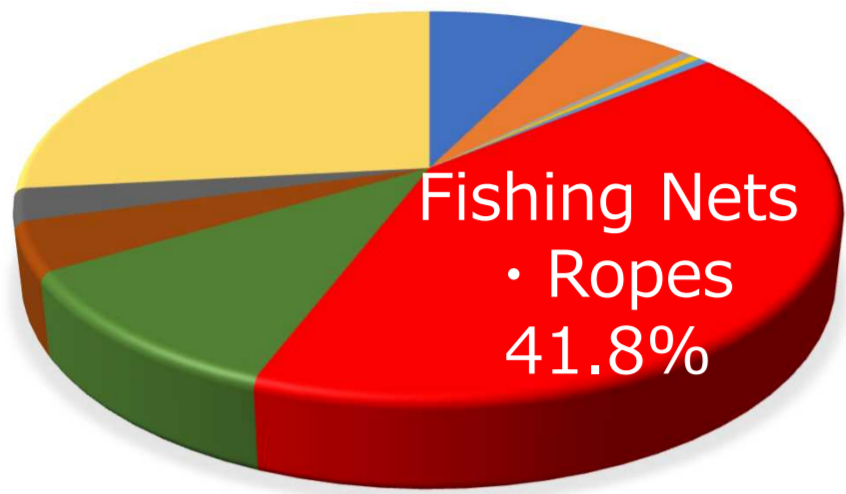
Summary

- We developed diene copolymer with multi-lock system cooperating with academia. ⇒POC, that improves biodegradability by more than 10 times, was achieved.
- Tough diene copolymer with multi-lock system was developed by introducing lock unit to diene rubber polymer chain to control tough and decomposition. ⇒More than 10 times higher degradability without sacrificing toughness was achieved.



➤ Introduction

<Classification of Drifted Plastics> <Problem of "Ghost Gear">



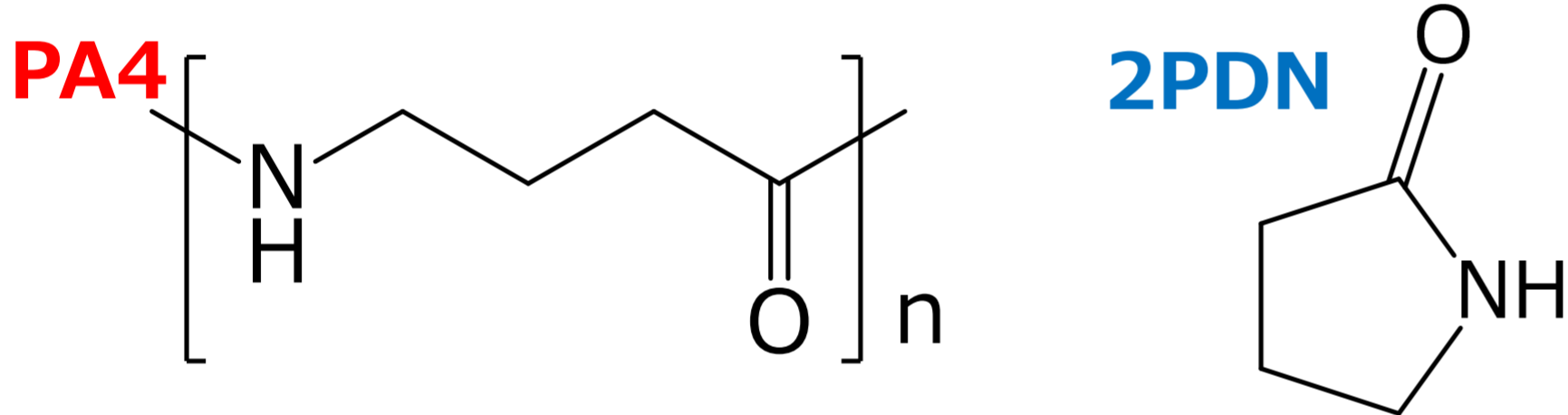
Cited from 「Recent trends in sea garbage」 by Ministry of the Environment



Cited from Wikipedia

- ✓ Fishing nets and ropes make up about 40% of the drifted plastics and they cause a problem called "Ghost Gear".
- ✓ Some biodegradable products are commercialized, however, they generally have inferior mechanical strength and degrade by hydrolysis during use.

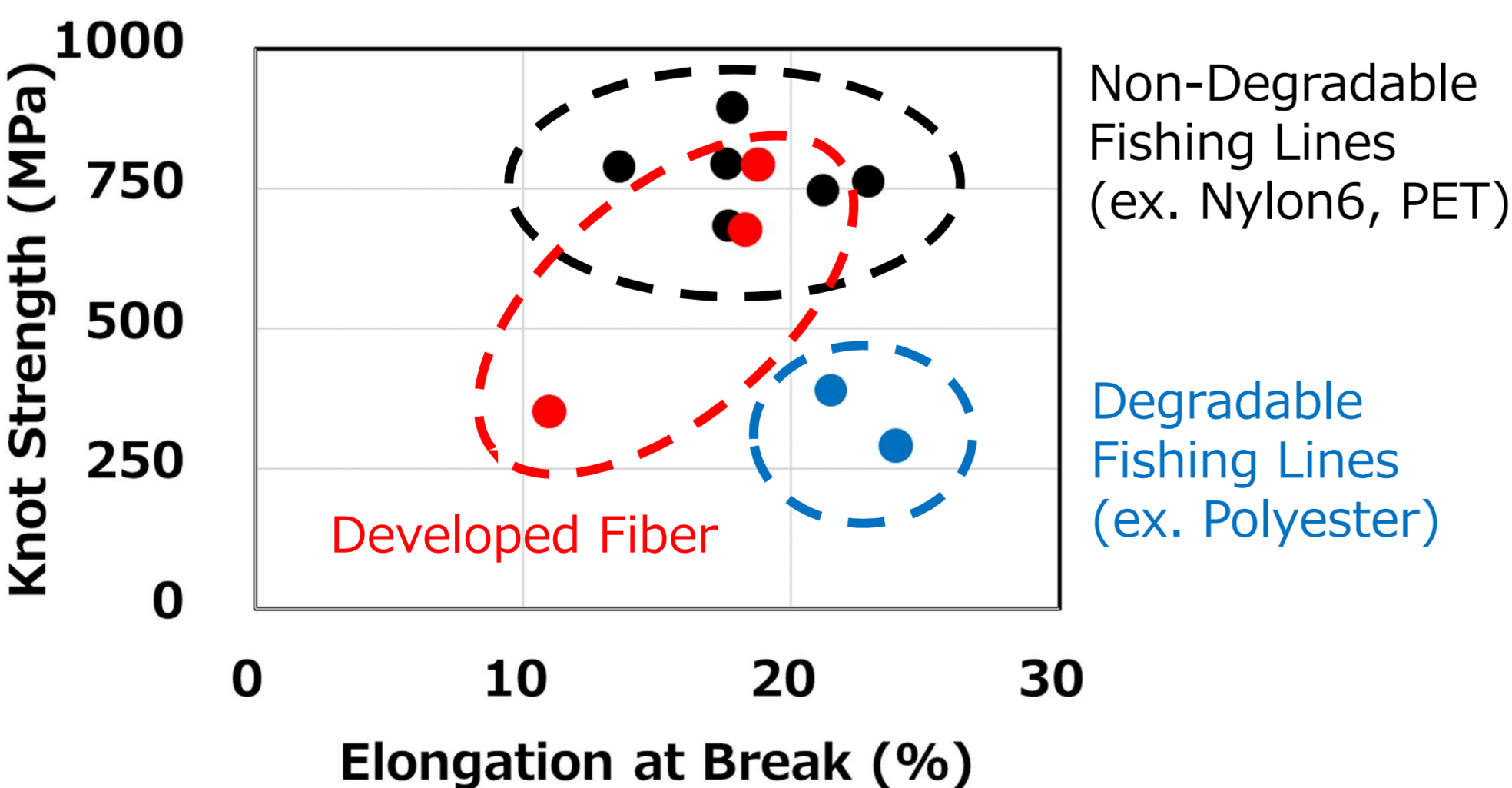
➤ Polyamide 4 (PA4)



- ✓ PA4 is an aliphatic polyamide (nylon) synthesized from 2-pyrrolidone (2PDN).
- ✓ PA4 degrades in natural environments such as soil, sea, etc.
- ✓ In general conditions, PA4 is stable and not hydrolyzed.
- ✓ The mechanical property of PA4 is superior to Nylon6, on the other hand, PA4 has poor processability because it's thermal decomposition point is close to melting point.

➤ Mechanical Strength of the Developed Fiber

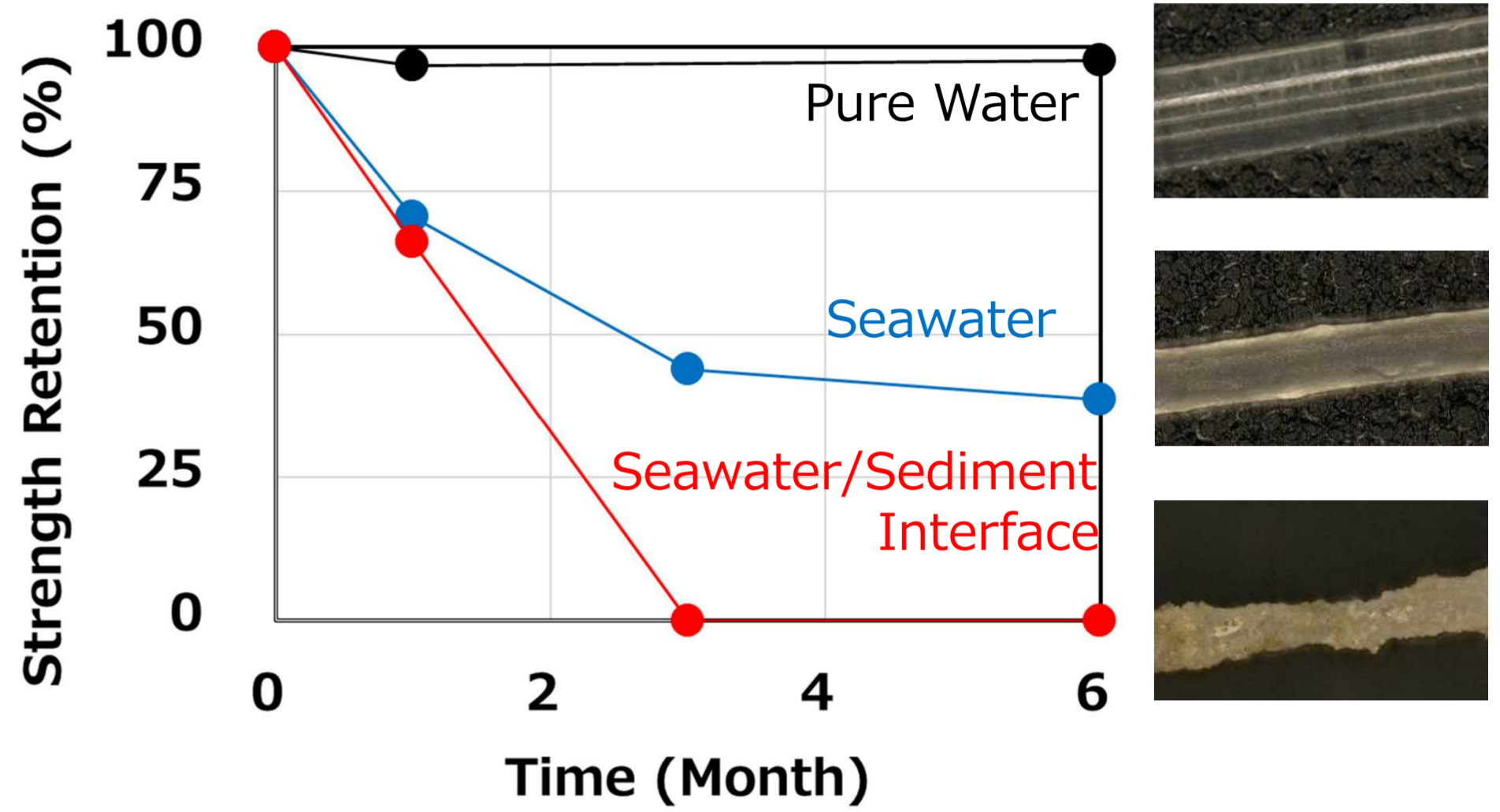
<Comparison of Knot Strength and Elongation>



- ✓ The strength and elongation of developed fiber is equivalent to commercial non-degradable fishing lines.

➤ Marine Biodegradability (Lab. Test)

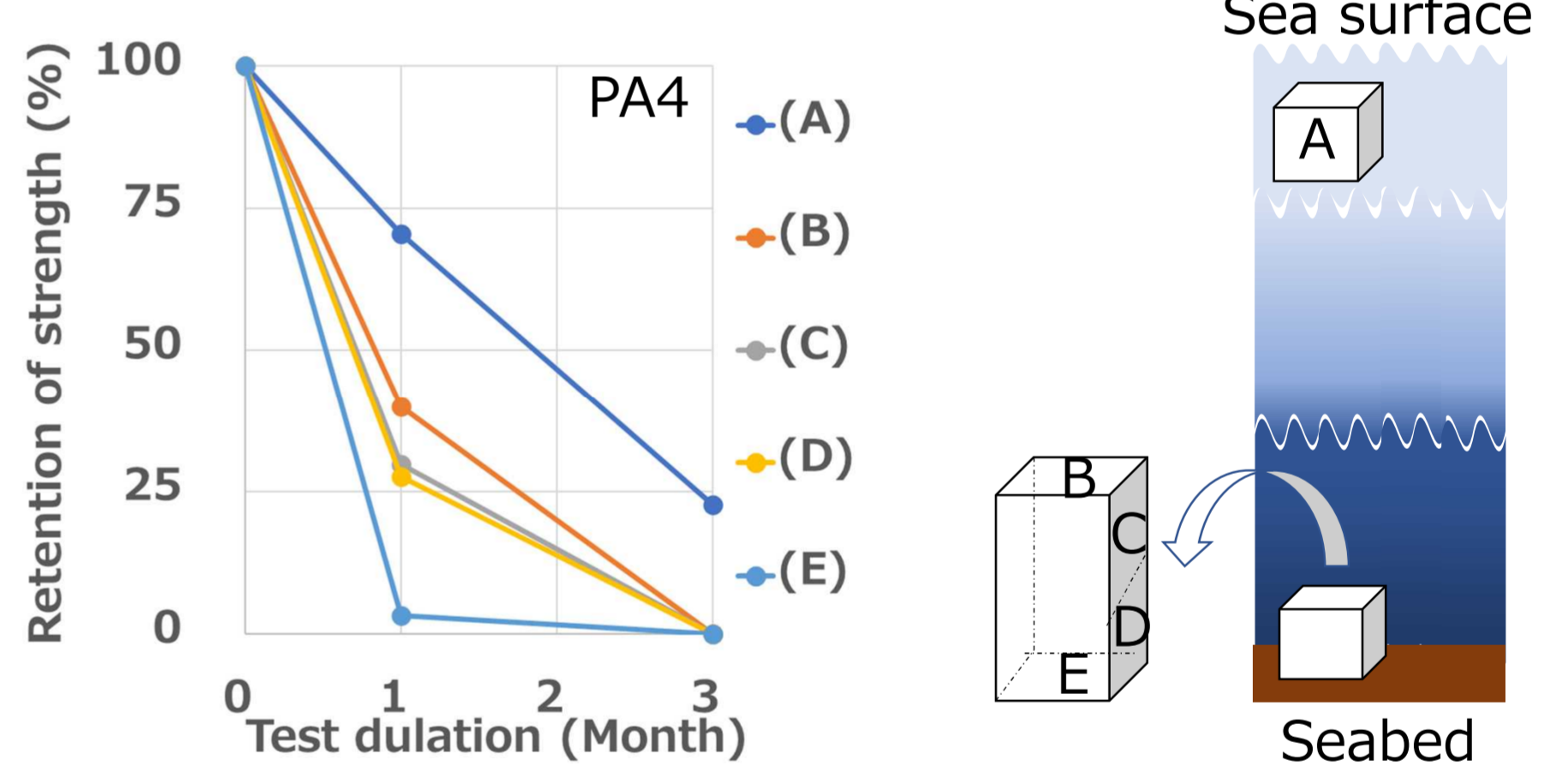
<Degradation Test of the Developed Fiber in Laboratory >
Seawater and sediment were collected from the Pacific Ocean (Fukushima pref.).



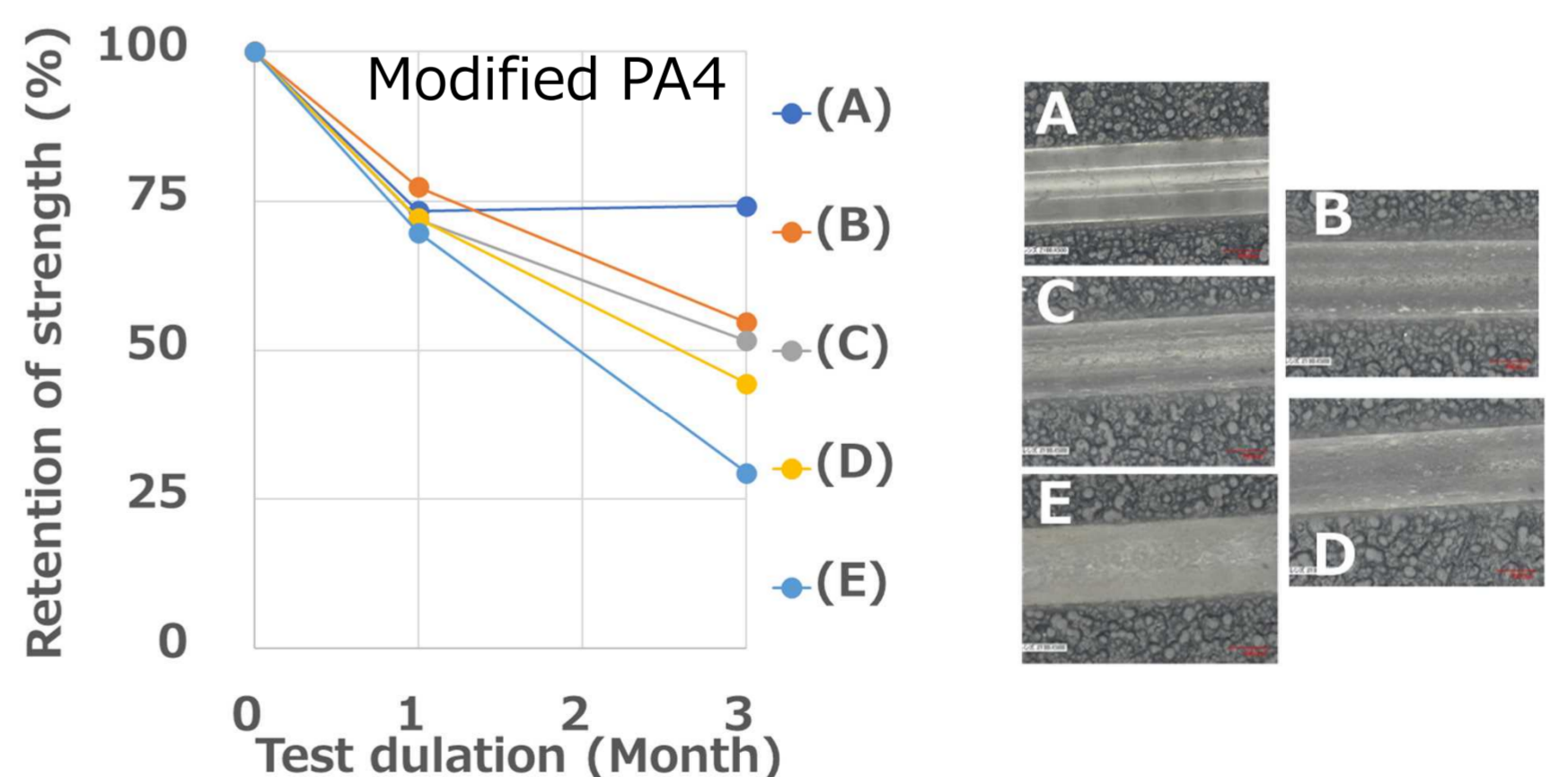
- ✓ In pure water, there was almost no reduction in strength.
- ✓ At seawater/sediment interface, degradation was faster than in seawater.

➤ Marine Biodegradability (Field Test)

Field test has been conducted at Ainancho (Ehime pref.).



➤ Control of Degradation

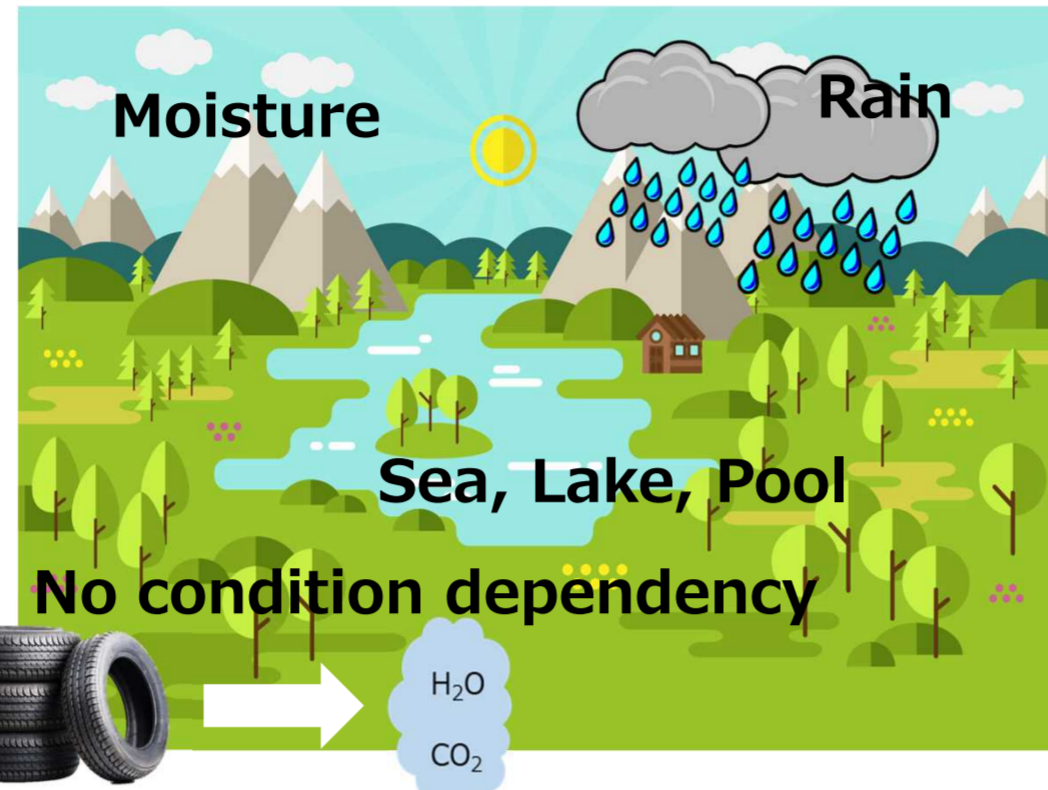
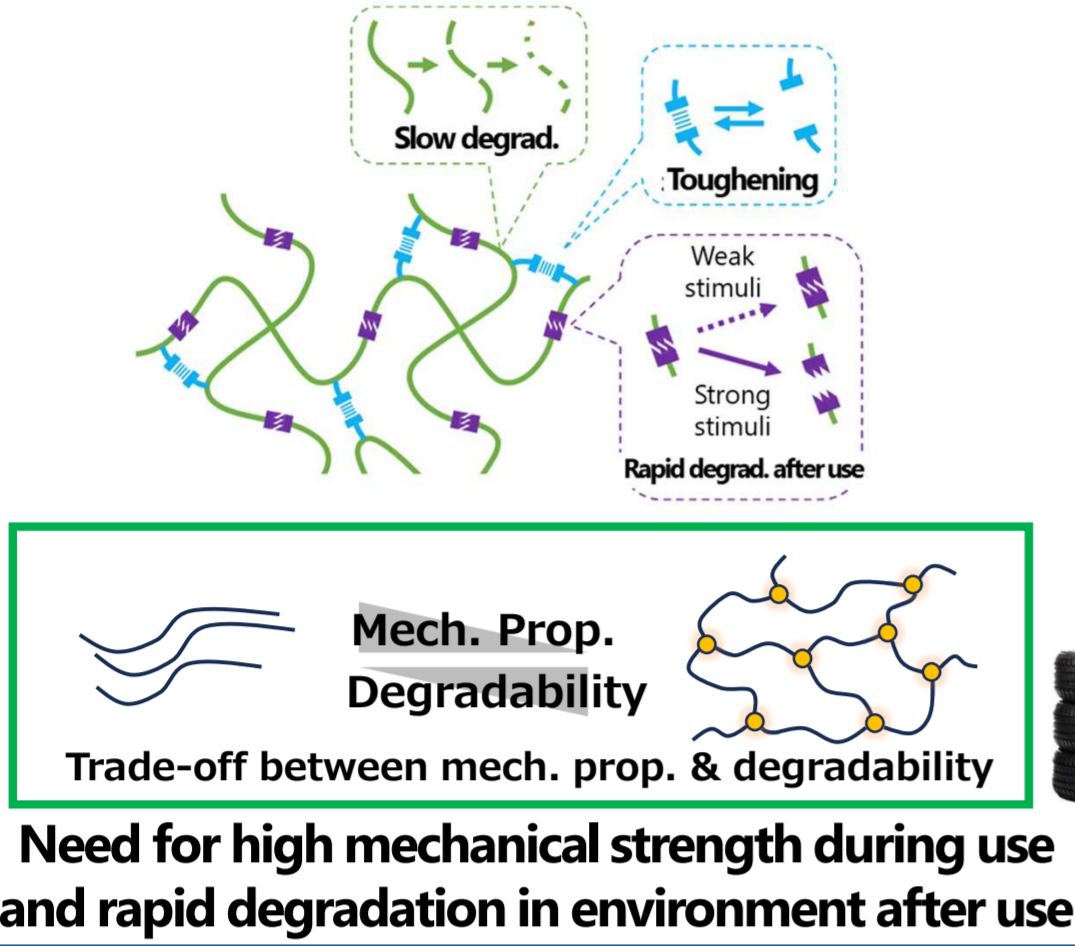


➤ Summary

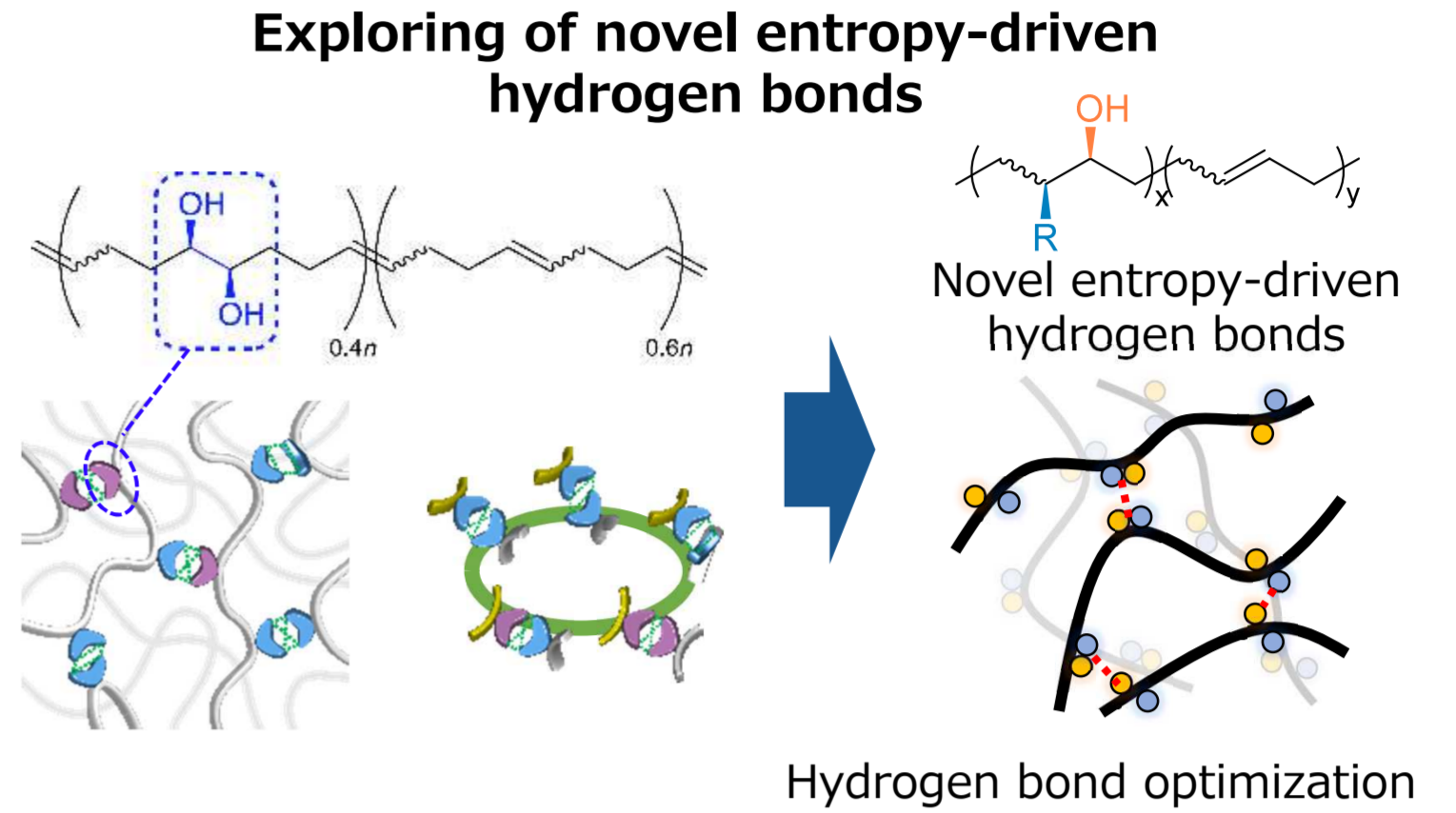
- ✓ We are developing biodegradable and tough biopolymer for fishing gears based on polyamide 4.
- ✓ The knot strength and elongation of developed fiber is equivalent to commercial non-degradable fishing lines.
- ✓ Marine biodegradability of the fiber was confirmed by lab. and field test.
- ✓ The degradation rate of modified fiber was reduced to 1/3.

Introduction

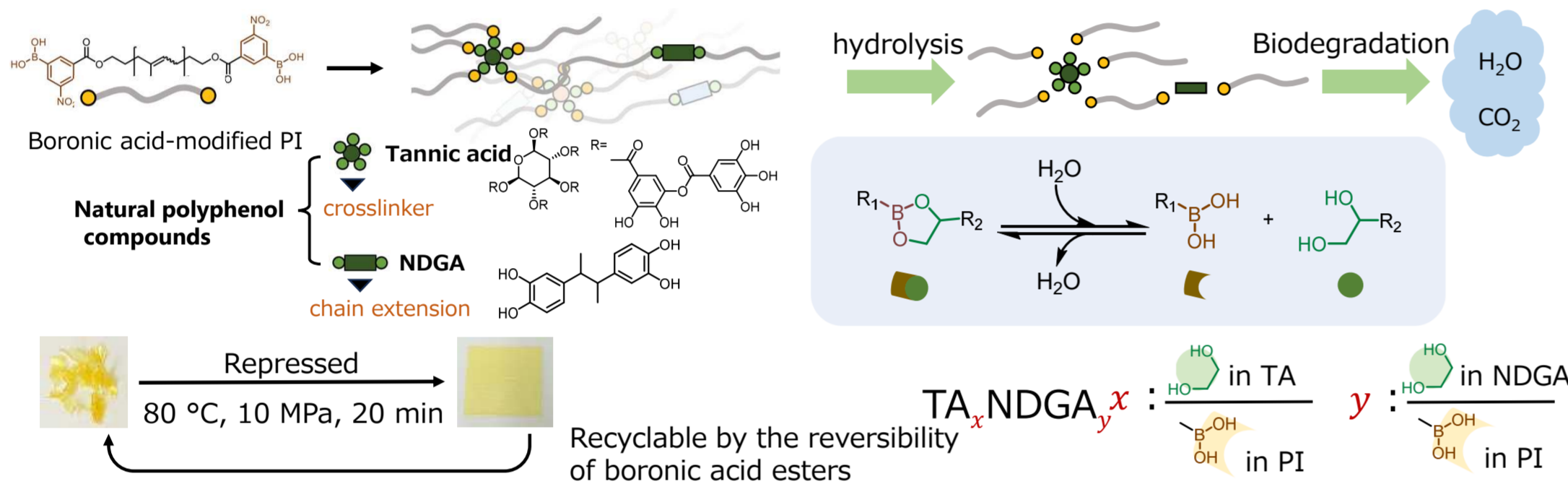
Topic 1: Development of multi-lock mechanism



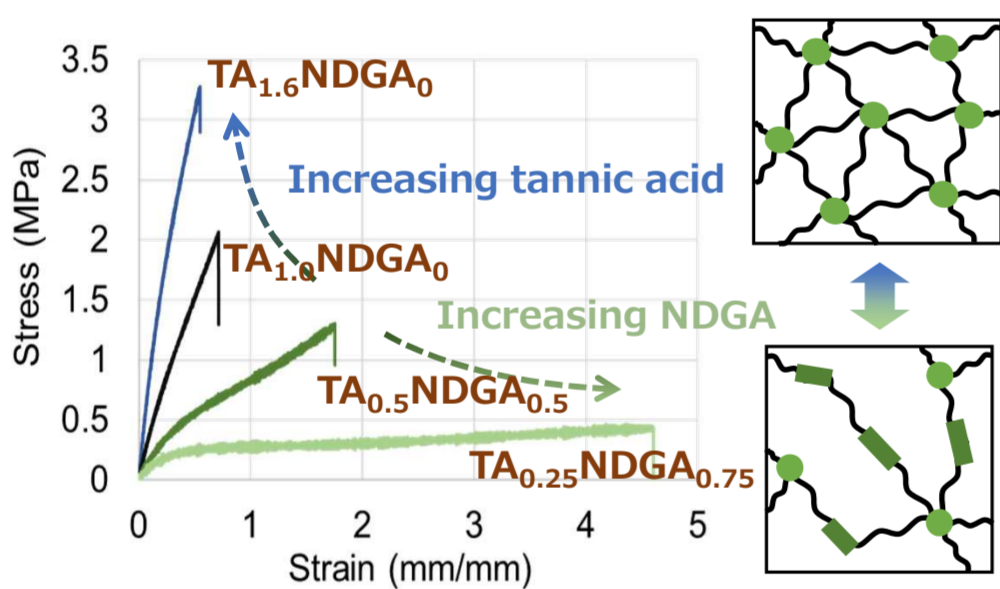
Topic 2: Combine crosslinking and degradability



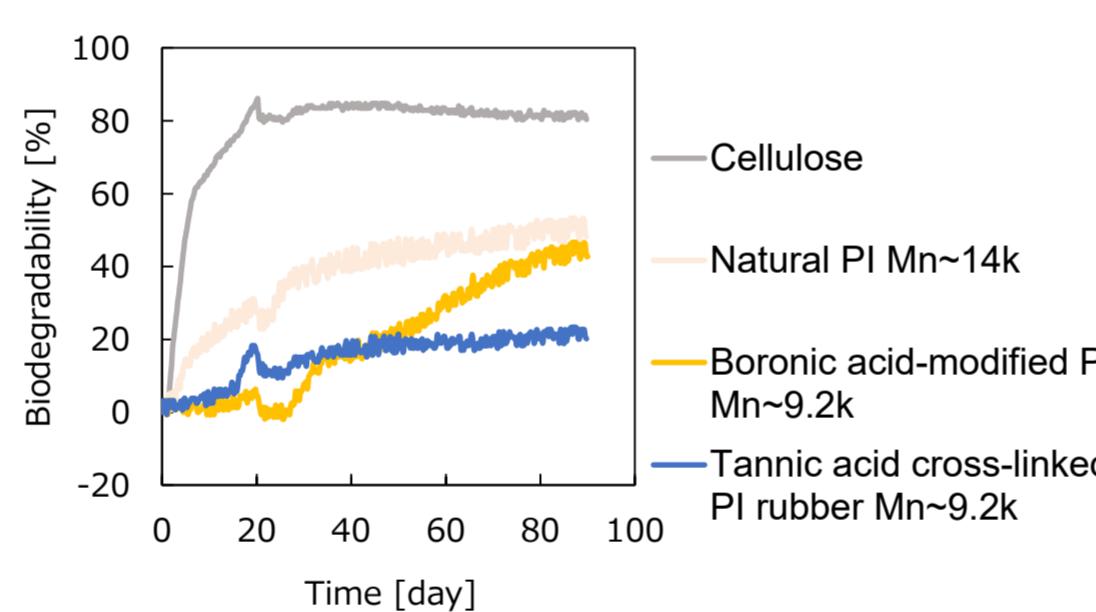
Tuning degradability using water-responsive dynamic bonds by using boronate esters



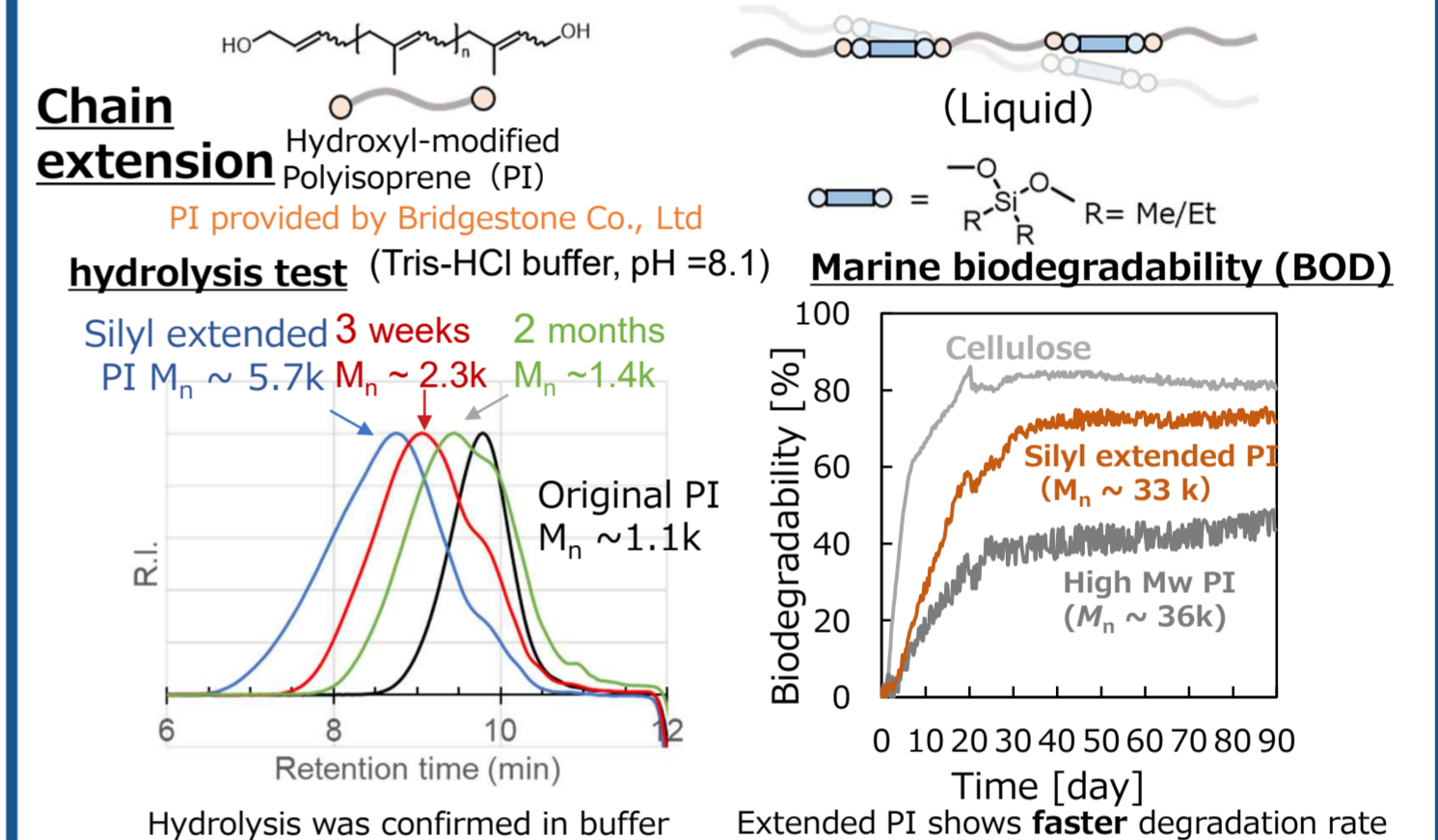
Uniaxial tensile test



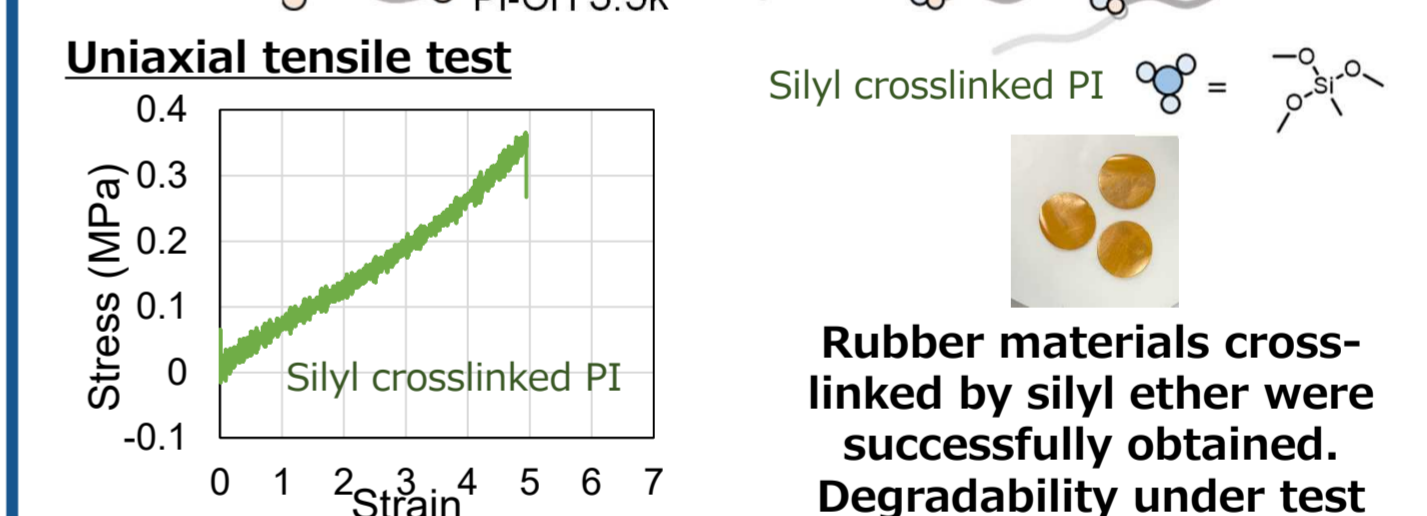
Marine biodegradability (BOD)



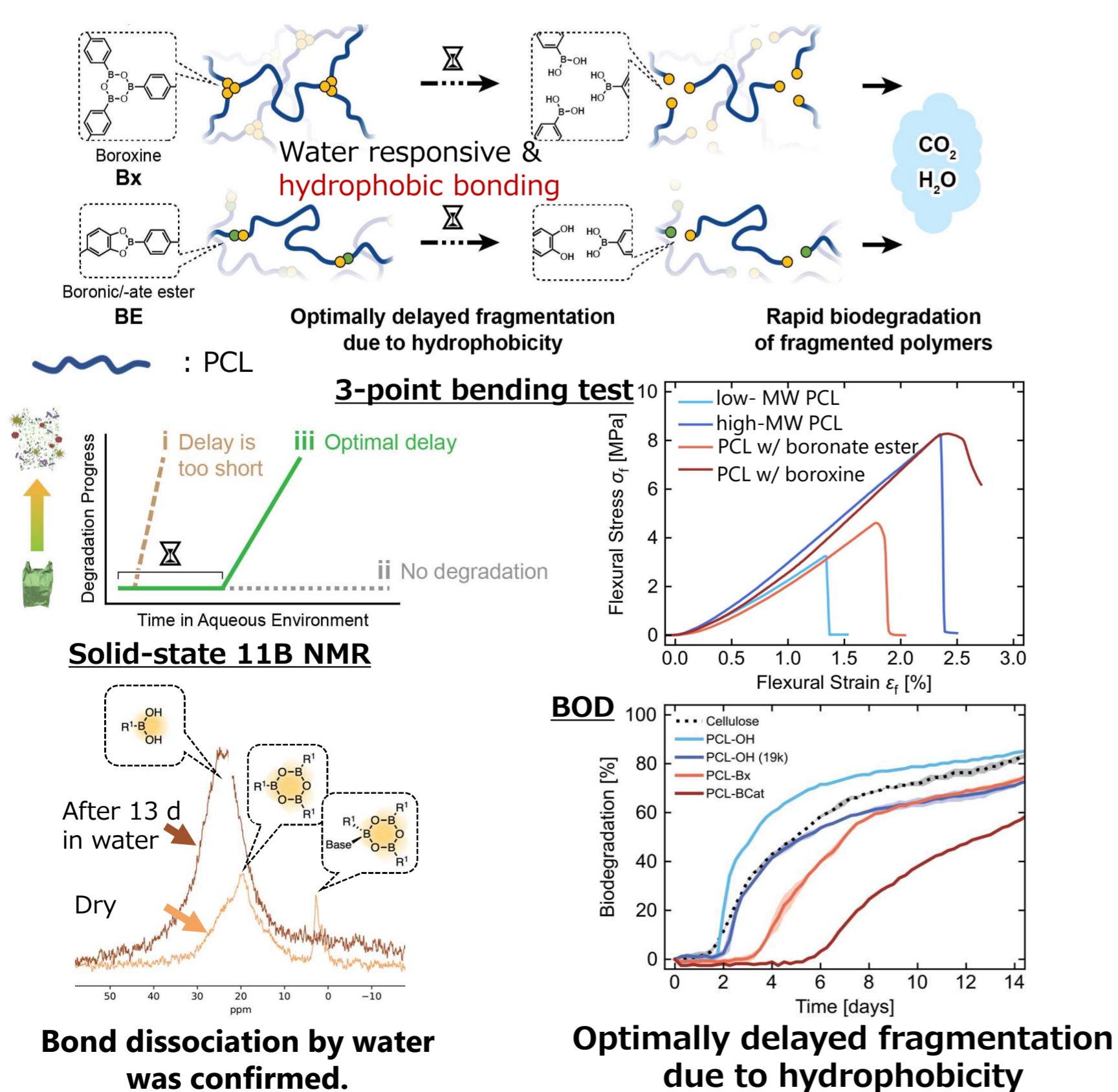
Utilization of silyl esters



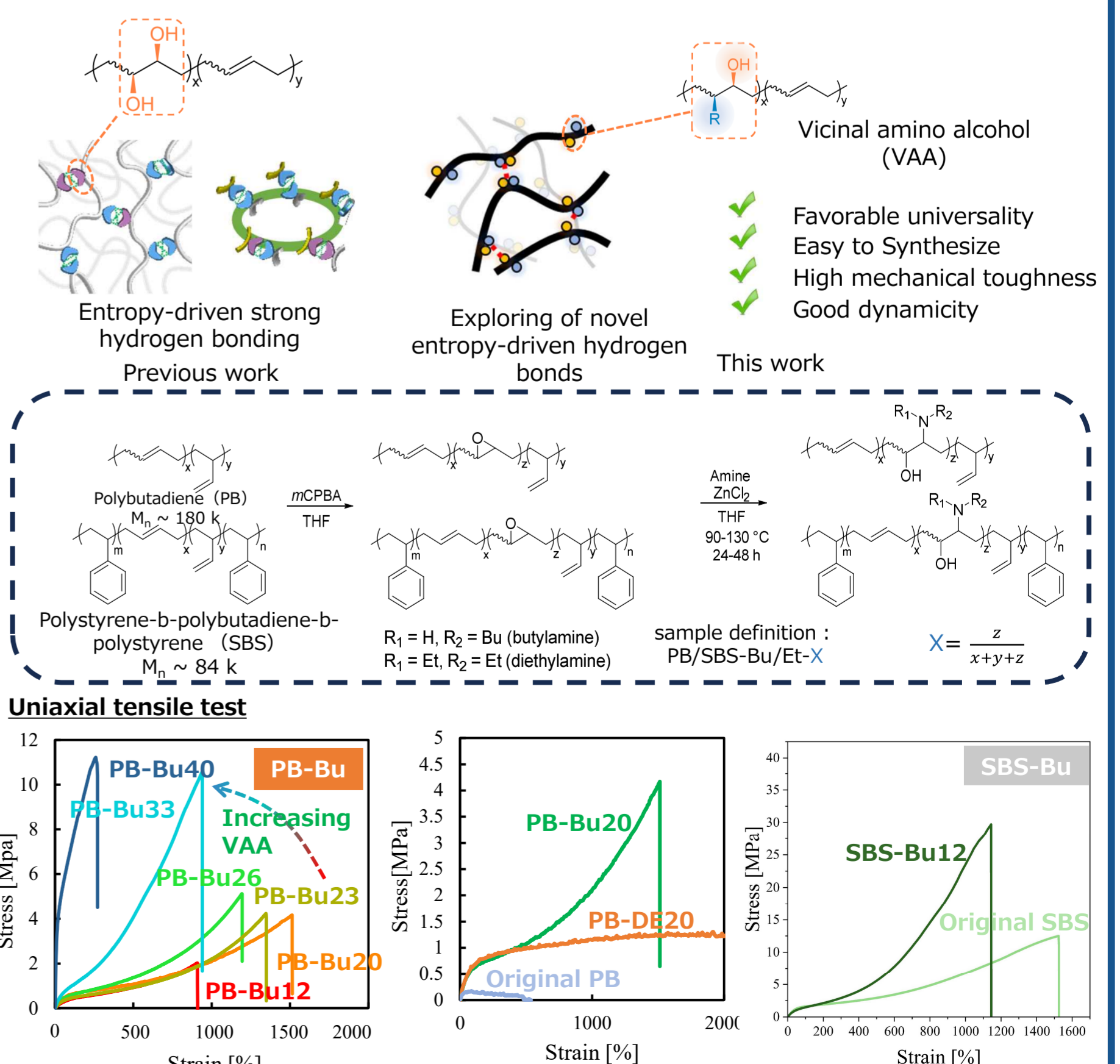
Crosslink



Tuning degradability of fast biodegradable PCL



Exploring of novel entropy-driven hydrogen bonds



1. Introduction

Our technology

- Multi-scale simulation technology for polymer materials from atomic to mesoscopic scales.
- Material analysis using informatics technology

Issues in development of multi-lock-type polymer materials

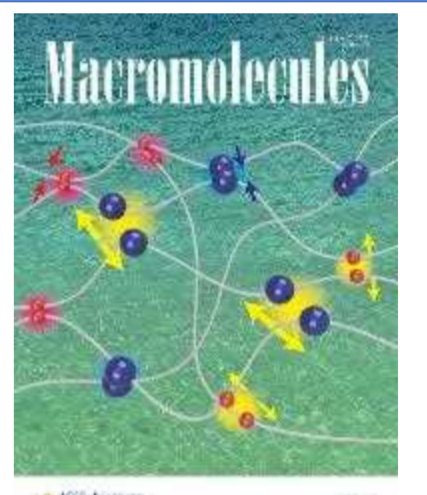
- Elucidation of degradation mechanisms for the development of multi-lock functions for polymers
- Material designs for toughened polymer with physical cross-linking

Objective until end of FY2024

- Model study of degradation of multi-lock polymer** : Yoshie et al developed the dynamic bond elastomer having the functions of toughness in use and degradation in marine. In this study, we made the model of dynamic bond elastomer to clarify the functions of both toughness and degradation. In the near future, we will design its material having those functions in high level.

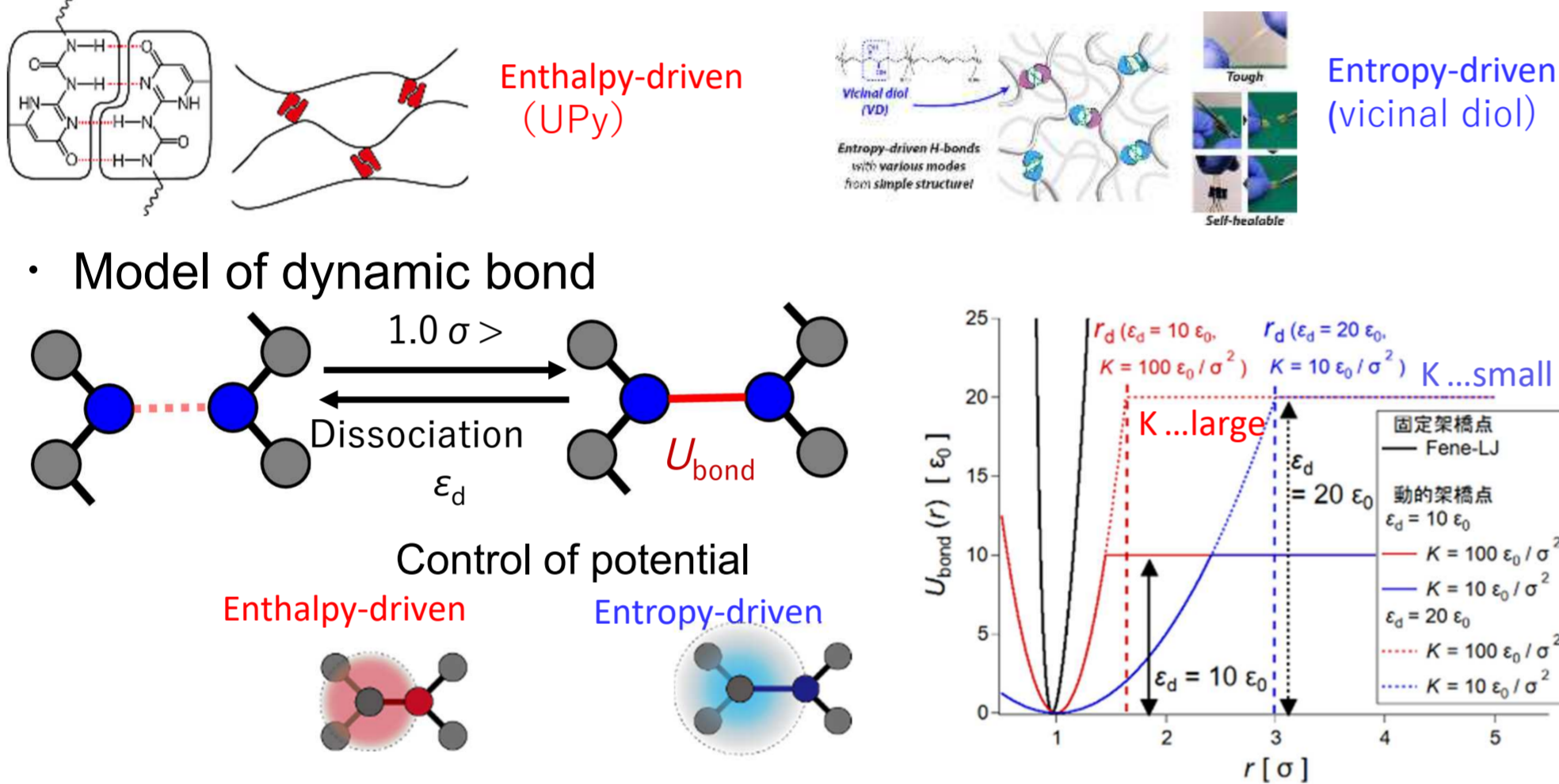
2. Model study of degradation of multi-lock polymer

Y. Yasuda, et al., Macromolecules, 2023, 56(18), 7432

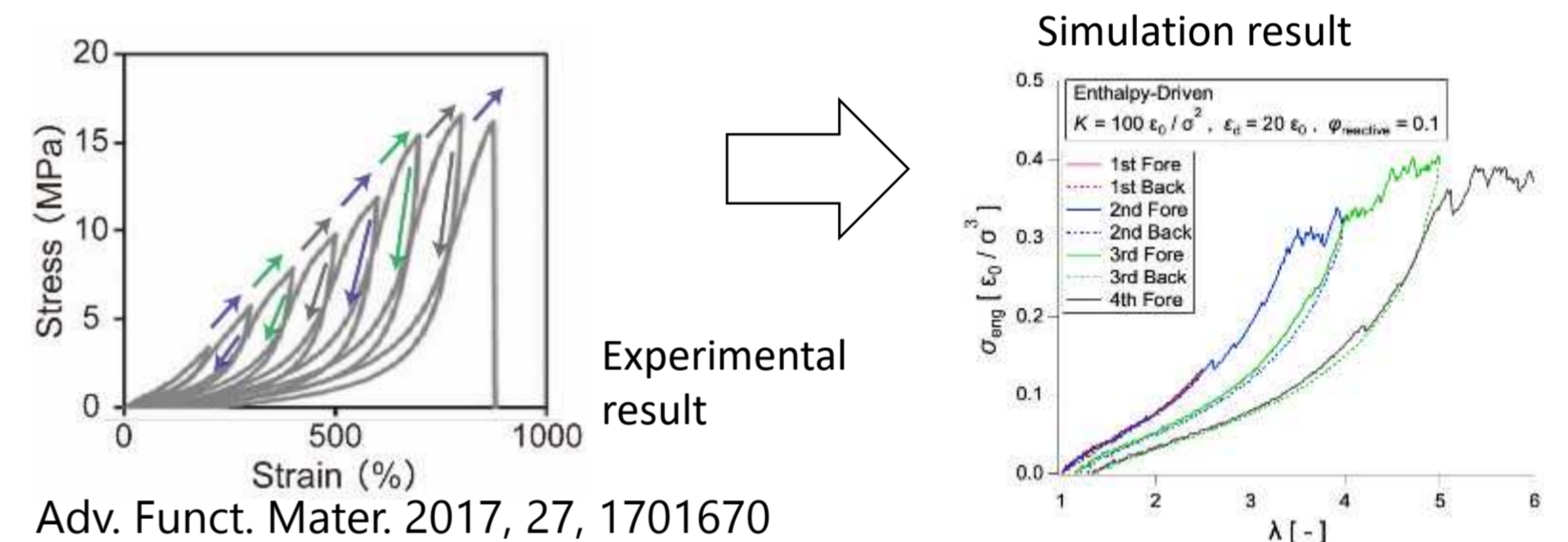


2.1 Modeling of Dynamic bond elastomer

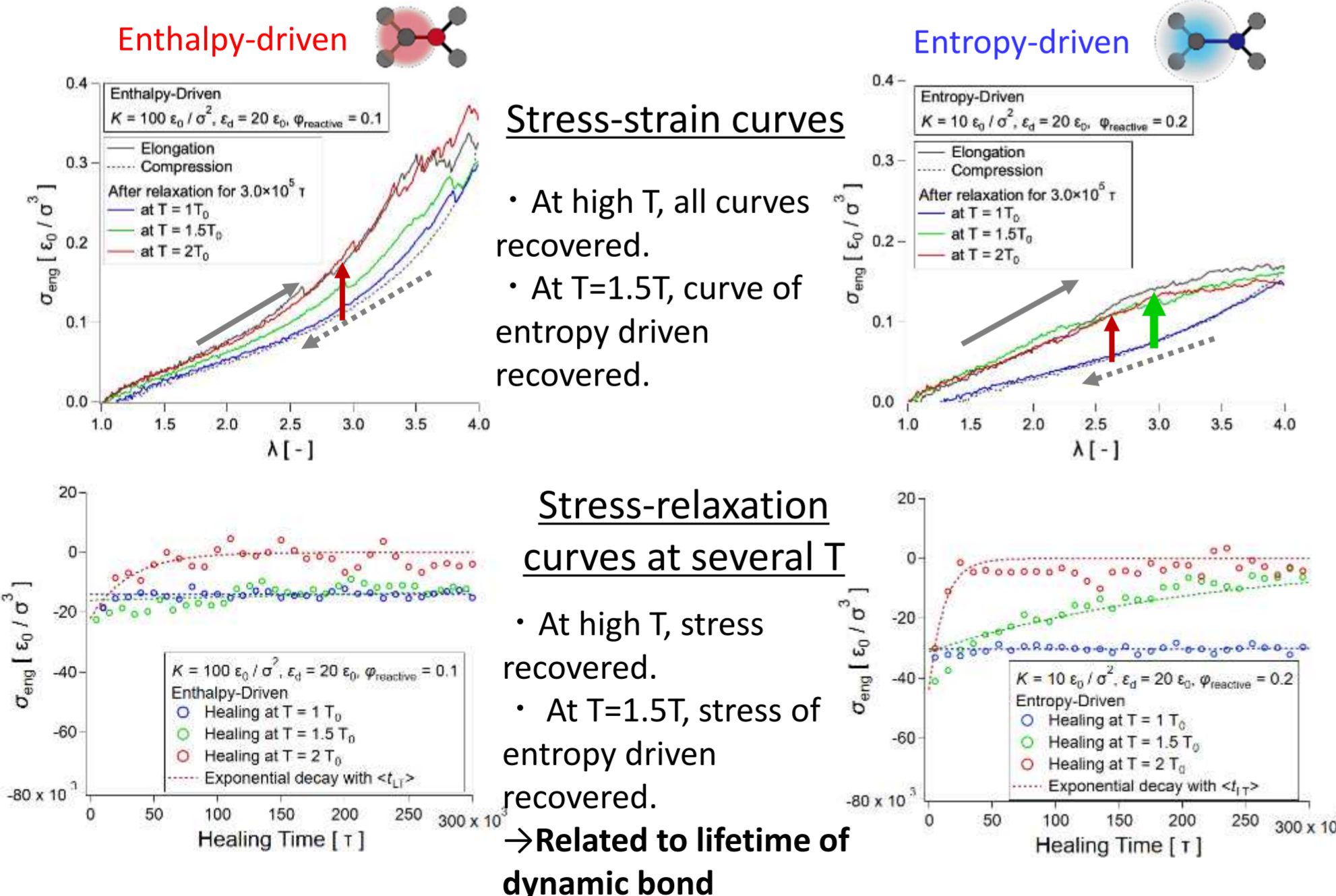
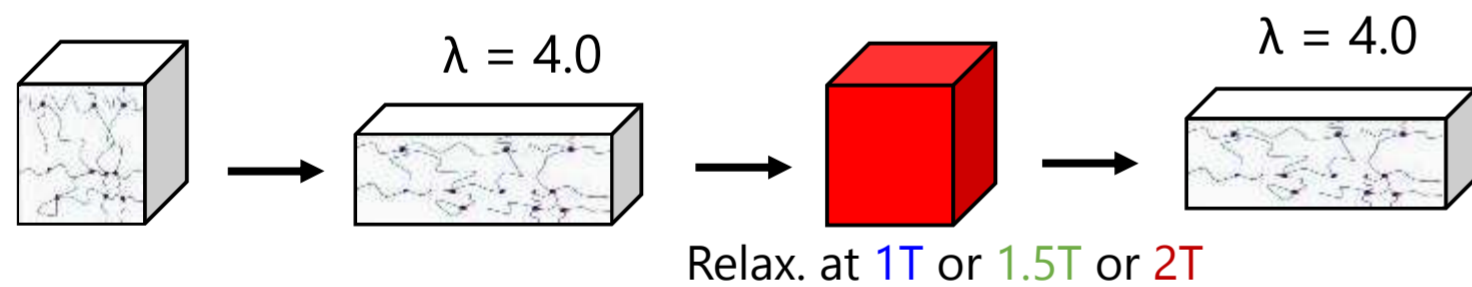
- Dynamic bond elastomer developed by Yoshie et al



- Reproducing experimental results

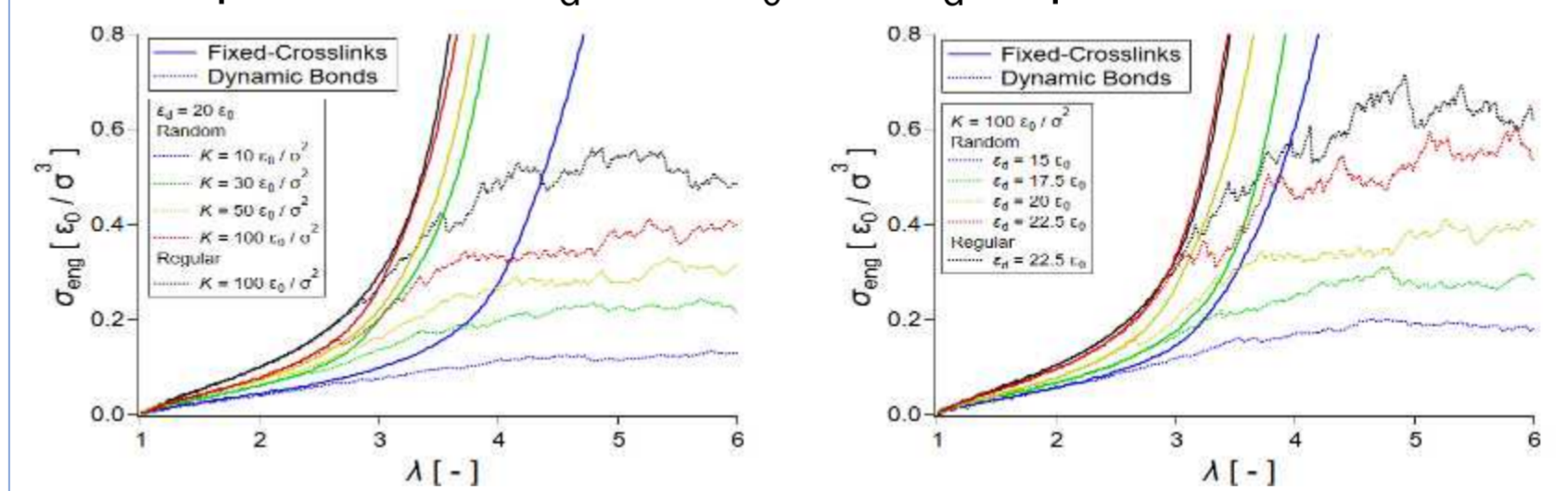


2.2 Stress recovery of dynamic bond elastomer

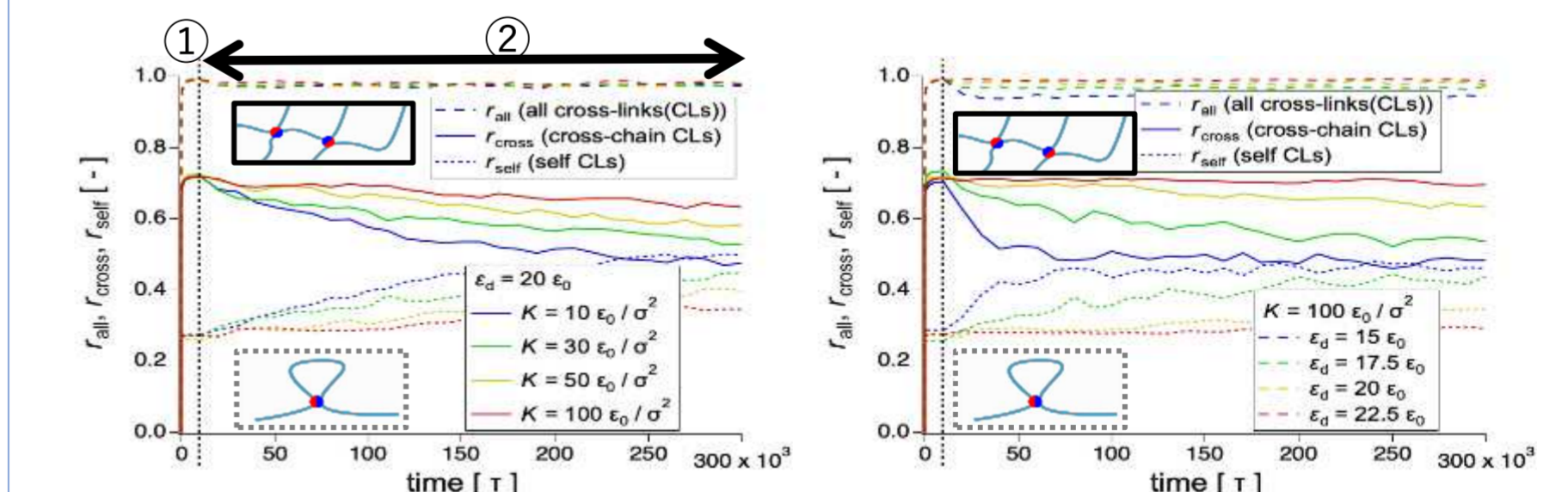


2.3 Change in the cross-linked structures of dynamic bond

- K dependence ($\epsilon_d = 20 \epsilon_0$)
- ϵ_d dependence ($K = 100 \epsilon_0 / \sigma^2$)



- in the case of small K (entropy driven) low ϵ_d (small dissociation energy), initial elastic modulus and maximum stress become smaller.



- At ① term (early term), network formed under kinetic control. At ② (following term), network reconfigured to thermodynamically stable structure.
- in the case of small K (entropy driven) low ϵ_d (small dissociation energy), number of self cross-link increased.

4. Future work

Model study of degradation of multi-lock polymer

- Refinement of the model along the corresponding experiment and simulation study using its model. Conducting a detailed analysis of the state of the association, which is not known by experimentation.
- In-depth collaborative research with companies by the simulations contributed to material design.

Biodegradable resin with polyrotaxane addition for both toughness and seawater degradability

Toughness

Biodegradable polymer

Marine biodegradation

Effect of adding PR

- Combination of toughness and biodegradability of resins
- Combines toughness and biodegradability through ester exchange reaction
- Achieved toughening and biodegradability of ester-exchange vitrimers

Multi-lock function

- Synthesis of end-stimulated degradative PR polymer chain
- Controlled structural parameters such as coverage ratio, axial molecular weight, and graft density

Social integration

- One-pot synthesis of PR to reduce production cost and control inclusion rate
- Reduced production cost by PR extraction and purification method
- Raw materials such as components, solvents, etc. Cost reduction

Achieved both toughness and seawater degradability (speed control). Aiming to elucidate point control and biodegradation mechanism using PR.

Toughening of epoxy vitrimer resins containing polyrotaxane

Homogeneously dispersed due to chemical bonding of PR and cross-linking network

Welding

Heating above T_g

Self-healing

schrotch depth: 0.1 mm

0 s, 10 s, 60 s

Reprocessability

200 °C, 20 kN, 2 h

Shape recovery and shape-memory editing

Initial shape, Deformed at T > T_{set}, Recovered at T > T_{set}, Fixed at T < T_{set}, Memory editing at T > T_{set}, Removal of tape at T < T_{set}, New equilibrated shape

Chemical degradation

VPR_10

12 h

Seawater biodegradation

Biodegradation by BOD (%)

Incubation time (day)

Toughened without changing hardness by adding PR

In addition to the toughening due to the addition of PR, the energy potential of the bond exchange reaction due to sliding motion is reduced.

S. Ando, M. Hirano, L. Watakabe, H. Hideaki, K. Ito, "Environmentally Friendly Sustainable Thermoset Vitrimer-Containing Polyrotaxane", ACS Materials Lett., 2023, 5, 3156.

Development of enzymatic switching degradation by utilizing suspected polyrotaxane nanosheets

A completely new type of nanosheet material consisting of the non-covalent assembly of cyclic oligosaccharides (cyclodextrins) and polyethers, discovered in our laboratory in 2019.

Size: 1-2 μm, Thickness: 16 nm

Monoclinic crystal: a = 1.91, b = 2.43, c = 1.57(nm), α = γ = 90°, β = 111°

Schematic illustration of this study

✓ Highly biocompatible

✓ Constituent molecules are biodegradable

✓ Easy preparation method

✓ Nano-scale thickness

✓ Polymer chains on the surface

✓ Enhanced stability due to end-capping with protein

✓ Protein releasing by the salt

✓ Controlled dissolution of PPRNS based on the concentration of salt

✓ Dissolution and degradation of PPRNS

✓ Promoted biodegradation of the plastic by the released enzyme

➤ Controlled switching of biodegradability which is triggered by salt concentration

Application in biodegradable plastics in ocean

✓ Toughening polymer by PPRNS filler

Wasted in ocean (+NaCl)

Biodegradable polymer

PPRNS / enzyme filler

Enzyme releasing by the sea salt

Corporate Collaboration (Mitsubishi Chemical)

Application of PBS and PBSA to agricultural coated fertilizer shells

Microcapsule (MC) Fertilizer

covering material

Fertilizer, Fertilizer aqueous, Fertilizer aqueous

Water absorption, Water (water vapor) Dissolution, Elution

Plastic film fertilizer coating can supply a fixed amount of fertilizer to plants for a long period of time. It is used in approximately 60% of rice paddies in Japan.

Coated fertilizer, After Elution

5 mm

Plastic shells collected at the beach

Requires material toughness to release fertilizer in constant amounts. PE is often used.

Expected to be replaced by biodegradable resin with the same level of toughness as PE.

PBS+PR S-カーブ

Stress (MPa) vs Elongation (%)

2.3 times up

PBSA+PPRNS S-カーブ

Stress (MPa) vs Strain (%)

Fracture toughness PPRNS Effect 1.5 times up

PE vs. 1.7 times up

PE mulch film

PBS+PR 海水生分解性

Biodegradability (%) vs Time (Days)

3-5 times up

PBSA+PPRNS 海水生分解性

Biodegradability (%) vs Time (Days)

1.2 times up

Toughness and biodegradability improved by addition of PR and PPRNS

Polybutylenesuccinate (PBS)

High strength but brittle

Low marine biodegradability

Polybutylenesuccinate-co-adipate (PBSA)

Softer and more ductile than PBS

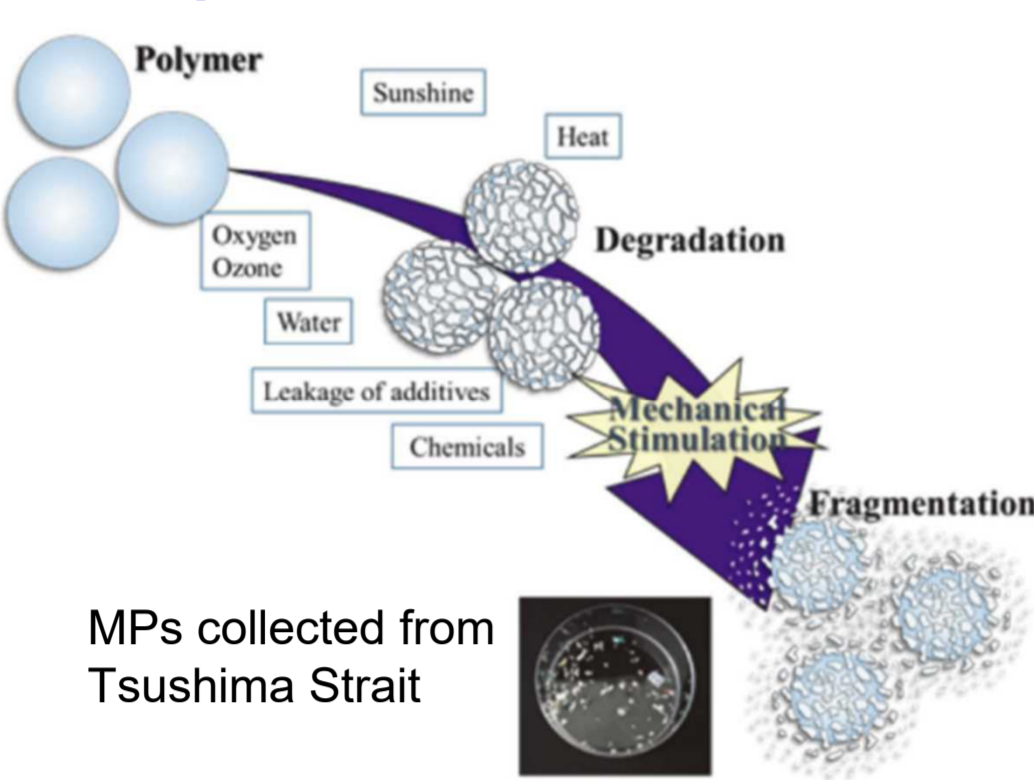
Highly marine biodegradable

Addition of PR with low inclusion rate improves ductility and biodegradability while maintaining strength

PPRNS addition improves strength while maintaining ductility and biodegradability

Photooxidation and Fragmentation Behaviors of *it*PP Films

Microplastic formation Process



Purpose: Investigate the photooxidation and fragmentation behaviors of *it*PP to simulate the MPs formation in ocean.

Experimental

Sample information		
<i>it</i> PP films	X_c (%)	f_{040}
Quenched	41.8	0
Gradually cooled	55.2	0
x2	56.0	0.379
x4	60.2	0.477

UV exposure test

Surface morphology

Quenched *it*PP films with lower crystallinity generated much denser cracks. Cracks were parallel to draw direction.

Surface chemical structure

- Photooxidation occurred in UV-exposed *it*PP films.
- Photooxidation was inhibited by the crystalline structure and oriented molecular chains.

Fragmentation test

Uniaxially oriented *it*PP films produced larger fragments than pressed films. The oriented structure was observed in fragments generated from oriented *it*PP films.

Summary

- Photooxidation likely occurs in the amorphous phase of *it*PP.
- Fragments with oriented crystalline structure were generated after fragmentation test.
- Ocean MPs were formed through photooxidation and fragmentation.

Environmental Degradation of Poly(ether-block-amide)

Purpose: Investigate the environmental degradation behaviors of Ny/PEG multi-block copolymers with different hard segments.

Experimental

Pebax	PEG (wt%)	Hard segment
1657	56	Ny6
1074	48	Ny12
1041	24	Ny12

Setup of BOD test: Compression molding, UV exposure test (125 W/m², BPT: 63°C), Biodegradation (BOD) test with extracted sea water.

UV exposure test

Surface morphology

Much denser cracks formed on poly(ether-block-amide) copolymer with higher PEG content.

Surface chemical structure (FTIR-ATR)

IR indicates degradation of PEG segment in Ny/PEG films by photooxidation.

Field Test of Polypropylene Containing Oxo-biodegradable Additives (P-Life)

Purpose: To establish advanced characterization method for field test samples

Field test (23.06-12)

Sample: PP containing oxo-biodegradable additives* (P-Life). Samples were fixed in frame and immersed in seawater. *Fatty acid Mn salt additives that promote oxidation under sunlight exposure were banned worldwide.

Surface Morphology

Surface morphology (2306-07)_Summer (1 month)

Surface morphology (2306-12)_Winter (6 months)

IR of P-Life/PP after field test (1months)

Additive was disappeared after 1 month field test.

Water contact angle after field test(6months)

Initial: 124.8°
Microorganisms attached: 0° (Highly hydrophilic)
Microorganisms were removed by acid and SDS: 70.3° hydrophilic

Surface became hydrophilic because of the attachment of microorganisms and photooxidation of polymer.

SAXS of P-Life/PP after field test

Disappearance of scattering from oxo-biodegradable additives.